



Stable cold polar molecules

Jun Ye

JILA, NIST and University of Colorado

Correlated States in Degenerate Atomic Gases, KITP, April 25, 2007

\$ Funding \$

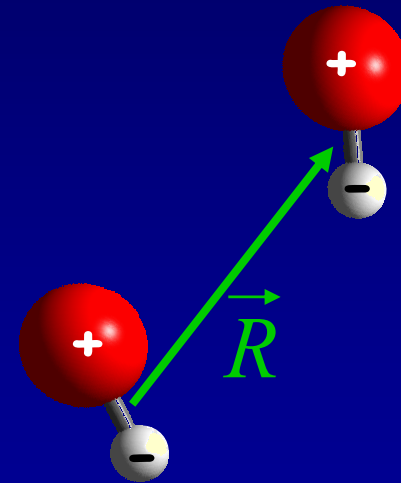
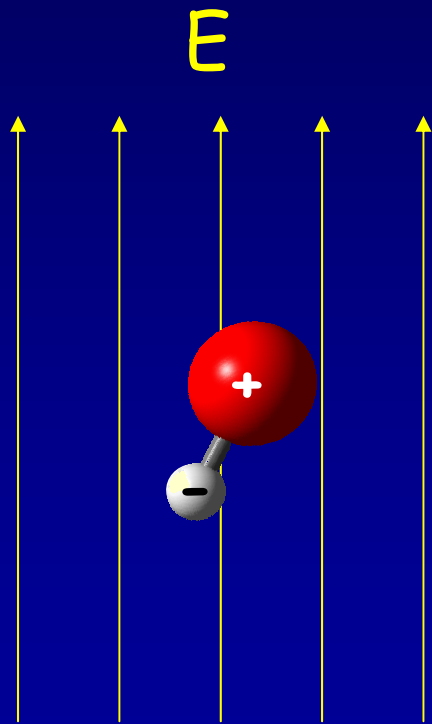
DOE, NIST, NSF



Why ultracold molecules?

J. Doyle *et al.*, *Eur. Phys. J. D* 31, 149 (2004).

Electric dipole moments: Orientation is a big deal !

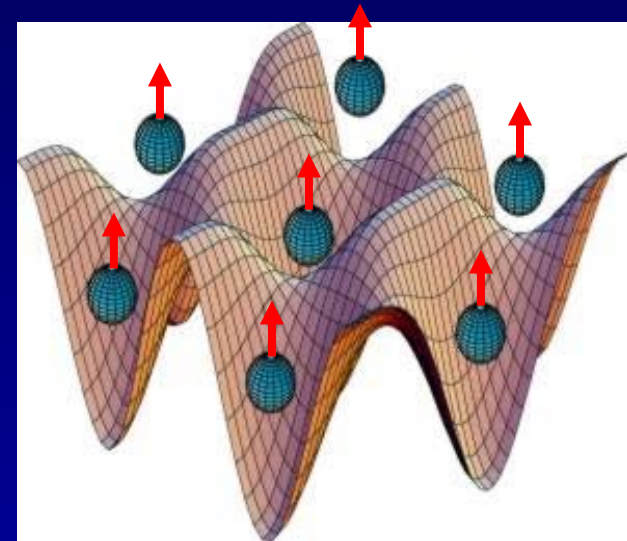


Manifested at or below
 μK temperatures

A. Avdeenkov and J. L. Bohn,
Phys. Rev. Lett. **90**, 043006 (2003).

Ultracold molecules: quantum physics

- **Quantum information**
(strong dipolar interactions, long coherence time)
- **Quantum degeneracy (e.g. BEC)**
(anisotropic interactions)
- **Dipolar phase transition**
(Condensed matter system)



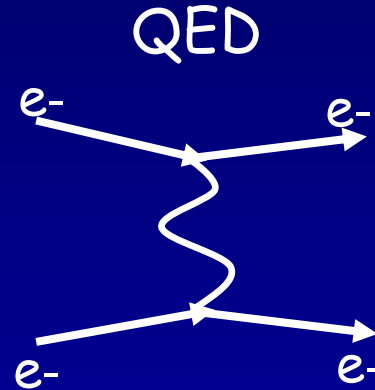
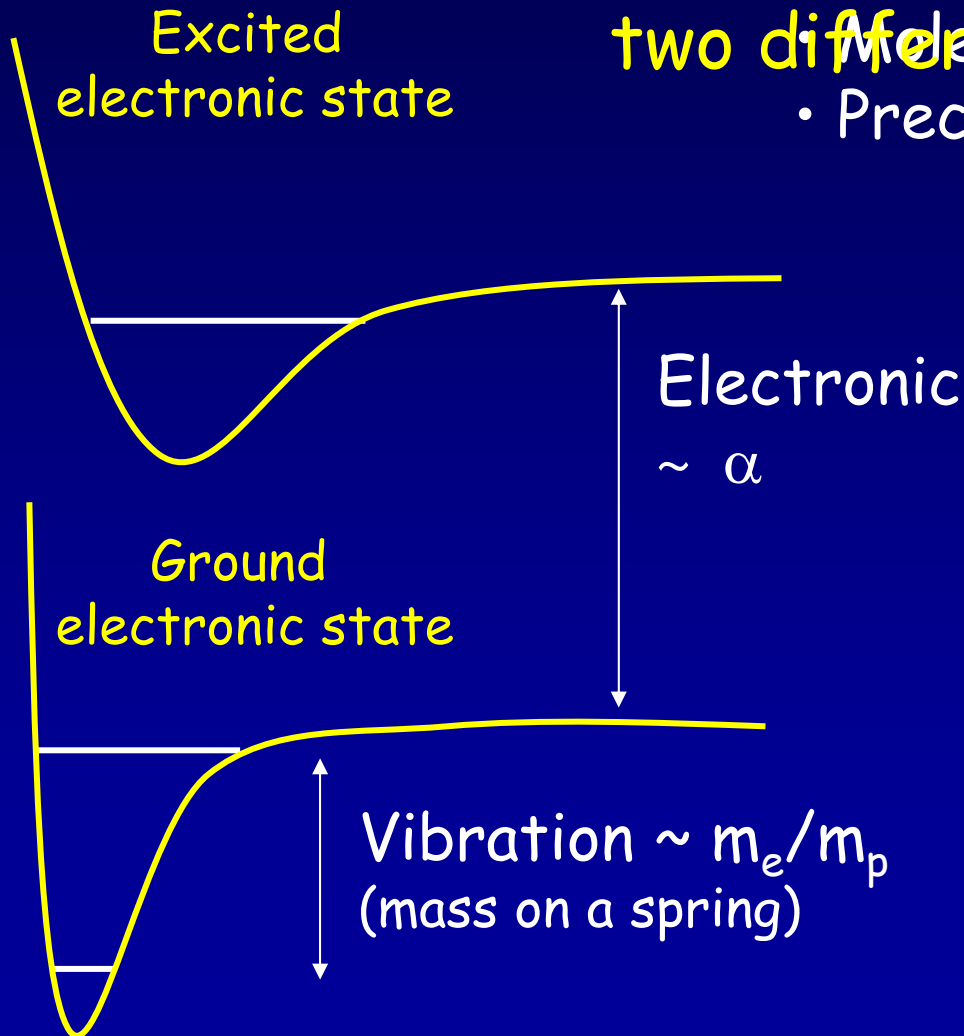
DeMille, Phys. Rev. Lett. **88**, 067901 (2002).

Griesmaier, Werner, Hensler, Stuhler, Pfau, Phys. Rev. Lett. **94**, 160401 (2005).

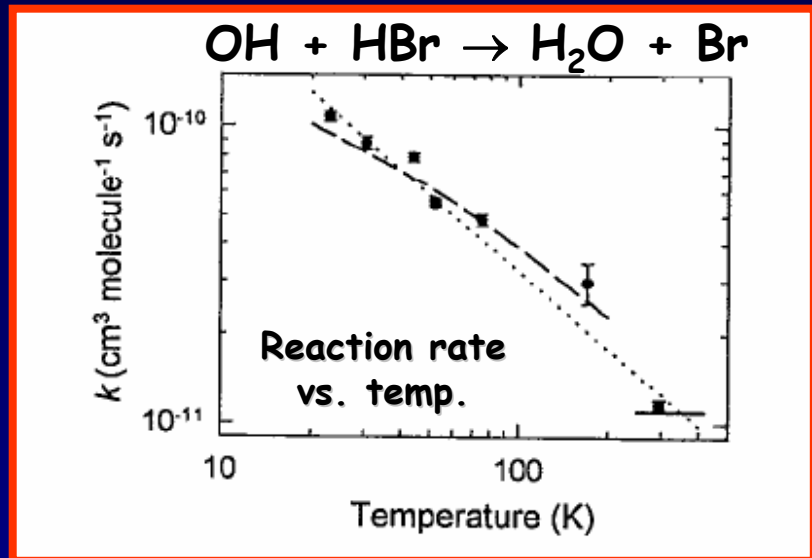
Micheli, Brennen, Zoller, Nature Physics **2**, 341 (2006). Hans Peter Büchler's talk

Ultracold molecules: Test fundamental principles

- Ultrahigh resolution spectroscopy
- Standard as wide spectral ranges
- **One system, two different fundamental forces!**
- Precision measurement

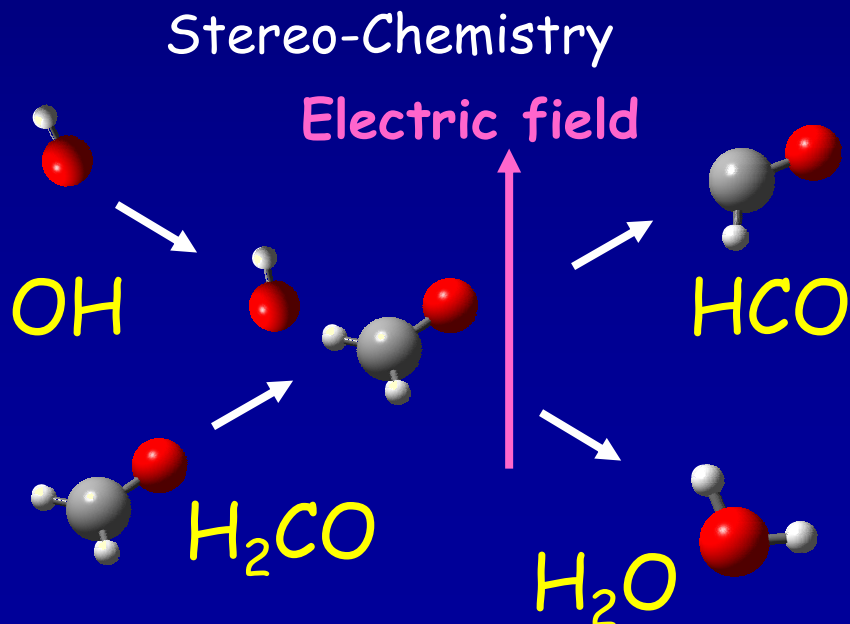


Ultracold molecules: Precision Chemistry



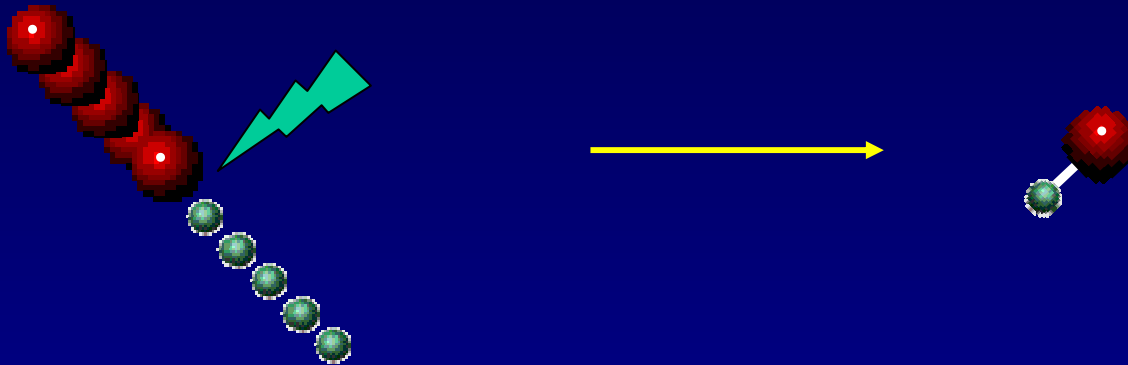
Controlled molecular collisions Ultracold chemical reactions

