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Some Perspectives on Ultracold  
Atomic and Molecular Interactions  
and their Control

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Looking for good students/postdocs  
Joint Quantum Institute  
NIST / University of Maryland

<http://www.jqi.umd.edu/>  
<http://physics.nist.gov/>

## Cold atoms and molecules

Widely used in forefront experiments

Bose gases and BEC

Fermi gases, BEC-BCS crossover

Lattices and reduced dimensional structures

Control scattering properties

by static or dynamic electromagnetic fields

depends on specific physical systems available

Complex calculations required

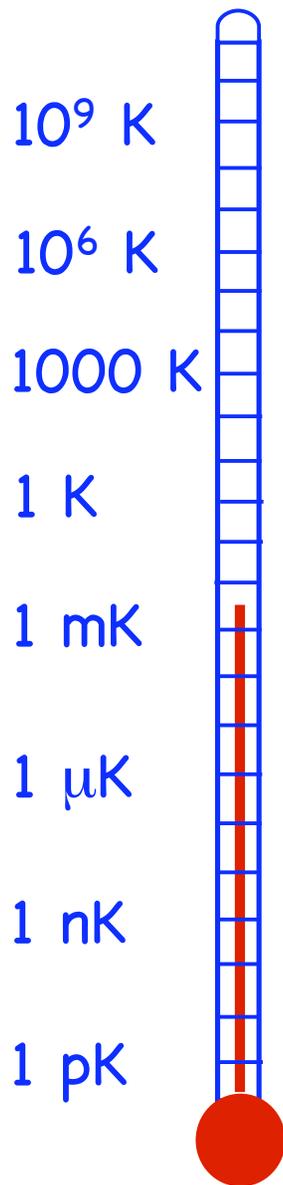
coupled channels scattering

*ab initio* structure and properties

But remain amenable to simple models

Universal--parameterized by  $a(B)$  and mass

Based on long range potentials



Interior of sun

Surface of sun  
Room temperature

Outer space (3K)  
Cold He

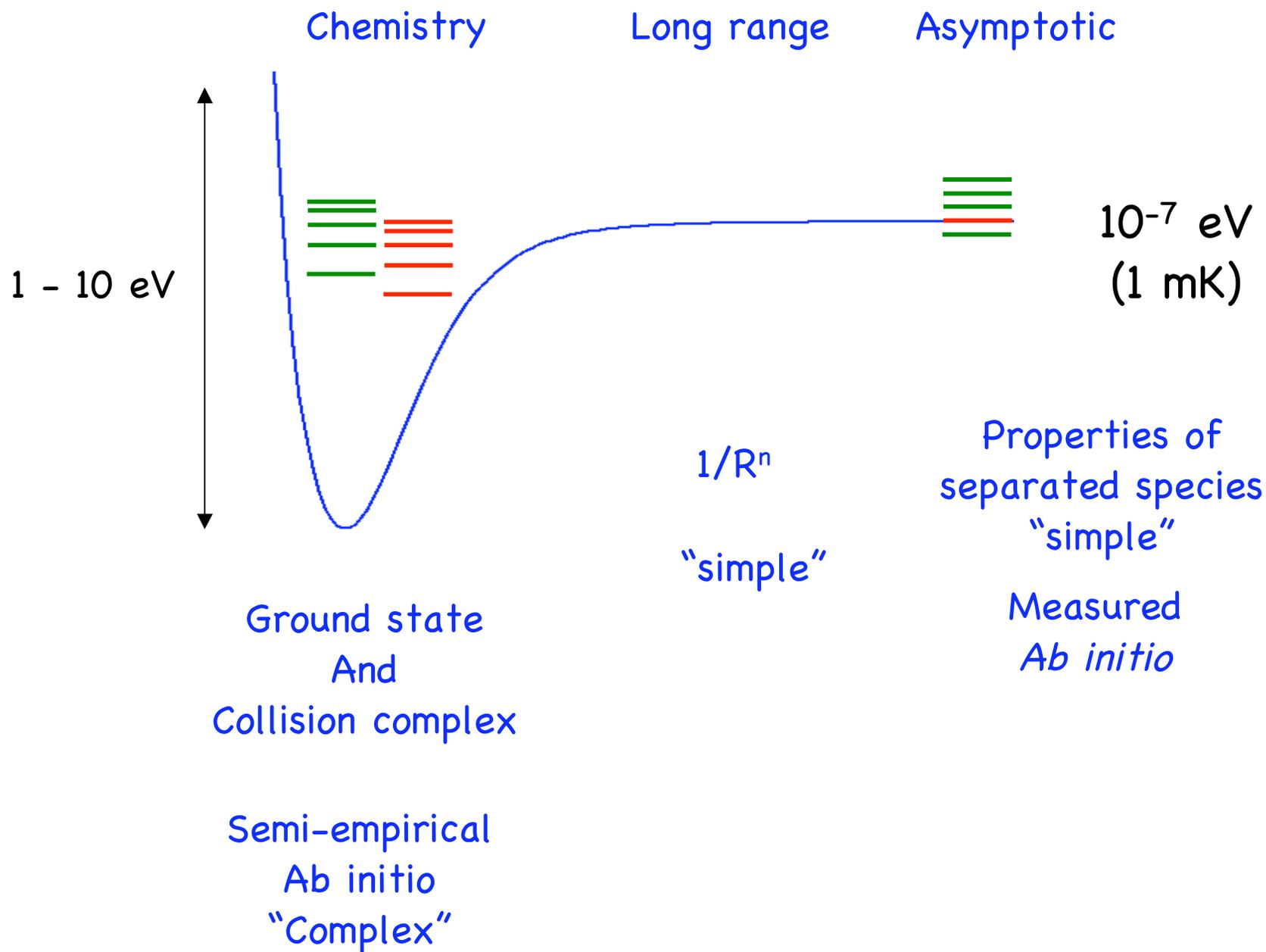
Laser cooled atoms

Atomic clock atoms

Fermionic quantum gases

Bose-Einstein condensates

Molecules  
↓  
Buffer gas cooling  
Decelerated beams  
Photoassociated atoms  
Feshbach molecules  
Molecular BEC



## s-wave Threshold Collisions Summary

Complex scattering length  $a - ib$   
( from exact  $E \rightarrow 0$  S-matrix element)

$$\text{Elastic collision } \sigma = 4\pi(a^2 + b^2)$$

$$\text{Inelastic collision } K = \sigma v = (2h/\mu)b$$

$$K_{\text{loss}} = \frac{2h}{\mu}b = 10^{-10} \text{ cm}^3/\text{s} \frac{b(a_0)}{\mu(\text{amu})}$$

Typical values: "Allowed"  $b \sim 10\text{-}100 a_0$

"Forbidden"  $b \ll 1 a_0$

Upper bound  $4b = k^{-1} = \lambda/2\pi$

How fast are inelastic cold collisions (Maxwell-Boltzmann)?

$$\frac{dn}{dt} = -2Kn^2 = -\frac{1}{\tau}n \quad \text{where} \quad \frac{1}{\tau} = 2Kn$$

$$K = \frac{1}{Q_T} \frac{k_B T}{h} \sum_{\ell} (2\ell + 1) \langle |S(E)|^2 \rangle$$

where  $\frac{1}{Q_T} = \Lambda_T^3 = \left( \frac{h}{2\pi\mu k_B T} \right)^{\frac{3}{2}}$

$Q_T$  = translational partition function

$\Lambda_T$  = thermal de Broglie wavelength

Probability  $|S|^2 < 1$   
Dynamical factor

$$\frac{1}{\tau} = 2Kn = 2 \underbrace{(n\Lambda_T^3)}_{\substack{\text{Phase Space} \\ \text{density}}} \underbrace{\frac{k_B T}{h}}_{\substack{\text{Upper} \\ \text{bound}}} \underbrace{f}_{\substack{\text{Dynamics}}}$$

$$\frac{k_B T}{h} = 21 \text{ kHz at } 1 \mu\text{K}$$

“Size” of  $-C_6/R^6$  van der Waals potential  $V(R)$

$$R_{\text{vdw}} = \frac{1}{2} \left( \frac{2\mu C_6}{\hbar^2} \right)^{\frac{1}{4}} \quad \text{or} \quad \bar{a} = \frac{\Gamma(3/4)}{\Gamma(5/4)} R_{\text{vdw}} = 0.956 R_{\text{vdw}}$$

G. F. Gribakin and V. V. Flambaum  
Phys. Rev. A 48, 546 (1993)

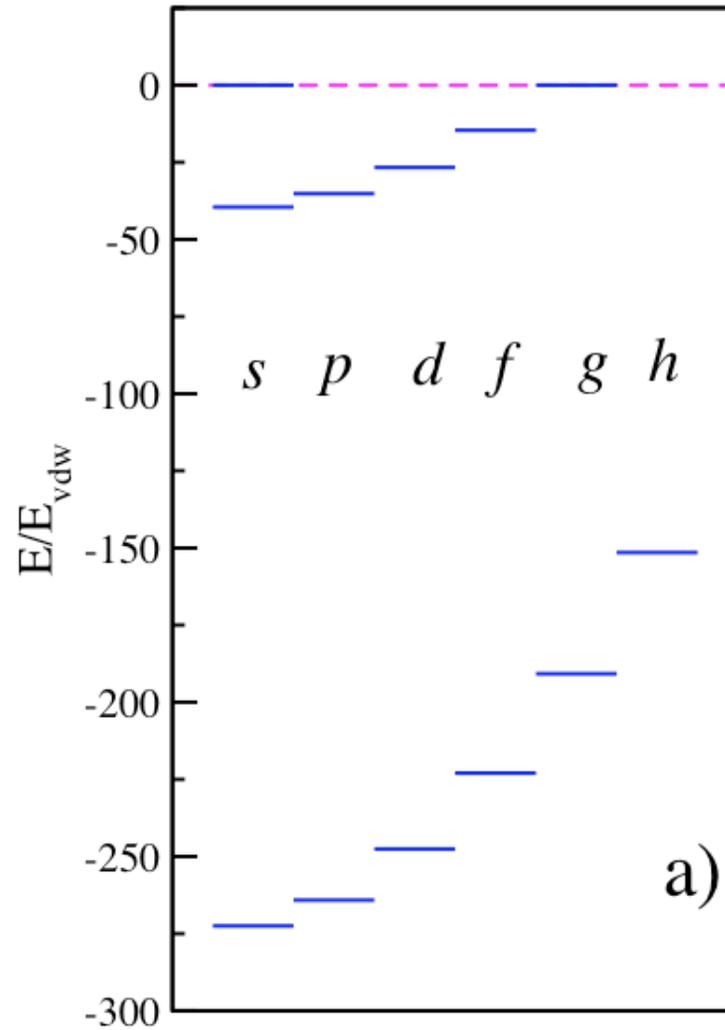
$$E_{\text{vdw}} = \frac{\hbar^2}{2\mu R_{\text{vdw}}^2}$$

	$R_{\text{vdw}}(a_0)$	$E_{\text{vdw}}(\text{mK})$
${}^6\text{Li}$	31	29
${}^{40}\text{K}$	65	1.0
${}^{85}\text{Rb}$	83	0.35
${}^{133}\text{Cs}$	101	0.13

See Jones, Lett, Tiesinga, Julienne, Rev. Mod. Phys. 78, 483 (2006)

# Bound states from van der Waals theory

$$a = \pm\infty$$



Adapted from Gao, Phys. Rev. A 62, 050702 (2000); Figure from E. Tiesinga

## Effective range and bound states of vdW $V(R)$

Effective range expansion of  $\eta(E)$

$$k \cot \eta(E) = -\frac{1}{a} + \frac{1}{2}r_0 k^2$$

$$r_0 = 2.918 \dots \bar{a} \left( 1 - 2\frac{\bar{a}}{a} + 2\left(\frac{\bar{a}}{a}\right)^2 \right)$$

