



Fluctuation relations for manipulated systems

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Motivations

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❖ Outline

Fluctuation relations
for the work

Applications

Fluctuation relations
for the entropy
production

- Remarkable, deceptively simple relations obeyed by fluctuation statistics in out-of-equilibrium systems
- Relevant for small systems, where fluctuations are large
- Provide tools for the investigation of equilibrium properties of small systems
- Provide structure to the “thermodynamic” properties of non-equilibrium steady states
- Could be relevant to understand the working of biological system, like biological motors



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- Fluctuation relations for the work:
 - ◆ Jarzynski's relation and its applications
 - ◆ Crooks's relation and its consequences
 - ◆ Experiments and applications to biomolecules
 - ◆ On the estimation of the free energy



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 - ✦ Crooks's relation and its consequences
 - ✦ Experiments and applications to biomolecules
 - ✦ On the estimation of the free energy
- Fluctuation relations for the entropy:
 - ✦ Entropy production in a single trajectory:
Seifert's relation and its consequences
 - ✦ Entropy production in nonequilibrium steady
states: The Gallavotti-Cohen relation



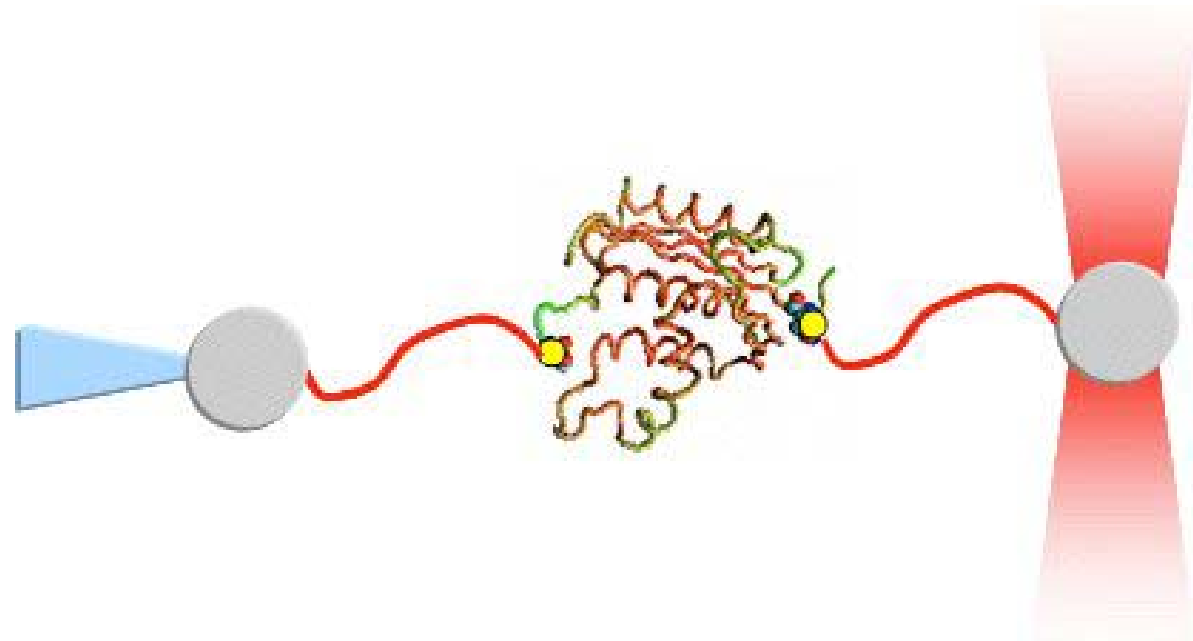
Nanomanipulation

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- ❖ Dissipation and irreversibility



Manipulation of a biopolymer by optical tweezers



Manipulating a system out of equilibrium

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❖ Dissipation and irreversibility

- A system at equilibrium at inverse temperature β :

$$Z_\mu = \text{Tr}_x e^{-\beta H(x,\mu)} = e^{-\beta F(\mu)}$$

- Manipulation protocol:

$$\mu : \mu = \mu(t); \quad \mu(0) = \mu_0; \quad \mu(t_f) = \mu_f$$

- Fluctuating work:

$$\mathcal{W}[x, \mu] = \int_0^{t_f} dt \dot{\mu}(t) \left. \frac{\partial H}{\partial \mu} \right|_{x(t), \mu(t)}$$



Manipulating a system out of equilibrium

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$$\mu : \mu = \mu(t); \quad \mu(0) = \mu_0; \quad \mu(t_f) = \mu_f$$

- Fluctuating work:

$$\mathcal{W}[x, \mu] = \int_0^{t_f} dt \dot{\mu}(t) \left. \frac{\partial H}{\partial \mu} \right|_{x(t), \mu(t)}$$
$$\langle \mathcal{W} \rangle \geq F(\mu_f) - F(\mu_0)$$



Jarzynski's relation (1997)

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What does nonequilibrium tell us of equilibrium?

$$\langle e^{-\beta \mathcal{W}} \rangle = e^{-\beta (F(\mu_f) - F(\mu_0))}$$

Average over all realizations of the system history $x(t)$:

$$\langle \dots \rangle = \text{Tr}_{\mathcal{X}} \mathcal{P}[x] \dots$$



Jarzynski's relation (1997)

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Average over all realizations of the system history $x(t)$:

$$\langle \dots \rangle = \text{Tr}_{\mathcal{X}} \mathcal{P}[\mathbf{x}] \dots$$

Assumptions:

- Evolution equation for $P(x, t)$: $\partial_t P = -\hat{\mathcal{H}}_{\mu(t)} P$
- Equilibrium distribution: $\hat{\mathcal{H}}_{\mu} P^{\text{eq}} = 0$

$$P^{\text{eq}}(x, \mu) = \frac{e^{-\beta H(x, \mu)}}{Z_{\mu}}$$



A simple proof

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$$\psi(x, t) = \int \mathcal{D}x \, \delta(x(t) - x) e^{-\beta \mathcal{W}[x]} \mathcal{P}[x]$$

Evolution equation:

$$\frac{\partial \psi}{\partial t} = -\hat{\mathcal{H}}_{\mu(t)} \psi - \beta \dot{\mu}(t) \frac{\partial H}{\partial \mu} \psi$$