- Molecules in single quantum states, under precise control, for internal & external motions
- Unprecedented study of fundamentally important reactions (Dial the rates):
 $\text{OH} + \text{HBr}$, $\text{OH} + \text{H}_2\text{CO}$, $\text{CN} + \text{O}_2$,
 $\text{OH} + \text{NO}$, $\text{OH} + \text{OH}$, $\text{CN} + \text{NH}_3$,
 $\text{OH} + \text{H}$
- Higher reaction rate at lower temperature (10 K, importance for interstellar chemistry)

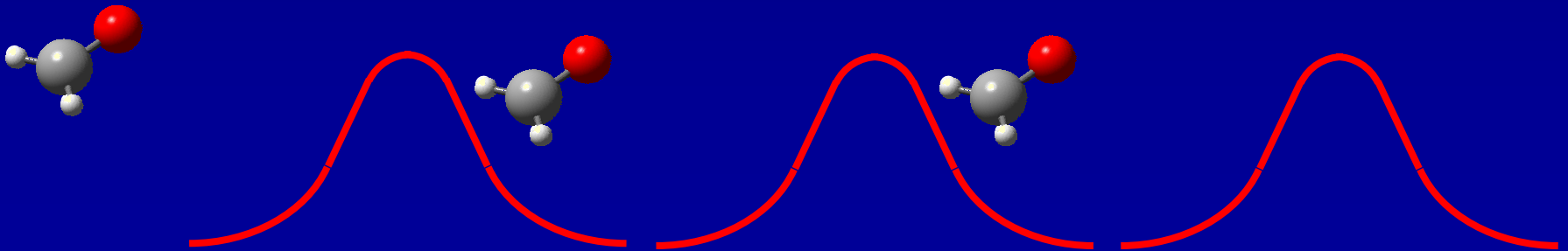


Ways to make cold polar molecules

- Start from ultracold atoms & pair them (Magneto-Photo-association)



- Start from ground-state molecules (Direct cooling of molecules)



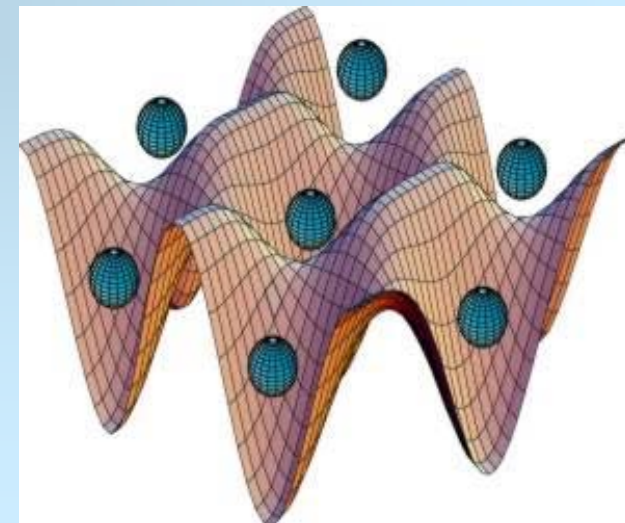
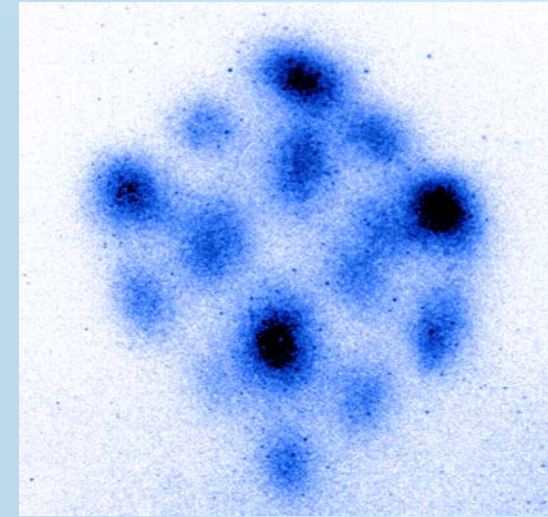
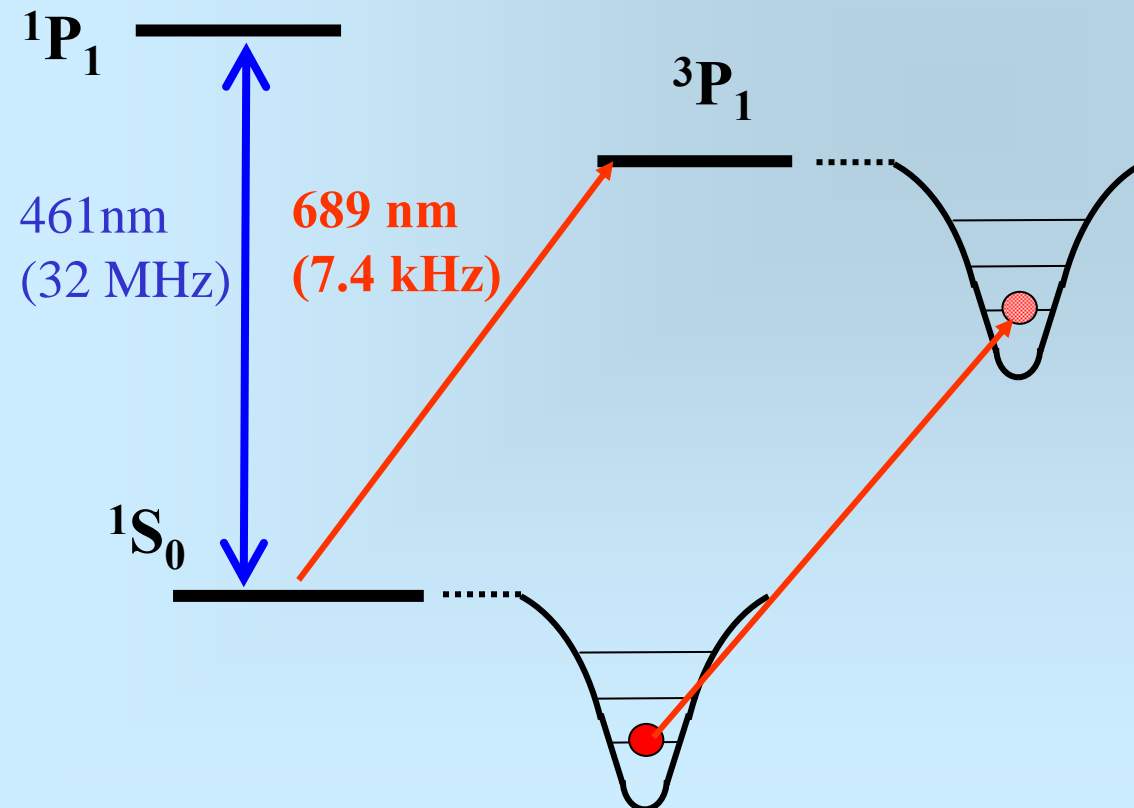
Cool Alkaline Earth – Strontium

Boyd, Zelevinsky, Ludlow, Foreman, Blatt, Ido, Ye, *Science* 314, 1430 (2006).

Boyd, Ludlow, Blatt, Foreman, Ido, Zelevinsky, Ye, *PRL* 98, 083002 (2007).

$T \sim 0.5$ photon recoil
 ~ 220 nK

Recoil-free
Potentials matched to $< 10^{-15}$
Extreme spectral resolution



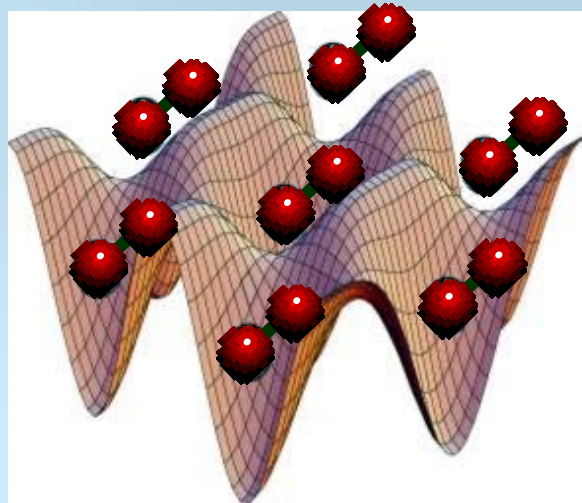
Ultracold Sr_2 Molecules in Lattice

Zelevinsky, Boyd, Ludlow, Ido, Ye, Ciurylo, Naidon, Julienne, PRL 96, 203201 (2006).

- Narrow lines
 - All bound states are resolved
- Favorable decay to electronic ground state
- Control of atomic collision with minimum loss

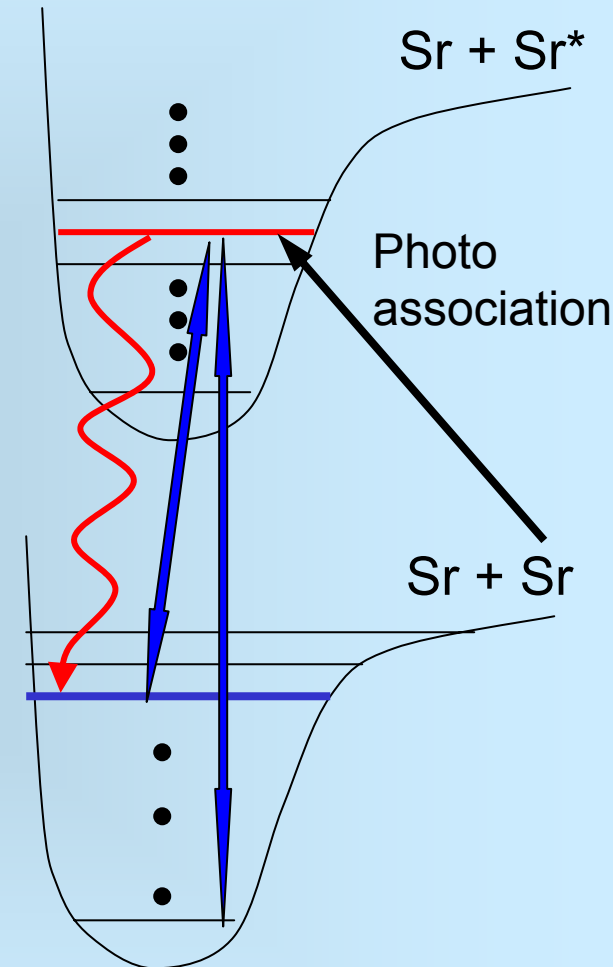
Raman transition for ground state production

Theory:



P. Julienne

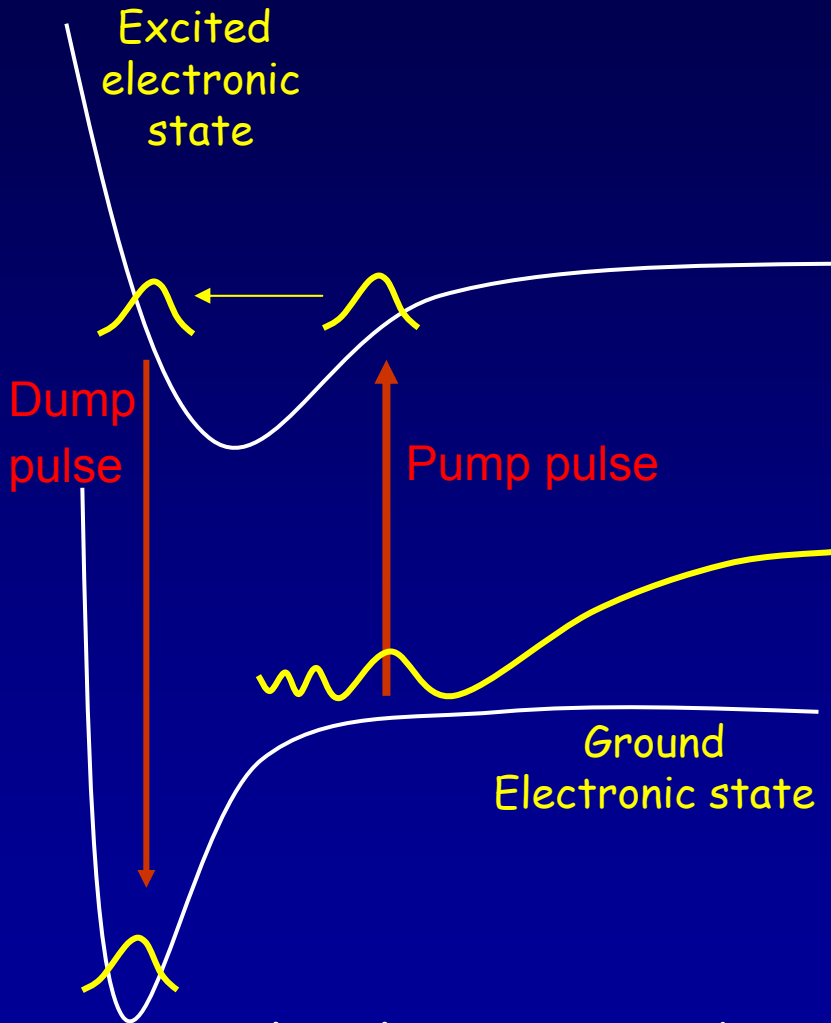
S. Kotochigova



Polar molecules KRb
in collaboration with D. Jin

Photo-Association via Coherent Accumulation of Pulses

Pe'er, Shapiro, Stowe, Shapiro, Ye,
Phys. Rev. Lett. 98, 113004 (2007).



