Effective interactions between macro-ions: valence-dependence of condensed counter-ions dealing with counter-ions in polyelectrolyte solutions

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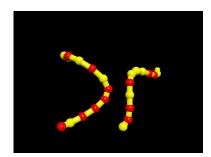


Pectin



Pectin in the fruit or additional pectin is responsible for some of the particular mechanical properties of certain types of jam.

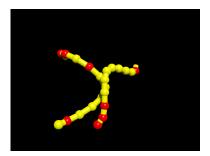


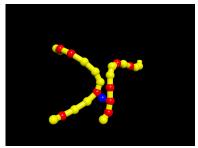


Chemistry of the chain and acidity of the environment control the degree of dissociation.



Linking by two methods





Gelation is thought to occur in different ways: hydrogen bonding with hydrophobicity and salt bridging by divalent ions (typically Ca^{2+}). Purification can be done with Cu^{2+} .



The ingredients

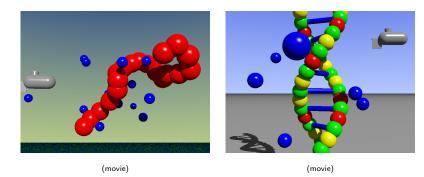
- Polymer chain
- Charge (and associated interactions)
- Gelation the formation of a (temporary) network of chains
- One needs at least divalent ions



Polyelectrolytes

Polyelectrolytes

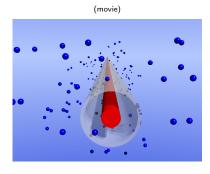
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- Macro-ions with charged groups that dissociate in polar solvent
- Macro-ions themselves have complicated behaviour of their conformations
- A variety of properties that have been difficult to understand



Counter-ion condensation (Manning & Oosawa)



- Non-interacting counter-ions in potential of an infinite rod
- Bjerrum length

$$\ell_{\mathsf{B}} = \frac{\mathsf{e}^2}{4\pi\epsilon \mathsf{k}_{\mathsf{B}}\mathsf{T}}$$

- Potential $\propto \ln r$
- Boltzmann weight

$$w \exp[-2\lambda \ell_{\rm B} r] = w r^{-2\lambda \ell_{\rm B}}$$

$$\Rightarrow N_{< r_0} = 2\pi \int_0^{r_0} dr \ r^{1-2\lambda \ell_B}$$

- Condensation when $\ell_{\mathsf{B}}\lambda > 1$.
- $f = 1 1/(\lambda \ell_B)$



Screening of electrostatic charge (Debye-Hückel)

- Poisson's equation $\nabla^2 \phi = -\varrho/\epsilon$
- Estimating the density by the Boltzmann weight for charges in an electric field leads to the Poisson-Boltzmann equation:

$$\nabla^2 \phi = -\sum_{s} c_{0,s}(ez_z) e^{ez_s \phi(r)/k_B T}$$

 Linearised to approximate gives Debye-Hückel screened potential:

$$V_{\mathsf{DH}}(r) = rac{1}{\epsilon |r|} e^{-\kappa |r|}$$

where
$$\kappa^2 = 4\pi \ell_{\rm B} \left(\sum_s c_{0,s} z_s^2\right)$$



Experimental and theoretical situation

- A long history.
- Debye-Hückel theory is the typical first approach to these systems.
- Nishida & Kanaya JCP (2001) phase diagram of salt-free polyelectrolytes; Zhang et al JCP (2001) and Dubois & Boué Macromol (2001) effects of counterion valences; Das et al PRL (2003) condensation profiles; Qiu et al PRL (2006) effective interaction as function of valence.
- Simulations have also addressed multivalence (e.g. Chang and Yethiraj JCP 2003).
- Counterion correlations in polyelectrolyte solutions, gelation, e.g. Ermoshkin & Olvera de la Cruz Macromol, PRL (2003)
- Strong coupling theory Netz and coworkers (e.g. Naji & Netz EPJE (2004))
- Two-fluid models, e.g. Lau, et al PRE (2002)



The questions

- How much condensation of counter-ion occurs on other objects (including flexible polymer chains)?
- Does the valence of counter-ions play a role here?
- In which ways would condensed counter-ions alter the pure Coulomb repulsion between macro-ions?



