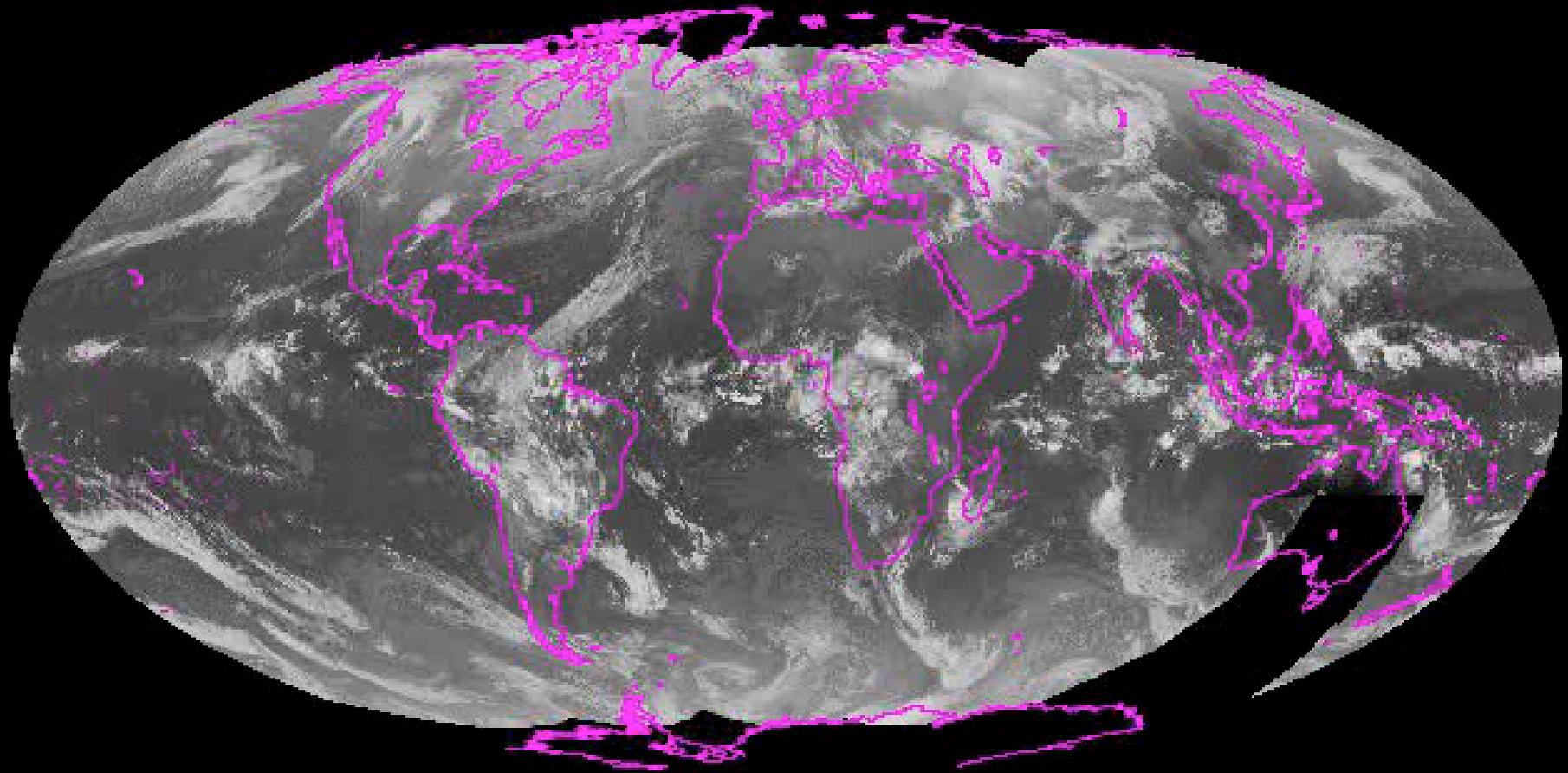


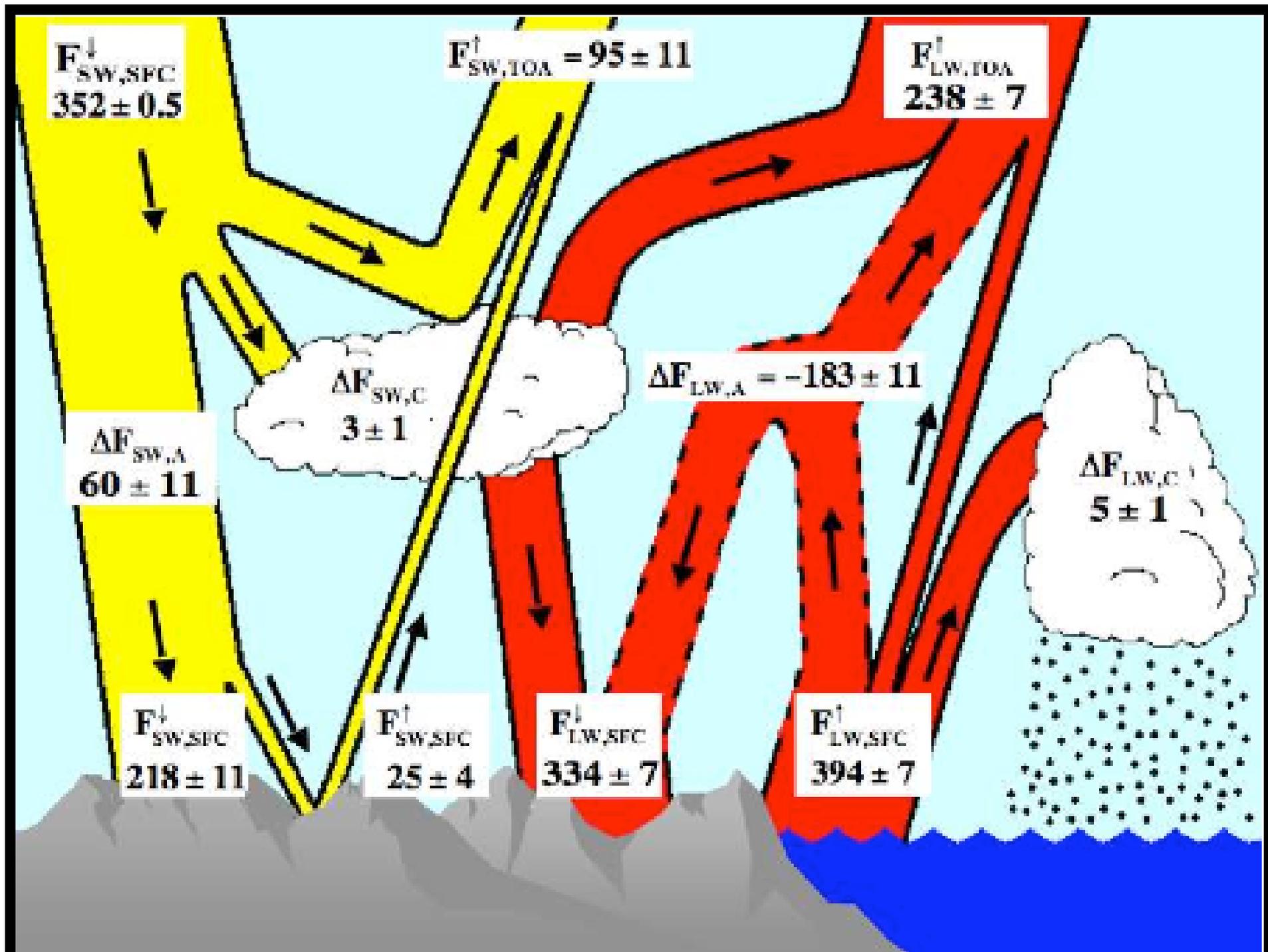
Entropy Production in the Cloud-Climate System