Initial condition:

$$\psi(x, 0) = e^{-\beta H(x, \mu(0))} / Z_{\mu(0)}$$

Thus

$$\psi(x, t) = \frac{1}{Z_{\mu(0)}} e^{-\beta H(x, \mu(t))}$$



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- Sudden change in the hamiltonian: $\mu_0 \longrightarrow \mu_1$

$$\begin{aligned}\langle e^{-\beta \mathcal{W}} \rangle &= \sum_x e^{-\beta(H(x, \mu_1) - H(x, \mu_0))} \frac{e^{-\beta H(x, \mu_0)}}{Z_{\mu_0}} \\ &= \frac{1}{Z_{\mu_0}} \sum_x e^{-\beta H(x, \mu_1)} = \frac{Z_{\mu_1}}{Z_{\mu_0}}\end{aligned}$$



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- Sudden change in the hamiltonian: $\mu_0 \longrightarrow \mu_1$

$$\begin{aligned}\langle e^{-\beta \mathcal{W}} \rangle &= \sum_x e^{-\beta(H(x, \mu_1) - H(x, \mu_0))} \frac{e^{-\beta H(x, \mu_0)}}{Z_{\mu_0}} \\ &= \frac{1}{Z_{\mu_0}} \sum_x e^{-\beta H(x, \mu_1)} = \frac{Z_{\mu_1}}{Z_{\mu_0}}\end{aligned}$$

- Adiabatic change: $\mu(t) = \ell(t/t_f)$,
 $\ell(0) = \mu_0$, $\ell(1) = \mu_1$

$$\begin{aligned}\mathcal{W}[\mathbf{x}] &\simeq \int_0^{t_f} dt' \left\langle \frac{\partial H(x(t'), \mu)}{\partial \mu} \right\rangle_{\mu=\mu(t')} \dot{\mu}(t') \\ &= \int_{\mu_0}^{\mu_1} d\mu \left\langle \frac{\partial H(x, \mu)}{\partial \mu} \right\rangle_{\mu} = W^{\text{th}}\end{aligned}$$



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- $\Delta F = F(\mu_f) - F(\mu_0)$ is the change in the equilibrium free energy with the corresponding values of the parameter μ



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- $\Delta F = F(\mu_f) - F(\mu_0)$ is the change in the equilibrium free energy with the corresponding values of the parameter μ
- One assumes that the system is initially at equilibrium



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- $\Delta F = F(\mu_f) - F(\mu_0)$ is the change in the equilibrium free energy with the corresponding values of the parameter μ
- One assumes that the system is initially at equilibrium
- The system is not at equilibrium at the end of the process



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- $\Delta F = F(\mu_f) - F(\mu_0)$ is the change in the equilibrium free energy with the corresponding values of the parameter μ
- One assumes that the system is initially at equilibrium
- The system is not at equilibrium at the end of the process
- The average is over all realizations of the process: If the process is deterministic, over the initial conditions



Derivation: A Markov chain

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$$x \in \{1, 2, \dots, q\} \quad t \in \{0, 1, 2, \dots\}$$

Equilibrium distribution determined by hamiltonian $H(x, \mu)$:

$$P^{\text{eq}}(x, \mu) = \frac{e^{-\beta H(x, \mu)}}{Z_\mu} \quad Z_\mu = \sum_x e^{-\beta H(x, \mu)} = e^{-\beta F(\mu)}$$

Evolution equation for the probabilities:

$$p(x, t + 1) = \sum_{x'} W_{xx'}(\mu(t)) p(x', t)$$

$$\sum_{x'} W_{xx'}(\mu) P^{\text{eq}}(x', \mu) = P^{\text{eq}}(x, \mu)$$



Work and heat in a reversible transformation

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$$E(\mu) = \langle H(\mu) \rangle_\mu \quad \langle A \rangle_\mu = \sum_x A(x) P^{\text{eq}}(x, \mu)$$

Manipulation: $\mu \longrightarrow \mu + d\mu$:

$$dE(\mu) = \underbrace{\sum_x dH(x, \mu) P^{\text{eq}}(x, \mu(t))}_{dW} + \underbrace{\sum_x H(x, \mu) dP^{\text{eq}}(x, \mu)}_{dQ}$$

$$\begin{aligned} dS^{\text{eq}} &= -k_B d \left(\sum_x P^{\text{eq}}(x, \mu) \ln P^{\text{eq}}(x, \mu) \right) \\ &= -k_B \sum_x \ln P^{\text{eq}}(x, \mu) dP^{\text{eq}}(x, \mu) \\ &= \frac{1}{T} \sum_x H(x, \mu) dP^{\text{eq}}(x, \mu) \end{aligned}$$



Path work and heat under manipulation

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Path $x = (x(0), x(1), x(2), \dots, x(t_f))$

Manipulation protocol: $\mu = (\mu(0), \mu(1), \dots, \mu(t_f - 1))$

$$\mathcal{W}[x] = \sum_{t=0}^{t_f-1} [H(x(t+1), \mu(t+1)) - H(x(t+1), \mu(t))]$$

$$\mathcal{Q}[x] = \sum_{t=0}^{t_f-1} [H(x(t+1), \mu(t)) - H(x(t), \mu(t))]$$

$$\Delta E = H(x(t_f), \mu(t_f)) - H(x(0), \mu(0)) = \mathcal{W} + \mathcal{Q}$$

N.B. \mathcal{W} and \mathcal{Q} depend on the whole path, ΔE depends only on initial and final states