Gao, Phys. Rev. A 62, 050702 (2000)  
Flambaum, Gribakin, and Harabati,  
Phys. Rev. A 59, 1998 (1999)

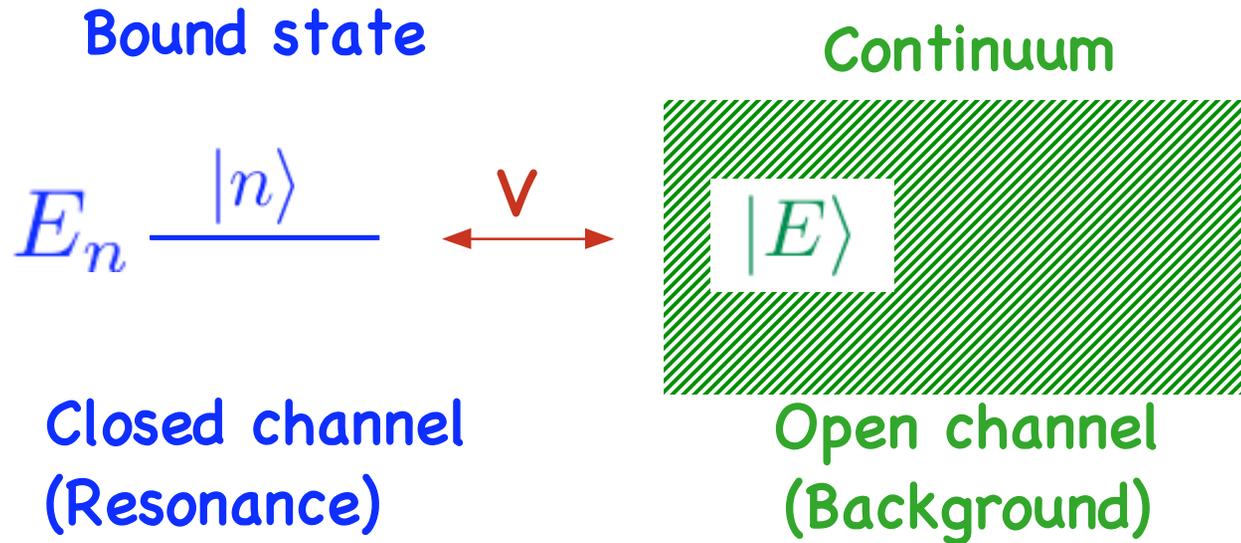
$$E_{-1} = \frac{\hbar^2}{2\mu(a - \bar{a})^2}$$

G. F. Gribakin and V. V. Flambaum  
Phys. Rev. A 48, 546 (1993)  
Gao, J. Phys. B 37, 4273 (2004)

See P. Naidon, E. Tiesinga, W. F. Mitchell, and P. S. Julienne, "Effective-range description of a Bose gas under strong one- or two-dimensional confinement," N. J. Phys. 9, 19 (2007).

# Resonant Scattering Picture

(U. Fano, Phys. Rev. 124, 1866 (1961))



$$\eta(E) = \eta_{\text{bg}} + \eta_{\text{res}}(E)$$

$$\eta_{\text{res}} = -\tan^{-1} \frac{\frac{1}{2}\Gamma_n}{E - E_n - \delta E_n}$$

Annotations for the resonance term:

- width**:  $\Gamma_n = 2\pi |\langle n|V|E\rangle|^2$
- shift**:  $\delta E_n$

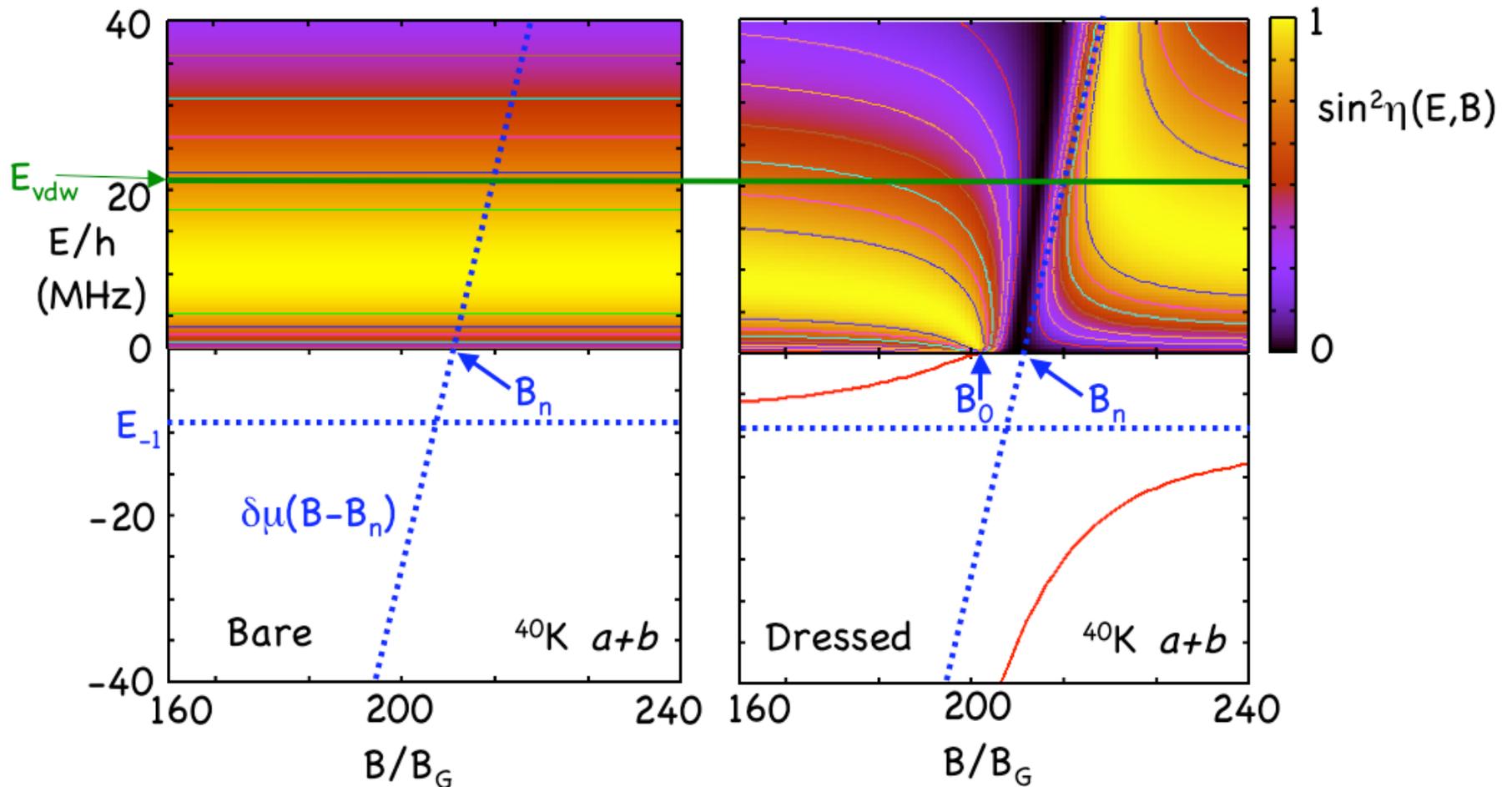
Analytic van der Waals theory reproduces ALL scattering versus E and B in the ultracold domain as compared to full coupled channels calculations.

Julienne and Gao, in *Atomic Physics 20* (AIP), p. 216, or physics/0609013

$$\eta_{\text{res}} = -\tan^{-1} \frac{\frac{1}{2}\Gamma(E)}{E - \delta\mu(B - B_n) - \delta E_n(E)}$$

$$\Gamma(E) = \frac{1}{2}\bar{\Gamma}C(E)^{-2}$$

$$\delta E_n(E) = -\frac{1}{2}\bar{\Gamma}\tan\lambda(E)$$



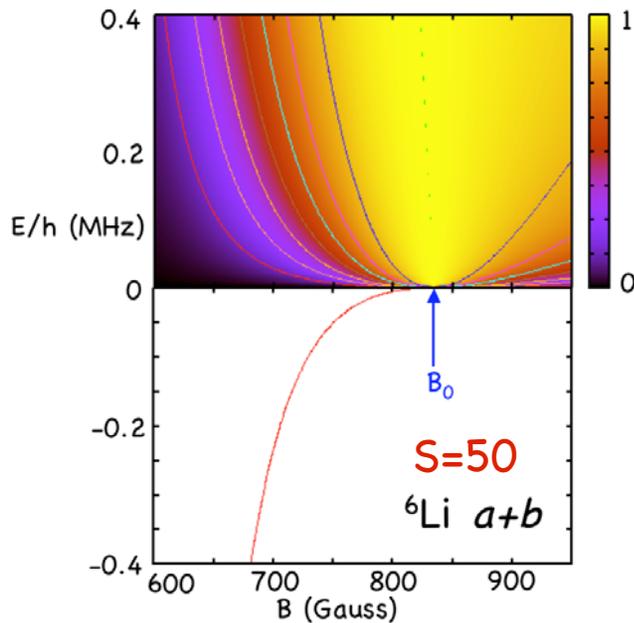
# Open or closed channel dominated?

$$S = \frac{a_{bg}}{\bar{a}} \frac{\delta\mu \Delta}{E_{vdW}}$$

Köhler, Goral, Julienne,  
Rev. Mod. Phys. 78, 1311 (2006)

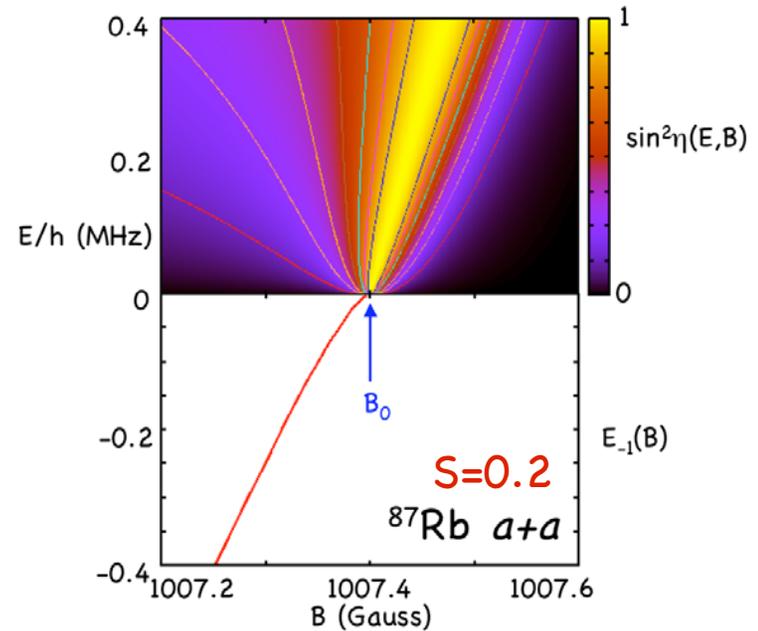
Open:  $S \gg 1$   
Universal,  $Z \ll 1$   
over  $\approx \Delta$