The problem – overlap

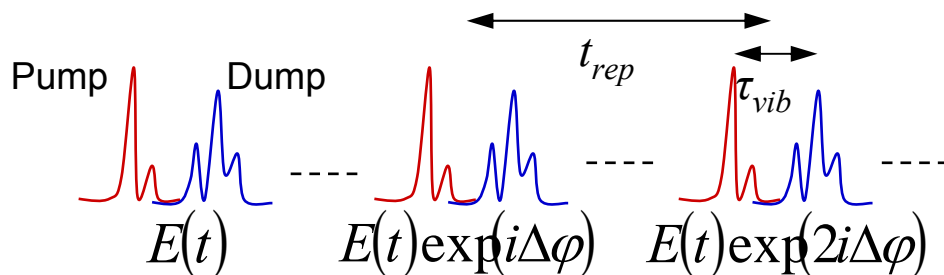
Molecular wave-packet dynamics bridge the overlap mismatch
Coherent accumulations resolve single quantum state

Making cold molecules with comb

Why pulse-train ?

Two inherent time scales:
molecular vibration & atomic coherence

Time Domain picture

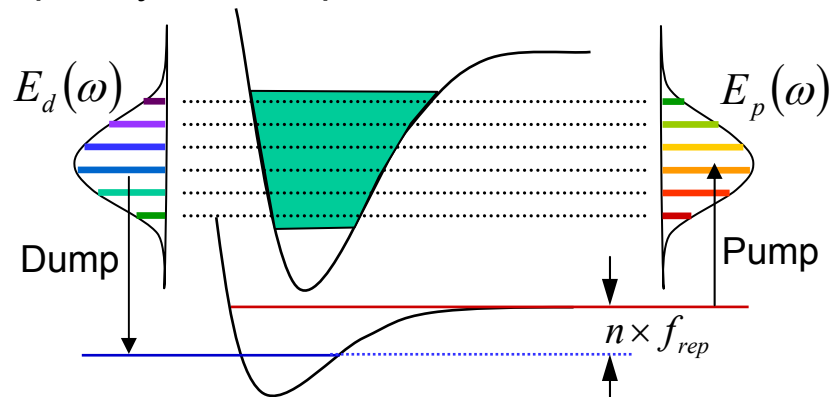


Pulse shaping:

- Positive chirp: "one way ticket"
- Robust to energy fluctuations
- Avoid ionization/dissociation

Coherent accumulation:
Weak field interaction
for strong field effects

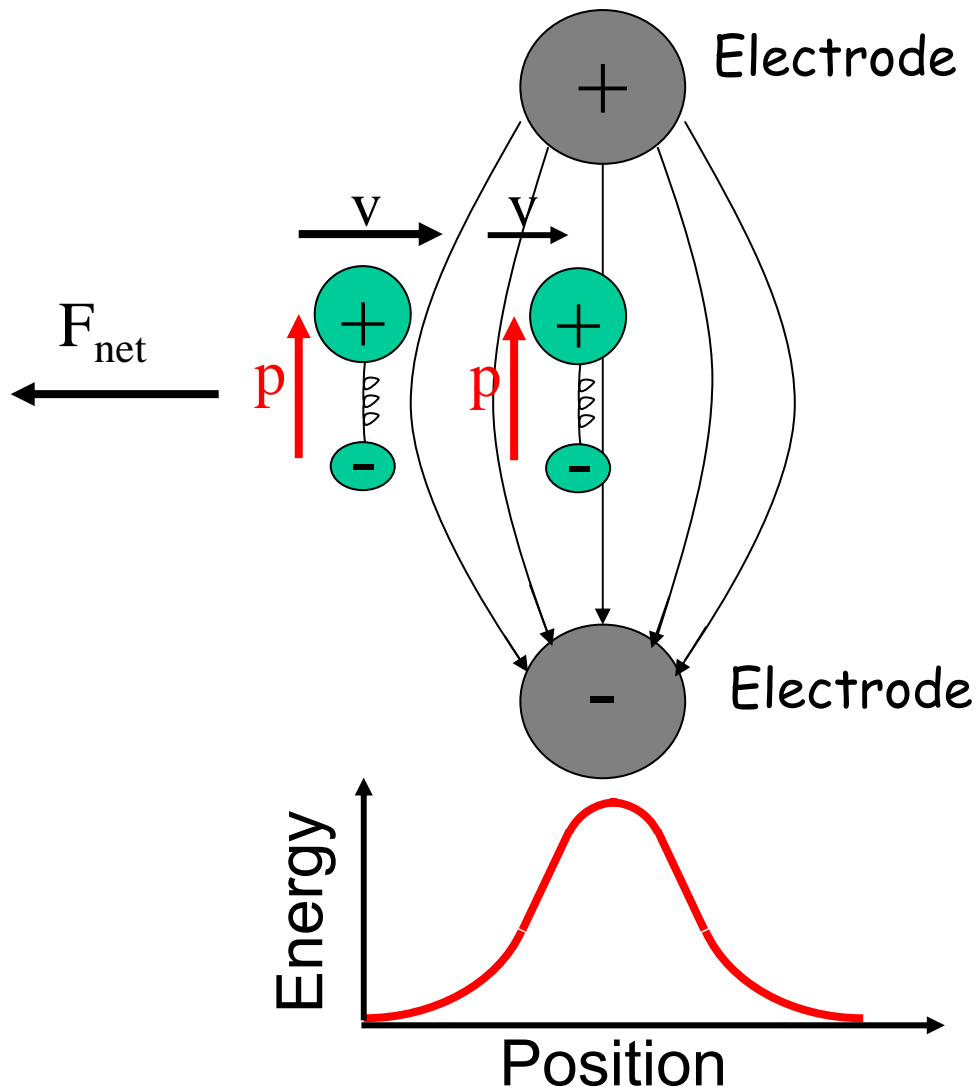
Frequency domain picture



Coherent control of single pulses – Coarse resolution (vibration).
Coherent accumulation of pulses – Fine resolution (rotation, hyperfine)

Stark deceleration

Direct manipulation of ground state molecules



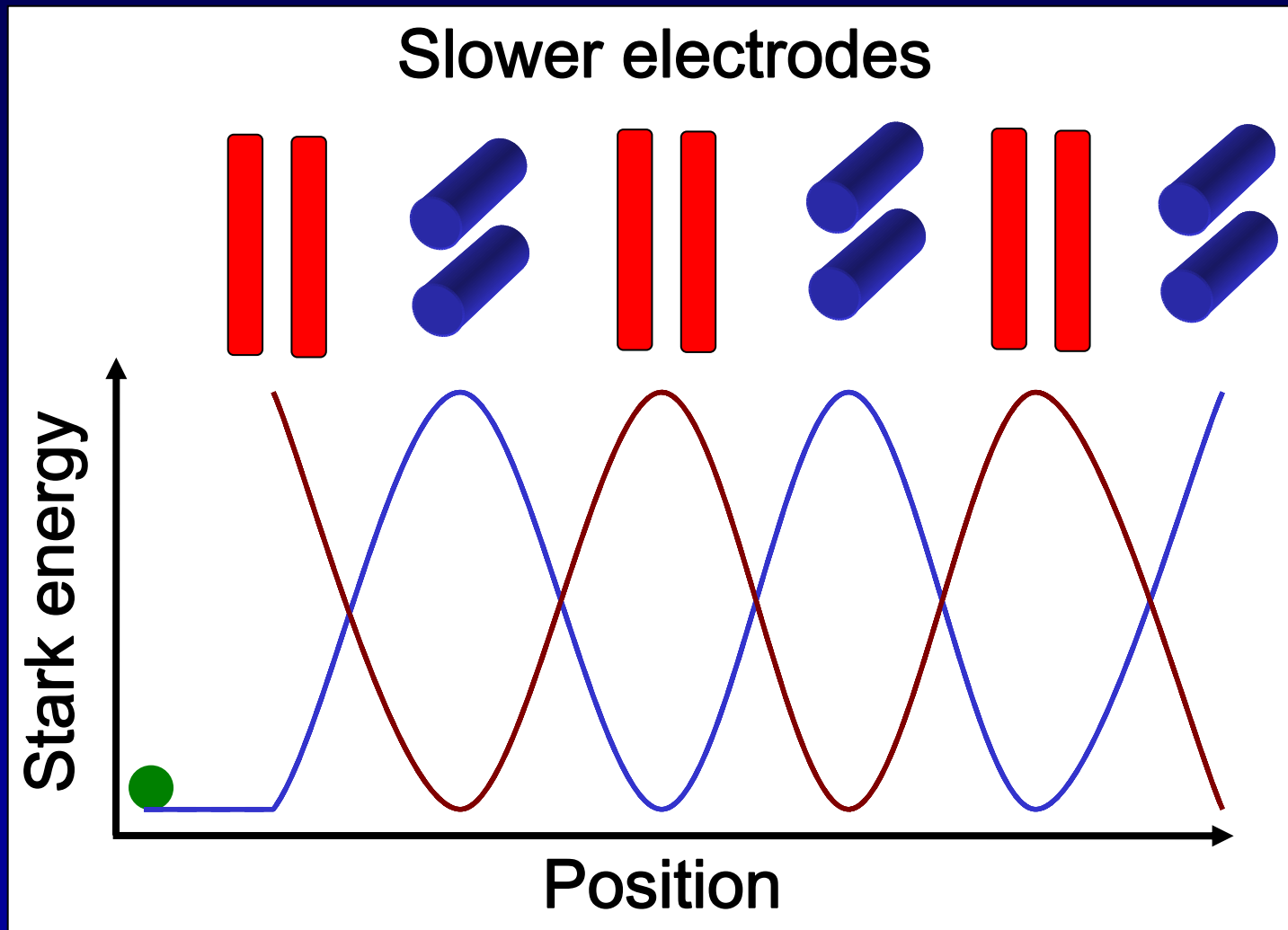
Initial cooling important
(supersonic jets: single internal
quantum state; external
temp. ~ 1 K in a moving frame)

Phase space selection (~ 10 mK)

Applicable to a large variety
of molecules

Bethlem, Berden, Meijer,
Phys. Rev. Lett. 83 1558 (1999).