Path probability

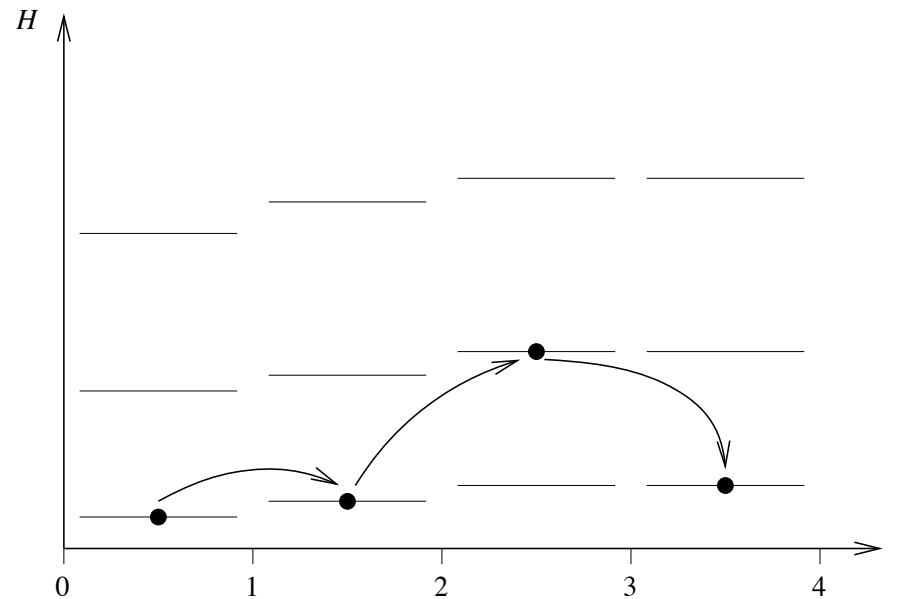
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$$\mathcal{P}[x \mid x(0)] = W_{x(t_f), x(t_f-1)}(\mu(t_f - 1)) \times \cdots \\ \cdots \times W_{x(2), x(1)}(1) W_{x(1), x(0)}(\mu(0))$$

N.B.: The transitions take place with the “old” probability $W_{xx'}(\mu(t - 1))$, then the energy changes to $H(x, \mu(t))$



Reversed path

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Time-reversal invariant states: $\mathcal{I}x = x$

$$\begin{aligned} \mathbf{x} &= (x(0), x(1), \dots, x(t_f - 1), x(t_f)) \\ \widehat{\mathbf{x}} &= (x(t_f), x(t_f - 1), \dots, x(1), x(0)) \\ \widehat{x}(t) &= x(t_f - t) = x(\widehat{t}) \end{aligned}$$

Time-reversed protocol:

$$\widehat{\mu} : \quad \widehat{\mu}(t) = \mu(t_f - t) = \mu(\widehat{t}) \quad t = 1, \dots, t_f$$

Probability of the reversed transition:

$$\widehat{W}_{xx'}(\mu) : \quad \widehat{W}_{xx'}(\mu) P^{\text{eq}}(x', \mu) = W_{x'x}(\mu) P^{\text{eq}}(x, \mu)$$



Reversed path

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Probability of the reversed transition:

$$\widehat{W}_{xx'}(\mu) : \quad \widehat{W}_{xx'}(\mu) P^{\text{eq}}(x', \mu) = W_{x'x}(\mu) P^{\text{eq}}(x, \mu)$$

(Detailed balance: $\widehat{W}_{xx'}(\mu) = W_{x'x}(\mu)$)



Probability of the reversed path

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$$W_{xx'}(\mu(t)) = W_{xx'}(t) \quad \text{etc.}$$

$$\widehat{W}_{xx'}(t) = W_{x'x}(t_f - t) \quad t = 1, \dots, t_f$$

$$\begin{aligned} \widehat{\mathcal{P}}[\widehat{\mathbf{x}} \mid \widehat{\mathbf{x}}(0)] &= \widehat{W}_{\widehat{\mathbf{x}}(t_f), \widehat{\mathbf{x}}(t_f-1)}(t_f) \cdots \widehat{W}_{\widehat{\mathbf{x}}(2), \widehat{\mathbf{x}}(1)}(2) \widehat{W}_{\widehat{\mathbf{x}}(1), \widehat{\mathbf{x}}(0)}(1) \\ &= W_{x(1), x(0)}(0) e^{\beta(H(x(1), \mu(0)) - H(x(0), \mu(0)))} \\ &\quad \times W_{x(2), x(1)}(1) e^{\beta(H(x(2), \mu(1)) - H(x(1), \mu(1)))} \times \dots \\ &\quad \times W_{x(t_f), x(t_f-1)}(t_f - 1) \\ &\quad \times e^{\beta(H(x(t_f), \mu(t_f-1)) - H(x(t_f-1), \mu(t_f-1)))} \\ &= \mathcal{P}[\mathbf{x} \mid x(0)] e^{\beta \mathcal{Q}[\mathbf{x}]} \end{aligned}$$



Crooks's relation (1999)

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$$\mathcal{P}[\mathbf{x}] = \mathcal{P}[\mathbf{x} \mid x(0)] P^{\text{eq}}(x(0), \mu(0))$$

$$\begin{aligned} \hat{\mathcal{P}}[\hat{\mathbf{x}}] &= \hat{\mathcal{P}}[\hat{\mathbf{x}} \mid \hat{x}(0)] e^{-\beta H(\hat{x}(0), \hat{\mu}(0))} / Z_{\hat{\mu}(0)} \\ &= \mathcal{P}[\mathbf{x} \mid x(0)] e^{\beta \{Q[\mathbf{x}] - H(x(t_f), \mu(t_f))\}} / Z_{\mu(t_f)} \\ &= \mathcal{P}[\mathbf{x}] e^{\beta \{Q[\mathbf{x}] - (H(x(t_f), \mu(t_f)) - H(x(0), \mu(0)))\}} (Z_{\mu(0)} / Z_{\mu(t_f)}) \end{aligned}$$

$$Q[\mathbf{x}] - \Delta H = -\mathcal{W}[\mathbf{x}] \quad Z_{\mu(0)} / Z_{\mu(t_f)} = e^{\beta \Delta F}$$

$$\hat{\mathcal{P}}[\hat{\mathbf{x}}] = \mathcal{P}[\mathbf{x}] e^{-\beta(\mathcal{W}[\mathbf{x}] - \Delta F)}$$

Summing over \mathbf{x} :

$$1 = \langle e^{-\beta(\mathcal{W} - \Delta F)} \rangle$$



Dissipation and irreversibility

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$$e^{\beta(\mathcal{W}[\mathbf{x}] - \Delta F)} = \frac{\mathcal{P}[\mathbf{x}]}{\hat{\mathcal{P}}[\hat{\mathbf{x}}]}$$