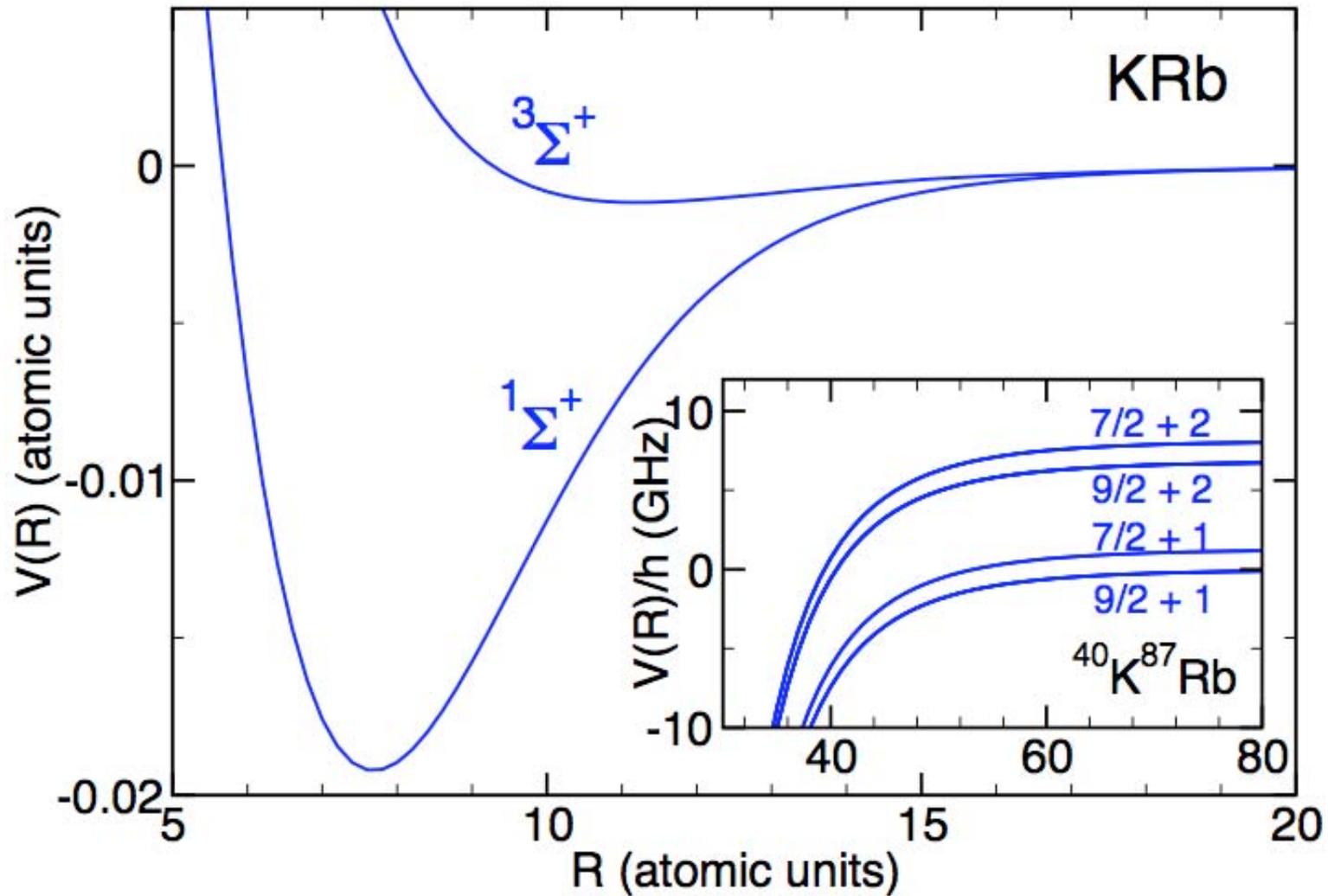
$\Gamma(E) \gg E$  when  $E < E_{vdw}$



Closed:  $S \ll 1$   
Not universal,  $Z \approx 1$   
over most of  $\Delta$

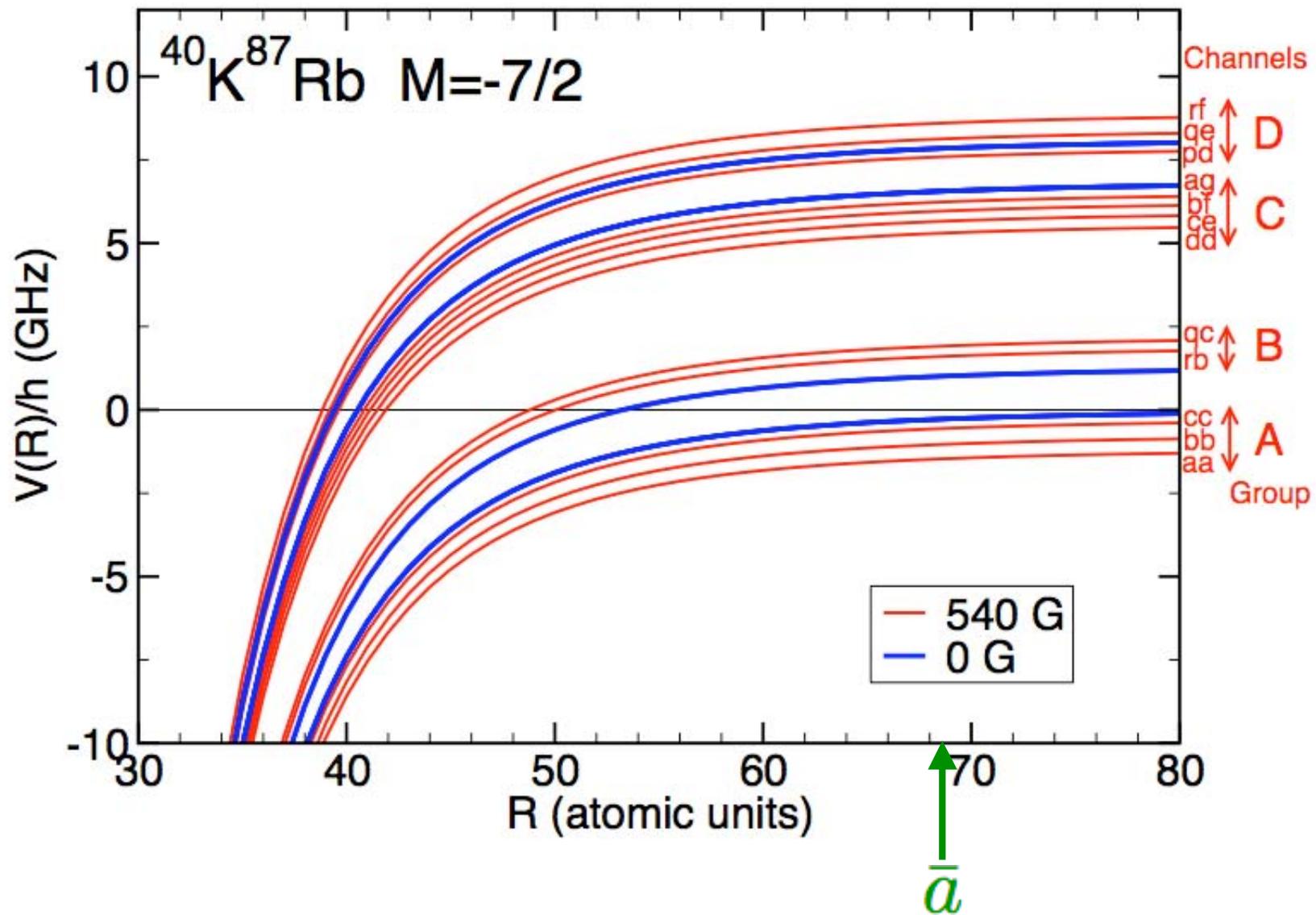
$\Gamma(E) \ll E$  when  $E < E_{vdw}$



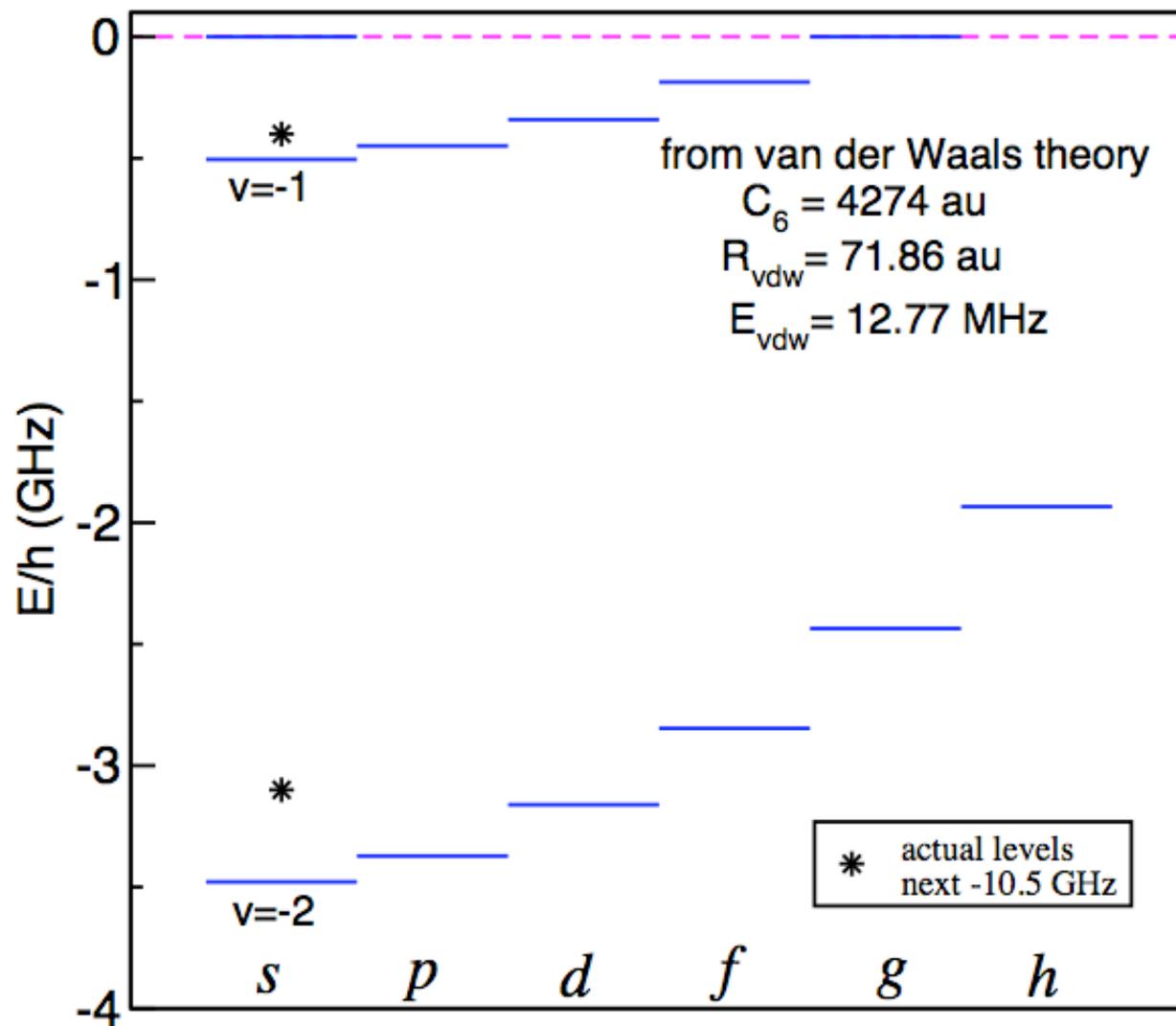


*Ab initio*: Kotochigova , Julienne, Tiesinga, Phys. Rev. A 68,022501 (2003).

