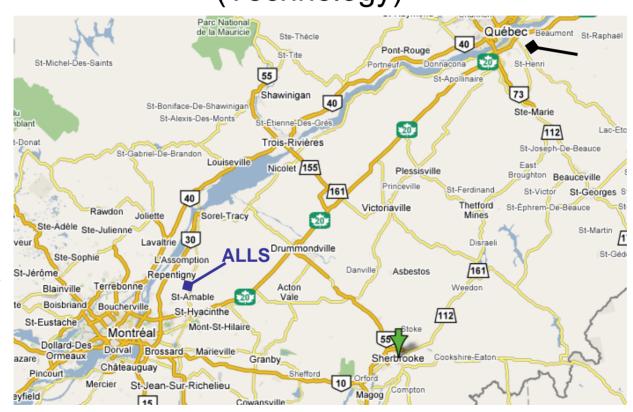
"Molecules in Intense Laser Fields – Femto to Attosecond Dynamics"

and / or

"FAST-Femto-Attosecond Simulation (Science) & Theory (Technology)"



NRC

(Ottawa)

André D, Bandrauk, PhD, FRSC, FAAAS
Canada Research Chair
Computational Chemistry & Molecular Photonics
Université de Sherbrooke

Potential energy:
$$V_o$$
: $\frac{e^2}{a_0} = 1$ Hartree = 27.2 eV, (1)

Electric field
$$E_o: \frac{e}{a_o^2} = 5 \times 10^9 \ V/cm$$
, (2)

Intensity
$$I_0 = cE_0^2/8\pi = 3.5 \times 10^{16} W/cm^2$$
, (3)

Distance
$$a_0 = 0.0529nm$$
, (4)

Time
$$t_0 = 24.2 \, as$$
. (5)

Table I

Evolution of Laser Parameters [1]

ſ	Time (s)		Intensity (Watts/cin ²)		Year
ı	Nano	10 ⁻⁹	Giga	10 ⁺⁹	1980
	Pico	10-12	Tera	10+12	1 9 <u>8</u> 5
	SERS				
	Femto	10-15	Peta	10+15	1990
	1a.u : 24 x 10 ⁻¹⁸		$I_0 = 3.5 \times 10^{+16}$		
. [
	Atto	10-18	Exa	10+18	2005
	Zepto	10-21	Zetta	10+31	2009
ا ۲	Yocto	10-24	Yotta	10+24	?

nd ns ate

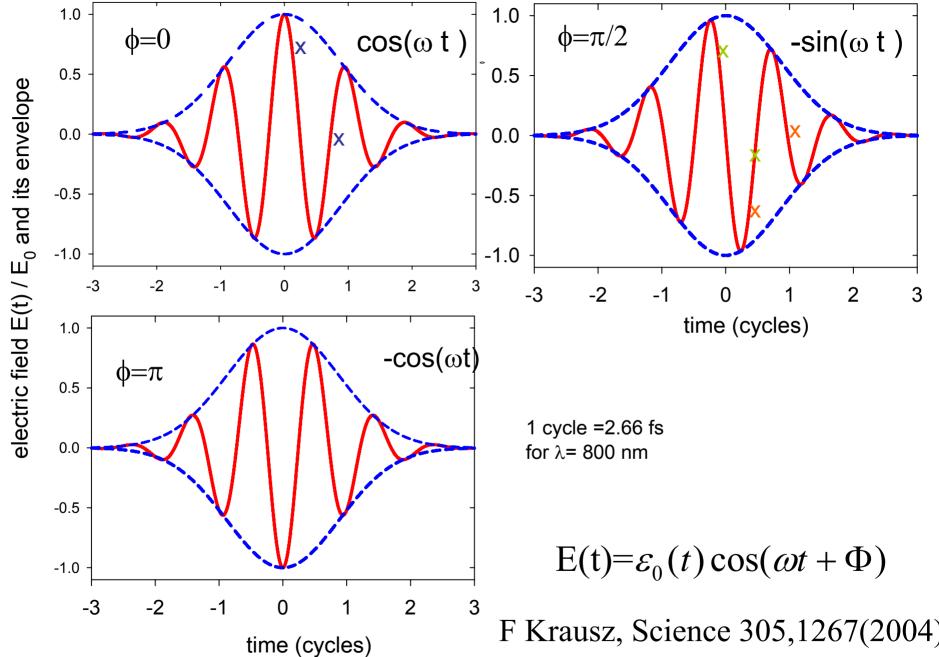
to field (1-3), t

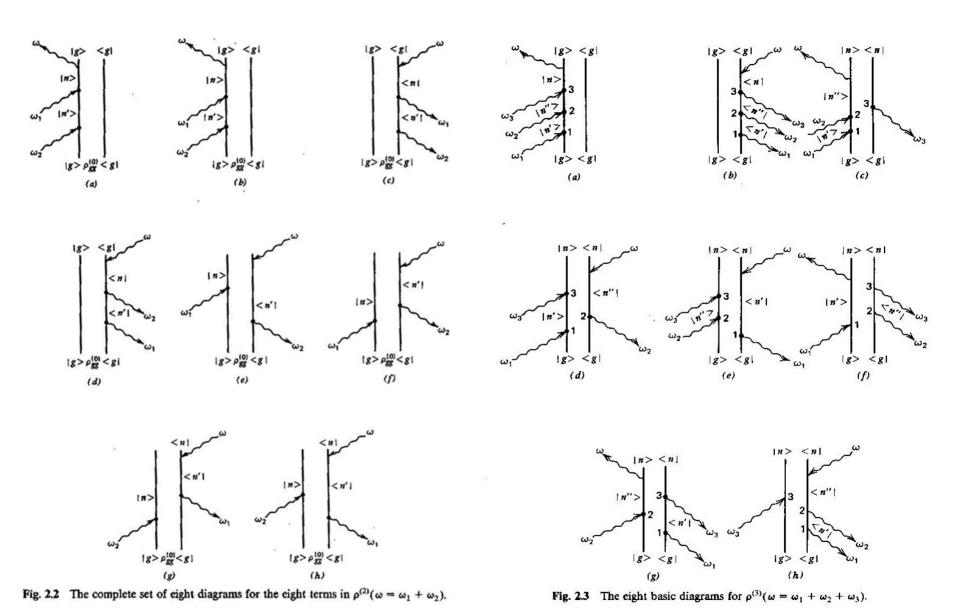
 10^{29}

representation these radiatively induced distortions creating LIMP's as discussed above lead to bond softening via laser-induced avoided crossing of molecular potentials [26-27]. At such intensities, one needs to consider further ionization and the remaining molecular ion potentials become LIMP's in the presence of intense laser pulses. The molecular ions, bound or dissociative can also undergo Above Threshold Dissociation, ATD, [20], [26-27],

Schwinger limit ~ 10**29 W/cm2)

Sunlight: 0.12 W/cm2





R Shen,"Nonlinear Optics"

MAXWELL - SCHROEDINGER

Classical

Quantum

MAXWELL

$$\frac{\partial^2 E}{\partial z^2} - \frac{1}{c^2} \frac{\partial^2 E}{\partial t^2} = \frac{4\pi}{c^2} \frac{\partial^2 P}{\partial t^2}$$

P = Medium Polarization = P(E)

 $(1^{st} \text{ Order } P = \alpha E)$

SCHROEDINGER

$$i\hbar \frac{\partial |\psi\rangle}{\partial t} = (\hat{H}_0 + \hat{V}(t))|\psi\rangle$$

$$P = P(E) = n_0 \langle \psi | \hat{\mu}_0 | \psi \rangle$$

$$\left|\Psi\right\rangle = \sum_{i} c_{i} e^{iE_{j}t/\hbar} \left|\Psi_{j}\right\rangle$$

$$V_{ii} = -P_{ii}(e(z,t)\cos(kz-\omega t))$$





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High-Dimensional Partial Differential Equations in Science and Engineering

Edited by: André Bandrauk, Université de Sherbrooke, QC, Canada, Michel C. Delfour, Université de Montréal, QC, Canada, and Claude Le Bris, École Nationale des Ponts et Chaussès, Marne La Vallée, France, and INRIA Rocquencourt, Le Chesnay, France

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CRM Proceedings & Lecture Notes

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Mang To Gers

Not yet published.

Expected publication date is July 14,

Suggest to a Colleague



High-dimensional spatio-temporal partial differential equations are a major challenge to scientific computing of the future. Up to now deemed prohibitive, they have recently become manageable by combining recent developments in numerical techniques, appropriate computer implementations, and the use of computers with parallel and even massively parallel architectures. This opens new perspectives in many fields of applications. Kinetic plasma physics equations, the many body Schrödinger equation, Dirac and Maxwell equations for molecular electronic structures and nuclear dynamic computations, options pricing equations in mathematical finance, as well as Fokker-Planck and fluid dynamics equations for complex fluids, are examples of equations that can now be handled.

The objective of this volume is to bring together contributions by experts of international stature in that broad spectrum of areas to confront their approaches and possibly bring out common problem formulations and research directions in the numerical solutions of highdimensional partial differential equations in various fields of science and engineering with special emphasis on chemistry and physics.

Titles in this series are co-published with the Centre de Recherches Mathématiques.

: Readership

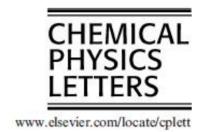
Graduate students and research mathematicians interested in numerical solutions of high-dimensional PDE's.



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Chemical Physics Letters 419 (2006) 346-350



Complex integration steps in decomposition of quantum exponential evolution operators

André D. Bandrauk 1, Effat Dehghanian *, Huizhong Lu

Laboratoire de Chimie Théorique, Faculté des Sciences, Université de Sherbrooke, Qué., Canada J1 K 2R1

Received 28 October 2005; in final form 1 December 2005 Available online 22 December 2005

Abstract

We generalize previous high-order exponential split operator methods for solving time-dependent Schroedinger equations [A.D. Bandrauk, H. Shen, Chem. Phys. Lett. 176 (1991) 428] by introducing complex integration steps (a + ib) with real positive part a. We show that this new procedure avoids real negative steps which occur generally in high-order split operator methods. New highly accurate splitting schemes are thus derived and the efficiency of these is demonstrated in the calculation of the eigenstates of the one-electron molecular ion H_2^+ . © 2005 Elsevier B.V. All rights reserved.