The strategy

Polymer +
 Counter-ions MIXED
 WITH Coulomb
 interaction

...Only in principle

Calculate the partition function.

$$Z = \sum_{ ext{polymers counter-ions}} e^{-eta H(ext{polymers, counter-ions} ext{ with Coulomb})}$$
 $= \sum_{ ext{polymers}} e^{-eta H_{ ext{effective}}(ext{polymer})}$

we do this in three stages:

- Stage 1: counter-ions as continuum
- Stage 2: counter-ions as discrete
- Stage 3: aspects of continuum discrete

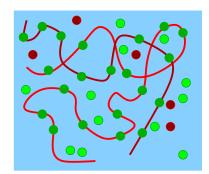


Charge condensation on flexible, linear chains

Work with Tannie Liverpool (Bristol)

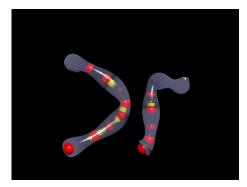
TB Liverpool & KKMN J. Phys. Condens. Matter (2006)

- ${\cal N}$ flexible polyelectrolyte chains of with linear charge density $-\lambda e$
- neutralized by N counterions of charge ze
- additional salt
- A fraction f of the number of counter-ions N is presumed to be condensed on the chain counter-ions split up into two species!





Seeing this...



There is much important work related to such ideas: Ha & Liu (late 1990's), Golestanian, et al.



where $C(N, M, f) = \frac{1}{M!} \frac{1}{[(1-f)N+M]!} \frac{1}{(fN)!}$

Method

$$\begin{split} Z(f) &= C(N,M,f)J_{\phi} \int [d\tilde{\mathbf{r}}(s)][d\tilde{\phi}(\mathbf{x})] \prod_{i=1}^{(1-f)N+M} d^{3}z_{i} \prod_{k=1}^{M} d^{3}z'_{k} \prod_{j=1}^{fN} ds_{j} \\ &\times \exp\left\{-\frac{3}{2} \sum_{i=1}^{N} \int_{0}^{L/\ell} ds \left(\frac{\partial \mathbf{r}_{i}}{\partial s}\right)^{2} - \frac{1}{8\pi z^{2} \tilde{\ell}_{\mathrm{B}}} \int_{\tilde{V}} d^{3}x \left(\nabla \tilde{\phi}(\mathbf{x})\right)^{2}\right\} \\ &\times \exp\left\{+i \int_{\tilde{V}} d^{3}x \left[\left(-\tilde{\lambda}/z + \tilde{c}_{0}\right) \varrho(\mathbf{x}) + \tilde{n}^{+}(\mathbf{x}) - \tilde{n}^{-}(\mathbf{x}) + \sum_{i=1}^{N} \int_{0}^{L/\ell} ds \, \delta\left(\mathbf{x} - \tilde{\mathbf{r}}_{i}(s)\right) \tilde{c}(s)\right] \tilde{\phi}(\mathbf{x})\right\}. \end{split}$$



Method (2): Quadratic order in density

$$\begin{split} Z_{\mathbf{q}}(f) &= \frac{1}{(\mathit{fN})!} \frac{1}{[(1-\mathit{f})\mathit{N}+\mathit{M}]!} J_{\phi} J_{\varrho} e^{[(1-\mathit{f})\mathit{N}+\mathit{M}] \ln \tilde{\mathit{V}} + \mathit{fN} \ln(\mathit{L}/\ell)} \\ &\times \int \left[\mathit{d}\varrho \right] \left[\mathit{d}\phi \right] \exp \left\{ -\frac{1}{2} \int \mathit{d}^3 x \, \mathit{d}^3 x' \varrho \, G_0^{-1} \varrho + i \int \mathit{d}^3 x \, \phi \left(-\frac{\tilde{\lambda}}{z} + \tilde{c}_0 \right) \varrho \right\} \\ &\times \exp \left\{ -\frac{1}{2z^2 \tilde{\ell}_{\mathrm{B}}} \int \mathit{d}^3 x \left[\frac{(\nabla \phi)^2}{4\pi} + z^2 \tilde{\ell}_{\mathrm{B}} \left(\frac{\tilde{\lambda}}{z} \Phi_0 + 2 \tilde{n}_0^- \right) \phi^2 \right] - \frac{1}{2} \tilde{c}_0 \int \mathit{d}^3 x \, \phi^2 \varrho \right\}. \end{split}$$