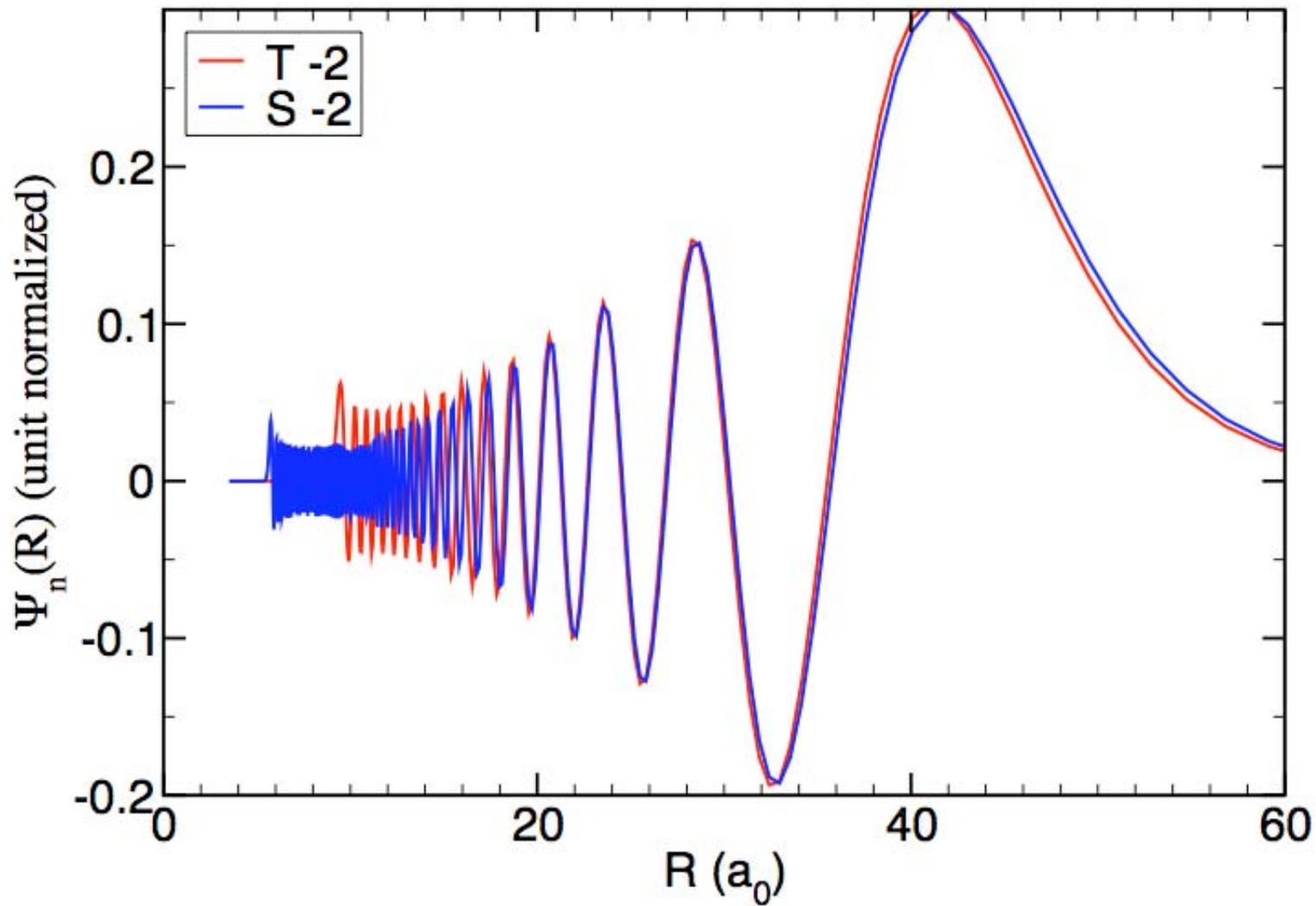
See also Aymar & Dulieu, JCP, 122, 204302(2005)



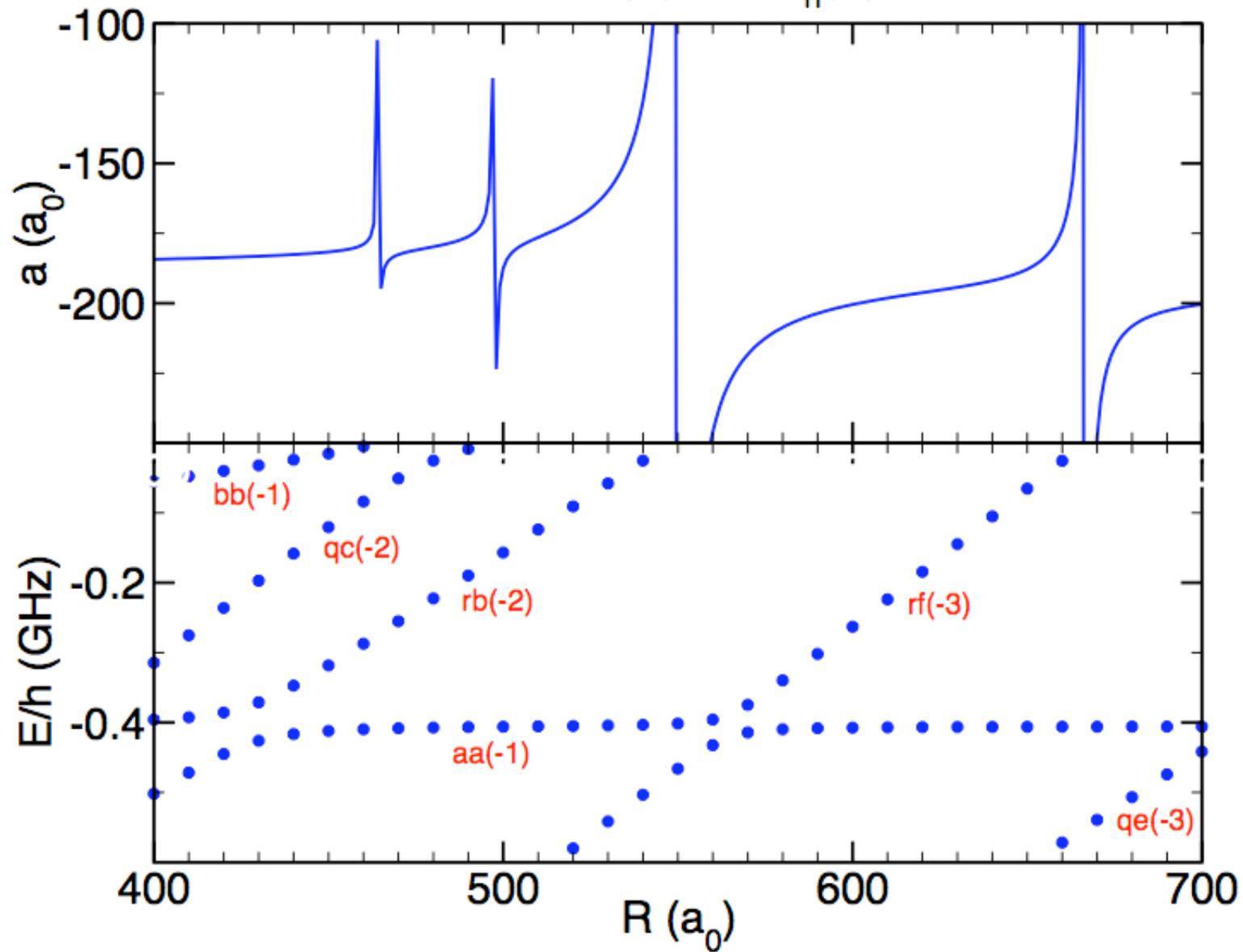
$^{40}\text{K}^{87}\text{Rb}$  energy level "bins"