Thus

$$W^d = \langle \mathcal{W} \rangle - \Delta F = \underbrace{\beta^{-1} \sum_{\mathbf{x}} \mathcal{P}[\mathbf{x}] \ln \frac{\mathcal{P}[\mathbf{x}]}{\hat{\mathcal{P}}[\hat{\mathbf{x}}]}}_{D(\mathcal{P}[\mathbf{x}] \| \hat{\mathcal{P}}[\hat{\mathbf{x}}])}$$

van den Broeck et al., 2007

Summing over x with a given value of W :

$$W^d = \beta^{-1} \int dW P(W) \ln \frac{P(W)}{\hat{P}(-W)} = \beta^{-1} D(P(W) \| \hat{P}(-W))$$



The work distribution

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Fluctuation relations
for the work

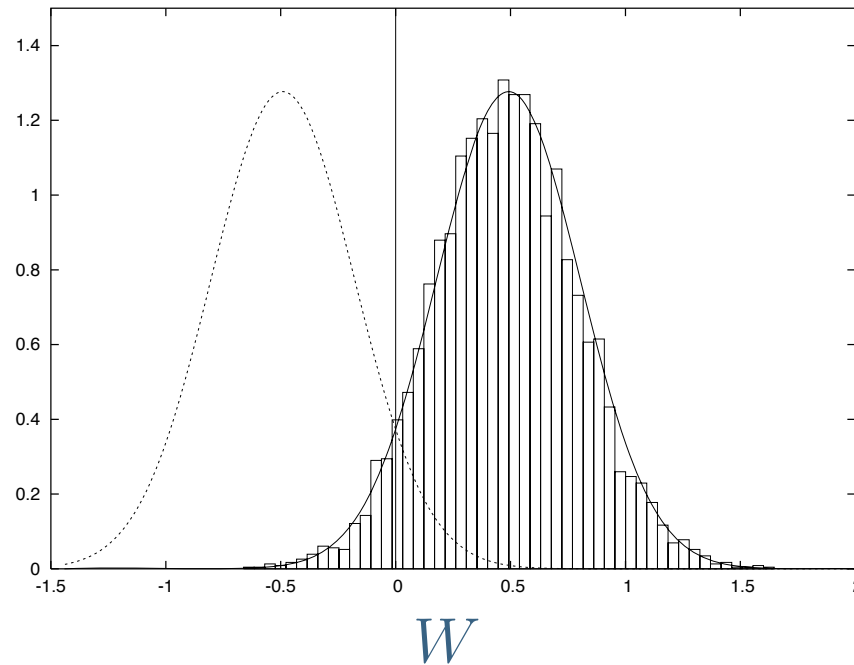
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❖ The work
distribution

- ❖ A RNA hairpin
- ❖ The Ig27 domain
of human titin
- ❖ The free-energy
landscape
- ❖ The Jarzynski
estimator: A REM
partition function

Fluctuation relations
for the entropy
production

Nonlinear oscillator: $H(x, \mu) = H_0(x) - \mu x$, $\hat{\mu} = -\mu$





The work distribution

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Fluctuation relations
for the work

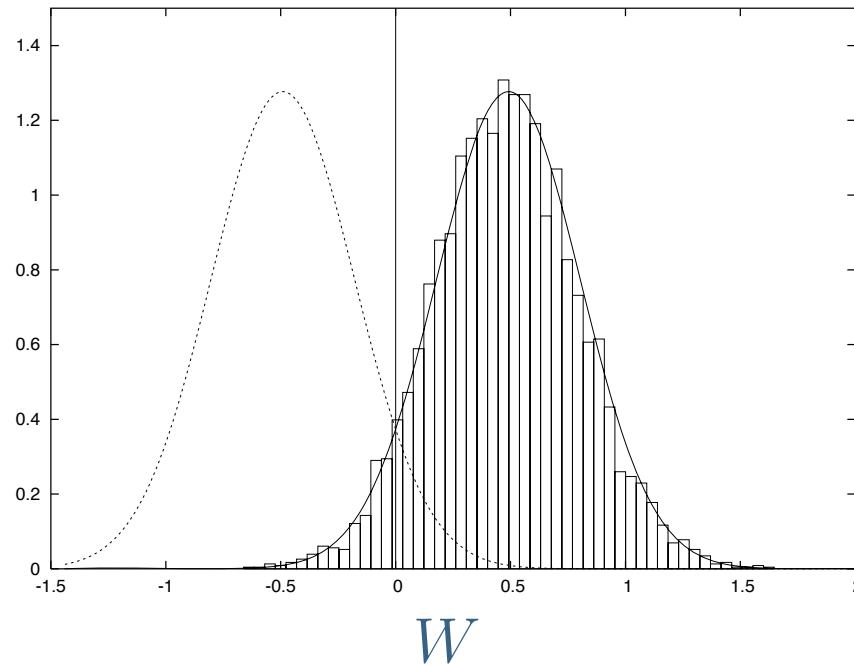
Applications

❖ The work
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- ❖ A RNA hairpin
- ❖ The Ig27 domain
of human titin
- ❖ The free-energy
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- ❖ The Jarzynski
estimator: A REM
partition function

Fluctuation relations
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Nonlinear oscillator: $H(x, \mu) = H_0(x) - \mu x$, $\hat{\mu} = -\mu$



Dotted line: $P(W) e^{-\beta W} = P(-W)$

The value of W that contributes most to Jarzynski's equality is the most probable in the inverse manipulation