Red parts: entropic counter-ion contributionDebye-Hückel interactioncontribution to $(1 - f)^2$ contribution to f^2 Leads to an **effective interaction** between macro-ions:

$$V_{\text{eff}} \equiv \left(\frac{\tilde{\lambda}}{z}\right)^2 \left[(1-f)^2 V_{\text{DH}} - \frac{f^2}{2} V_{\text{DH}}^2 \right].$$

with Debye-Hückel interaction $V_{\mbox{DH}}({\bf r}) = z^2 \tilde{\ell}_{\mbox{B}} e^{-\tilde{\kappa} |\tilde{\bf r}|}/|\tilde{\bf r}|$ where

$$ilde{\kappa}^2 = 4\pi z^2 ilde{\ell}_{\mathrm{B}} \left[2 ilde{n}_0^- + rac{ ilde{\lambda}}{z} \Phi_0
ight]$$



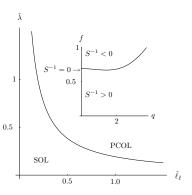


Figure: $\tilde{\ell}_{\rm B}$ and $\tilde{\lambda}$ where chains always in solution (SOL). Outside — possible chain collapse (PCOL). Inset: PCOL — $S_f^{-1}(q)$ break down of formalism. ($\tilde{R}_{\rm g}^2=10.0,\ \Phi_0=0.05.$ Main graph: $n_0=0.0,\$ Inset: $\tilde{\lambda}=8.0,\ \tilde{\ell}_{\rm B}=0.5$ and $n_0=0.01.$)

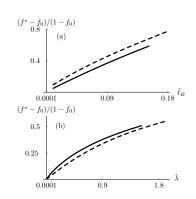


Figure: Numerical solutions of f^* : (a) for $\tilde{n}_0=0.0$ (solid) and $\tilde{n}_0=0.03$ (dashed), with $\Phi_0=0.1$, $\tilde{\lambda}=1.0$, $\tilde{R}_g^2=10.0$. In (b) for $\Phi_0=0.1$ (solid) and $\Phi_0=0.05$ (dashed) where $\tilde{\ell}_{\rm B}=0.1$ ($\tilde{n}_0=0.0$, $\tilde{R}_\sigma^2=10.0$.

Results (2): Valence

Table: We list values of f^* for increasing valences in systems where $\tilde{\ell}_{\rm B}=0.05,~\tilde{\lambda}=1.0,~\tilde{R}_{\rm g}^2=10,~\Phi_0=0.1,~{\rm and}~\tilde{n}_0=0.0.$ As expected the results indicate that increasing valence causes stronger condensation (and by implication, in our model, a significant increase in the attraction of the effective chain-chain interaction.)

Z	1	2	3
f*	0.292	0.556	0.884
q_{peak}	0.743	0.507	0.0

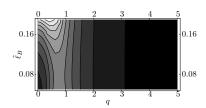


Figure: Contour plot of structure factor as $\tilde{\ell}_{\rm B} \propto 1/T$ varies. Shows the polyelectrolyte peak (lighter areas — greater scattering intensity). $\tilde{\lambda}=1.0,~\Phi_0=0.1,~\tilde{n}_0^+=0.0,$ and $\tilde{R}_{\rm g}^2=10.0.$

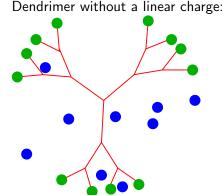


Introduction: discrete condensing counter-ions

Related to ideas by Arenzon et al. Eur. Phys. J B (1999)



Condensed charge, valence 2





The model

Work done with Thomas Vilgis, Max-Planck-Institut für Polymerforschung, Mainz.

Fixed sites with labels

$$i \in \{1, ..., N\}$$
 with positions $\{\mathbf{r}_1, ..., \mathbf{r}_N\} \subset \mathbb{R}^3$.