$^{40}\text{K}^{87}\text{Rb}$   $n=-2$  levels  
single channel  $\Psi_n(R)$

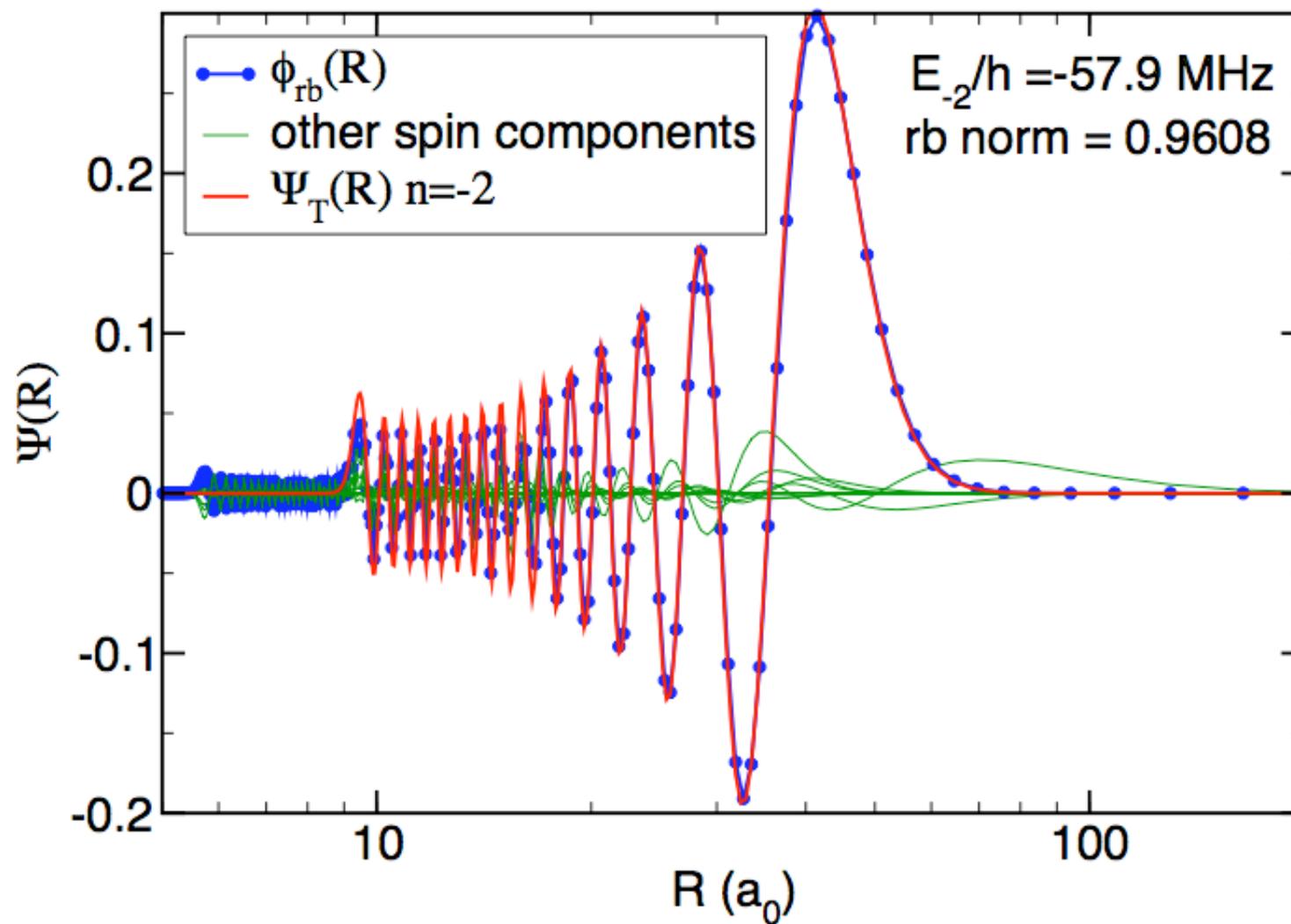


$^{40}\text{K}^{87}\text{Rb}$   $a(B)$  and  $E_n(B)$  vs.  $B$

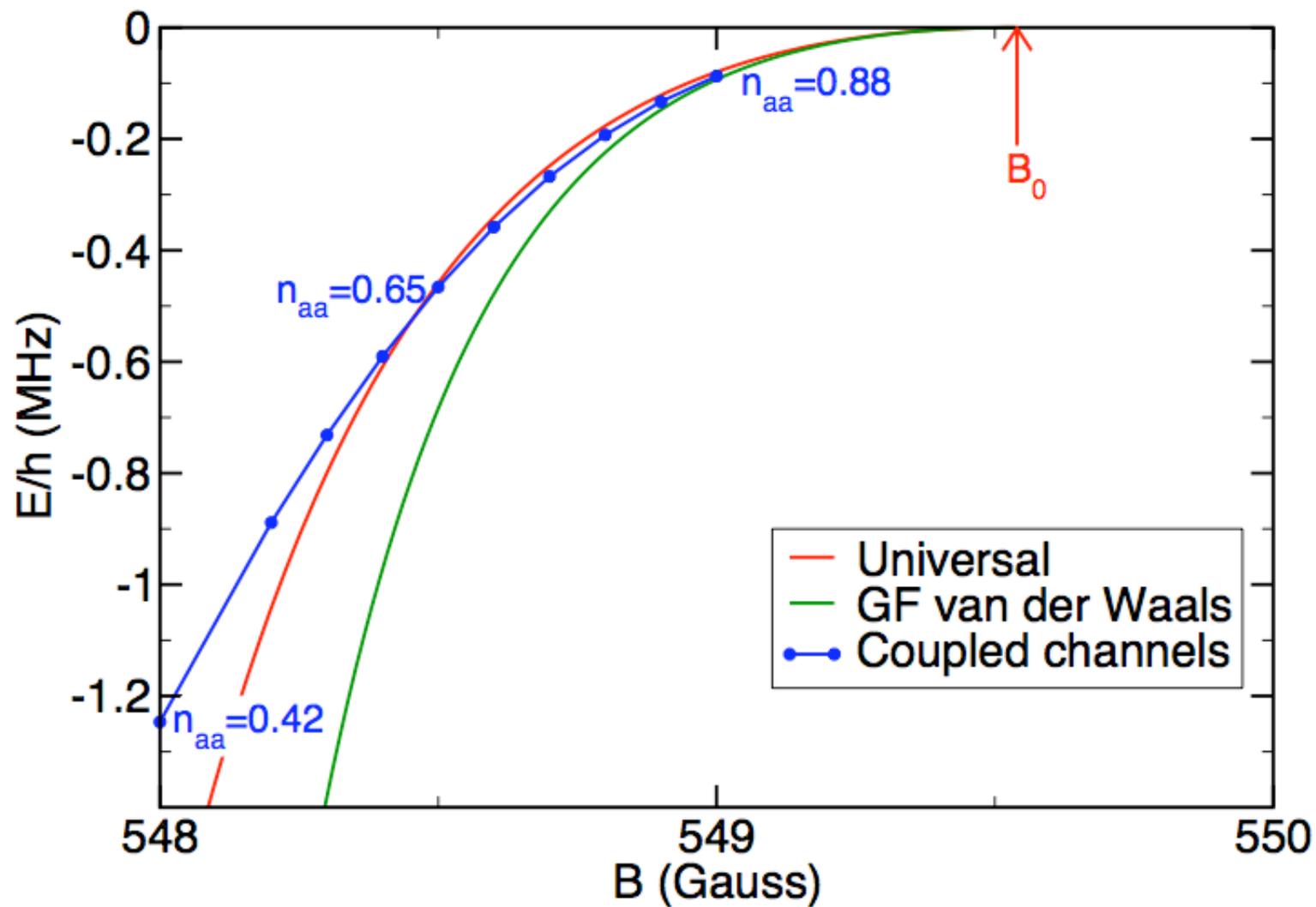


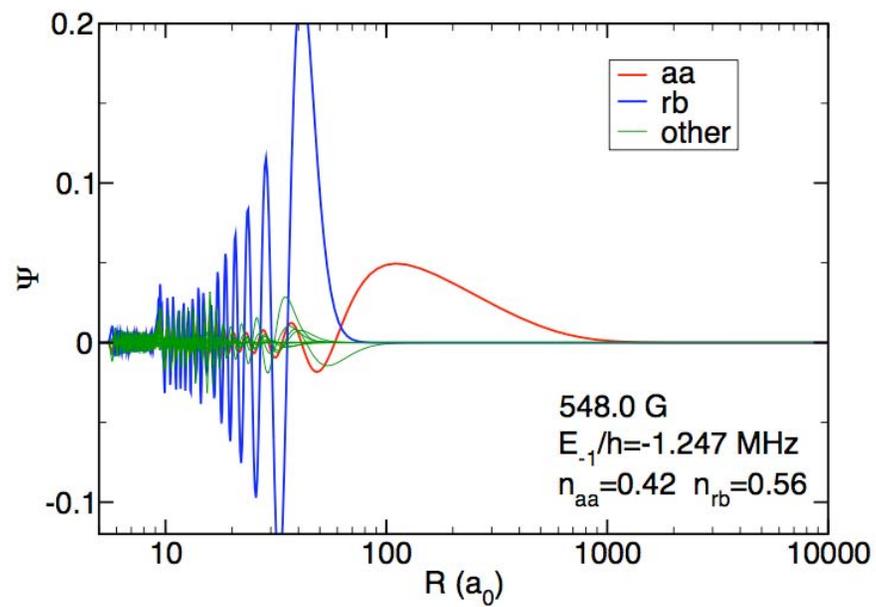
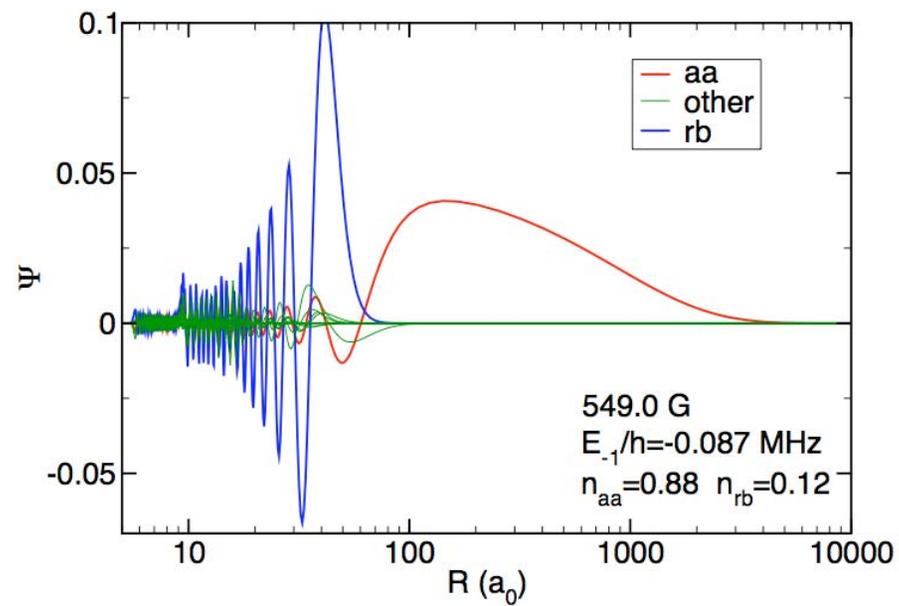
# $^{40}\text{K}^{87}\text{Rb}$ rb(-2) level 530 G

Coupled channels calculation with 12 spin channels



$^{40}\text{K}^{87}\text{Rb}$  {aa(s0), rb(-2s0)} M=-7/2 bound state





# PERIODIC TABLE Atomic Properties of the Elements

**NIST**

National Institute of Standards and Technology  
Technology Administration, U.S. Department of Commerce

Physics Laboratory  
physics.nist.gov  
Standard Reference Data Group  
www.nist.gov/srd