The work distribution

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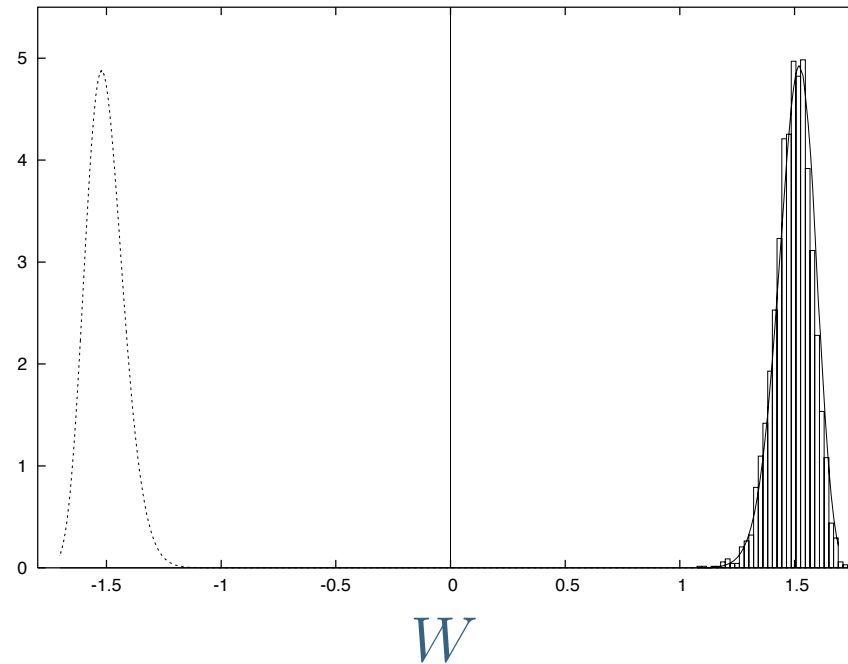
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Fluctuation relations
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Nonlinear oscillator: $H(x, \mu) = H_0(x) - \mu x$, $\hat{\mu} = -\mu$



For larger systems, fluctuations which contribute to Jarzynski's equality are exceedingly rare



Fluctuation relations for the entropy production





The Ig27 domain of human titin

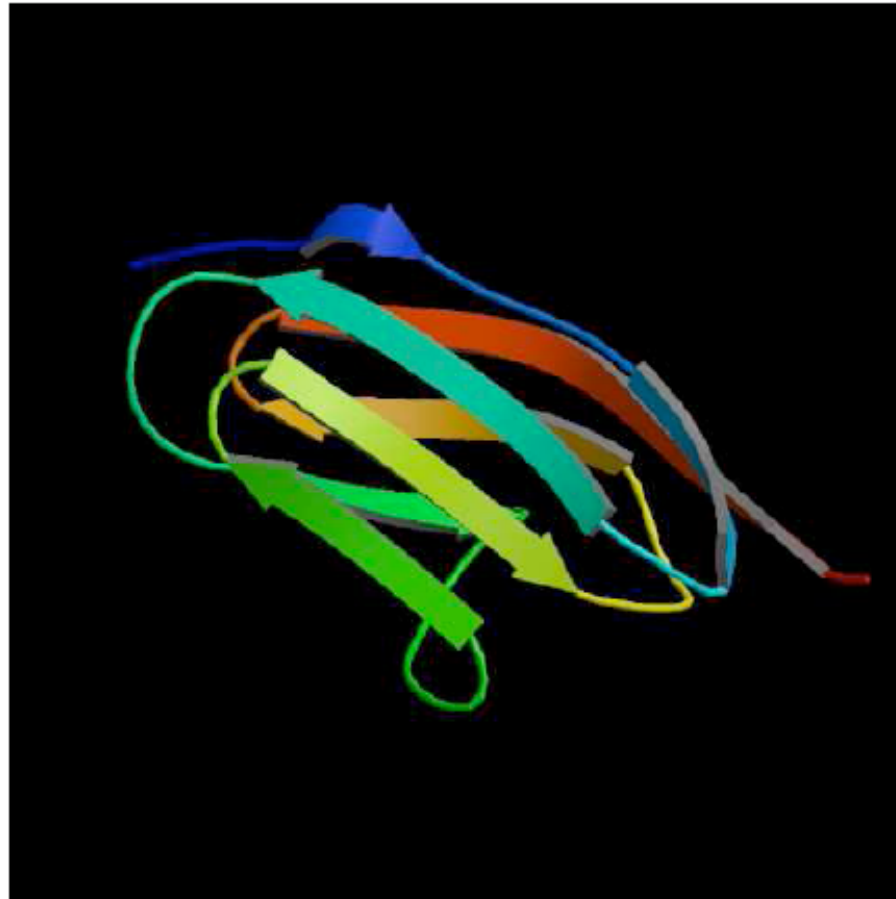
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89 aminoacids, $\epsilon/k_B = 43$ K