Two-state variables at sites

$$au_i = \left\{ egin{array}{ll} 1, & ext{counter-ion on site } i \longrightarrow ext{ charge at site } = e(1-z) \\ 0, & ext{no counter-ion on site } i \longrightarrow ext{ charge at site } = e \end{array} , orall in the state of the sta$$

- Coulomb interaction $q_i \times q_j \frac{1}{4\pi\epsilon_0|\mathbf{r}_i \mathbf{r}_i|}$.
- Partition function for counter-ions condensing at the sites

$$\begin{split} Z\left(\left\{\mathbf{r}_{i}\right\}\right) &= \sum_{\left\{\tau_{i}\right\}} \exp\left(-\beta H\left(\left\{\tau_{i}\right\}\right)\right) \\ \text{with } -\beta H\left(\left\{\tau_{i}\right\}\right) &= \mu \sum_{i=1}^{N} \tau_{i} - \ell_{B} \sum_{i < j} (1 - z\tau_{i})(1 - z\tau_{j}) V_{ij}. \end{split}$$



Effective potential

Interaction potential may be expanded in a systematic, virial technique.

$$Z = Z_{\mu} \exp \left\{ \sum_{i < j} \left[-\tilde{V}_{ij} + \ln \left(1 + \underbrace{i}_{j} + \underbrace{i}_{j} + \underbrace{i}_{j} + \underbrace{i}_{j} + \underbrace{i}_{j} \right) \right] + \sum_{i < j < k} w_{ijk}^{(3)} \right\}$$

The effective potential

$$w_{ij}^{(2)} \equiv \tilde{V}_{ij} - \ln \left[(1 - f)^2 + 2f(1 - f)e^{z\tilde{V}_{ij}} + f^2e^{-(z^2 - 2z)\tilde{V}_{ij}} \right]$$

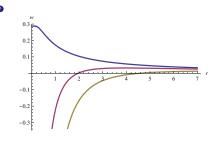


Effective potential (2)

Limiting behaviour

$$w^{(2)}(r) = \frac{1}{2} - \ln\left[(1-f)^2 + 2f(1-f)e^{z/r} + f^2e^{(-z^2+2z)/r} \right]$$

$$\simeq \begin{cases} \frac{(1-zf)^2}{r}, & r \text{ large } -\text{repulsion} \\ \frac{1-z^r}{r} - \delta_{z,1} \ln\left(2f - f^2\right) & r \text{ small } -\text{attraction for } z > 1 \end{cases}$$



blue — monovalentpurple — divalentbrown — trivalent



Polyelectrolytes

Effective potential (3)

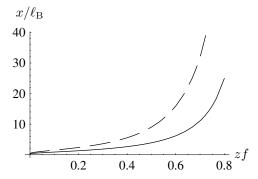
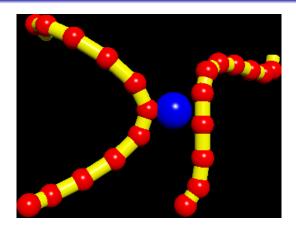


Figure: Where the sign changes as a function of distance and neutralized charge (in noninteracting system). $\frac{\partial w_{ij}^{(2)}}{\partial x} = 0$ indicating crossover for 2nd virial contribution from repulsion at large distances to attraction at closer distances; z > 1. Dashed line: trivalent case. Solid line: divalent case. zf = 1 neutrality in *non-interacting* system.

Reversible networking

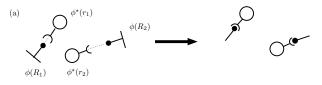


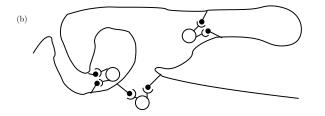
- Kornyshev & Leikin PRL 1999 zipping of DNA
- Salt bridges
- Also suggested by experiments (Dubois & Boué; Zhang)



Networking by means of crosslinkers

- crosslinkers with some functionality
- places where they can stick







Basic model

- Reversible networking, with crosslinkers as the high valence counter-ion.
- Competition
 - Crosslink reduces translational freedom of polymer chains
 - Potentially is of energetic advantage
 - However closer chains also bring about energetic penalty (like charges). However, with the condensation renormalising the charge and the much weakened interactions, this can become feasible.
- Need a suitable formalism: Edwards developed a field theory formulation for polymer networks: linking by cross-linkers of some functionality. J Phys France (1988)