The 1-D electron TDSE in atomic units, a.u. $(e = \hbar = m = 1)$ is written as

$$i\frac{\partial\psi(x,t)}{\partial x} = \left[-\frac{1}{2}\frac{\partial^2}{\partial x^2} + V(x)\right]\psi(x,t) = (A+B)\psi(x,t), \quad (2)$$

where $A = -\frac{1}{2} \frac{\partial^2}{\partial x^2}$; B = V(x). For time independent potentials, the solution is

$$\psi(x, t + \Delta t) = e^{+\lambda(A+B)}\psi(x, t), \quad \lambda = -i\Delta t,$$

$$= S(\lambda)\psi(x, t),$$
(3)

whereas for time-dependent V(x,t) one must introduce time-ordering operators [3]. A symmetric fourth-order accurate decomposition of $S = \exp[\lambda(A+B)]$, S_4 is obtained as a product of three second-order S_2 (Eq. (1)) operators with time steps $\gamma\lambda$ and $(1-2\gamma)\lambda$, [2]

$$S_4^B = e^{\gamma \lambda B/2} e^{\gamma \lambda A} e^{(1-\gamma)\lambda B/2} e^{(1-2\gamma)\lambda A} e^{(1-\gamma)\lambda B/2} e^{\gamma \lambda A} e^{\gamma \lambda B/2}. \tag{4}$$

Permuting A with B gives another fourth-order operator S_4^A . Both involve seven exponential operators with three kinetic energy terms, A = T in S_4^B but four such terms in S_4^A . The accuracy of this decomposition is obtained from

$$S - S_4^B = \frac{\lambda^3}{24} [A + 2B, [A + B]] (2\gamma^3 + (1 - 2\gamma)^3) + \frac{\lambda^4}{24}$$

$$\times (3\gamma^3 - 6\gamma^2 + 3\gamma - 1/2) [-2A^2 [A, B]$$

$$- A[A, B]B \dots] + O(\lambda^5). \tag{5}$$

Cancellation of corrections of $O(\lambda^3)$ and $O(\lambda^4)$ occur simultaneously for real values of $\gamma = (2 - 2^{1/3})^{-1}$, thus ensuring fourth-order accuracy. Recently following Suzuki's work [9], Chin has also obtained fourth-order accuracy We note that S_4^B , a seven exponential operator, Eq. (5) can be written as

$$S_4^B = e^{\gamma \lambda B/2} e^{\gamma \lambda A} e^{\gamma^* \lambda B/2} e^{(1-2\gamma)\lambda A} e^{\gamma^* \lambda B/2} e^{\gamma \lambda A} e^{\gamma \lambda B/2} + O(\lambda^5),$$

$$\gamma^* = 1 - \gamma, \quad \gamma = (2 - 2^{1/3})^{-1}.$$
(6)

Based on this observation we construct a new five exponential third-order operator S_3^B , (exchanging A with B gives S_3^A)

$$S_3^B = e^{\gamma \lambda B/2} e^{\gamma \lambda A} e^{\lambda B/2} e^{\gamma^* \lambda A} e^{\gamma^* \lambda B/2}, \tag{7}$$

$$S - S_3^B = \frac{\lambda^3}{6} (-3\gamma^2 + 3\gamma - 1) \left(\frac{1}{4} [[A, B], B] - \frac{1}{2} [A, [A, B]] \right) + \mathcal{O}(\lambda^4).$$
(8)

The $O(\lambda^3)$ error is cancelled by choosing $\gamma = \frac{1}{2}(1 \pm i/\sqrt{3})$.

It is to be noted that the new S_3^B , Eq. (7) with $\gamma = a \pm ib$, $a = \frac{1}{2}$, $b = \frac{1}{2\sqrt{3}}$, can be written equivalently as the product of two S_2' 's with complex γ 's,

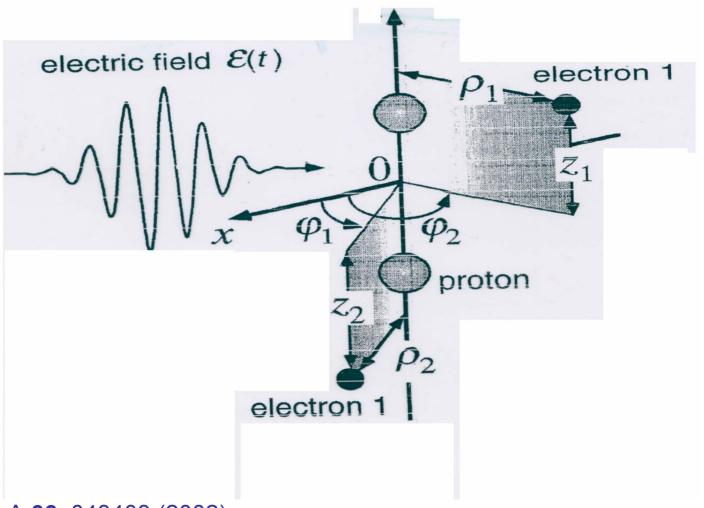
$$S_3^B = (e^{\gamma \lambda B/2} e^{\gamma \lambda A} e^{\gamma \lambda B/2})(e^{\gamma^* \lambda B/2} e^{\gamma^* \lambda A} e^{\gamma^* \lambda B/2}), \qquad (9)$$

= $S_2^{B'}(\gamma) * S_2^{B'}(\gamma^*) + O(\lambda^4), \qquad (10)$

since $y + y^* = 1$.

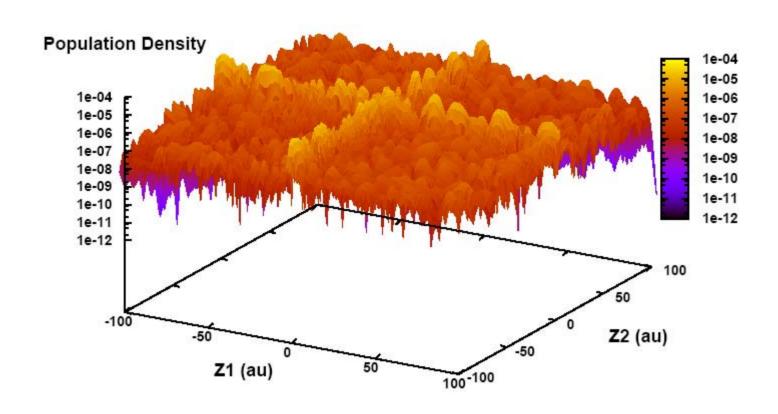
Eqs. (6) and (10) bring out the central role of the secondorder operators $S_2^B = e^{\gamma \lambda B} e^{\lambda A} e^{\gamma^* \lambda B}$, $S_2^A = e^{\gamma \lambda A} e^{\lambda B} e^{\gamma^* \lambda A}$ which are second-order accurate $(O(\lambda^3))$ as in Eq. (1) with $\gamma = \frac{1}{2}$. We show next that as suggested by Eqs. (7) and (10), combinations of these S_2 's with complex $\gamma = a \pm ib$, and a > 0, gives new high-order evolution operators.

$$N_P = \underbrace{10^3}_{z_1} \times \underbrace{10^3}_{z_2} \times \underbrace{10^2}_{\zeta_1} \times \underbrace{10^2}_{\zeta_2} \times \underbrace{10^2}_{\phi} \simeq 10^{12} \times \underbrace{10^z}_{P} \simeq 10^{14}$$



Phys. Rev. A 66, 043403 (2002)

$H_2 (X^1 \Sigma_g^+)$: I=3x10¹⁵W/cm², λ =800nm, T = 6 cycles



Vol.440 S.Auy 2007

NEWS

The petaflop challenge

IBM's announcement on 26 fune that it was about to much the petalog open between outbreald a new era in computing. But unless the research and computing communities get their pergramming on together they rich having few scientific applications that can take advantage of this huge increase in power, lay seperits.

Called Blue Gene¹I) the first of the new highpowers do computers should be operational near year, and IBM has already lined up potential customers in the United States. Germany and Britain. The largest planned configuration for the machines would run continually at 1,000 trillian floating point apparations per second, or terrallops, and be capable of peak speech of up to 2,000 terrallops (3 petallops). That would rusto it between three and ten times faster than the machine that tops the latest I OP 500 list of supercomputers, the IBM Blue Gene/L at Lawrence Livermore National Laboratory in California, which peaks at 500 terrallops.

"Traver yezcited," says Ray Bair, held of the Argonne Leadership Competing Facility at the Argonne National Laboratory in Hinsei, which has worked with IBM on developing the Blas Gene'll; and which will heat averaion of the hardware. "We're croming a threshold." Bair adds, chaiming that the increased power will at his allow researchers to build and ranno data in the way they have always wasted.

At prices of around \$100 a gigaflop, petadop computers will start off in the \$100-million range. A great deal of the added speed that aroung will buy simply comes from more procursors. Supercomputer designess began taking parallel procursing seriously in the 1990, but few machines have been designed to work with more than 10,000 procursors. In 2003. Blue Gene II. which is almost three times a fracturity as easiest competition, marked a significant step up with 131,072 procursors. A 3-petaflop Blue Gene IP will boust \$58,736 — a multitude that brings withit problems as well as promise.

The rapid recent growth in supercomputing power — Blue Gene'll is a powerful arche whole of 2000's TOP200 fart constitued — has come month from increase in the performance of the computative component processors. But

a fewyeain ago that process "caree to a grinding hab", says line Section, head of Blue Gene applications at IBM's Thomas I, Watson Research Center is Yorkdown Heights, New York. As the processors gotfaster and faster they began suffering from disproportionate increases.

in power contemption and best output.

To combat this chipmakers made a historic switch. Since 2004, they have concentrated on increasing the number of processors on a chip. allowing the speed at which individual procesnors operate to plateau. So the dual and quadchips now common in laptops, offering two or four processors, may well be appreded to 128 or 256 corer by 2015. This means that the cheap Linux supers computing chatters common in universities would have bundreds of thourands of processors, and dedicated supercompotent might have handreds of millions. The processors will not necessarily be blackingly fast. Blue Gosso Ve 850-MH schiou are little finter. than a Pantium III from 1999. But with their numerical advantage they would have to be.

This new reliance on parallel processing for

increased performance means that companies from Microsoft to Nintendo will have to rethink their software — and so will ncientists. A few scientific applications. full into the tireoured. reducet called temberreminely parallel. problemá Genome analysis using BLAST notification to companie. requescer and marr spectroscopy for professionale are sea-

enally fairly easy to parallelize, any Le roy He-od, president of the Institute for Systems Biology in Seattle, Washington, Each processor can take on a specific task without much reference to what all the others are doing. But other sorts of problem, in which many of the calculations depend on other calculations being done diswhere, are not so tractable.

For the moderate levels of parallelism seen to date, it is possible to get by with the content practice of writing code, and designing models, in terms of linear sequences of interac-

tion and then parallelizing once satisfied. To get the most out of mature parallel clusters and machines, that will no longer be no option. "Coding models remaining across as many as a million processors is a new challenge we have to most," says I'm Palmer of the European Cantre for Medium-Lunge Weather Forecasts in Bending, UK, who is inserested in petallop mechines for climate modelling. "We have an choice but to follow the fundy are treath."