The free-energy landscape

A. Imparato, S. Luccioli, A. Torcini, 2007

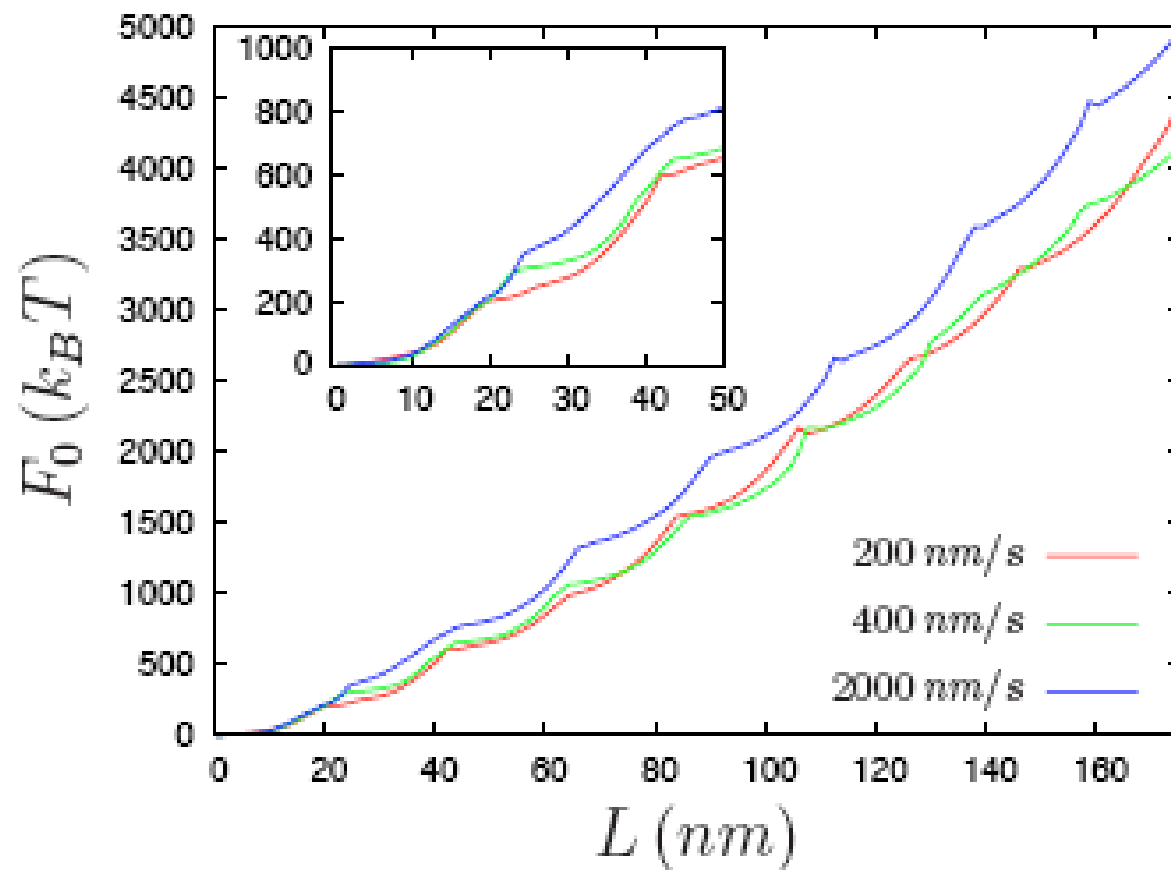
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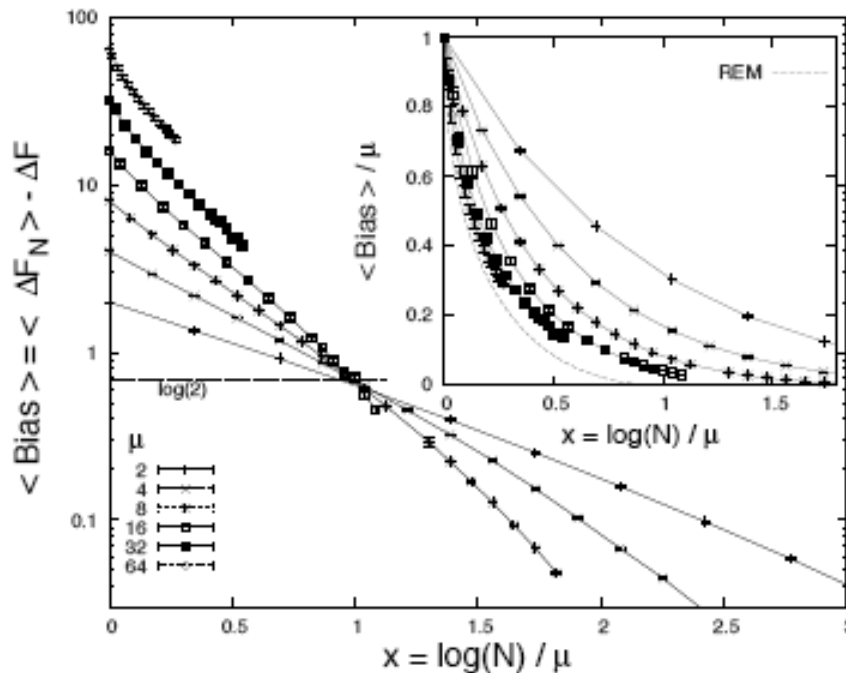
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Palassini and Ritort, 2008



$$\Delta F_N = -\frac{1}{N\beta} \sum_{k=1}^N e^{-\beta W_k}$$



The Jarzynski estimator: A REM partition function

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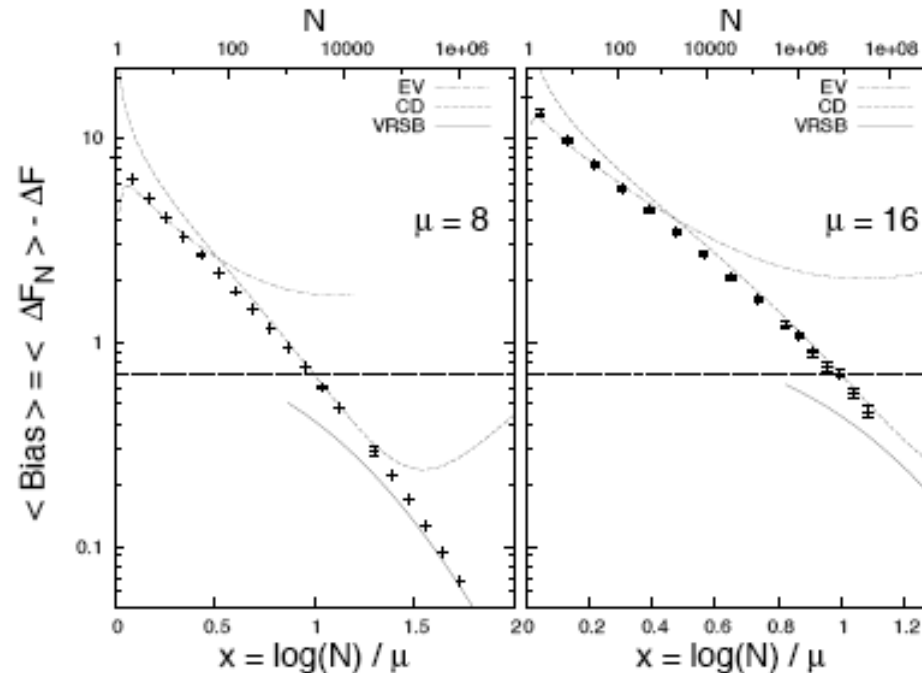
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$$\Delta F_N = -\frac{1}{N\beta} \sum_{k=1}^N e^{-\beta W_k}$$



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Detailed balance:

$$W_{xx'} e^{-\beta H(x')} = W_{x'x} e^{-\beta H(x)}$$

$$\Downarrow$$

$$\frac{W_{xx'}}{W_{x'x}} = e^{-\beta(H(x)-H(x'))} = e^{-\beta Q_{xx'}} = e^{\Delta S_{xx'}^{\text{bath}}/k_B}$$