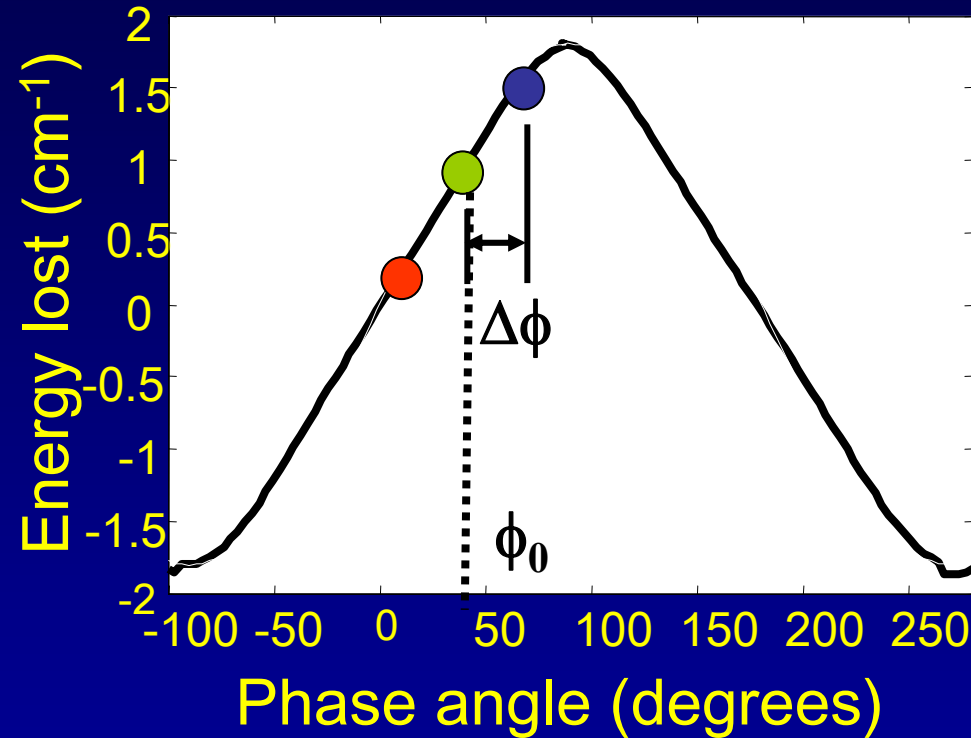
Stark Decelerator



Mathematical description of slowing

Hudson, Bochinski, Lewandowski, Ye, Eur. Phys. J. D 31, 351 (2004).

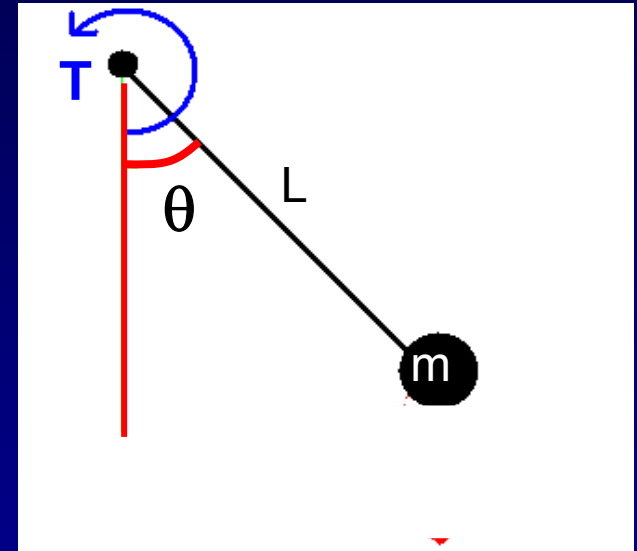
Change in kinetic energy



$$\frac{d^2 \Delta\phi}{dt^2} + \frac{\pi W}{mL^2} [\sin(\phi_0 + \Delta\phi) - \sin \phi_0] = 0$$

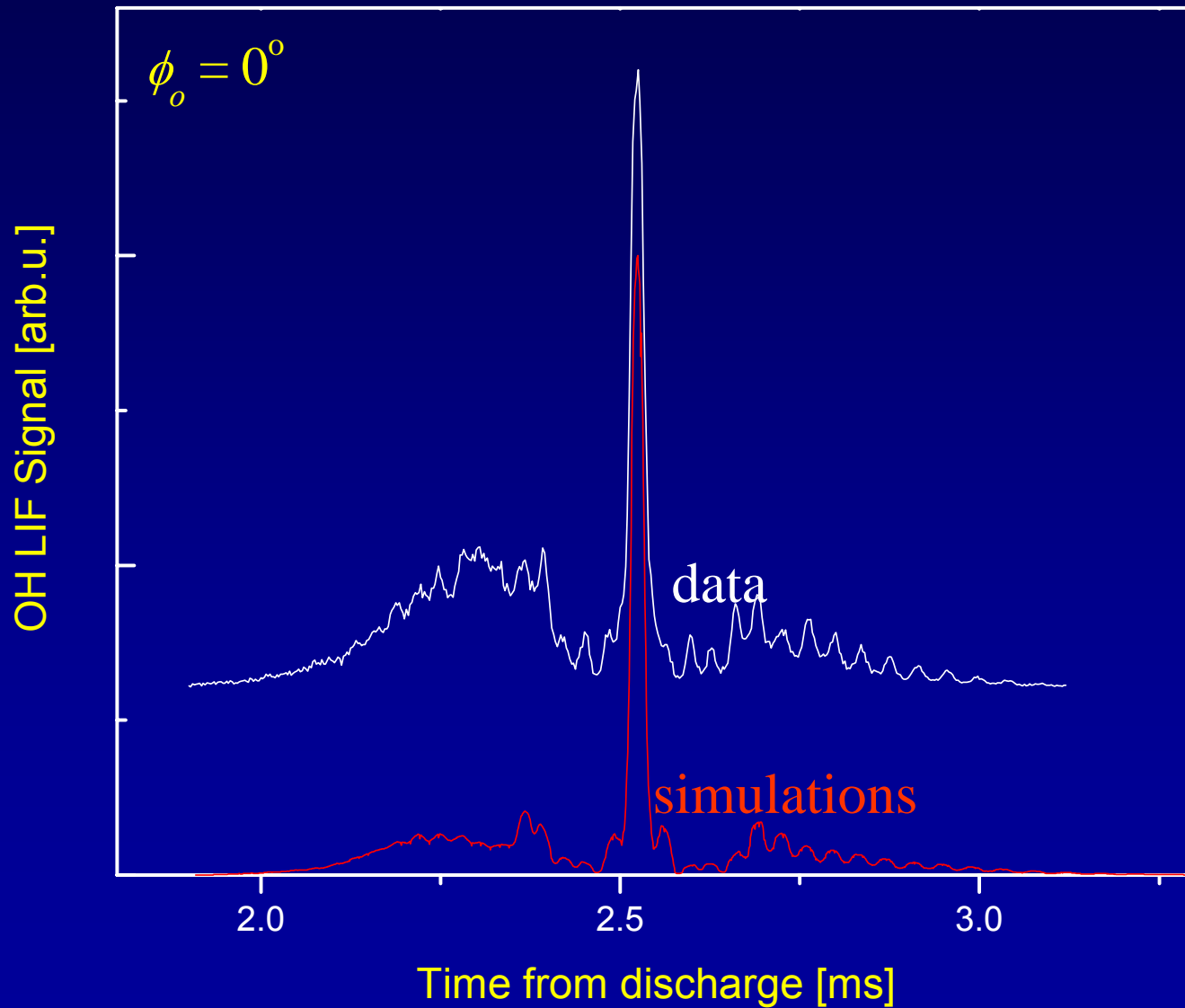
Equilibrium position

Pendulum driven by constant torque



$$\frac{d^2 \theta}{dt^2} - \left[\frac{g}{L} \sin(\theta) - \frac{T}{mL^2} \right] = 0$$

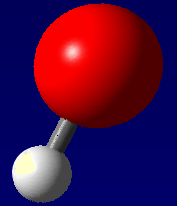
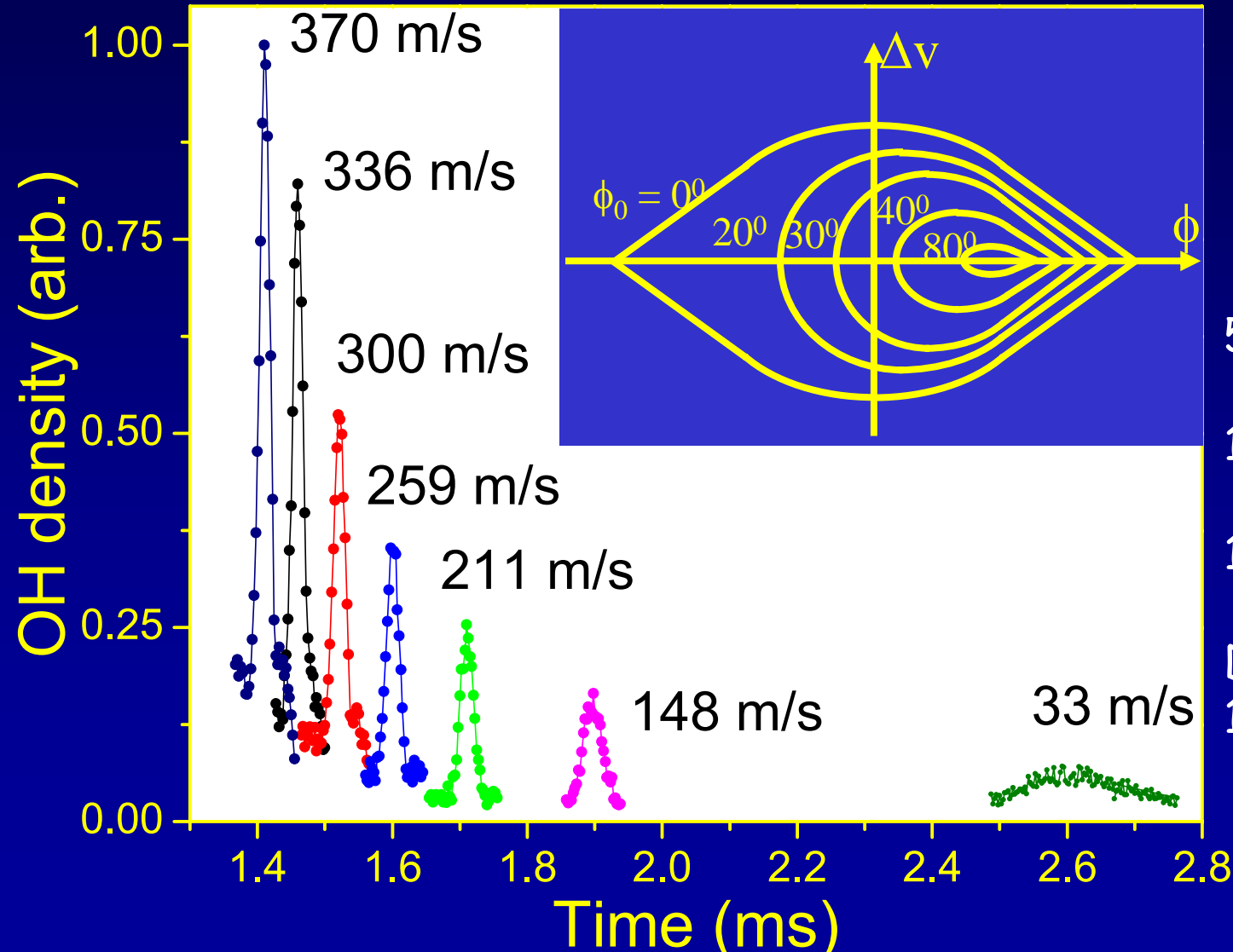
Experiment & Theory



Cold OH molecules

Bochinski, Hudson, Lewandowski, Meijer, Ye, Phys. Rev. Lett. **91**, 243001 (2003).

Bochinski, Hudson, Lewandowski, Ye, Phys. Rev. A **70**, 043410 (2004).