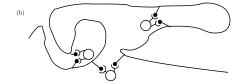


Counting the number of ways in which one can link

Based on two Gaussian field and appropriate grouping. Produces crosslinkers and we do the "statistical physics thing" by counting all the possible ways to do this via the field theory

$$\mathcal{N}'' \int [d\phi] [d\phi^*] \left\{ \prod_i^M \phi(\mathbf{R}_i) \right\} \left[\int d^3 r \left(\phi^*(\mathbf{r}) \right)^f \right]^{M'} \exp \left(- \int \phi \phi^* \right).$$

$$(a) \qquad \phi^*(r_1) \qquad \phi(R_2) \qquad \qquad \phi(R_2) \qquad \qquad \phi(R_1) \qquad \phi^*(r_2) \qquad \qquad \phi(R_2) \qquad \qquad \phi(R$$





The model

• Take polymer and counter-ions with Coulomb interaction

$$Z(\{\mathbf{R}_i\}) = \frac{1}{M!} \int d^3 r_1 \dots d^3 r_M \int [d\psi]$$

$$\exp \left\{ + \frac{1}{2\ell_B} \int d^3 r \ \psi(\mathbf{r}) \nabla^2 \psi(\mathbf{r}) - i \sum_{i=1}^N \psi(\mathbf{R}_i) + iz \sum_{j=1}^M \psi(\mathbf{r}_j) \right\}$$

A model for the polymer

$$P(\mathbf{R}_i) = G(\mathbf{R}_1, \mathbf{R}_2) G(\mathbf{R}_2, \mathbf{R}_3) \dots G(\mathbf{R}_{N-1}, \mathbf{R}_N).$$

Allow for linking (there are some options here)



Polyelectrolytes 000000000

- with different functionality
- with caps





Grand canconical partition function

The following partition function now applies

$$\begin{split} Z_{\text{II}} &= \frac{\mathcal{N}}{M!} \int \left[d\phi \right] \left[d\phi^{\star} \right] \left[d\psi \right] d^{3}R_{1} \dots d^{3}R_{M} \\ & \left(1 + \phi(\textbf{R}_{1}) \right) e^{-i\psi(\textbf{R}_{1})} G\left(\textbf{R}_{1}, \textbf{R}_{2} \right) \left(1 + \phi(\textbf{R}_{2}) \right) e^{-i\psi(\textbf{R}_{2})} G\left(\textbf{R}_{2}, \textbf{R}_{3} \right) \times \dots \\ & \times \left(\int d^{3}r \ \left(\phi^{\star}(\textbf{r}) \right)^{z} e^{iz\psi(\textbf{r})} \right)^{M} e^{\int d^{3}r \ \phi(\textbf{r})} \\ & \times \exp \left[+ \frac{1}{2} \int d^{3}r \ \psi(\textbf{r}) \nabla^{2}\psi(\textbf{r}) - \int d^{3}r \ \phi(\textbf{r}) \phi^{\star}(\textbf{r}) \right]. \end{split}$$

In the grand canonical formalism for the counter-ions

$$\begin{split} \Xi_{\mathsf{II}} &= \int \left[d\phi \right] \left[d\phi^{\star} \right] \left[d\psi \right] \left\{ \prod d^{3}R \right\} \left\{ \prod G \right\} \\ &\times \exp \left\{ + \int \phi - i \int \psi \rho + \int \rho \ln(1+\phi) - \int \phi \phi^{\star} + \frac{1}{2} \int \psi \nabla^{2}\psi + \mathfrak{z} \int e^{iz\psi} \phi^{\star z} \right\} \\ &= \int \left[d\phi \right] \left[d\phi^{\star} \right] \left[d\psi \right] \left\{ \prod d^{3}R \right\} \left\{ \prod G \right\} \exp \left\{ + \mathcal{F}_{\mathsf{II}} \left[\phi, \phi^{\star}, \psi \right] \right\} \end{split}$$