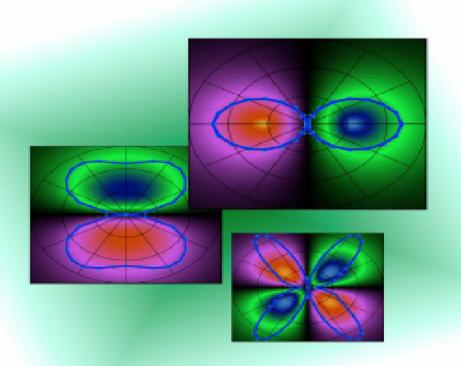
Scientists need to thirt to thinking in panellel from the outset, designing hypotheses and code accordingly, says Horst Simon, associate lab director for computing at Lawrence Berkeley National Laboratory California, Be describes what is needed as positions short of a "orroktion is scientific programming". Such a revolution has been browing for decades, but there ham't bean much storming of the barricades. "The high-performance comparing community has been working on the parallel-programming problem for over 25 years. And fraulds we don't have much to show for it," wrote Intel researcher Timothy Matteon on his company's research blog last week. "On the nofewherefront, it's chaos?"



DE CALCUL DE HAUTE PERFORMANCE



Rapport annuel 2008 1^{er} janvier au 31 décembre



NEW TECHNIQUE: LIED LASER INDUCED ELECTRON DIFFRACTION

T.Zuo, A.D. Bandrauk, and P.B. Corkum, Chem. Phys. Lett., 313 (1996).

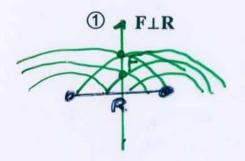
H₂⁺ in a δ- pulse (attosecond)

1as=10⁻¹⁸

$$E(t) = F\delta(O)$$

$$\varphi(\vec{p}) = \phi_{1\sigma_g}(\vec{p} + \vec{F})$$

$$= 2 \cos \left[(\vec{p} + \vec{F}) \cdot \frac{\vec{R}}{2} \right] \phi_{1s} (\vec{p} + \vec{F})$$



$$\varphi(p) = 2 \cos\left(\frac{\vec{p} \cdot \vec{R}}{2}\right) \phi_{1s} (\vec{F} + \vec{p})$$

$$\underline{\mathbf{max}} \text{ at } p = \frac{2m\pi}{R\cos\theta}$$

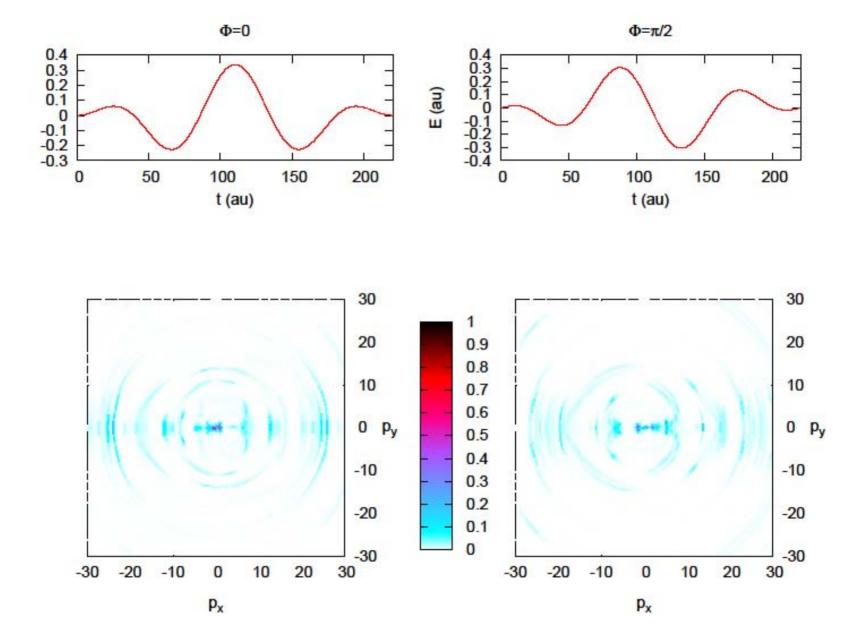
$$\underline{\min} \text{ at } p = \frac{(2 m+1) \pi}{R \cos \theta}$$

$$\varphi(p) = 2 \cos \left[\frac{FR}{2} \left(1 + \frac{p \cos \theta}{F} \right) \right] \phi_{1s} (\vec{F} + \vec{p})$$



Small p:
$${max \atop min} FR = {2m\pi \atop (2m+1)\pi}$$

large p:
$$(FR + pR \cos \theta) \rightarrow \frac{2m\pi}{(2m+1)\pi}$$



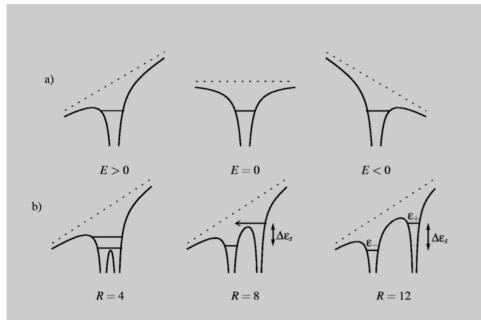


Figure 1. (a) Atomic electronic potentials distorted respectively by the presence of a strong positive, null, and negative external electric field. (b) Electronic molecular ptoentials in a positive electric field, as a function of the internuclear distance R. The Stark shift $\Delta \epsilon_{\rm S}$ between the two localized molecular orbitals ϵ_{+} and $\epsilon_{-} \simeq E_{\rm max} R$.

P. B. Corkum, PRL,71, 1994 (1993).

T. Zuo and A. D. Bandrauk, PRA, 52, R2511 (1995).

Nuclear fusion from explosions of femtosecond laser-heated deuterium clusters

T. Ditmire, J. Zweiback, V. P. Yanovsky, T. E. Cowan, G. Hays & K. B. Wharton

Laser Program, L-477, Lawrence Livermore National Laboratory, Livermore, California 94550, USA

As a form of matter intermediate between molecules and bulk solids, atomic clusters have been much studied¹. Light-induced processes in clusters can lead to photo-fragmentation^{2,3} and Coulombic fission⁴, producing atom and ion fragments with a few electronvolts (eV) of energy. However, recent studies of the photoionization of atomic

cluster) are ionized, electrons undergo rapid collisional heating for the short time (<1 ps) before the cluster disassembles in the laser field 19. Through various collective and nonlinear processes, the laser rapidly heats the electrons to a non-equilibrium state (with mean

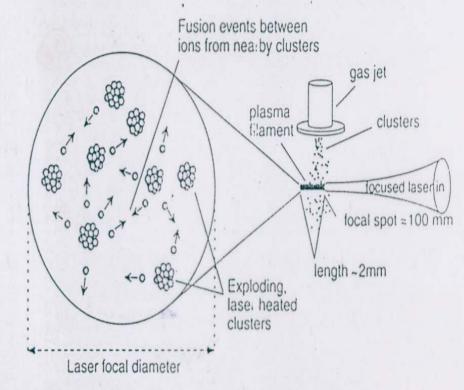
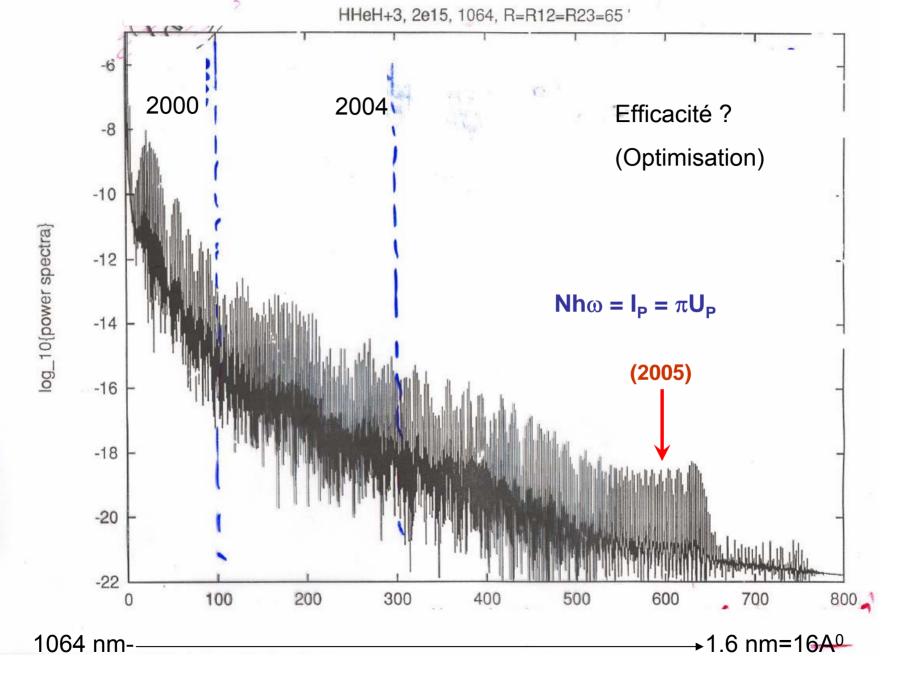


Figure 1 Layout of the deuterium cluster fusion experiment.



A.D. Bandrauk and H.S. Nguyen, Phys. Rev. A 66, 031401 (2002)

Step 1: get spectrum

$$a(t) = \langle \psi(t) | \dot{z} | \psi(t) \rangle$$

$$a(\omega) = \frac{1}{2\pi} \int_{-\infty}^{+\infty} a(t)e^{-i\omega t} dt$$

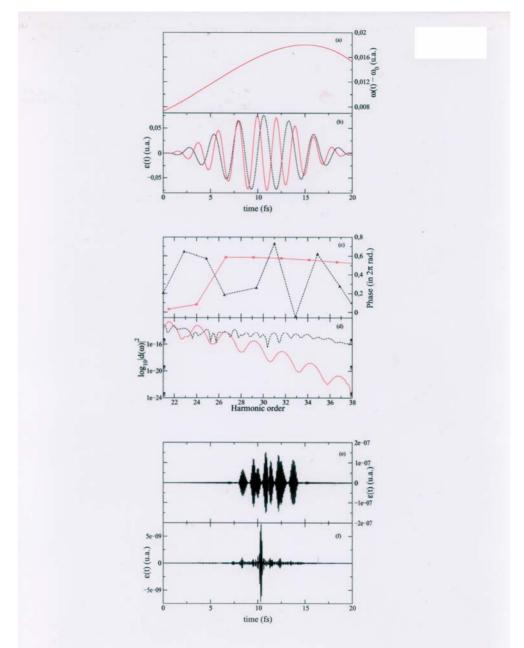
Same: select frequency region between $\omega_1 < \omega < \omega_2$

Step3: come back to time domain

$$\widetilde{a}(t) = \int_{\omega_1}^{\omega_2} a(\omega) e^{i\omega t} d\omega$$