18  
VIII A

**Group II**

**Ytterbium**

1 IA	2 IIA											13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	18 VIII A
1 <sup>1</sup> S <sub>1/2</sub> <b>H</b> Hydrogen 1.00794 1s																	2 <sup>1</sup> S <sub>1/2</sub> <b>He</b> Helium 4.002602 1s <sup>2</sup> 24.5874
2 <sup>2</sup> S <sub>1/2</sub> <b>Li</b> Lithium 6.941 1s <sup>2</sup> 2s <sup>1</sup> 5.3917	<sup>2</sup> S <sub>1/2</sub> <b>Be</b> Beryllium 9.012182 1s <sup>2</sup> 2s <sup>2</sup> 9.3227											5 <sup>2</sup> P <sub>1/2</sub> <b>B</b> Boron 10.811 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>1</sup> 8.2980	6 <sup>2</sup> P <sub>3/2</sub> <b>C</b> Carbon 12.0107 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>2</sup> 11.2603	7 <sup>2</sup> S <sub>3/2</sub> <b>N</b> Nitrogen 14.0067 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 14.5341	8 <sup>2</sup> P <sub>3/2</sub> <b>O</b> Oxygen 15.9994 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>4</sup> 13.0181	9 <sup>2</sup> P <sub>3/2</sub> <b>F</b> Fluorine 18.9984032 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>5</sup> 17.4226	10 <sup>2</sup> S <sub>1/2</sub> <b>Ne</b> Neon 20.1797 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 21.5645
3 <sup>3</sup> S <sub>1/2</sub> <b>Na</b> Sodium 22.989770 [He]3s <sup>1</sup> 7.1391	<sup>3</sup> S <sub>1/2</sub> <b>Mg</b> Magnesium 24.3050 [He]3s <sup>2</sup> 7.6462											13 <sup>3</sup> P <sub>1/2</sub> <b>Al</b> Aluminum 26.981538 [Ne]3s <sup>2</sup> 3p <sup>1</sup> 5.6659	14 <sup>3</sup> P <sub>3/2</sub> <b>Si</b> Silicon 28.0855 [Ne]3s <sup>2</sup> 3p <sup>2</sup> 8.1517	15 <sup>3</sup> S <sub>3/2</sub> <b>P</b> Phosphorus 30.973761 [Ne]3s <sup>2</sup> 3p <sup>3</sup> 10.4857	16 <sup>3</sup> P <sub>3/2</sub> <b>S</b> Sulfur 32.065 [Ne]3s <sup>2</sup> 3p <sup>4</sup> 10.3600	17 <sup>3</sup> P <sub>3/2</sub> <b>Cl</b> Chlorine 35.453 [Ne]3s <sup>2</sup> 3p <sup>5</sup> 12.9676	18 <sup>3</sup> S <sub>1/2</sub> <b>Ar</b> Argon 39.948 [Ne]3s <sup>2</sup> 3p <sup>6</sup> 15.7586
4 <sup>4</sup> S <sub>1/2</sub> <b>K</b> Potassium 39.0983 [Ar]4s <sup>1</sup> 4.3407	<sup>4</sup> S <sub>1/2</sub> <b>Ca</b> Calcium 40.078 [Ar]4s <sup>2</sup> 6.1132	3 <sup>3</sup> D <sub>3/2</sub> <b>Sc</b> Scandium 44.955910 [Ar]3d <sup>1</sup> 4s <sup>2</sup> 6.5615	4 <sup>3</sup> F <sub>2</sub> <b>Ti</b> Titanium 47.867 [Ar]3d <sup>2</sup> 4s <sup>2</sup> 6.6281	5 <sup>3</sup> F <sub>3/2</sub> <b>V</b> Vanadium 50.9415 [Ar]3d <sup>3</sup> 4s <sup>2</sup> 6.7462	6 <sup>3</sup> F <sub>3/2</sub> <b>Cr</b> Chromium 51.9961 [Ar]3d <sup>5</sup> 4s <sup>1</sup> 6.7665	7 <sup>3</sup> F <sub>3/2</sub> <b>Mn</b> Manganese 54.938049 [Ar]3d <sup>5</sup> 4s <sup>2</sup> 7.4340	8 <sup>3</sup> D <sub>5/2</sub> <b>Fe</b> Iron 55.845 [Ar]3d <sup>6</sup> 4s <sup>2</sup> 7.9024	9 <sup>3</sup> F <sub>3/2</sub> <b>Co</b> Cobalt 58.933200 [Ar]3d <sup>7</sup> 4s <sup>2</sup> 7.6810	10 <sup>3</sup> F <sub>3/2</sub> <b>Ni</b> Nickel 58.6934 [Ar]3d <sup>8</sup> 4s <sup>2</sup> 7.5398	11 <sup>3</sup> S <sub>1/2</sub> <b>Cu</b> Copper 63.546 [Ar]3d <sup>10</sup> 4s <sup>1</sup> 7.7264	12 <sup>3</sup> S <sub>1/2</sub> <b>Zn</b> Zinc 65.408 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 7.942	13 <sup>3</sup> P <sub>1/2</sub> <b>Ga</b> Gallium 69.723 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>1</sup> 5.9993	14 <sup>3</sup> P <sub>3/2</sub> <b>Ge</b> Germanium 72.64 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>2</sup> 7.6994	15 <sup>3</sup> S <sub>3/2</sub> <b>As</b> Arsenic 74.92160 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>3</sup> 9.7666	16 <sup>3</sup> P <sub>3/2</sub> <b>Se</b> Selenium 78.96 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 9.7524	17 <sup>3</sup> P <sub>3/2</sub> <b>Br</b> Bromine 79.904 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup> 11.8138	18 <sup>3</sup> S <sub>1/2</sub> <b>Kr</b> Krypton 83.796 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>6</sup> 13.9956
5 <sup>5</sup> S <sub>1/2</sub> <b>Rb</b> Rubidium 85.4678 [Kr]5s <sup>1</sup> 4.1771	<sup>5</sup> S <sub>1/2</sub> <b>Sr</b> Strontium 87.62 [Kr]5s <sup>2</sup> 5.6949	3 <sup>3</sup> D <sub>3/2</sub> <b>Y</b> Yttrium 88.90505 [Kr]4d <sup>1</sup> 5s <sup>2</sup> 6.2173	4 <sup>3</sup> F <sub>2</sub> <b>Zr</b> Zirconium 91.224 [Kr]4d <sup>2</sup> 5s <sup>2</sup> 6.6330	5 <sup>3</sup> F <sub>3/2</sub> <b>Nb</b> Niobium 92.90638 [Kr]4d <sup>4</sup> 5s <sup>1</sup> 6.7568	6 <sup>3</sup> F <sub>3/2</sub> <b>Mo</b> Molybdenum 95.94 [Kr]4d <sup>5</sup> 5s <sup>1</sup> 7.0924	7 <sup>3</sup> F <sub>3/2</sub> <b>Tc</b> Technetium (98) [Kr]4d <sup>5</sup> 5s <sup>2</sup> 7.26	8 <sup>3</sup> D <sub>5/2</sub> <b>Ru</b> Ruthenium 101.07 [Kr]4d <sup>7</sup> 5s <sup>1</sup> 7.3605	9 <sup>3</sup> F <sub>3/2</sub> <b>Rh</b> Rhodium 102.90550 [Kr]4d <sup>8</sup> 5s <sup>1</sup> 7.4593	10 <sup>3</sup> D <sub>5/2</sub> <b>Pd</b> Palladium 106.42 [Kr]4d <sup>10</sup> 8.3369	11 <sup>3</sup> S <sub>1/2</sub> <b>Ag</b> Silver 107.8682 [Kr]4d <sup>10</sup> 5s <sup>1</sup> 7.5782	12 <sup>3</sup> S <sub>1/2</sub> <b>Cd</b> Cadmium 112.411 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 8.9938	13 <sup>3</sup> P <sub>1/2</sub> <b>In</b> Indium 114.818 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>1</sup> 5.7864	14 <sup>3</sup> P <sub>3/2</sub> <b>Sn</b> Tin 118.710 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>2</sup> 7.3436	15 <sup>3</sup> S <sub>3/2</sub> <b>Sb</b> Antimony 121.750 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>3</sup> 8.6064	16 <sup>3</sup> P <sub>3/2</sub> <b>Te</b> Tellurium 127.60 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>4</sup> 9.6095	17 <sup>3</sup> P <sub>3/2</sub> <b>I</b> Iodine 126.90447 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>5</sup> 10.4513	18 <sup>3</sup> S <sub>1/2</sub> <b>Xe</b> Xenon 131.293 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>6</sup> 12.1268
6 <sup>6</sup> S <sub>1/2</sub> <b>Cs</b> Cesium 132.90545 [Xe]6s <sup>1</sup> 5.8039	<sup>6</sup> S <sub>1/2</sub> <b>Ba</b> Barium 137.327 [Xe]6s <sup>2</sup> 5.2117		4 <sup>4</sup> F <sub>2</sub> <b>Hf</b> Hafnium 178.49 [Xe]4f <sup>14</sup> 5d <sup>2</sup> 6s <sup>2</sup> 6.8251	5 <sup>4</sup> F <sub>3/2</sub> <b>Ta</b> Tantalum 180.9479 [Xe]4f <sup>14</sup> 5d <sup>3</sup> 6s <sup>2</sup> 7.5466	6 <sup>4</sup> F <sub>3/2</sub> <b>W</b> Tungsten 183.84 [Xe]4f <sup>14</sup> 5d <sup>4</sup> 6s <sup>2</sup> 7.8640	7 <sup>4</sup> F <sub>3/2</sub> <b>Re</b> Rhenium 186.207 [Xe]4f <sup>14</sup> 5d <sup>5</sup> 6s <sup>2</sup> 7.8355	8 <sup>4</sup> D <sub>5/2</sub> <b>Os</b> Osmium 190.23 [Xe]4f <sup>14</sup> 5d <sup>6</sup> 6s <sup>2</sup> 8.4382	9 <sup>4</sup> F <sub>3/2</sub> <b>Ir</b> Iridium 192.217 [Xe]4f <sup>14</sup> 5d <sup>7</sup> 6s <sup>2</sup> 8.6670	10 <sup>4</sup> D <sub>5/2</sub> <b>Pt</b> Platinum 195.078 [Xe]4f <sup>14</sup> 5d <sup>9</sup> 6s <sup>1</sup> 8.9588	11 <sup>4</sup> S <sub>1/2</sub> <b>Au</b> Gold 196.96657 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>1</sup> 8.2255	12 <sup>4</sup> S <sub>1/2</sub> <b>Hg</b> Mercury 200.59 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 10.4375	13 <sup>4</sup> P <sub>1/2</sub> <b>Tl</b> Thallium 204.3833 [Hg]6p <sup>1</sup> 6.1082	14 <sup>4</sup> P <sub>3/2</sub> <b>Pb</b> Lead 207.2 [Hg]6p <sup>2</sup> 7.4167	15 <sup>4</sup> S <sub>3/2</sub> <b>Bi</b> Bismuth 208.98038 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>3</sup> 8.555	16 <sup>4</sup> P <sub>3/2</sub> <b>Po</b> Polonium (209) [Hg]6p <sup>4</sup> 8.414	17 <sup>4</sup> S <sub>3/2</sub> <b>At</b> Astatine (210) [Hg]6p <sup>5</sup>	18 <sup>4</sup> S <sub>1/2</sub> <b>Rn</b> Radon (222) [Hg]6p <sup>6</sup> 10.7485
7 <sup>7</sup> S <sub>1/2</sub> <b>Fr</b> Francium (223) [Rn]7s <sup>1</sup> 4.0727	<sup>7</sup> S <sub>1/2</sub> <b>Ra</b> Radium (226) [Rn]7s <sup>2</sup> 5.4936		4 <sup>4</sup> F <sub>2</sub> <b>Rf</b> Rutherfordium (261) [Rn]5f <sup>14</sup> 6d <sup>2</sup> 7s <sup>2</sup> 6.0	5 <sup>4</sup> F <sub>3/2</sub> <b>Db</b> Dubnium (262)	6 <sup>4</sup> F <sub>3/2</sub> <b>Sg</b> Seaborgium (266)	7 <sup>4</sup> F <sub>3/2</sub> <b>Bh</b> Bohrium (264)	8 <sup>4</sup> D <sub>5/2</sub> <b>Hs</b> Hassium (277)	9 <sup>4</sup> F <sub>3/2</sub> <b>Mt</b> Meitnerium (268)	10 <sup>4</sup> G <sub>1/2</sub> <b>Uun</b> Ununium (281)	11 <sup>4</sup> F <sub>3/2</sub> <b>Uuu</b> Unununium (272)	12 <sup>4</sup> G <sub>3/2</sub> <b>Uub</b> Ununbium (285)	13 <sup>4</sup> G <sub>1/2</sub> <b>Uuq</b> Ununquadium (289)	14 <sup>4</sup> G <sub>3/2</sub> <b>Uuh</b> Ununhexium (293)	15 <sup>4</sup> G <sub>3/2</sub> <b>Uuq</b> Ununseptium (297)	16 <sup>4</sup> G <sub>3/2</sub> <b>Uuh</b> Ununnonium (301)	17 <sup>4</sup> G <sub>3/2</sub> <b>Uuq</b> Ununquadrium (305)	18 <sup>4</sup> G <sub>1/2</sub> <b>Uuq</b> Ununseptium (309)
		Lanthanides	57 <sup>5</sup> D <sub>3/2</sub> <b>La</b> Lanthanum 138.9055 [Xe]5d <sup>1</sup> 6s <sup>2</sup> 5.5789	58 <sup>5</sup> G <sub>5/2</sub> <b>Ce</b> Cerium 140.90765 [Xe]4f <sup>1</sup> 5d <sup>1</sup> 6s <sup>2</sup> 5.9473	59 <sup>5</sup> F <sub>7/2</sub> <b>Pr</b> Praseodymium 140.90765 [Xe]4f <sup>2</sup> 6s <sup>2</sup> 5.473	60 <sup>5</sup> F <sub>5/2</sub> <b>Nd</b> Neodymium 144.24 [Xe]4f <sup>3</sup> 6s <sup>2</sup> 5.5250	61 <sup>5</sup> G <sub>7/2</sub> <b>Pm</b> Promethium (145) [Xe]4f <sup>4</sup> 6s <sup>2</sup> 5.582	62 <sup>5</sup> F <sub>5/2</sub> <b>Sm</b> Samarium 150.36 [Xe]4f <sup>6</sup> 6s <sup>2</sup> 5.8437	63 <sup>5</sup> G <sub>7/2</sub> <b>Eu</b> Europium 151.964 [Xe]4f <sup>7</sup> 6s <sup>2</sup> 5.9704	64 <sup>5</sup> D <sub>3/2</sub> <b>Gd</b> Gadolinium 157.25 [Xe]4f <sup>7</sup> 5d <sup>1</sup> 6s <sup>2</sup> 6.1498	65 <sup>5</sup> G <sub>7/2</sub> <b>Tb</b> Terbium 158.92534 [Xe]4f <sup>9</sup> 6s <sup>2</sup> 5.8988	66 <sup>5</sup> F <sub>7/2</sub> <b>Dy</b> Dysprosium 162.500 [Xe]4f <sup>10</sup> 6s <sup>2</sup> 5.9389	67 <sup>5</sup> F <sub>7/2</sub> <b>Ho</b> Holmium 164.93032 [Xe]4f <sup>11</sup> 6s <sup>2</sup> 6.0215	68 <sup>5</sup> G <sub>7/2</sub> <b>Er</b> Erbium 167.259 [Xe]4f <sup>12</sup> 6s <sup>2</sup> 6.1077	69 <sup>5</sup> F <sub>7/2</sub> <b>Tm</b> Thulium 168.9342 [Xe]4f <sup>13</sup> 6s <sup>2</sup> 6.1845	70 <sup>5</sup> G <sub>7/2</sub> <b>Yb</b> Ytterbium 173.04 [Xe]4f <sup>14</sup> 6s <sup>2</sup> 6.2542	71 <sup>5</sup> D <sub>3/2</sub> <b>Lu</b> Lutetium 174.967 [Xe]4f <sup>14</sup> 5d <sup>1</sup> 6s <sup>2</sup> 5.4259
		Actinides	89 <sup>6</sup> D <sub>3/2</sub> <b>Ac</b> Actinium (227) [Rn]5f <sup>7</sup> 6d <sup>1</sup> 7s <sup>2</sup> 5.17	90 <sup>6</sup> F <sub>5/2</sub> <b>Th</b> Thorium 232.0381 [Rn]5f <sup>0</sup> 6d <sup>2</sup> 7s <sup>2</sup> 6.3067	91 <sup>6</sup> K <sub>11/2</sub> <b>Pa</b> Protactinium 231.03688 [Rn]5f <sup>1</sup> 6d <sup>1</sup> 7s <sup>2</sup> 5.89	92 <sup>6</sup> F <sub>5/2</sub> <b>U</b> Uranium 238.02891 [Rn]5f <sup>3</sup> 6d <sup>1</sup> 7s <sup>2</sup> 6.1941	93 <sup>6</sup> G <sub>7/2</sub> <b>Np</b> Neptunium (237) [Rn]5f <sup>4</sup> 6d <sup>1</sup> 7s <sup>2</sup> 6.2657	94 <sup>6</sup> F <sub>5/2</sub> <b>Pu</b> Plutonium (244) [Rn]5f <sup>6</sup> 7s <sup>2</sup> 6.0280	95 <sup>6</sup> G <sub>7/2</sub> <b>Am</b> Americium (247) [Rn]5f <sup>7</sup> 7s <sup>2</sup> 5.9738	96 <sup>6</sup> D <sub>3/2</sub> <b>Cm</b> Curium (247) [Rn]5f <sup>8</sup> 6d <sup>1</sup> 7s <sup>2</sup> 5.9914	97 <sup>6</sup> G <sub>7/2</sub> <b>Bk</b> Berkelium (247) [Rn]5f <sup>9</sup> 7s <sup>2</sup> 6.1979	98 <sup>6</sup> F <sub>5/2</sub> <b>Cf</b> Californium (251) [Rn]5f <sup>10</sup> 7s <sup>2</sup> 6.2617	99 <sup>6</sup> G <sub>7/2</sub> <b>Es</b> Einsteinium (252) [Rn]5f <sup>11</sup> 7s <sup>2</sup> 6.42	100 <sup>6</sup> F <sub>5/2</sub> <b>Fm</b> Fermium (257) [Rn]5f <sup>12</sup> 7s <sup>2</sup> 6.50	101 <sup>6</sup> G <sub>7/2</sub> <b>Md</b> Mendelevium (258) [Rn]5f <sup>13</sup> 7s <sup>2</sup> 6.58	102 <sup>6</sup> F <sub>5/2</sub> <b>No</b> Nobelium (259) [Rn]5f <sup>14</sup> 7s <sup>2</sup> 6.65	103 <sup>6</sup> G <sub>7/2</sub> <b>Lr</b> Lawrencium (262) [Rn]5f <sup>14</sup> 6d <sup>1</sup> 7s <sup>2</sup> 4.97