Generalize by definition: (units $k_B = 1$)

$$\Delta S_{xx'}^{\text{bath}} = \ln \frac{W_{xx'}}{\widehat{W}_{x'x}}$$

$$\frac{\widehat{\mathcal{P}}[\widehat{x} \mid \widehat{x}(0)]}{\mathcal{P}[x \mid x(0)]} = \prod_{t=0}^{t_f-1} \frac{\widehat{W}_{x(t)x(t+1)}}{W_{x(t+1)x(t)}} = e^{-\Delta \mathcal{S}^{\text{bath}}[x]}$$



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Define $\pi_0(x)$ and $\hat{\pi}_0(x)$ arbitrary but normalized,
 $\mathcal{P}[x] = \mathcal{P}[x \mid x(0)]\pi_0(x(0))$ etc.:

$$\ln \frac{\hat{\mathcal{P}}[\hat{x}]}{\mathcal{P}[x]} = -\Delta \mathcal{S}^{\text{bath}}[x] + \ln \frac{\hat{\pi}_0(\hat{x}(0))}{\pi_0(x(0))}$$

Seifert 2005



Fluctuation theorem I (Seifert)

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Let $p(x, t)$ be the solution of

$$p(x, t + 1) = \sum_{x'} W_{xx'}(t) p(x', t) \quad p(x, 0) = p_0(x)$$

$$\pi_0(x) = p_0(x) \quad \hat{\pi}_0(x) = p(x, t_f)$$

Define $S(x, t) = -\ln p(x, t)$, then

$$\ln \frac{\hat{\pi}_0(\hat{x}(0))}{\pi_0(x(0))} = \ln \frac{p(x(t_f), t_f)}{p(x(0), 0)} = -\Delta S$$

$$\Delta \mathcal{S}^{\text{tot}} = \Delta \mathcal{S}^{\text{bath}} + \Delta S$$

$$\hat{\mathcal{P}}[\hat{x}] = \mathcal{P}[x] e^{-\Delta \mathcal{S}^{\text{tot}}[x]} \Rightarrow \boxed{\langle e^{-\Delta \mathcal{S}^{\text{tot}}} \rangle = 1}$$



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- Entropy becomes a fluctuating quantity, related to a single trajectory rather than to an ensemble



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- Entropy becomes a fluctuating quantity, related to a single trajectory rather than to an ensemble
- There is a nonzero probability for $\Delta S^{\text{tot}} < 0$!



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- Entropy becomes a fluctuating quantity, related to a single trajectory rather than to an ensemble
- There is a nonzero probability for $\Delta S^{\text{tot}} < 0$!
- Jensen's inequality: $\langle e^X \rangle \geq e^{\langle X \rangle}$:

$$\langle \Delta S^{\text{tot}} \rangle \geq -\ln \langle e^{-\Delta S^{\text{tot}}} \rangle = 0$$



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- $\Delta S < 0$ can be interpreted as a “transient violation” of the 2nd principle



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- Entropy becomes a fluctuating quantity, related to a single trajectory rather than to an ensemble
- There is a nonzero probability for $\Delta S^{\text{tot}} < 0$!
- Jensen's inequality: $\langle e^X \rangle \geq e^{\langle X \rangle}$:

$$\langle \Delta S^{\text{tot}} \rangle \geq -\ln \langle e^{-\Delta S^{\text{tot}}} \rangle = 0$$

- $\Delta S < 0$ can be interpreted as a “transient violation” of the 2nd principle
- Transient violations are related to Loschmidt's reversibility paradox



Fluctuation theorem II (Jarzynski)

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Let $\pi_0(x) = P^{\text{eq}}(x, \mu_0)$, $\hat{\pi}_0(x) = P^{\text{eq}}(x, \mu_1)$, $\beta = 1/k_B T$:

$$\ln \frac{\hat{\pi}_0(\hat{x}(0))}{\pi_0(x(0))} = e^R = e^{\beta \Delta F - \beta (H(x(t_f), \mu_1) - H(x(0), \mu_0))}$$

$$\Delta \mathcal{S} = -\beta \mathcal{Q}$$

$$\begin{aligned} R &= \beta [\mathcal{Q} + \Delta F - \Delta H(x, \mu)] \\ &= -\beta (\mathcal{W}[x] - \Delta F) \end{aligned}$$

$$\hat{\mathcal{P}}[\hat{x}] = \mathcal{P}[x] e^R \Rightarrow \boxed{\langle e^{-\beta \mathcal{W}} \rangle = e^{-\beta \Delta F}}$$



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$$\widehat{W}_{x'x} = W_{x'x}$$

Detailed balance is violated:

$$W_{xy}W_{yz}W_{zx} \neq W_{zy}W_{yx}W_{xz} \quad \exists x, y, z$$

Thus $\nexists H(x) : W_{x'x}/W_{xx'} = e^{-\beta(H(x')-H(x))}$

Steady state distribution:

$$\sum_{x'} W_{xx'} p^{\text{ss}}(x') = p^{\text{ss}}(x)$$



Fluctuation theorem III (Evans-Searles)

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Choose $\pi_0(x) = \hat{\pi}_0(x) = p^{\text{ss}}(x)$

$$\ln \frac{\hat{\mathcal{P}}[\hat{x}]}{\mathcal{P}[x]} = -\Delta\mathcal{S}^{\text{bath}}[x] - \Delta\mathcal{S}^{\text{ss}}$$

Total entropy production:

$$\Delta\mathcal{S}^{\text{tot}} = \Delta\mathcal{S}^{\text{bath}}[x] + \Delta\mathcal{S}^{\text{ss}}$$

Summing over all paths x with a given value of $\Delta\mathcal{S}^{\text{tot}}$
yields the **fluctuation theorem**:

$$\frac{p(-\Delta\mathcal{S}^{\text{tot}})}{p(\Delta\mathcal{S}^{\text{tot}})} = e^{-\Delta\mathcal{S}^{\text{tot}}}$$



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- The fluctuation theorem holds for finite times, but starting from the steady state