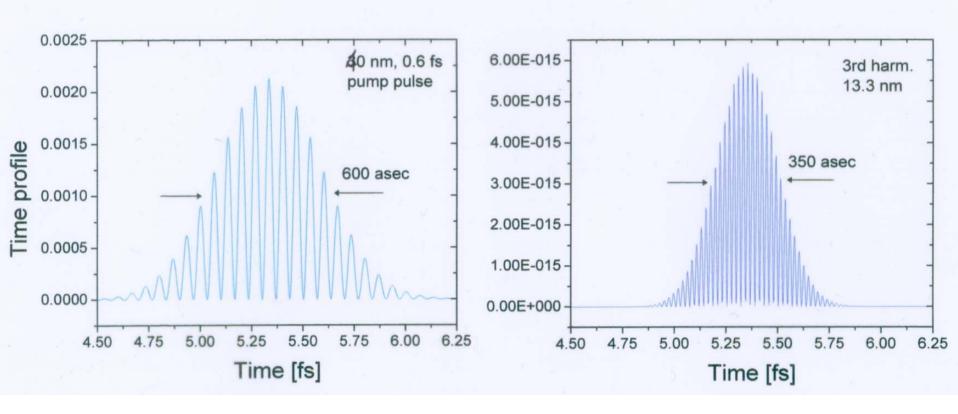
$$\Delta \omega = \omega_2 - \omega_1$$

$$\tau \quad \Box \quad \frac{1}{\Lambda \omega} \rightarrow \quad attosecond \quad = 3 \text{ Angstroms}(10^{**}-8cm) / c(3x10^{**}10cm/s-1))$$



Yedder, LeBris, Chelkowski, Bandrauk, PRA 69, 041802 (2004)

Frequency-up conversion, 1st -->3rd harmonics



ADB et al, J Molec Str. 735,203(2004)

Effect of Nuclear Motion on Molecular High-order Harmonics and on Generation of Attosecond Pulses in Intense Laser Pulses

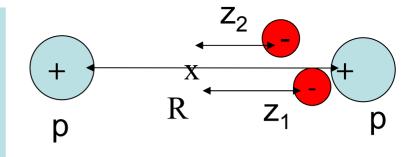
André D. Bandrauk, Szczepan Chelkowski, Shinnosuke Kawai, and Huizhong Lu Département de Chimie, Université de Sherbrooke, Sherbrooke, Qc, J1K 2R1 Canada

Abstract

We calculate harmonic spectra and shapes of attosecond pulse trains using numerical solutions of Non-Born Oppenheimer time-dependent Shrödinger equation for 1-D H₂ molecules in an intense laser pulse. A very strong signature of nuclear motion is seen in the time profiles of high order harmonics. In general the nuclear motion shortens the part of the attosecond pulse train originating from the first electron contribution but it may enhance the second electron contribution for longer pulses. The shape of time profiles of harmonics can thus be used for monitoring the nuclear motion.

PACS numbers: 42.65.Ky, 42.65.Re, 42.50.Hz, 32.80.Rm

Phys Rev Lett(2008,)101,153901 J Phys B –to appear 2009 La dynamique de 4-particules: p+p+e-+e- décrite par l'éq. de Schrödinger solutionneé numériquement



Équation de Schrödinger dependant de temps pour une molecule H₂

éxposée au champ laser intense decrit par : E(t)=ε(t) cos($ω_L t$)

(polarisation linéaire)

a.u.
$$e = \hbar = m_e = 1$$

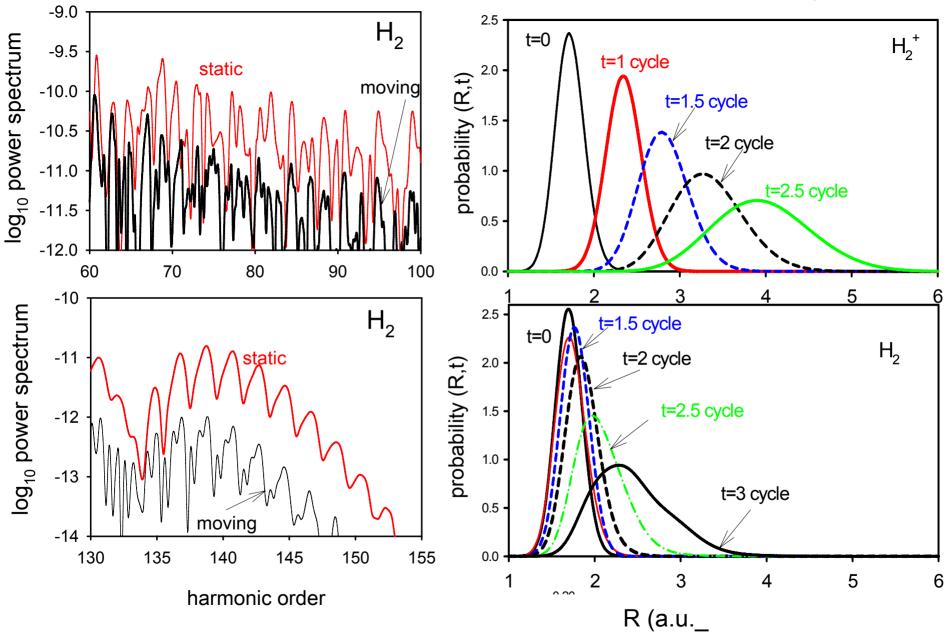
$$i\frac{\partial \psi(z_1, z_2, R, t)}{\partial t} = [H_e + H_N + V(z_1, z_2, t)]\psi(z_1, z_2, R, t),$$
 (1)

$$H_e = \sum_{i=1}^{2} \left[-\frac{1}{2} \frac{\partial^2}{\partial z_i^2} - \frac{1}{[(z_i + R/2)^2 + c]^{1/2}} - \frac{1}{[(z_i - R/2)^2 + c]^{1/2}} \right] + V_{rep}(z_1, z_2)$$
 (2)

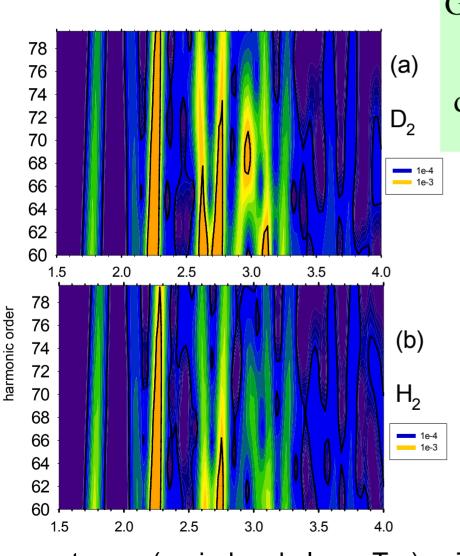
$$V_{rep}(z_1, z_2) = \frac{1}{[(z_1 - z_2)^2 + d]^{1/2}} , ; H_N = -\frac{1}{2M} \frac{\partial^2}{\partial R^2} + \frac{1}{R}$$
 (3)

$$V(z_1, z_2, t) = (z_1 + z_2)\varepsilon(t)\cos(\omega_L t) \tag{4}$$

$$d(t) = \langle z_1 + z_2 \rangle = \int_{-\infty}^{\infty} dz_1 \int_{-\infty}^{\infty} dz_2 \int_{0}^{\infty} dR \ \psi^*(t) \ (z_1 + z_2) \ \psi(t)$$



Analyse en temps-frequence de Gabor (ondelettes):

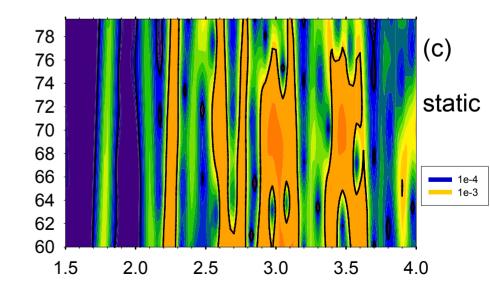


 $= d_G(t,\omega) = \int_{-\infty}^{\infty} dt' G(t,t') \exp(-i\omega t) d(t)$

G(t-t') = exp[
$$-\frac{(t-t')^2}{2\sigma_0^2}$$
], σ_0 =0.1 fs

$$d_{G}(t,\omega) = \operatorname{cte} \int_{-\infty}^{\infty} d\omega' e^{-b(\omega-\omega')^{2}} e^{i\omega't} d_{F}(\omega')$$

 $|d_G(t,\!\omega)|$ - profile temporaire des impulsions atto dans un train , $\Delta\omega{\sim}5\text{-}10~\omega_L$



temps (periodes du laser T_{las}), T_{las} =2.67 fs

Journal of Physics B

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Volume 42 Number 7 14 April 2009

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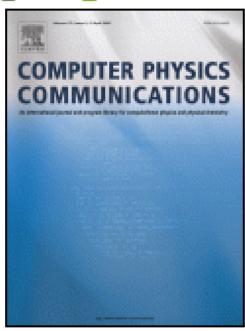
ScienceDirect Top 25 Articles Overall

October - December 2007









A numerical Maxwell-Schrödinger model for intense laser-matter interaction and propagation

E. Lorina, b, -, -, S. Chelkowski and A. Bandrauk

^aCentre de Recherche en Mathématiques de Montréal, University of Montréal, Canada

^bUniversity of Ontario, Institute of Technology, Oshawa, Canada

^cUniversité de Sherbrooke, laboratoire de chimie théorique, Canada

d'Université de Sherbrooke, laboratoire de chimie théorique, Centre de Recherche en Mathématiques de Montréal, Canada Research Chair, Canada

Model

Coupling of macroscopic Maxwell's equations with many TDSE's.

Lorin, Chelkowski, Bandrauk, Comput. Phys. Comm. vol. 177 (2007)

$$\begin{cases} \partial_{t}\mathbf{B}(\mathbf{r},t) & = -\nabla \times \mathbf{E}(\mathbf{r},t) \\ \partial_{t}\mathbf{E}(\mathbf{r},t) & = \nabla \times \mathbf{B}(\mathbf{r},t) - 4\pi\partial_{t}\mathbf{P}(\mathbf{r},t) \\ \nabla \cdot \mathbf{B}(\mathbf{r},t) & = 0 \\ \nabla \cdot \left(\mathbf{E}(\mathbf{r},t) + \mathbf{P}(\mathbf{r},t)\right) & = 0 \end{cases}$$

$$\begin{cases} \mathbf{P}(\mathbf{r},t) = n(\mathbf{r}) \sum_{i=1}^{m} \mathbf{P}_{i}(\mathbf{r},t) & = n(\mathbf{r}) \sum_{i=1}^{m} \chi_{\Omega_{i}}(\mathbf{r}) \int_{\mathbb{R}^{3}} \psi_{i}\mathbf{r}'\psi_{i}^{*} \\ i\partial_{t}\psi_{i}(\mathbf{r}',t) & = -\frac{\triangle_{\mathbf{r}'}}{2}\psi_{i} + \mathbf{r}' \cdot \mathbf{E}_{\mathbf{r}_{i}}\psi_{i} + V_{c}\psi_{i}, \\ \forall i \in \{1,...,m\} \end{cases}$$

The numerical model is the one presented in [19], where the gas domain is divided in small cells of gas denoted by Δv (corresponding the Ω_i 's of Section 2) and in which we solve 1 TDSE, representing the $n\Delta v$ molecules of the cell. In practice 3d Maxwell's equations are solved in parallel with $\sim 140,000$ 1d TDSE's, see Fig. 5 and [17]. We then represent at