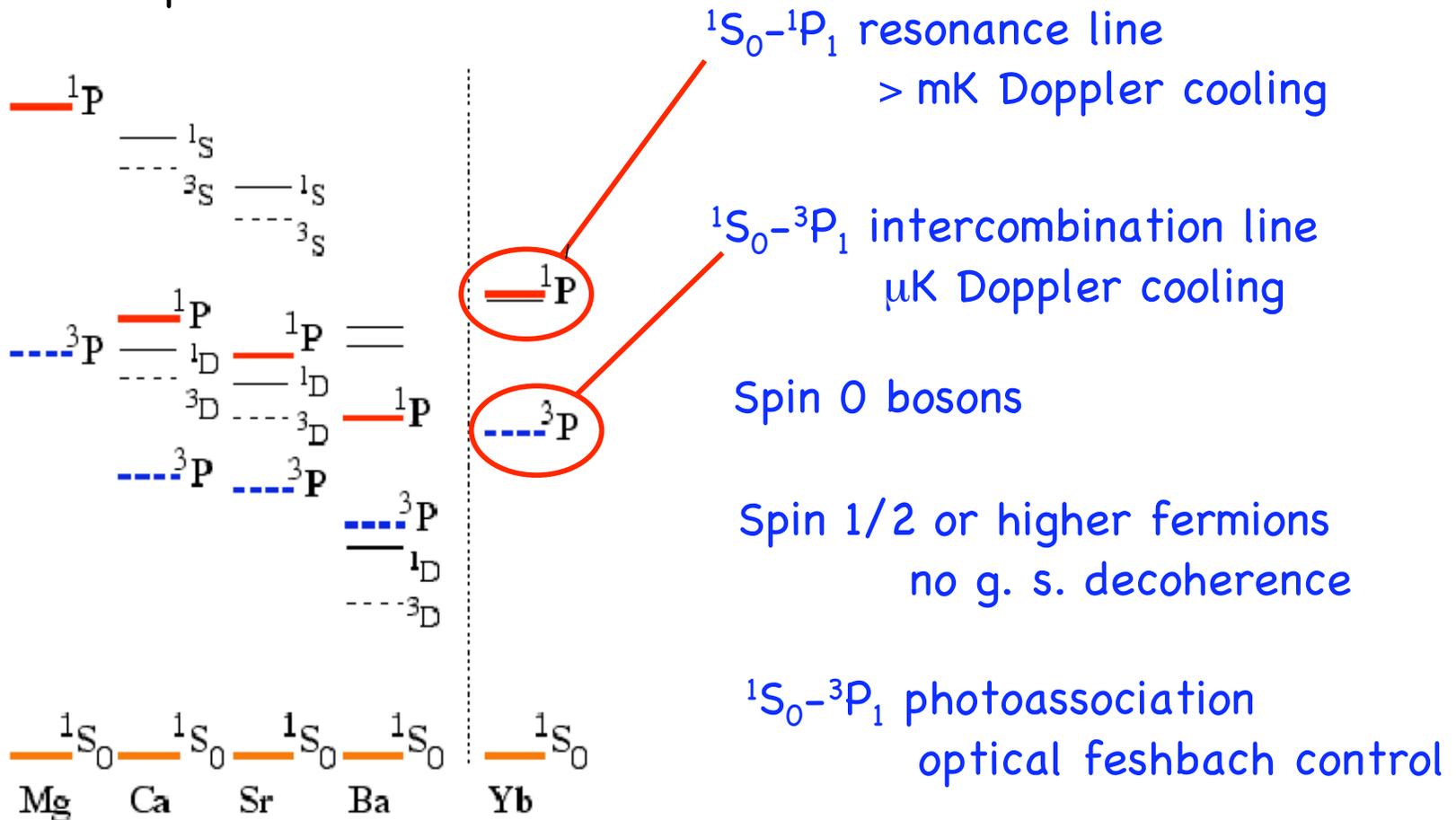
Atomic Number: 58  
Ground-state Level:  $4f^1$   
Symbol: **Ce**  
Name: Cerium  
Atomic Weight: 140.116  
Ground-state Configuration:  $[Xe]4f^1 6s^2$   
Ionization Energy (eV): 5.5387

<sup>1</sup>Based upon <sup>12</sup>C. ( ) indicates the mass number of the most stable isotope.

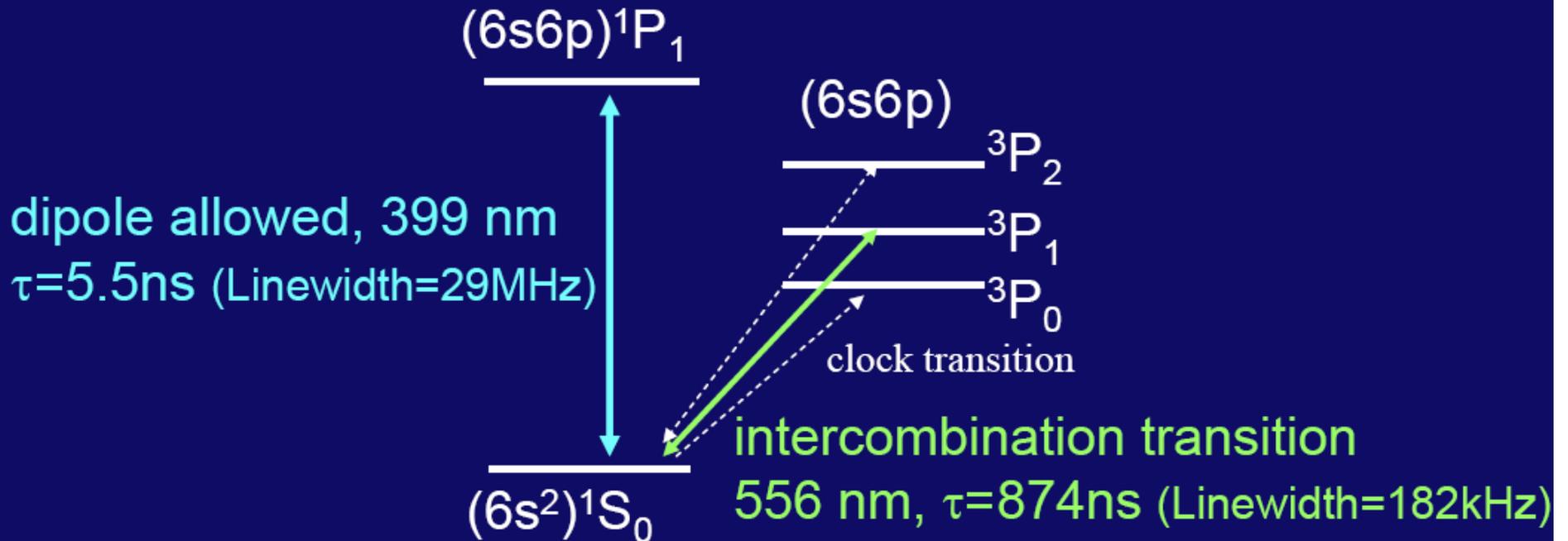
For a description of the data, visit [physics.nist.gov/data](http://physics.nist.gov/data)

NIST SP 966 (September 2003)