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- The fluctuation theorem holds for finite times, but starting from the steady state
- Since ΔS is bounded, but ΔS^{bath} grows, we have for large t_f

$$\Delta S^{\text{tot}} \simeq \Delta S^{\text{bath}}$$



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- The fluctuation theorem holds for finite times, but starting from the steady state
- Since ΔS is bounded, but ΔS^{bath} grows, we have for large t_f

$$\Delta S^{\text{tot}} \simeq \Delta S^{\text{bath}}$$

- Large-deviation function $\phi(s)$:

$$p(\Delta S^{\text{tot}}) \propto e^{-t_f \phi(\Delta S^{\text{tot}}/t_f)}$$



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- The fluctuation theorem holds for finite times, but starting from the steady state
- Since ΔS is bounded, but ΔS^{bath} grows, we have for large t_f

$$\Delta S^{\text{tot}} \simeq \Delta S^{\text{bath}}$$

- Large-deviation function $\phi(s)$:

$$p(\Delta S^{\text{tot}}) \propto e^{-t_f \phi(\Delta S^{\text{tot}}/t_f)}$$

Gallavotti-Cohen relation:

$$\phi(-s) = \phi(s) + s$$



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Stationary system:

$$\begin{aligned}\Delta\mathcal{S}_{x'x} &= \ln \frac{W_{x'x}}{W_{xx'}} \\ &= \underbrace{\ln \frac{W_{x'x}p^{\text{ss}}(x)}{W_{xx'}p^{\text{ss}}(x')}}_{\Delta\mathcal{S}^{\text{hk}}} - \underbrace{\ln \frac{p^{\text{ss}}(x)}{p^{\text{ss}}(x')}}_{\Delta\mathcal{S}^{\text{ex}}}\end{aligned}$$

N.B.: If detailed balance is satisfied:

$$W_{xx'}p^{\text{ss}}(x') = W_{x'x}p^{\text{ss}}(x)$$

then

$$-\Delta\mathcal{S}_{x'x}^{\text{hk}} = 0 \quad \forall x, x'$$



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$$\varphi_{x'x} = \text{Prob}^{\text{ss}}(x \longrightarrow x') = W_{x'x} p^{\text{ss}}(x)$$

$$\varphi_{x'x} \geq 0 \quad \sum_{x'x} \varphi_{x'x} = 1$$

$$\hat{\varphi}_{x'x} = \varphi_{xx'} \quad \sum_{x'x} \hat{\varphi}_{x'x} = \sum_{xx'} \varphi_{xx'} = 1$$

$$\begin{aligned} \langle \Delta \mathcal{S}^{\text{hk}} \rangle^{\text{ss}} &= \sum_{xx'} \ln \frac{W_{x'x} p^{\text{ss}}(x)}{W_{xx'} p^{\text{ss}}(x')} W_{x'x} p^{\text{ss}}(x) \\ &= \sum_{xx'} \ln \frac{\varphi_{x'x}}{\varphi_{xx'}} \varphi_{x'x} = D(\hat{\varphi} \| \varphi) \geq 0 \end{aligned}$$



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Since

$$\Delta S_{x'x}^{\text{hk}} = \ln \frac{\varphi_{x'x}}{\varphi_{xx'}}$$

we have

$$\begin{aligned} \left\langle e^{-\Delta S^{\text{hk}}} \right\rangle &= \sum_{xx'} \frac{\varphi_{xx'}}{\varphi_{x'x}} \varphi_{x'x} \\ &= \sum_{xx'} \varphi_{xx'} = 1 \end{aligned}$$

Speck and Seifert, 2005



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- Fluctuation relations for the entropy are more general than the corresponding ones for the work



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- Fluctuation relations for the entropy are more general than the corresponding ones for the work
- They provide hope for a characterization of non equilibrium steady states



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- Fluctuation relations for the entropy are more general than the corresponding ones for the work
- They provide hope for a characterization of non equilibrium steady states
- The field is still unsettled



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General review: F. Ritort, arXiv: 0705.0455

- C. Bustamante, J. Liphardt, F. Ritort, arXiv:cond-mat/0511629 (2005); F. Ritort, Poincar Seminar **2**, 193-226 (2003)
- C. Jarzynski, Phys. Rev. Lett. **78**, 2690 (1997); Phys. Rev. E **56**, 5018 (1997); J. Stat. Phys. **96**, 77 (2000)
- G. E. Crooks, Phys. Rev. E **60** 2721 (1999); **61**, 2361 (2000); Thesis (UC Berkeley, 1999)
- R. Kawai, J. M. R. Parrondo, C. van den Broeck, Phys. Rev. Lett. **98**, 080602 (2007); A. Gomez-Martin, J. M. R. Parrondo, C. van den Broeck, arXiv:0710.4290 (2007)
- G. N. Bochkov, Y. E. Kuzovlev, Physica A **106** 443, 480 (1981)
- M. Palassini and F. Ritort, Communication to the APS March Meeting, 2008



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- ❖ Entropy produced in a transition

- ❖ Seifert's relation

- ❖ Fluctuation theorem I (Seifert)

- ❖ Comments

- ❖ Fluctuation theorem II (Jarzynski)

- ❖ Steady states out of equilibrium

- ❖ Fluctuation theorem III (Evans-Searles)

- ❖ Comments

- ❖ Housekeeping entropy production (heat)

- ❖ Average housekeeping heat

- ❖ Fluctuation theorem for the housekeeping heat

- J. Kurchan, arXiv:cond-mat/0511073
- U. Seifert, arXiv:cond-mat/0710.1187
- D. J. Evans, E. G. D. Cohen, G. P. Morriss, Phys. Rev. Lett. **71**, 2401 (1993); D. J. Evans, D. J. Searles, Phys. Rev. E **50**, 1645 (1994)
- G. Gallavotti, E. G. D. Cohen, J. Stat. Phys. **80**, 931 (1995); G. Gallavotti, Phys. Rev. Lett. **77**, 4334 (1996)
- J. Kurchan, J. Phys. A: Math. Gen. **31**, 3719 (1998); J. L. Lebowitz, H. Spohn, J. Stat. Phys. **95**, 333 (1999); C. Maes, J. Stat. Phys. **95**, 367 (1999)
- U. Seifert, Phys. Rev. Lett. **95**, 040602 (2005); T. Speck, U. Seifert, Phys. Rev. E **70**, 066112 (2006); C. Tietz, S. Schuler, T. Speck, U. Seifert, J. Wrachtrup, Phys. Rev. Lett. **97**, 050602 (2006)