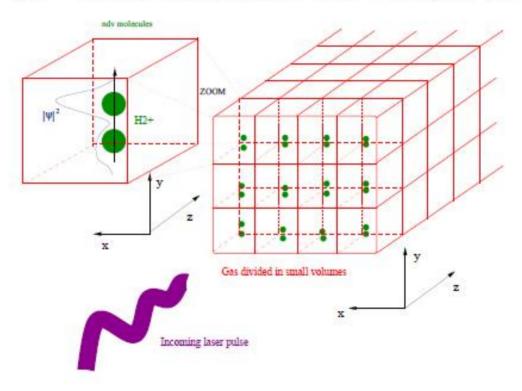


Figure 5: Numerical geometry

Beyond the paraxial & SVEA approximation

Under the paraxial and SVEA approximation equation on $\mathbf{E}(x,y,z,t) = \mathcal{E}(x,y,z,t)e^{\mathbf{i}(kz-\omega t)}\mathbf{e}_y$ with \mathcal{E} the envelope Bergé et al.. Physica D, 176 (2003), Couairon, Mysyrovicz, Phys. Report 441 (2007):

$$\partial_{z}\mathcal{E} = \frac{\mathrm{i}}{2k} \triangle_{\perp}\mathcal{E} + \mathrm{i}k_{0}n_{2}|\mathcal{E}|^{2}\mathcal{E} - \frac{\mathrm{i}k_{0}}{2n_{0}\rho_{c}}\rho\mathcal{E}, \qquad \partial_{t}\rho = \sigma_{K}|\mathcal{E}|^{2K}\rho_{\mathsf{atm}}$$

- K the number of photons in multi-photon ionization, $K = \langle U_i/\hbar\omega_0 + 1 \rangle$ (U_i ionization potential of the medium)
- \bullet σ_K the coefficient of the multi-photon ionization rate
- \bullet ρ_{atm} the neutral atom density

Focusing and defocusing effects clearly identified numerically and theoretically for long pulses (Bergé, Couairon, Ginibre, Fibich, Sulem, etc)

Numerical data

Numerical data

- ho \sim 140,000 1D TDSE's solved in parallel
- 3D Maxwell's equations
- $\sim 30h$ on 128 processors of mammouth (RQCHP) Lorin, Bandrauk, IEEE proceed. (2008) We represent at time different times t_k the transversal cut of the pulse at z_k such that $|Ey(0,0,z,t_k)|$ is maximal at z_k on (Oz). In other words

We represent transversal cuts of the beam

- At t_k fixed, we denote $|E_y|_{\infty} = \max_z |E_y(0,0,z,t_k)|$ the maximal value on the (Oz) axis reached in z_k
- We draw $E_y(x, 0, z_k, t_k)|/|E_y|_{\infty}$ in order to have normalized graphs to compare with vacuum

Results I - $I \sim 2 \times 10^{16} \text{W} \cdot \text{cm}^{-2}$, $n_0 \sim 3 \times 10^{20} \text{mol} \cdot \text{cm}^{-3}$

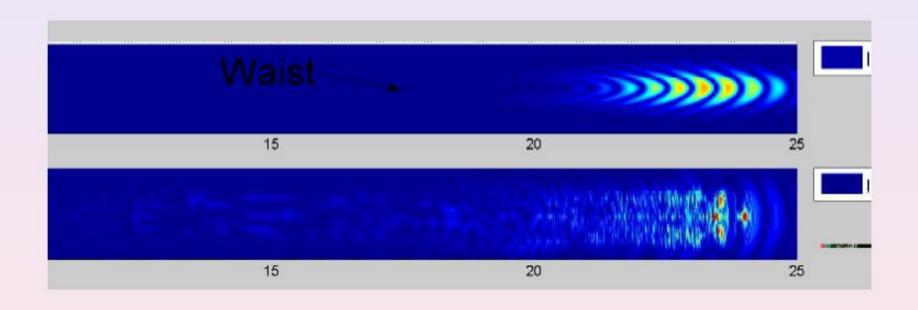


Figure: $|E_y|^2$ - 4.5 μm after the waist in vacuum and gas

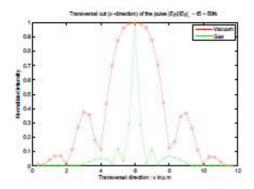


Figure 11: Pulse thickness during propagation $7\mu m$ after the waist

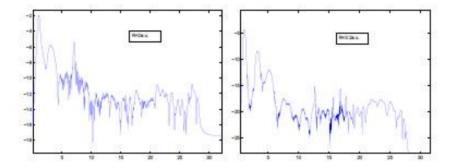


Figure 12: Electric field harmonics $R=2.a.u,\,R=3.2a.u.$

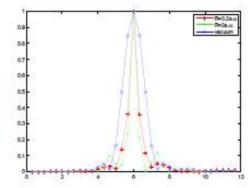
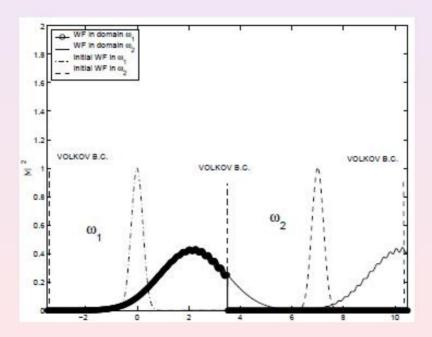
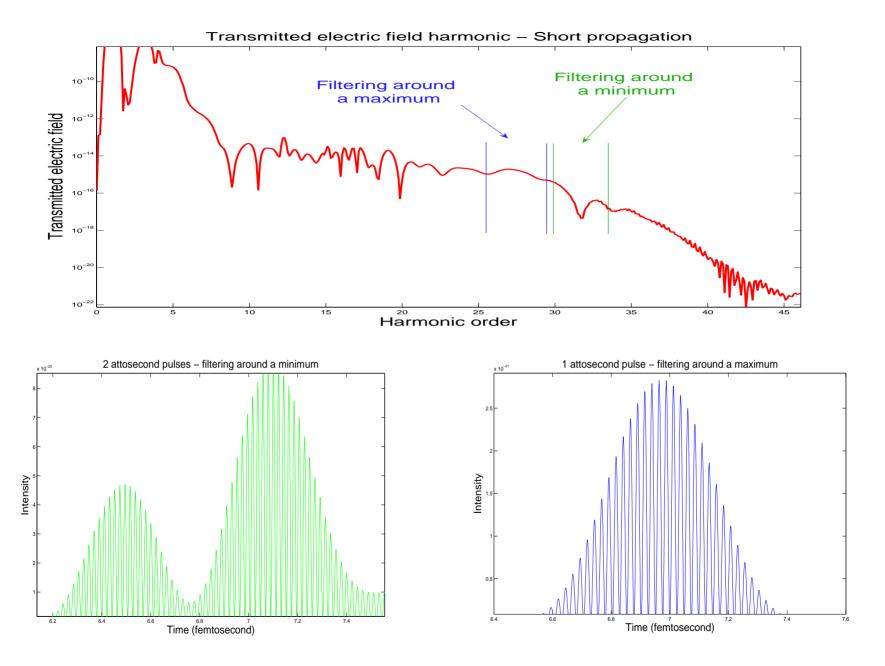


Figure 13: Electric field transversal cut $R=2.a.u,\,R=3.2a.u.$

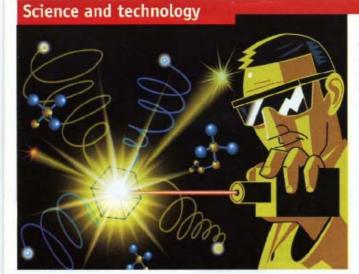
Improvement of the model I - microscopic approach

Another approach is presented in Lorin, Bandrauk, Chelkowski, Num. Methods for Partial Diff. Eq., (2008). A method to transmit free electron from a molecule to another. Based on a particular choice of boundary conditions (Volkov)





Two (minimum) or one (maximum) attosecond pulses



Attosecond science

The fast show

Extremely short laser pulses can illuminate electrons in motion

ON THE atomic scale, things move mind-bogglingly quickly. Electrons jump between orbits or escape the nucleus altogether in attoseconds-that is, million, million, millionths of a second, Indeed, one attosecond is to one second what one second is to the age of the universe. Seeing such acrobatics takes wit and ingenuity, but it is possible. Moreover, if such processes could be manipulated-and the early signs are that they can be-then it would have applications in fields as far apart as computing and medicine.

A report just drafted by America's National Research Council, "Controlling the Quantum World", outlines how scientists might manipulate the inner workings of a molecule. A long-term workshop at the Kayli Institute for Theoretical Physics, part of the University of California, Santa Barbara, is also investigating how this might be achieved. And, at a conference held recently at the institute, Ferenc Krausz of the Max Flanck Institute of Quantum Optics in Garching, Germany, and Marc Vrakking of the FOM Institute for Atomic and Molecular Physics in Amsterdam described one way that it could be done.

Lasers work by creating a chain reaction in which photons of light prompt the generation of further photons. These photons are emitted in bursts. Shortening each burst sufficiently is what makes atto-

second science possible. The two researchers employed what they call "high harmonic pulse generation" to create pulses a few hundred attoseconds long. They did this by using a laser that emits short pulses of light to drive a second laser that then emits even shorter pulses. In fact, the pulses are so rapid that they come close to the limit imposed by Heisenberg's famous uncertainty principle, which states that the precision of a time measurement is limited by the precision of a corresponding energy measurement.

Atto boys

Dr Krausz and Dr Vrakking fired their laser at a molecule of deuterium. Deuterium, also known as heavy hydrogen, is a simple molecule, consisting of two atomic nuclei and two electrons. The sample under investigation became positively charged because zapping it with the laser removed one of the electrons. The researchers found that the molecule then separated into a deuterium atom, consisting of a nucleus and an electron, and a deuterium ion, consisting of a nucleus.

Using conventional laser pulses causes atoms and ions to be ejected to the right and left at random. Using ultrafast laser pulses, though, makes the atoms fly off to the right and the ions to the left. The researchers were thus able, in effect, to conAlso in this section

72 Negative databases

72 Supernovas

73 Gene therapy and cancer

73 The virtues of cider

trol on which of the two deuterium atoms the electron resides at the end of their experiment. That is to say, they had separated the atoms from the ions.

Exactly how this werks is complicated-not least because all of the atoms are interacting simultaneously with the laser and with each other. But the researchers think that the laser pushes the electron, which initially binds the two atoms together, back and forth between the two ions until, at some point, the distance between the two gets too large and it is no longer able to jump from one to the other.

The ability to manipulate electrons in this way is important because electronsharing is essential to chemical bonding. Ultrafast lasers could thus be used to change the outcome of chemical reactions. Proponents point to possible applications in magnetic information-storage devices, which would lead to much more powerful computers. Other possibilities include the development of compact, portable x-ray lasers for medical imaging that needs to be done outside hospital radiology departments, and bright ultrafast x-ray lasers for use within those departments.