## Group II Atoms



# Level Diagram of Yb



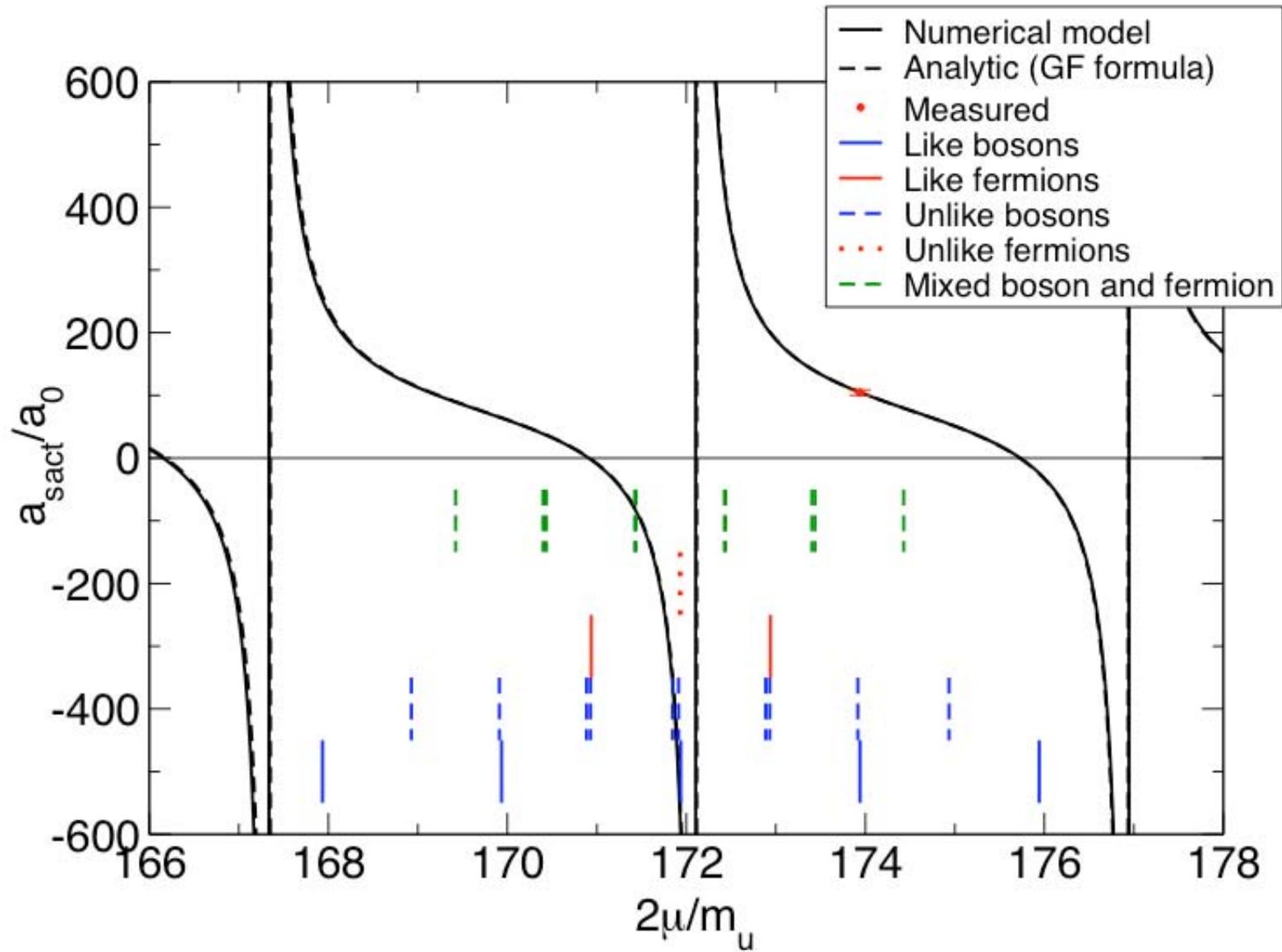
From Enomoto et al

Mass number	168	170	171	172	173	174	176
Nuclear spin $i$	0	0	1/2	0	5/2	0	0
Abundance(%)	0.13	3.05	14.3	21.9	16.2	31.8	12.7

## Scattering lengths for Yb ground state model (in $a_0$ units)

	168	170	171	172	173	174	176
168	252(6)	117(1)	89(1)	65(1)	39(1)	2(2)	-360(30)
170	117	64(1)	37(1)	-2(2)	-81(4)	-520(50)	209(4)
171	89	37	-3(2)	-84(5)	-580(60)	430(20)	142(2)
172	65	-2	-84	-600(60)	420(20)	201(3)	106(1)
173	39	-81	-580	420	199(3)	139(2)	80(1)
174	2	-520	430	201	139	105(1)	55(1)
176	-360	209	142	106	80	55	-24(2)

# Variation of scattering length with mass



Model: LJ 6-12 +  $C_8$  van der Waals  
 1 potential + reduced mass

$C_6=1932(4)$  au     $C_8=1.9(5)\times 10^5$  Exactly 72 bound states in  $^{174}\text{Yb}_2$

Isotope	vib, rot	Eb [MHz]	Error [MHz]	Eb [MHz]	Dev [MHz]
N	176 v=1,J=0	-70.404	0.011	-70.405	-0.001
	v=1,J=2	-37.142	0.013	-37.118	0.024
$C_6$	174 v=1,J=0	-10.612	0.038	-10.642	-0.030
	v=2,J=0	-325.607	0.018	-325.607	0.000
$C_8$	v=2,J=2	-268.575	0.021	-268.576	-0.001
	173 v=1,J=0	-1.539	0.074	-1.613	-0.074
	172 v=1,J=0	-123.269	0.026	-123.349	-0.080
	v=1,J=2	-81.786	0.019	-81.879	-0.093
	171 v=1,J=0	-64.418	0.040	-64.548	-0.130
	v=1,J=2	-31.302	0.050	-31.392	-0.090
	170 v=1,J=0	-27.661	0.023	-27.755	-0.094
	v=1,J=2	-3.651	0.026	-3.683	-0.032

# Gribakin and Flambaum

Phys. Rev. A 48, 546 (1993)

$$a = \bar{a} \left( 1 - \tan \left( \Phi - \frac{\pi}{8} \right) \right)$$

$$\bar{a} = \frac{1}{2^{3/2}} \frac{\Gamma(3/4)}{\Gamma(5/4)} \left( \frac{2\mu C_6}{\hbar^2} \right)^{1/4}$$

$$\Phi = \int_{r_{in}}^{\infty} \left( \frac{2\mu}{\hbar^2} (-V(R)) \right)^{1/2} dR$$

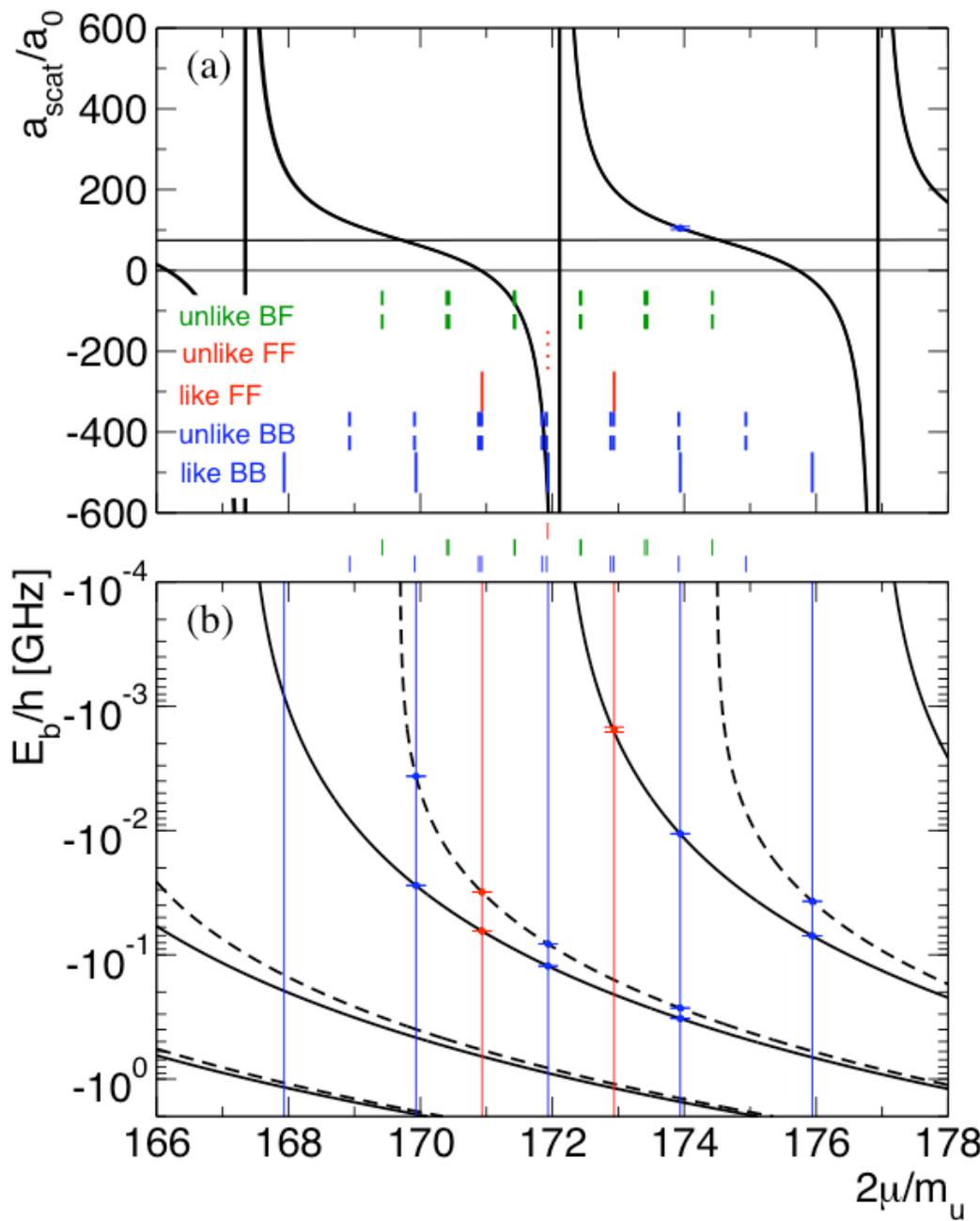
$$\text{Number of bound states in } V(R) = \text{Int} \left[ \frac{\Phi}{\pi} - \frac{5}{8} \right] + 1$$

# Pure van der Waals theory

(GF and B. Gao)

$$V(R) = \infty \text{ for } 0 < R \leq R_{in}$$

$$V(R) = -\frac{C_6}{R^6} \text{ for } R_{in} < R \leq \infty$$



Scattering  
length

Bound  
states

Species	Spin	% Abundance
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$^{84}\text{Sr}$	0	0.56
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$^{86}\text{Sr}$	0	9.86
------------------	---	------

$^{87}\text{Sr}$	9/2	7.00
------------------	-----	------

$^{88}\text{Sr}$	0	82.58
------------------	---	-------

$^{196}\text{Hg}$	0	0.15
-------------------	---	------

$^{198}\text{Hg}$	0	9.98
-------------------	---	------

$^{199}\text{Hg}$	1/2	16.87
-------------------	-----	-------

$^{200}\text{Hg}$	0	23.10
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$^{201}\text{Hg}$	3/2	13.19
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$^{202}\text{Hg}$	0	29.86
-------------------	---	-------

$^{204}\text{Hg}$	0	6.87
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$^{130}\text{Ba}$	0	0.11
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$^{132}\text{Ba}$	0	0.10
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$^{134}\text{Ba}$	0	2.42
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$^{135}\text{Ba}$	3/2	6.59
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$^{136}\text{Ba}$	0	7.85
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$^{137}\text{Ba}$	3/2	11.23
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$^{138}\text{Ba}$	0	71.71
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