550 m/s to rest

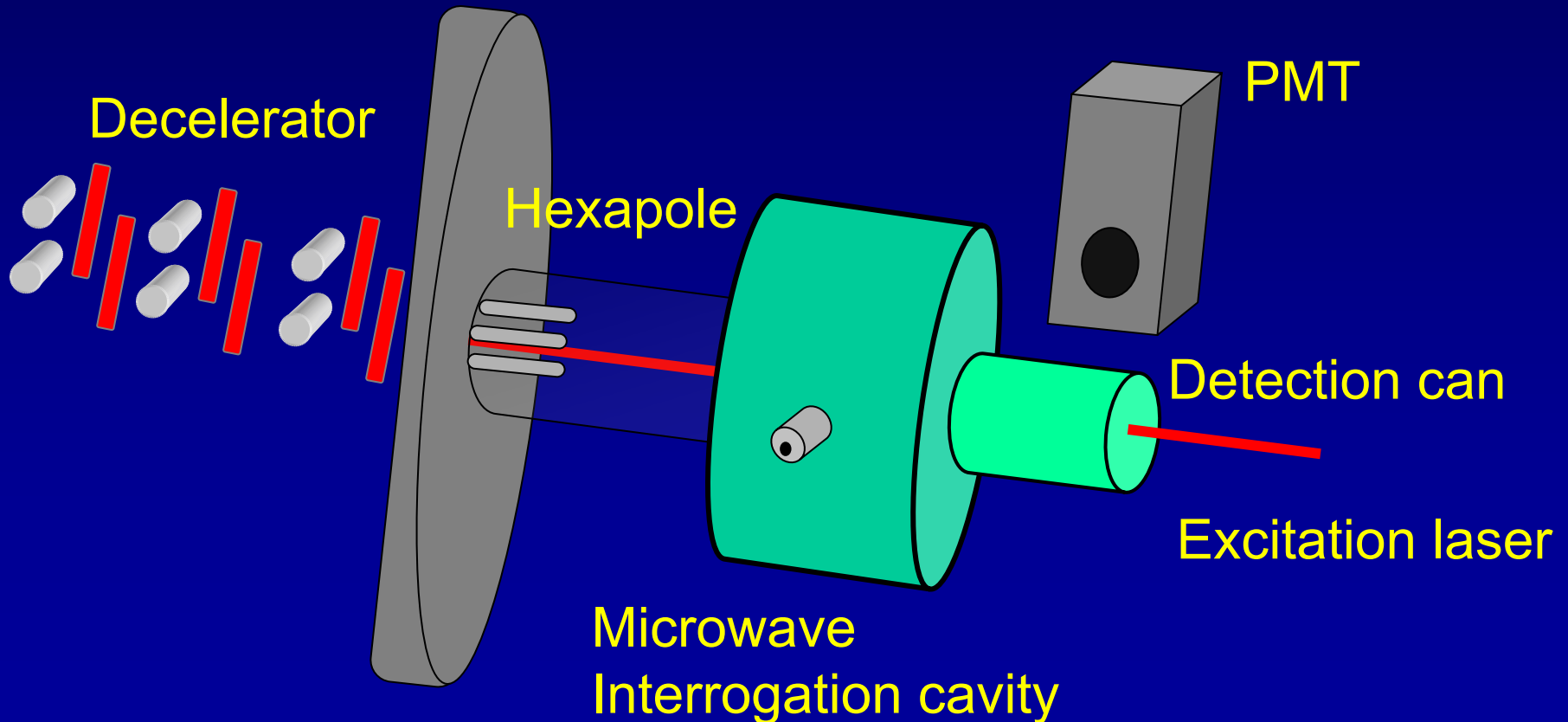
1 K to 10 mK

$10^4 - 10^6$ molecules

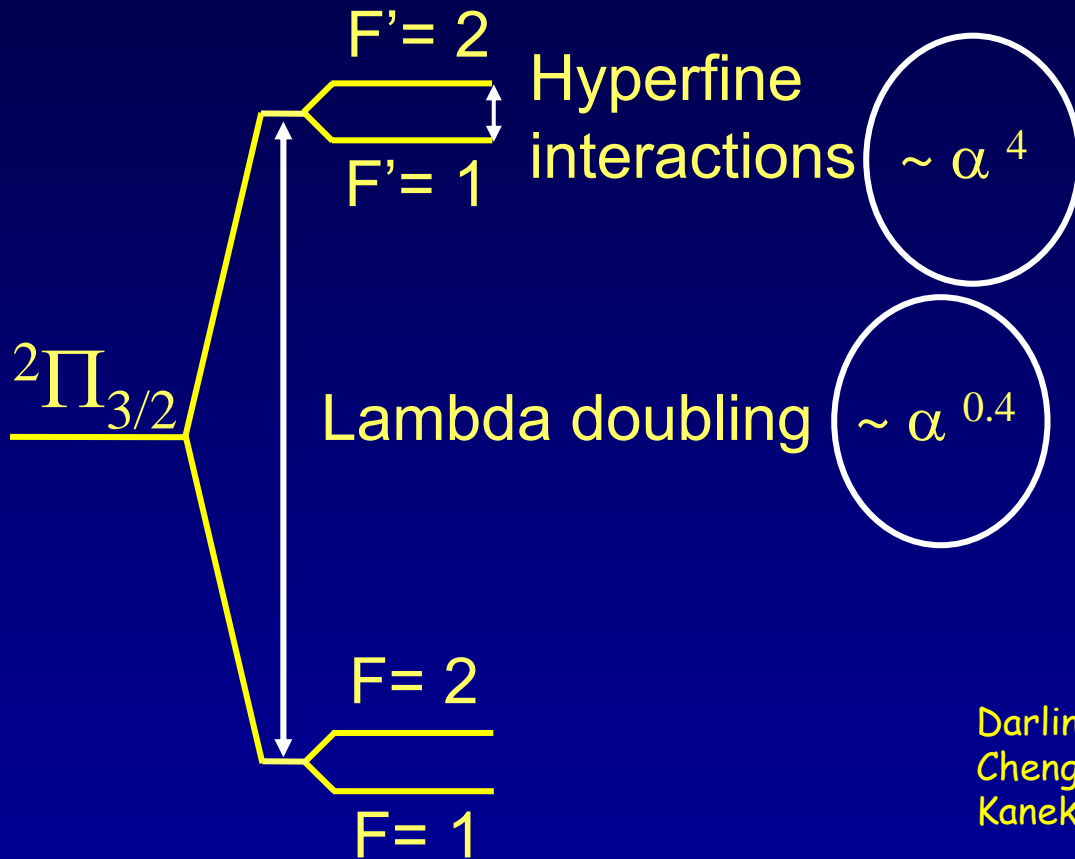
Density:
 $10^5 - 10^7 / \text{cm}^3$

Cold molecule based precision spectroscopy

- Rabi or Ramsey interrogation
- High resolution and precision
- Systematic evaluations



Cold OH molecules to constrain $\dot{\alpha}$



OH megamasers



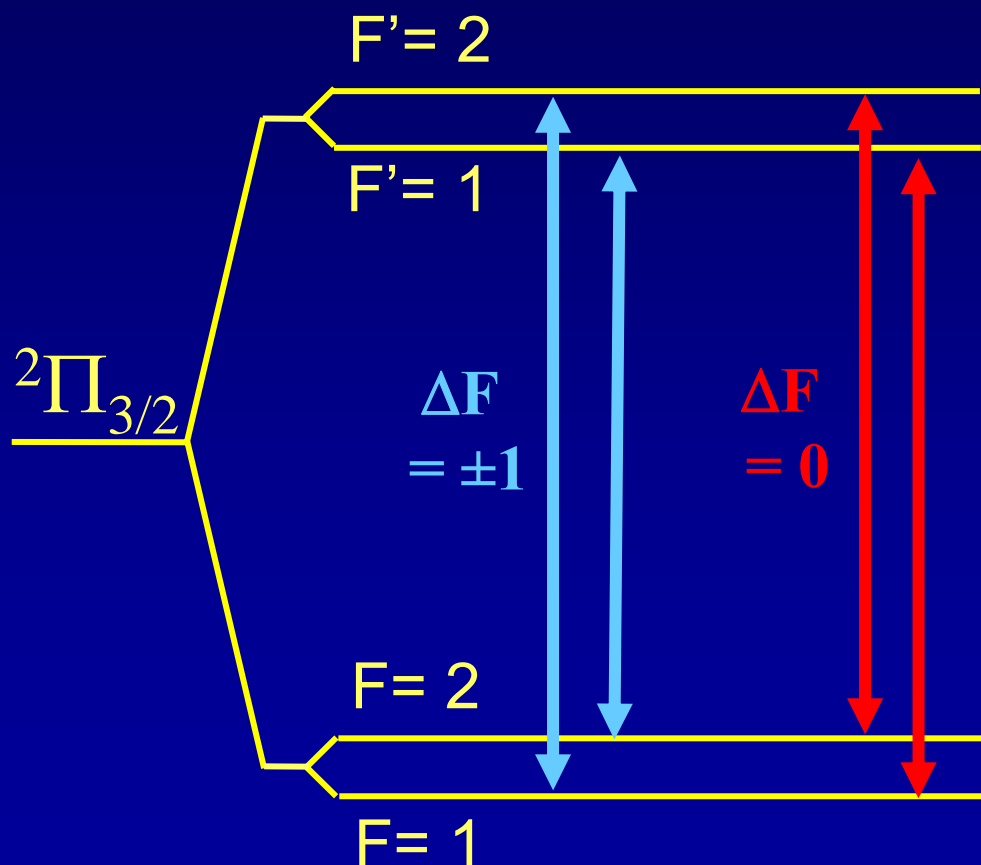
Darling, *Phys. Rev. Lett* **91**, 011301 (2003).
Chengalur *et al.*, *Phys. Rev. Lett.* **91**, 241302 (2003).
Kanekar *et al.*, *Phys. Rev. Lett.* **93**, 051302 (2004).

Multiple transitions from the same gas cloud
(Self check on systematics)

Precision measurement of OH ground structure

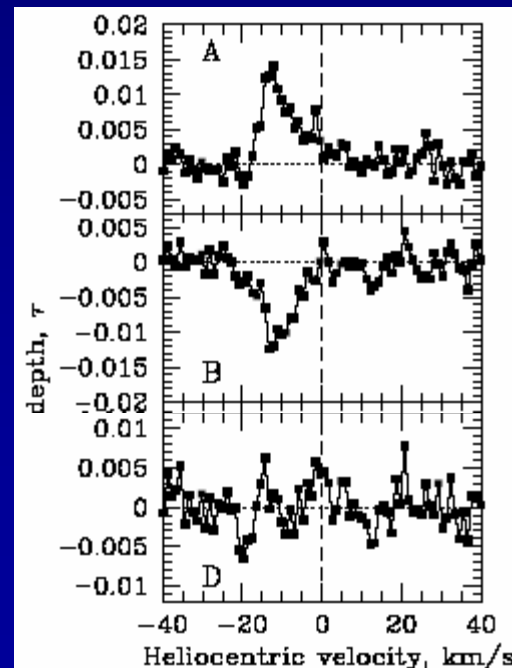
Measurement accuracy for all four lines: 4 - 10 Hz

Hudson, Lewandowski, Sawyer, Ye PRL 96, 143004 (2006).
Lev, Meyer, Hudson, Sawyer, Bohn, Ye, PRA 74, 061402 (2006).

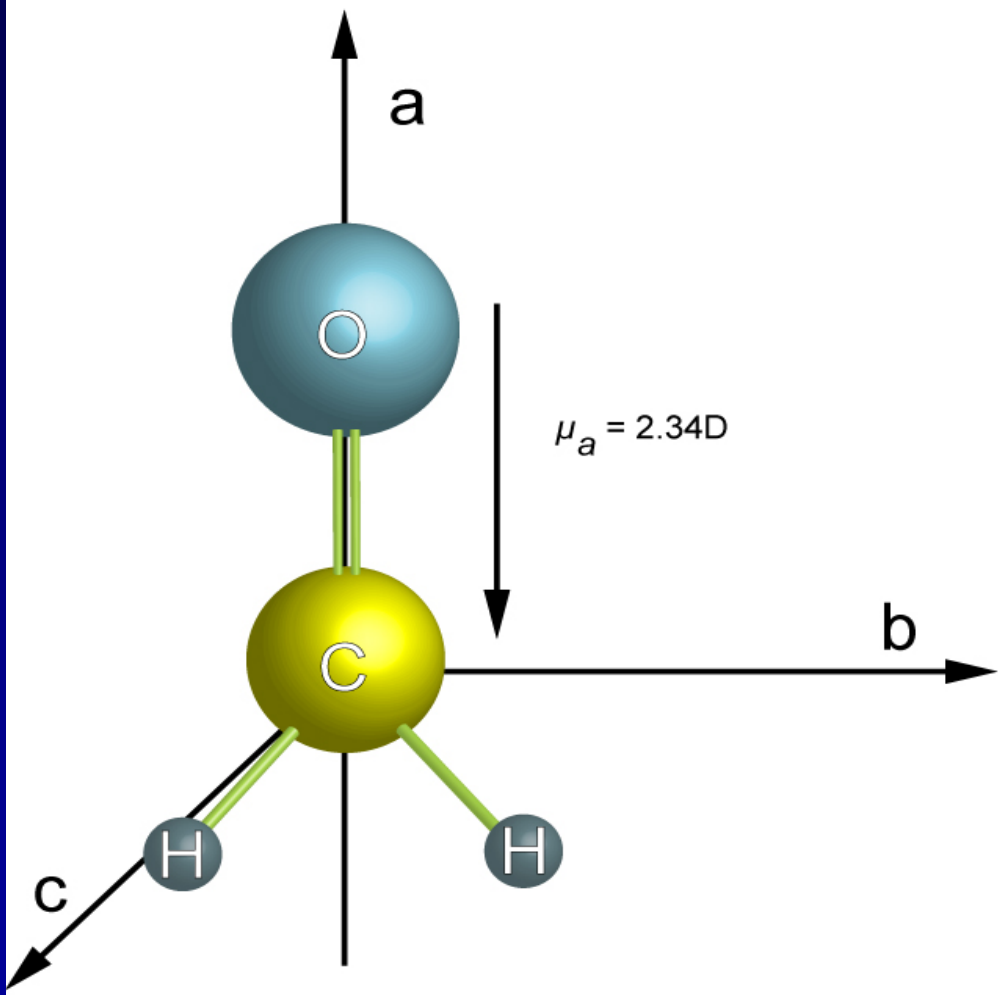


Kanekar *et al.*,
PRL 93, 051302
(2004).