The motion of electrons is the fundamental basis of chemistry. Watching the steps in the dance of the electrons will help chemists work out why some atoms bind when others do not, why reactions take the time that they do, and why some molecules bend one way and not the other. Brighter x-ray lasers could also be used to reveal the atomic details of chemical catalysis or the way that light energy is absorbed and stored during photosynthesis, according to the National Research Council report Knowing exactly how to capture sunlight and turn it into chemical energy would be a prize indeed.

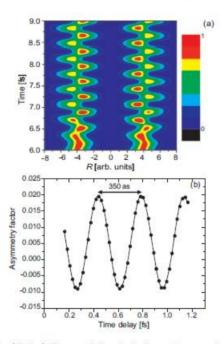


FIG. 53. (Color) Proposal for inducing attosecond electron wave-packet dynamics by a 0.8-fs, 115-nm VUV pump pulse in H_2^+ and probing it with a time-delayed 0.1-fs, 20-nm XUV pulse (Bandrauk *et al.*, 2004). Both pulses are polarized parallel to the molecular axis. (a) Contour plot of the electron probability distribution along the molecular axis for an internuclear distance of eight atomic units vs pump-probe delay. (b) Asymmetry factor $(P_- - P_+)/(P_- + P_+)$ vs delay, where P_+ and P_- represent the probability of observing the electron liberated by the XUV probe in the positive or negative direction (along the molecular axis), respectively. Adapted from Bandrauk *et al.*, 2004.

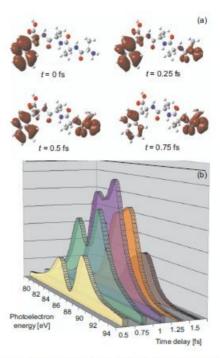


FIG. 54. (Color) Computed ultrafast positive charge (hole) migration in a tryptophane-terminated tetrapeptide (Remacle and Levine, 2006a, 2007). (a) The hole density shown in red indicates that the charge swings across the entire peptide from the aromatic amino acid on the left to the N end on the right within less than one femtosecond, following excitation of the electronic wave packet on an attosecond time scale. This hyperfast charge migration is proposed to be probed by measuring the kinetic energy distribution of photoelectrons released by a time-delayed sub-fs XUV pulse. (b) A series of such freeze-frame spectra calculated for a 250-as, 95-eV probe pulse at different pump-probe delays. From Remacle and Levine.

To exploit attosecond technology

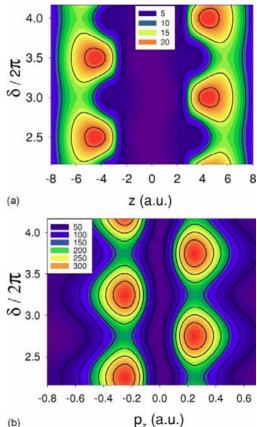
Measuring electron wave packets

1. Attosecond pulses are fast enough to observe

electron wave packets.

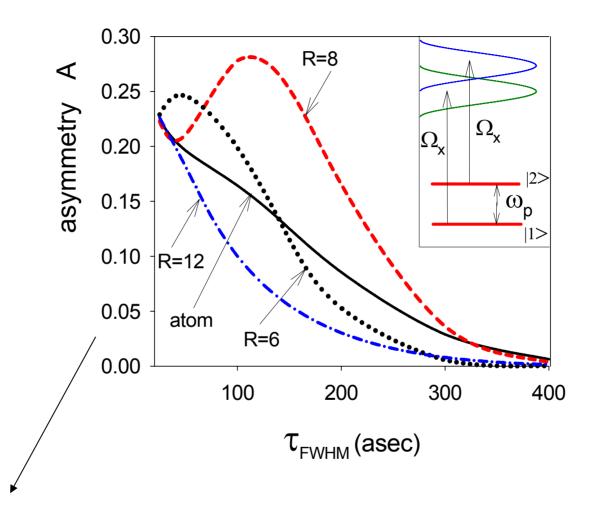
- 2. Electron wave packets are resolved through changes to the photo-electron spectrum as a function of pump-probe time delay.
- 3. The attosecond pulse projects the momentum distribution into the continuum.

Yudin et al, Phys Rev A 72, 51401(R) (2005)



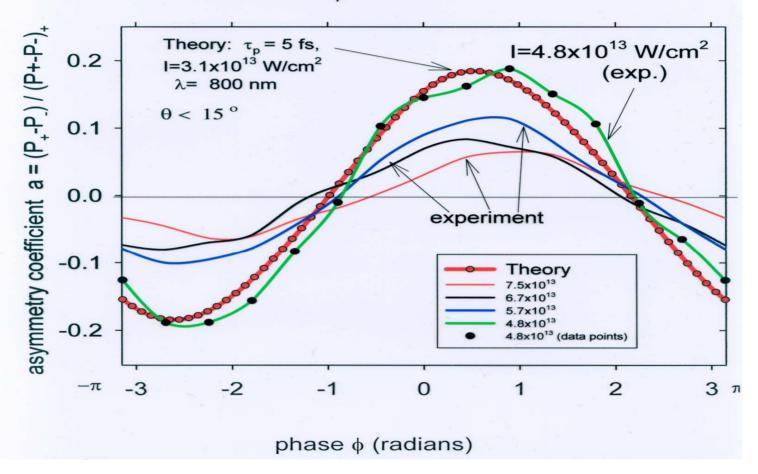
J. Phys. <u>B39</u>, S409 (2006)

Atom: 1s+2p and $\bullet_g 1s + \bullet_u^* 2p$ in H_2^+



<u>no</u> asymmetry in atomic 1s +2s and in molelcular: ${}^{\bullet}_{g}1s + {}^{\bullet}_{g}{}^{*}2s$, but there asymmetry in ${}^{\bullet}_{a}1s + {}^{\bullet}_{u}{}^{*}2s$

Experimental asymmetries (Garching). F. Lindner, Ph.D. Thesis. λ =760 nm, $\tau_{\rm p}$ =5 fs .

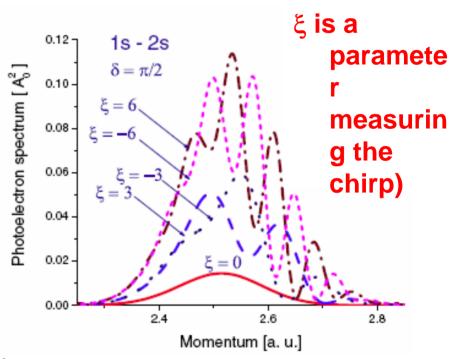


- •Phys. Rev. A, 70, 013815 (2004)
- •Opt. Lett. 29, 1557 (2004)

To exploit attosecond technology

Chirped pulses (formed naturally with attosecond pulse technology) are as effective as transform limited pulses of the same bandwidth.

1. With a chirped pulse, all dynamic information is gained with only a single pump-probe time delay



Yudin et al., Phys. Rev. Lett. 96, 063002 (2006)

Attosecond photoionization of a coherent superposition of bound and dissociative molecular states: effect of nuclear motion

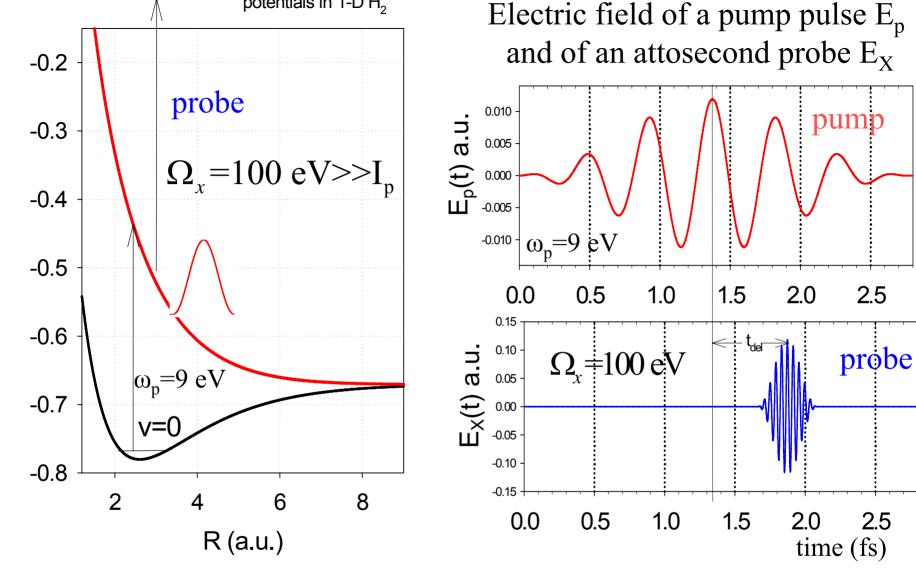
by

André D. Bandrauk,

with S.Chelkowski, G.L. Yudin, Université de Sherbrooke, Canada

> P.B. Corkum (Ottawa), J. Manz (Berlin)