Saddle point approximation

Determine saddle point

$$\frac{\delta \mathcal{F}_{\text{II}}}{\delta \phi(\mathbf{x})}\bigg|_{\overline{\phi},\overline{\phi^{\star}},\overline{\psi}} = 0, \ \frac{\delta \mathcal{F}_{\text{II}}}{\delta \phi^{\star}(\mathbf{x})}\bigg|_{\overline{\phi},\overline{\phi^{\star}},\overline{\psi}} = 0 \ \text{and} \quad \frac{\delta \mathcal{F}_{\text{II}}}{\delta \psi(\mathbf{x})}\bigg|_{\overline{\phi},\overline{\phi^{\star}},\overline{\psi}} = 0.$$

such that

$$\begin{array}{rcl} \displaystyle \frac{\rho}{1+\overline{\phi}}+1 & = & \overline{\phi^{\star}} \\ \\ z_3 e^{iz\overline{\psi}} \overline{\phi^{\star}}^{z-1} & = & \overline{\phi} \\ \\ -i\rho + iz_3 e^{iz\overline{\psi}} \overline{\phi^{\star^2}} + \nabla^2 \overline{\psi} & = & 0 \text{ (Poisson-Boltzmann of sorts)} \end{array}$$

A little rearrangement of "Poisson-Boltzmann"-like term

$$-i\left(rac{1}{1+\overline{\phi}}
ight)
ho+i\overline{\phi}+
abla^2\overline{\psi}=0.$$

— a bulk contibution and renormalisation of polymer charge.



Monovalent solution

- Linearise Poisson-Boltzmann equation (remember this gives a screened interaction in general — also here)
- Substitute, keep terms quadratic in ρ
- Effective polymer-polymer interaction term

$$-\frac{1}{2}\int \frac{\rho}{1+\mathfrak{z}} V_{\mathsf{DH}} \frac{\rho}{1+\mathfrak{z}}$$

We find: charge renormalised and screened interaction



Divalent solution

- Two solutions of fields, one is physical.
- Linearise as before
- Substitute
- Effective polymer-polymer interaction term

$$\int \rho \left[\frac{\mathfrak{z}}{(1+2\mathfrak{z})^2} - \frac{1}{2} \left(\frac{1-2\mathfrak{z}}{1+2\mathfrak{z}} \right)^2 V_{\mathsf{DH}} \right] \rho.$$

- We find: charge renormalised, screened interaction and short-ranged attraction
- counter-ion density profile more subtle



- Quadratic fluctuations $\Delta\phi$, $\Delta\phi^{\star}$ and $\Delta\psi$ around $\overline{\phi}$, $\overline{\phi^{\star}}$ and $\overline{\psi}$
- Leads to

Polyelectrolytes

$$\begin{split} \Xi & \approx & \int \left[d\Delta \psi \right] \exp \left\{ -\frac{1}{2} \mathrm{Tr} \; \ln \left[1 + 2 \mathfrak{J} \frac{\rho}{\left(1 + \overline{\phi} \right)^2} \mathrm{e}^{2 i \overline{\psi}} \right] \right\} \\ & \exp \left\{ -\frac{1}{2} \int \Delta \psi \left[-\nabla^2 + 4 \mathfrak{J} \overline{\phi^*}^2 \mathrm{e}^{2 i \overline{\psi}} \left(1 + \frac{4 \mathrm{e}^{4 i \overline{\psi}} \mathfrak{J}^2 \left(1 + \overline{\phi} \right)^2}{2 \mathrm{e}^{2 i \overline{\psi}} \mathfrak{J} \rho + \left(1 + \overline{\phi} \right)^2} \right) \right] \Delta \psi \right\} \end{aligned}$$



Conclusions and outlook

- Condensation of counter-ions on non-cylindrical objects.
- Self-consistent calculation with fluctuating counter-ion density along chain backbone.
- Discrete model: role of valence becomes very clear.
- Formalism for bridging provides a combination of discrete nature and continuum formalism.
- Field theory allows far more intricate and subtle dependencies on polymer shape to be investigated.
- Other work: counter-ions in resticted geometries around a macro-ion. (With L Boonzaaier, F Scholtz)



Polyelectrolytes

- Hosts at KITP
- Thomas Vilgis (MPI-P, Mainz), Tanniemola Liverpool (Bristol), Przemysław Tarasewicz (Bydgoszcz), Lee Boonzaaier (Stellenbosch), Frederik Scholtz (Stellenbosch)
- Funding agency: National Research Foundation of South Africa



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