- SUM (2 satellites)
= SUM (2 main lines)
- Satellites calibrate B
- Observed satellites conjugate



Cold Chemical Reactions



H₂CO – near symmetric rotor:

Formaldehyde (H₂CO):

Bring this most general class of molecules to rest in laboratory

H₂CO – OH:

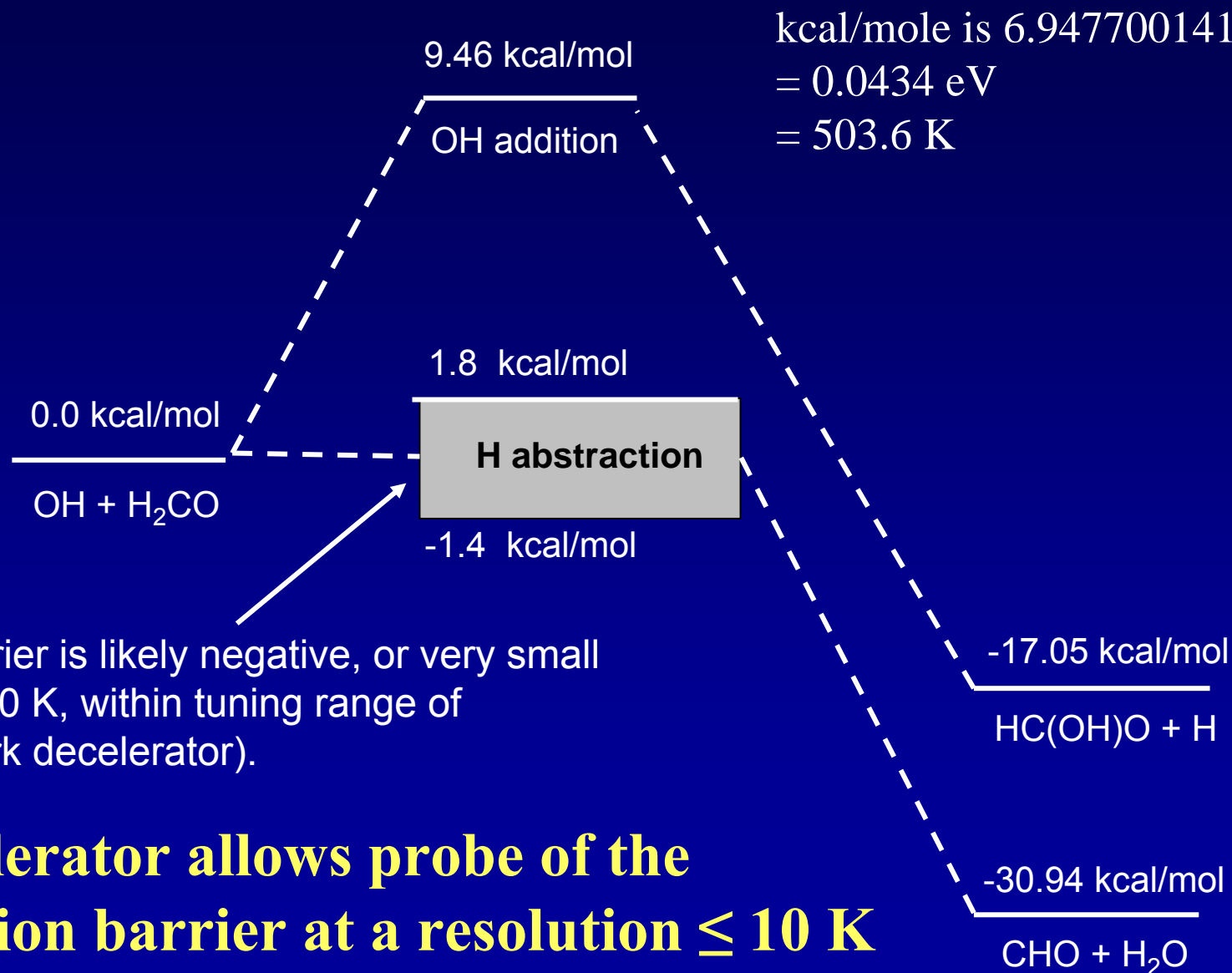
Fundamental reaction dynamics

Atmospheric chemistry

Combustion dynamics

Pollutant monitoring

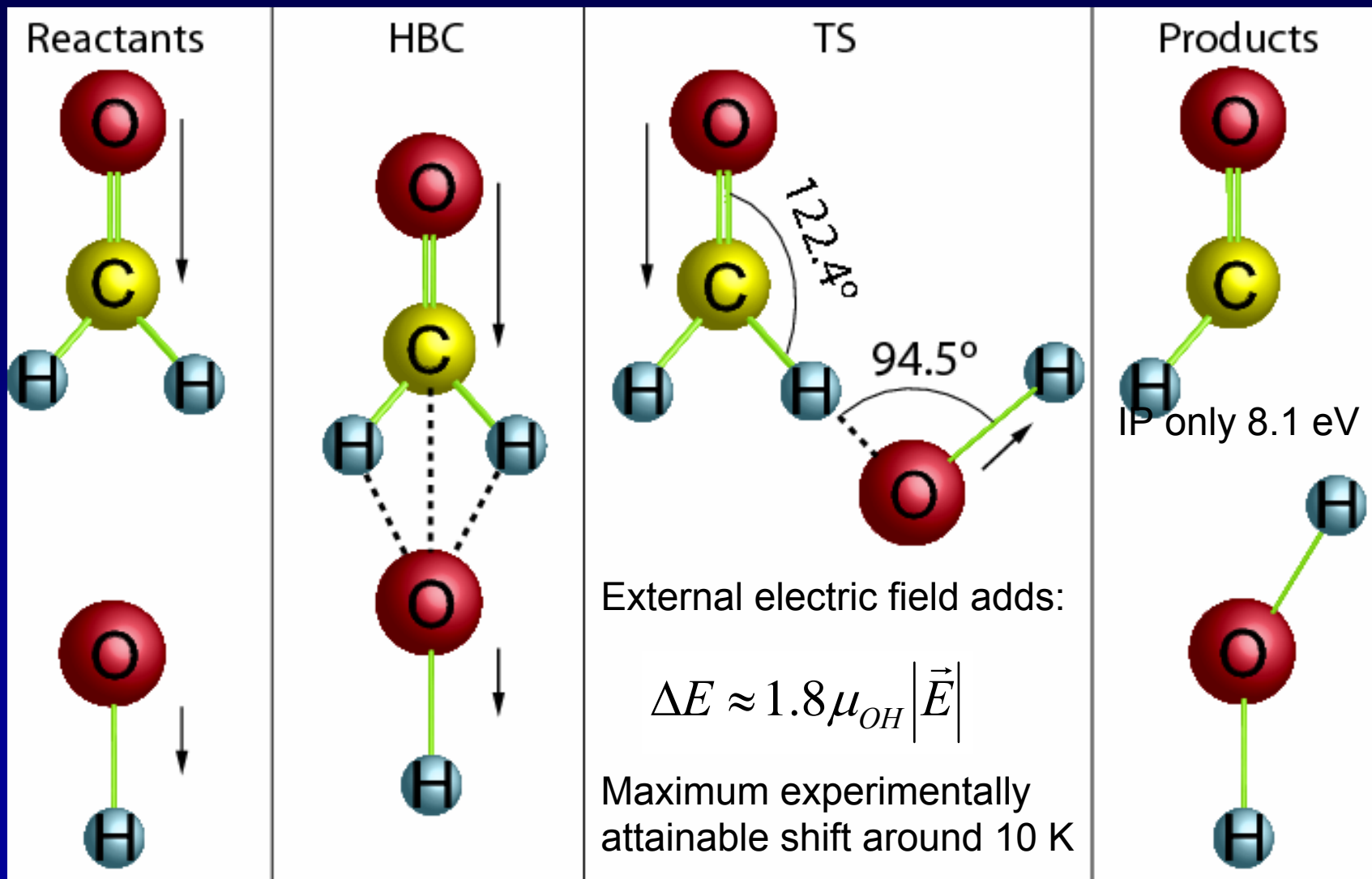
OH + H₂CO reaction pathways



Decelerator allows probe of the reaction barrier at a resolution ≤ 10 K

External electric field tunes reaction barrier

Hudson, Ticknor, Sawyer, Taatjes, Lewandowski, Bochinski, Bohn, Ye,
Phys. Rev. A 73, 063404 (2006).



Control of cold chemical reactions; Unique dipolar interaction dynamics

A pressing requirement

Enhancement of the phase space density of cold molecules!

Sympathetic cooling in a trap

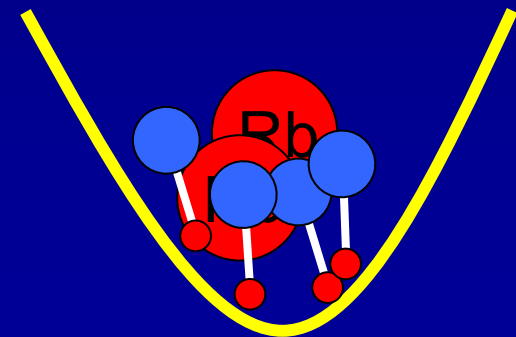
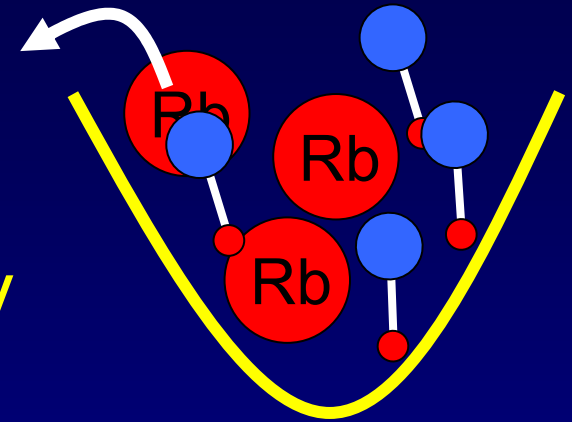
Molecules and atoms co-located in a trap.

Selectively remove highest energy Rb atoms.