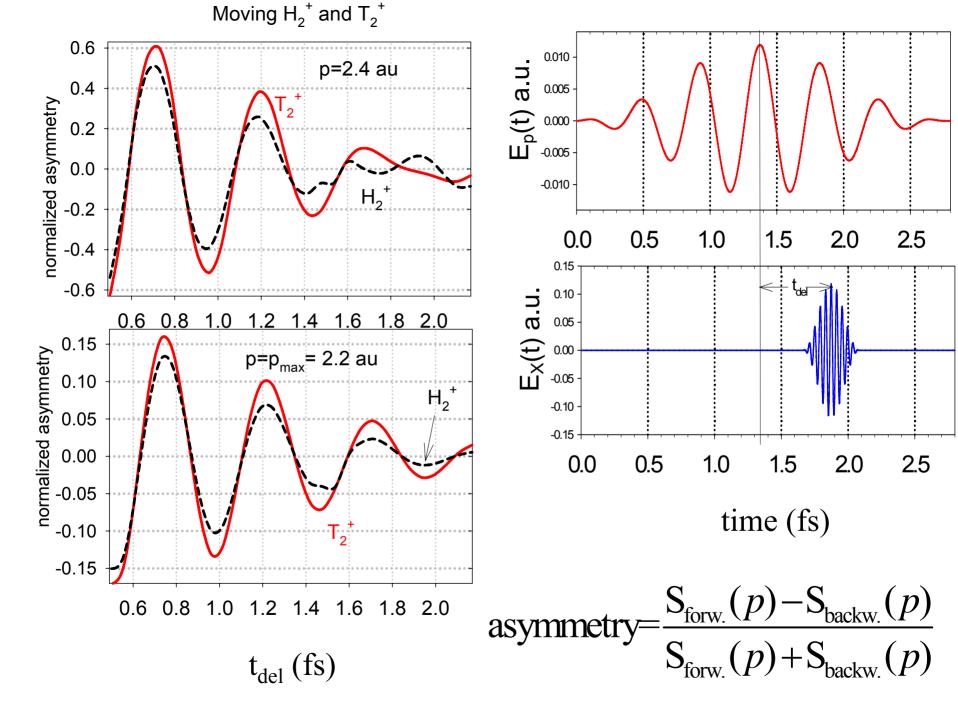
> > to appear J Phys B 2009

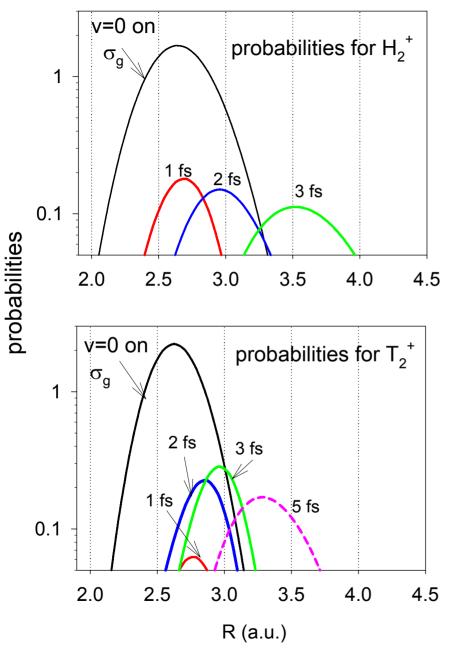


potentials in 1-D H₂⁺

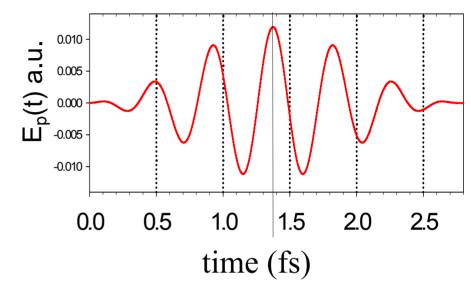
potential (a.u.)

We solved the TDSE for a series of delays: $t_{del} = 0.5 \text{fs} + k \text{ T}_1 / 8$, k=0,1,.. $T_1=2\pi/\omega_p$. We calculated the forward and backward) photoelectron spectra S(p) T1/8 = 0.46 fs





Wave packet motion induced by the pump shown below:

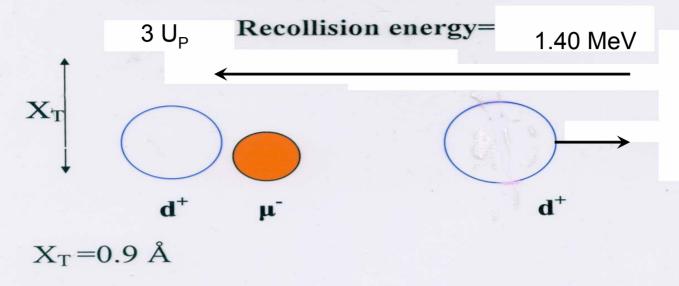


We show the initial v=0 vibrational state and the dissociating packets on σ_{u}

Conclusion: at t>3 fs (t_{del} > 1.5fs we loose the overlap in H_2^+ . This agrees with the attenuation seen in the previous slide

Decoherence ?: see Zurek, PRD 47,488(1993)

RECOLLISION OF d⁺ WITH A μd ATOM INITIATED BY A SUPER-INTENSE LASER



ELECTRIC FIELD E(t) OF THE LASER

 $\lambda = 800 \text{ nm}$

THUS THE LASER CAN INITIATE

A NUCLEAR REACTION, e.g.

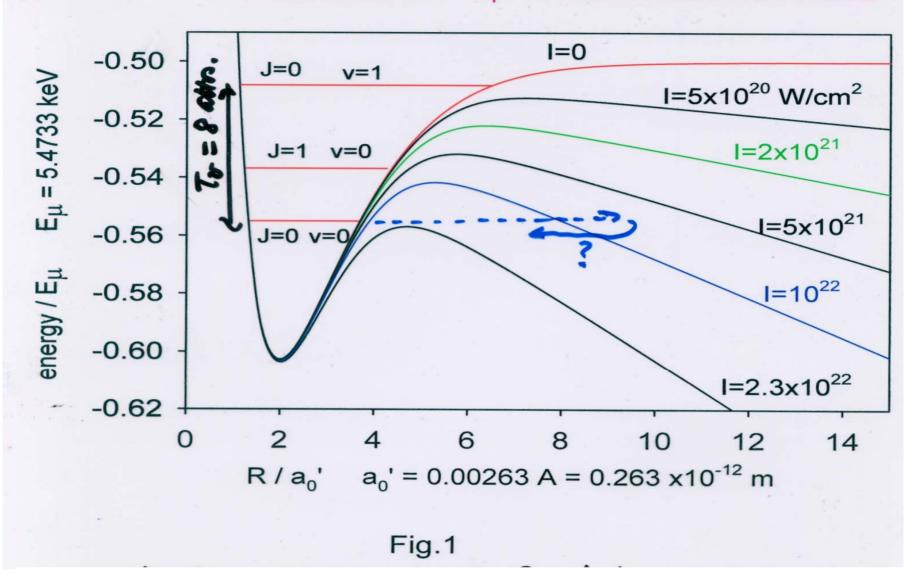
 $I - 3 \times 10^{22} \text{ W/cm}^2$

$$d + d \rightarrow n (2.45 \text{ MeV}) + ^{3}\text{He} (0.82 \text{ MeV})$$

or

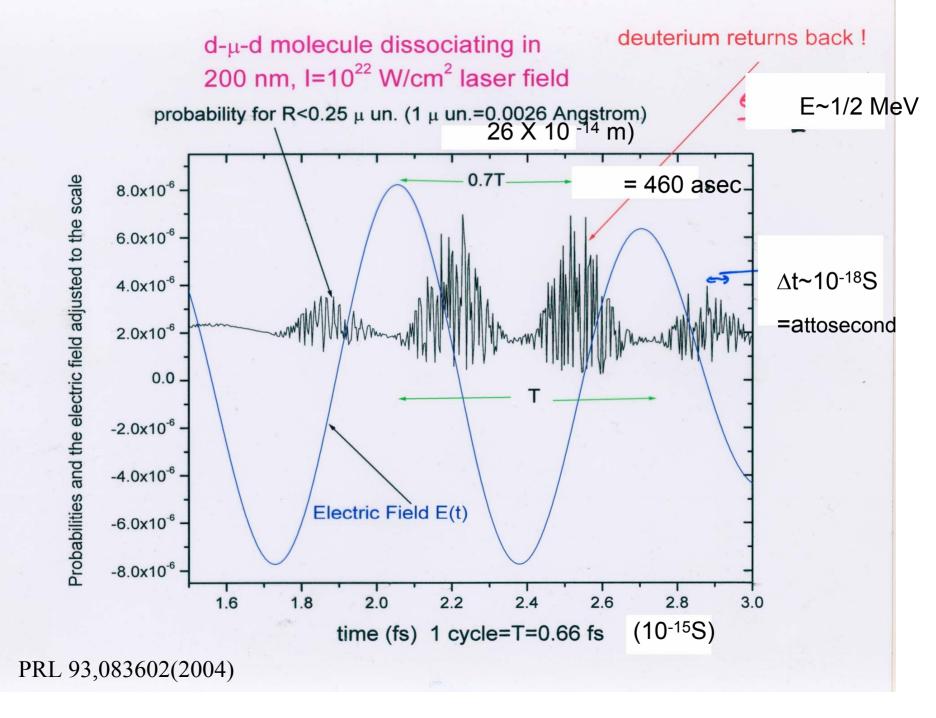
$$d + t \rightarrow n (14.1 \text{ MeV}) + ^{4}\text{He} (3.5 \text{ MeV})$$

POTENTIALS OF d-µ-d IN DC ELECTRIC FIELD



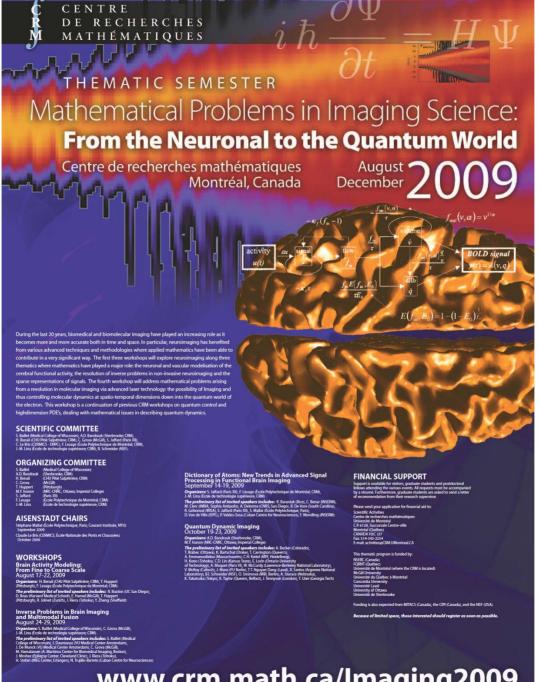
H+₂ (Bond Softening / PH. Bucksbaum PR '90-92

A_{R2}⁺ (Bond Softening / ADB (JCP 1981)



Mathematical Problems

- 1. High order SOT
- 2. Multiscale time frequency analysis
- 3. Infinite D Optimal Control theory
- 4. High order NLS
- 5. Relativistic QM
- 6. Molecular movies (Dynamic Imaging of Electrons-Nuclei)



www.crm.math.ca/lmaging2009

SCIENCE / TECHNOLOGY

APRIL 13, 2009 VOLUME 87, NUMBER 15 PP. 50-51
ACS MEETING NEWS

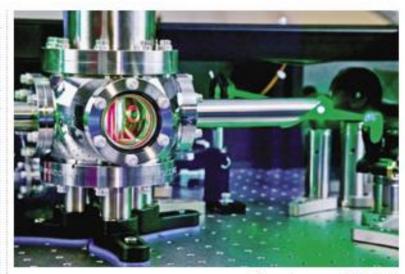
Tracking Electrons

Attosecond science opens the door to real-time observation and control of electron dynamics

Jyllian Kemsley

TEN YEARS AGO, Ahmed H. Zewail won the Nobel Prize for using femtosecond spectroscopy to study atomic motions during chemical reactions. Emerging now from Zewail's pioneering work is the ability to use femtosecond laser pulses to monitor attosecond-scale electron dynamics, which was the focus of a Divison of Physical Chemistry symposium on attosecond science at the American Chemical Society national meeting in Salt Lake City last month.

"There's a whole class of processes associated with electron dynamics that occur at a femtosecond timescale or less," Daniel M. Neumark, a chemistry professor at the University of California, Berkeley, said at the meeting. "These are electron dynamics processes that don't require nuclear motion. To probe them you need attosecond-scale pulses."

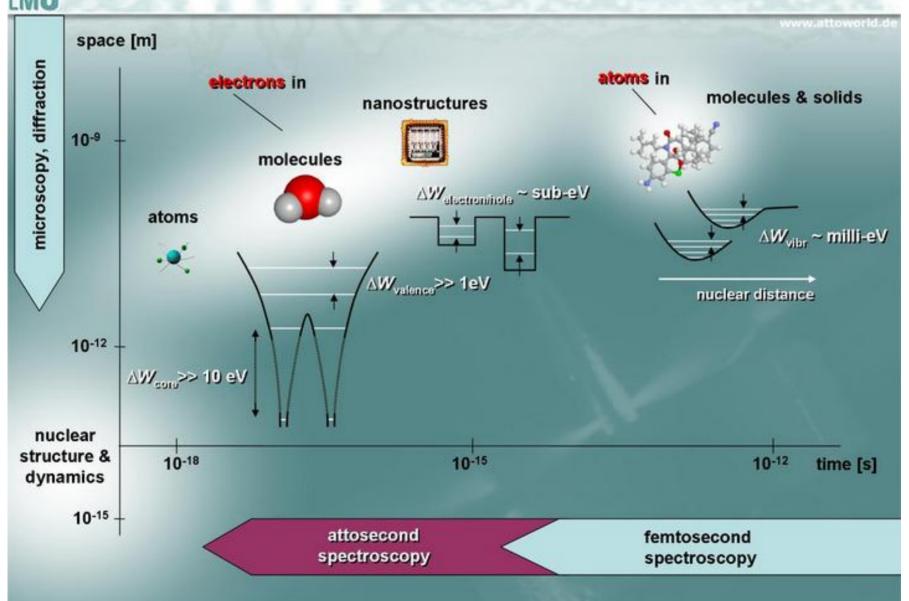


Gary Larson © 2004

ULTRAFAST A cryostat contains a cooled Ti:sapphire laser amplifier crystal that is used to generate high-power femtosecond pulses for attosecond experiments.



structure and dynamics in the microcosm



Quantum mechanics for plants

Graham R. Fleming and Gregory D. Scholes

To what extent do photosynthetic organisms use quantum mechanics to optimize the capture and distribution of light? Answers are emerging from the examination of energy transfer at the submolecular scale.

he first law of photosynthetic economics is: "A photon saved is a photon earned." Research into the factors behind this principle has been burgeoning, and has recently culminated in a paper in *Physical Review Letters* by Jang et al.¹ in which the authors look at photosynthetic energy transfer at the quantum level.

Plants use solar antennae to capture incident photons and transmit the excitation energy to reaction centres, where it is used to initiate the primary electron transfer reactions of photosynthesis. These antennae are one of nature's supreme examples of nanoscale engineering, and are constructed from specialized light-har vesting complexes formed of proteins that bind chlorophylls and carotenoids. Photon collection involves up to several hundred light-absorbing molecules, or chromophores. Hundreds of energy-transfer steps over a hierarchy of time scales and distances, which often occur with near-perfect efficiency2, are therefore required to collect and trap solar energy.

More than 50 years ago, Theodore Förster described a method for calculating the rate of energy transfer between molecules from the overlap of the donor molecule's fluorescence spectrum and the acceptor molecule's absorption spectrum³⁴. The theory has had an enormous impact on biology, chemistry and physics. Collectively, high-resolution structural models, ultrafast spectroscopy and quantum chemical calculations have helped to expose the complex and, in some cases, subtle relationships between structure and light-harvesting in photosynthetic systems. Indeed, it has turned out that there are only a

few cases in which the energy transfer within photosynthetic light-harvesting complexes can be correctly characterized by conventional Förster theory. Moreover, the realization that the concepts elucidated during the study of light-harvesting proteins are general chromophoricassemblies.

To understand the dynamics of lightharvesting and light-trapping in photosynthesis, certain design features must be taken into account. For example, the distances between the molecules are often smaller than the overall size of each molecule. In this confined geometry, energy transfer is governed by how the donor 'sees' the acceptor on the submolecular scale at which the fine differences in the shape of the wavefunctions between the ground and excited states at the donor-acceptor junction become significant (Fig. 1). At this level of spatial confinement, transitions and energy levels that would be ineffective, or even inoperative, in

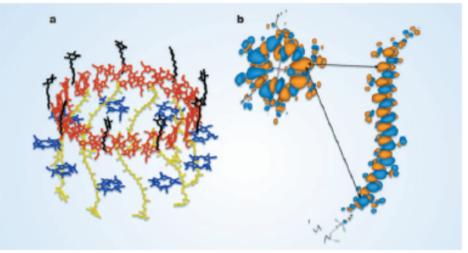
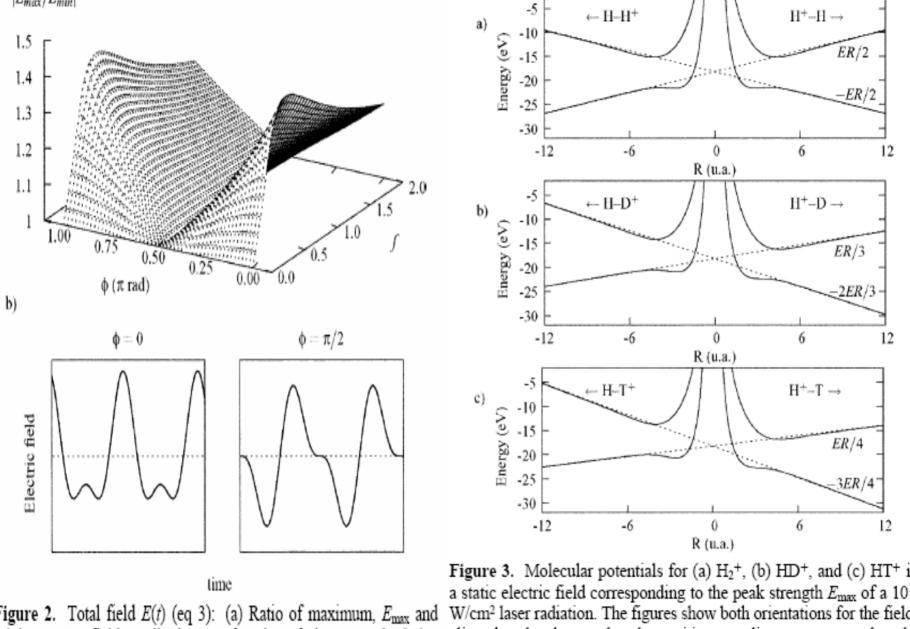


Figure 1 Designs for energy transfer. a, Chromophores in a model of light-harvesting complex (LH) 2 from the bacterium *Rhodopseudomonas acidophila* (Fig. 2), radius 3.4 nm. B800 bacteriochlorophyll molecules (blue) are widely spaced and constitute simple donors, but the B850 molecules (red) interact strongly and constitute a complex acceptor in a confined geometry. Through such interactions between molecules, photosynthetic organisms employ quantum mechanics to funnel absorbed photons to the reaction centre. On time scales of less than 1 picosecond, energy flows from the 800-nm-absorbing B800 molecules to the 850-nm-absorbing B850 molecules, and from the carotenoids (yellow) to both B800 and B850. b, A real-space picture of electronic interactions between molecules on a submolecular scale, as seen in the transition densities of LH2 bacteriochlorophyll (left) and carotenoid (right) molecules calculated from ground- and excited-state wavefunctions. The different colours represent the sign of the electron density. Instead of one average separation between donor and acceptor defining the energy transfer rate, as in Förster theory, there are clearly many length scales (examples arrowed) over which the various parts of the donor and acceptor electron densities interact.



 E_{max}/E_{min}

Figure 2. Total field E(t) (eq 3): (a) Ratio of maximum, E_{\max} and minimum E_{\min} field amplitudes as a function of phase ϕ and relative amplitude f, (b) ratio $E(t)/E_0$ for phases $\phi=0$ and $\phi=\pi/2$, $I_0=4.4$ Show the potentials for (a) H_2^+ , (b) HD^+ , and (c) HT^+ in a static electric field corresponding to the peak strength E_{\max} of a 10^{14} W/cm² laser radiation. The figures show both orientations for the field-aligned molecules, so that the positive coordinates correspond to the proton being upfield (H_2^+, DH^+, TH^+) , while the negative coordinates show the potentials for the orientation with the proton downfield (H_2^+, DH^+, TH^+) .

We enumerate the pairs of integrators with second, third and fourth-order accuracies for later numerical comparison:

$$S_2^A = e^{\gamma \lambda A} e^{\lambda B} e^{\gamma^* \lambda A} + (O(\lambda^3)), \quad \gamma = \frac{1}{2}, \tag{18}$$

$$S_2^B = e^{\gamma \lambda B} e^{\lambda A} e^{\gamma^* \lambda B} + (O(\lambda^3)), \quad \gamma = \frac{1}{2},$$
 (19)

$$S_3^{\text{B}} = e^{\gamma \lambda B/2} \, e^{\gamma \lambda A} \, e^{\lambda B/2} \, e^{\gamma^* \lambda A} \, e^{\gamma^* \lambda B/2} + (\mathbf{O}(\lambda^4)),$$

$$\gamma = \frac{1}{2} \left(1 \pm \frac{i}{\sqrt{3}} \right), \tag{20}$$

$$S_3' = \frac{1}{2} (e^{\gamma \lambda B} e^{\lambda A} e^{\gamma^* \lambda B} + e^{\gamma \lambda A} e^{\lambda B} e^{\gamma^* \lambda A}) + (O(\lambda^4)),$$

$$\gamma = \frac{1}{2} \left(1 \pm \frac{i}{\sqrt{3}} \right),\tag{21}$$

$$S_4^B = e^{\gamma \lambda B/2} e^{\gamma \lambda A} e^{(1-\gamma)\lambda B/2} e^{(1-2\gamma)\lambda A} e^{(1-\gamma)\lambda B/2} e^{\gamma \lambda A} e^{\gamma \lambda B/2} + (O(\lambda^5)), \quad \gamma = (2-2^{1/3})^{-1},$$
(22)

$$S_4' = \frac{1}{2} \left(e^{\gamma \lambda B/2} e^{\lambda A/2} e^{\gamma^* \lambda B} e^{\lambda A/2} e^{\gamma \lambda B/2} + e^{\gamma \lambda A/2} e^{\lambda B/2} e^{\gamma^* \lambda A} e^{\lambda B/2} e^{\gamma \lambda A/2} \right)$$

$$+ (O(\lambda^5)), \quad \gamma = \frac{1}{2} \pm \frac{i}{2\sqrt{3}}.$$
 (23)

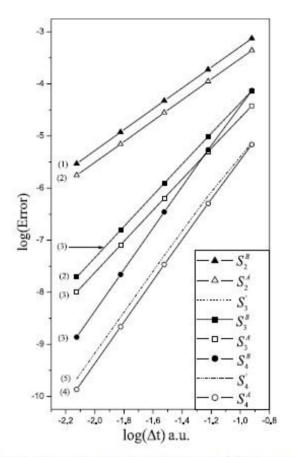


Fig. 1. Test of convergence for ground state of H₂⁺ (Eqs. 25 and 26). m corresponds to number of FFT's, N_{FFT}.

Japanese connection