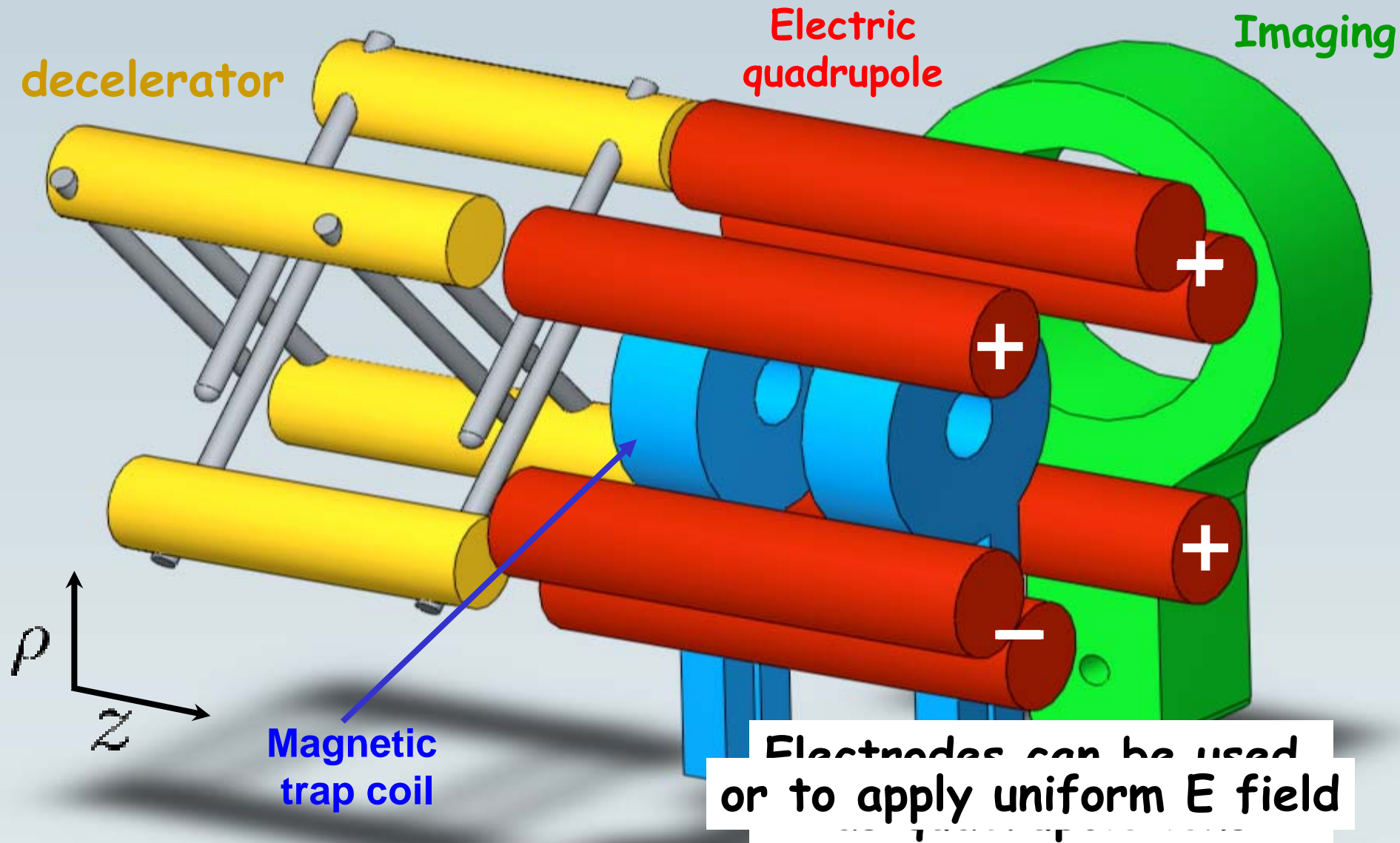
Atoms and molecules rethermalize, through collisions, to a lower temperature.

Requires favorable collision rates

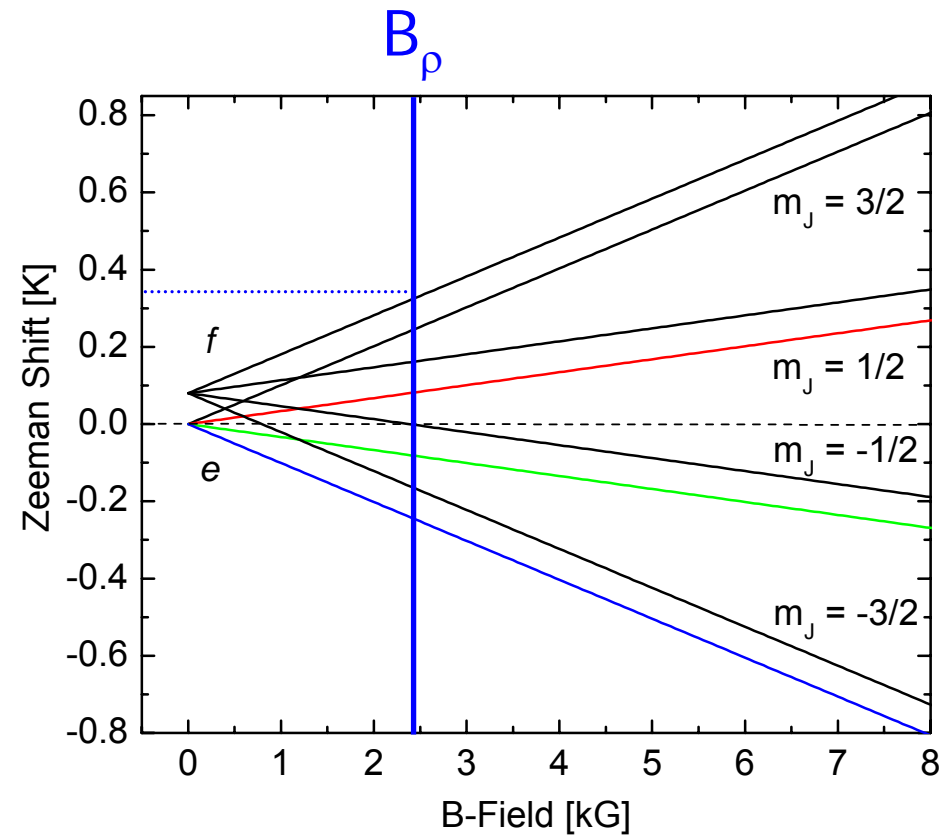
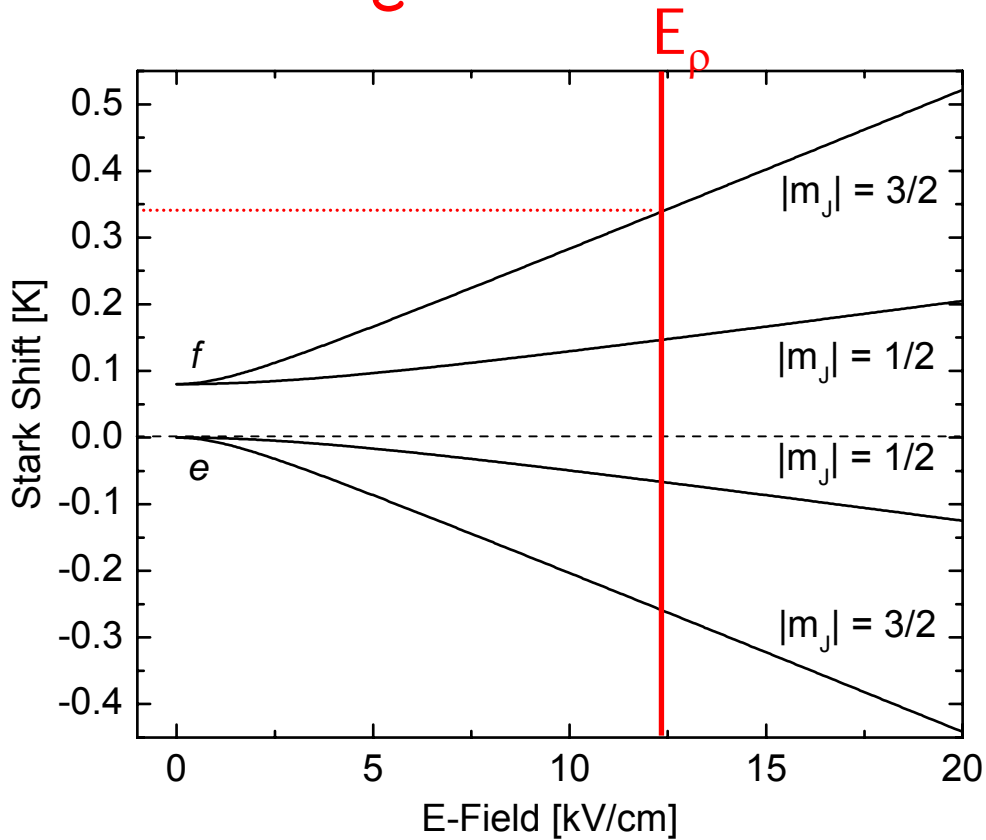
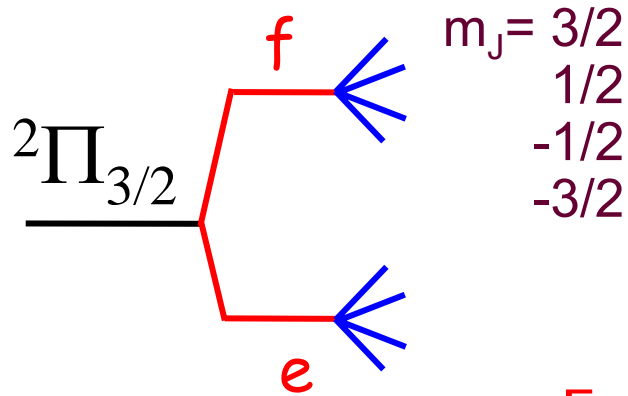
Jeremy Hutson's talk (Friday)



Magnetic trapping of OH



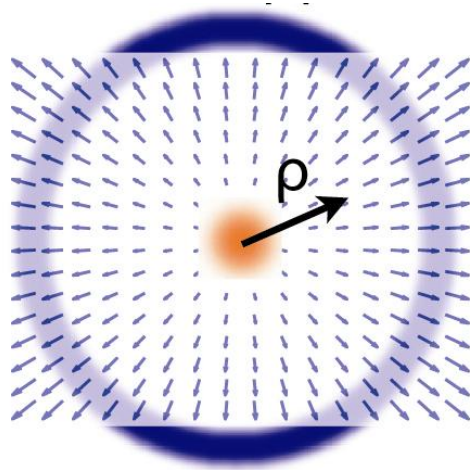
OH Stark and Zeeman effects



R. Krems' Talk Friday

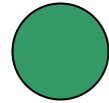
Trapping Scheme

End view



Magnetic
Quadrupole

20 m/s

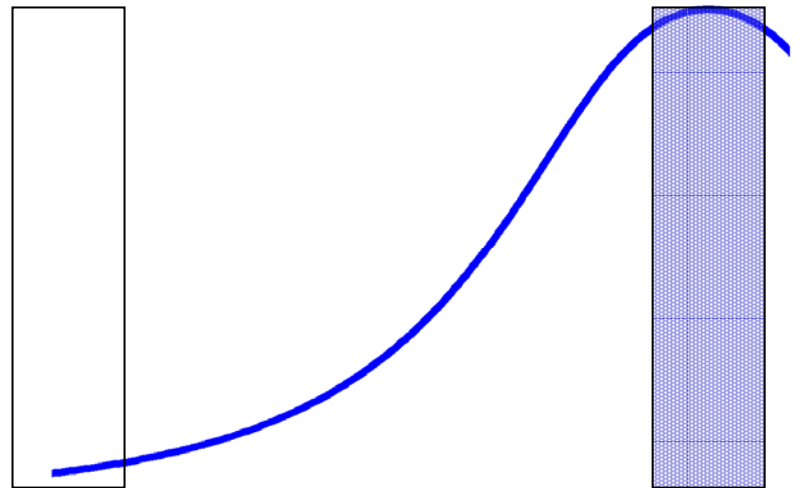
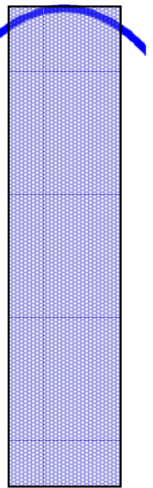


Side view

0 A



2000 A

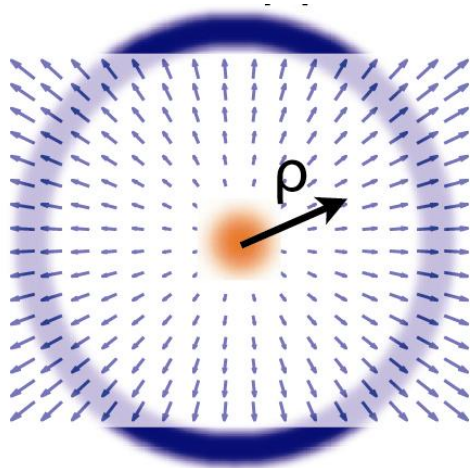


z

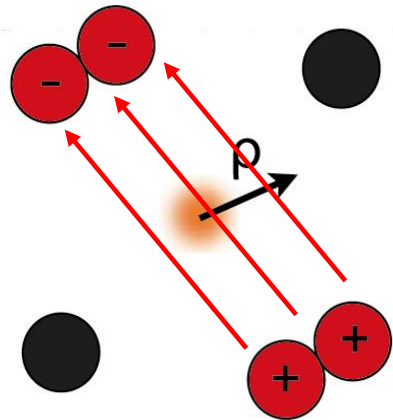
Trapping Scheme

End view

Magnetic
Quadrupole



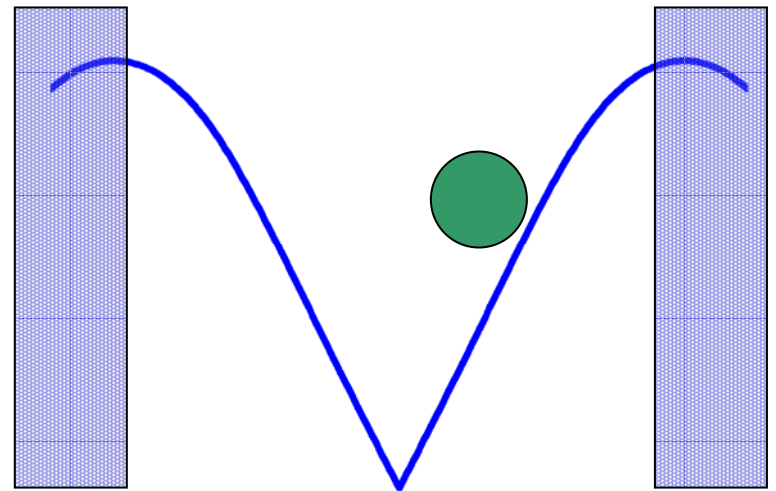
Uniform
Electric Field



Side view

-1500 A

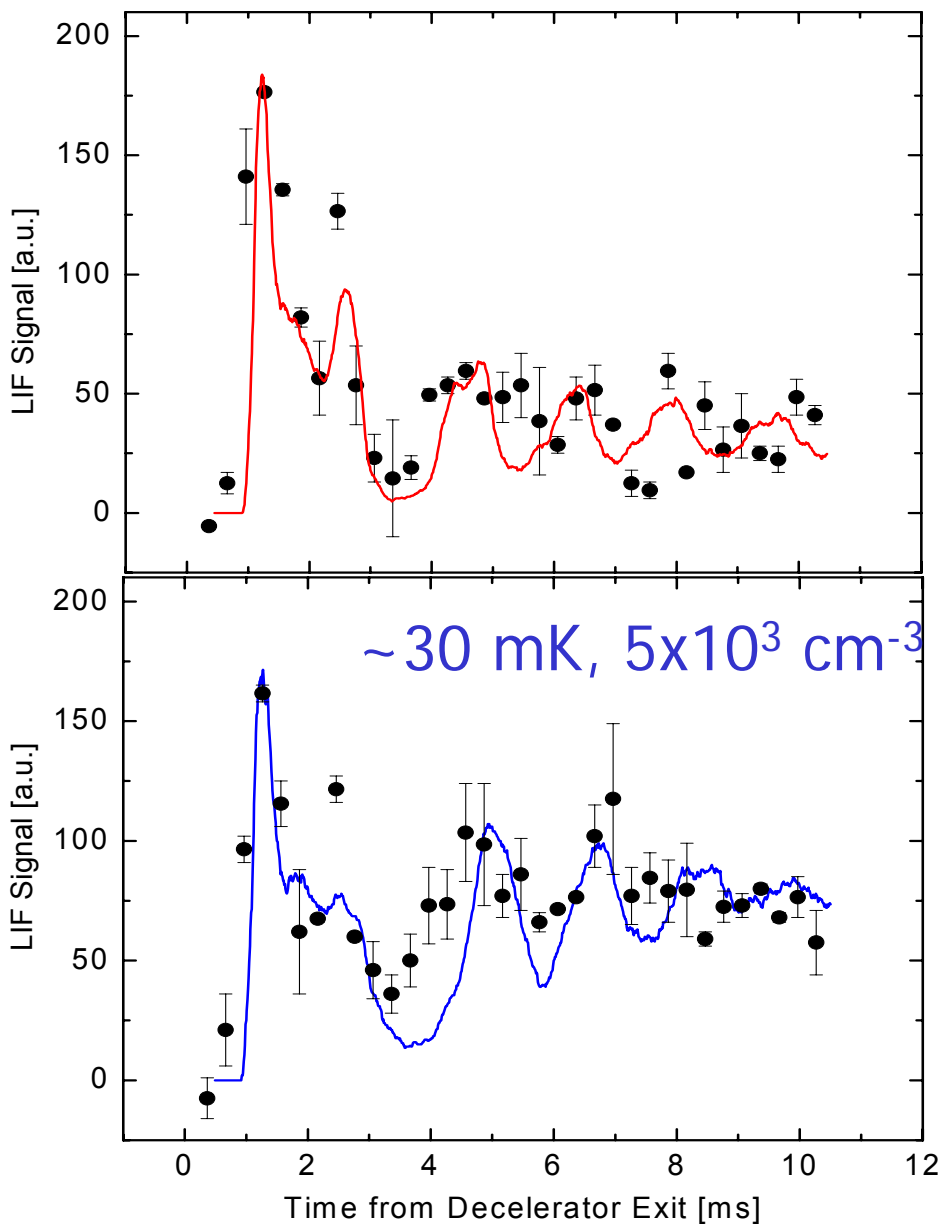
1500 A



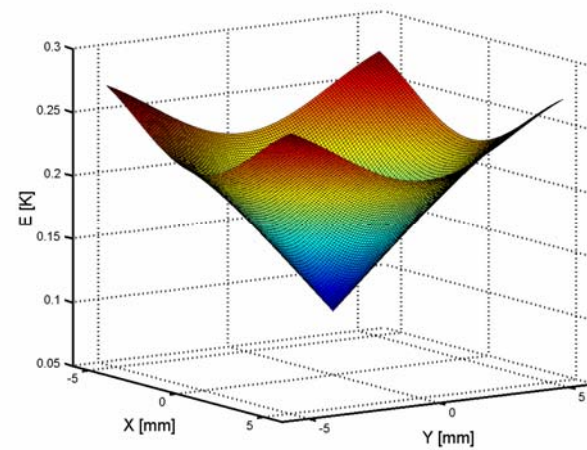
z

Trap dynamics

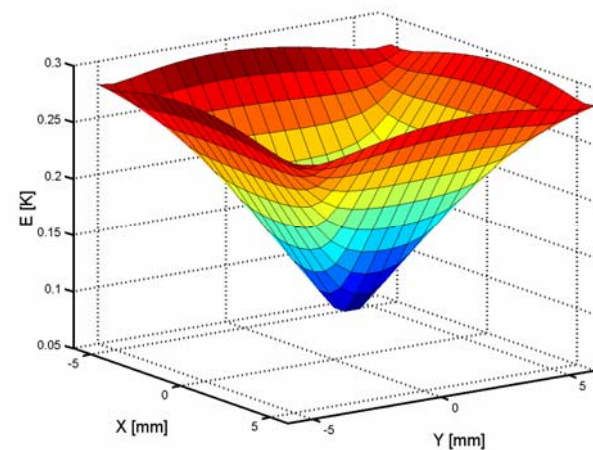
Sawyer, Lev, Hudson, Stuhl, Lara, Bohn, Ye, physics/0702146, PRL in press (2007).



Quadrupole
B

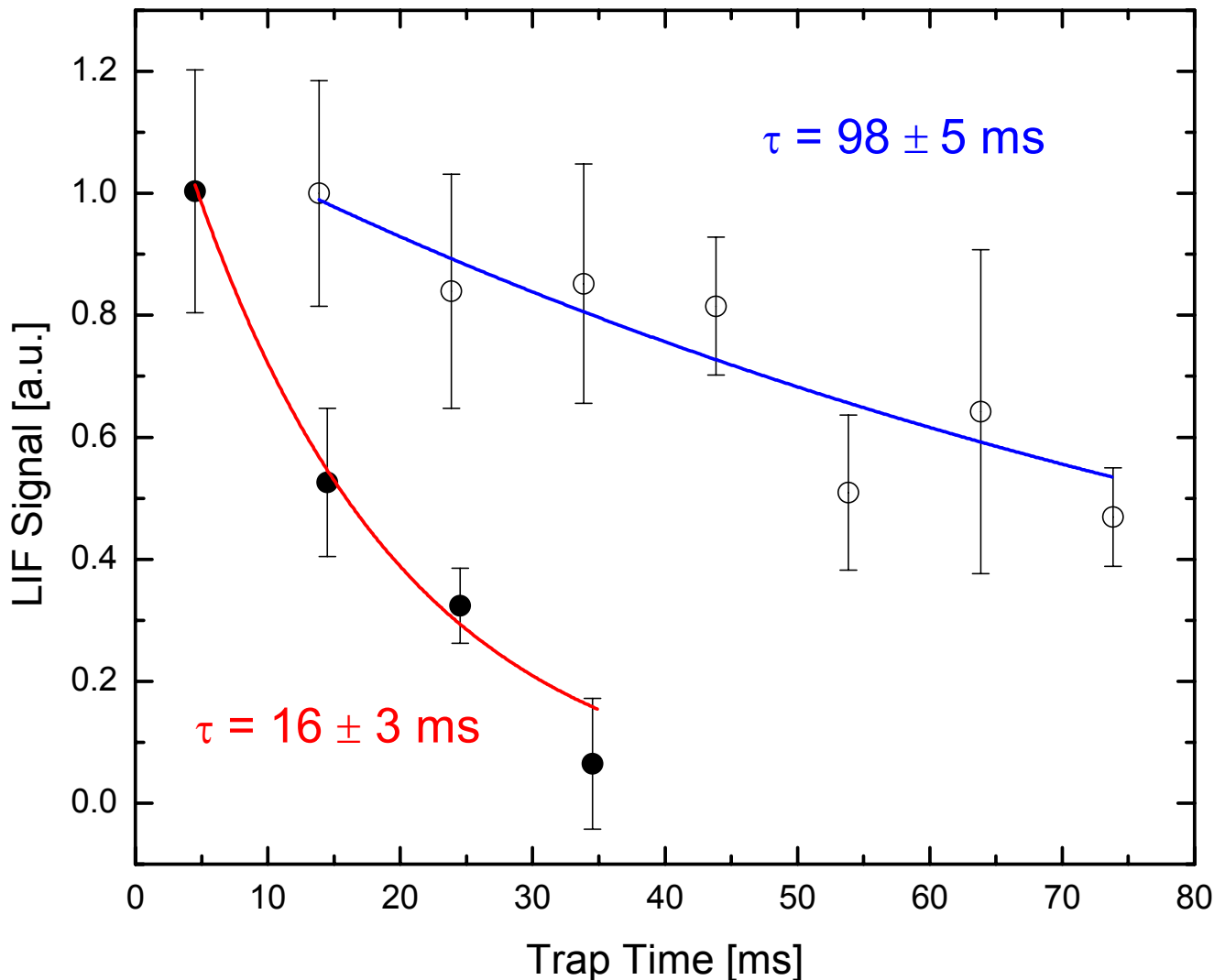


Quadrupole
E, B



OH - N₂ collisions

Collision-limited lifetimes



1×10^{-6} Torr N₂
 20 ± 5 ms;

4×10^{-8} Torr N₂
 500 ± 100 ms

OH - N₂
cross section:

500 ± 100 Å²

Soon to study dipolar collisions!

Special thanks

<http://jilawww.colorado.edu/YeLabs>

Ultracold Sr & Sr₂

M. Boyd
A. Ludlow
S. Blatt
Dr. T. Zelevinsky
Dr. G. Campbell
Dr. T. Zanon

Cold Polar Molecules

B. Sawyer
Dr. B. Lev
B. Stuhl

Dr. S. Ospelkaus

Femtosecond comb & quantum control

S. Foreman
M. Thorpe
D. Hudson
D. Yost
M. Stowe
Dr. T. Schibli
Dr. A. Pe'er

Collaborators

J. Bohn, D. Jin (JILA); E. Hudson, H. Lewandowski, J. Bocinski (former members)
P. Julienne, S. Kotochigova (NIST)
M. Shapiro, E. Shapiro (UBC)