Tim Garrett
University of Utah

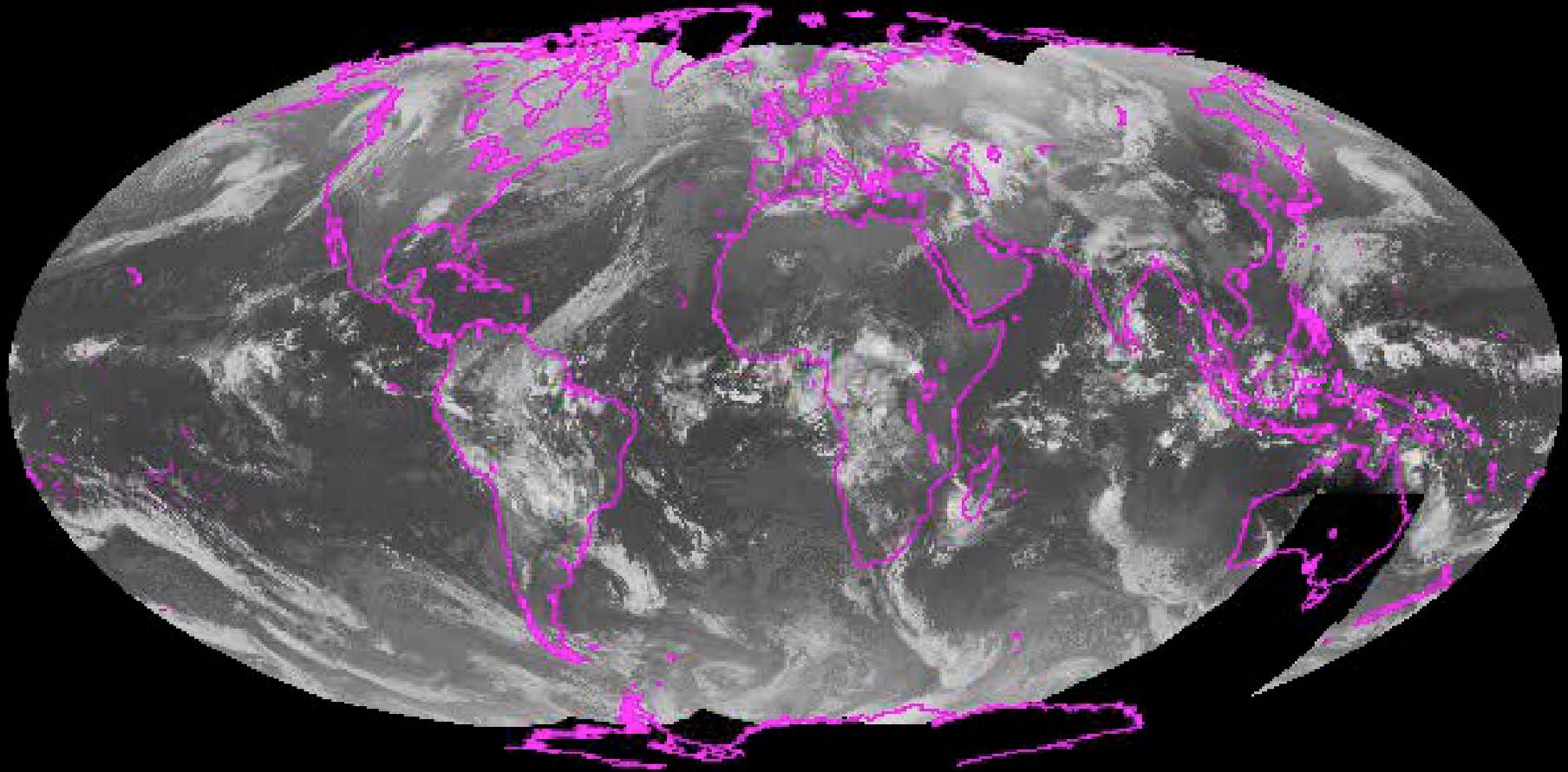
INFRARED COMPOSITE FROM 11 MAR 08 AT 03:00 UTC (SSEC:UW-MADISON)



1 INFRARED COMPOSITE FROM 11 MAR 08 AT 03:00 UTC (SSEC:UW-MADISON) KIDAS



INFRARED COMPOSITE FROM 11 MAR 08 AT 03:00 UTC (SSEC:UW-MADISON)

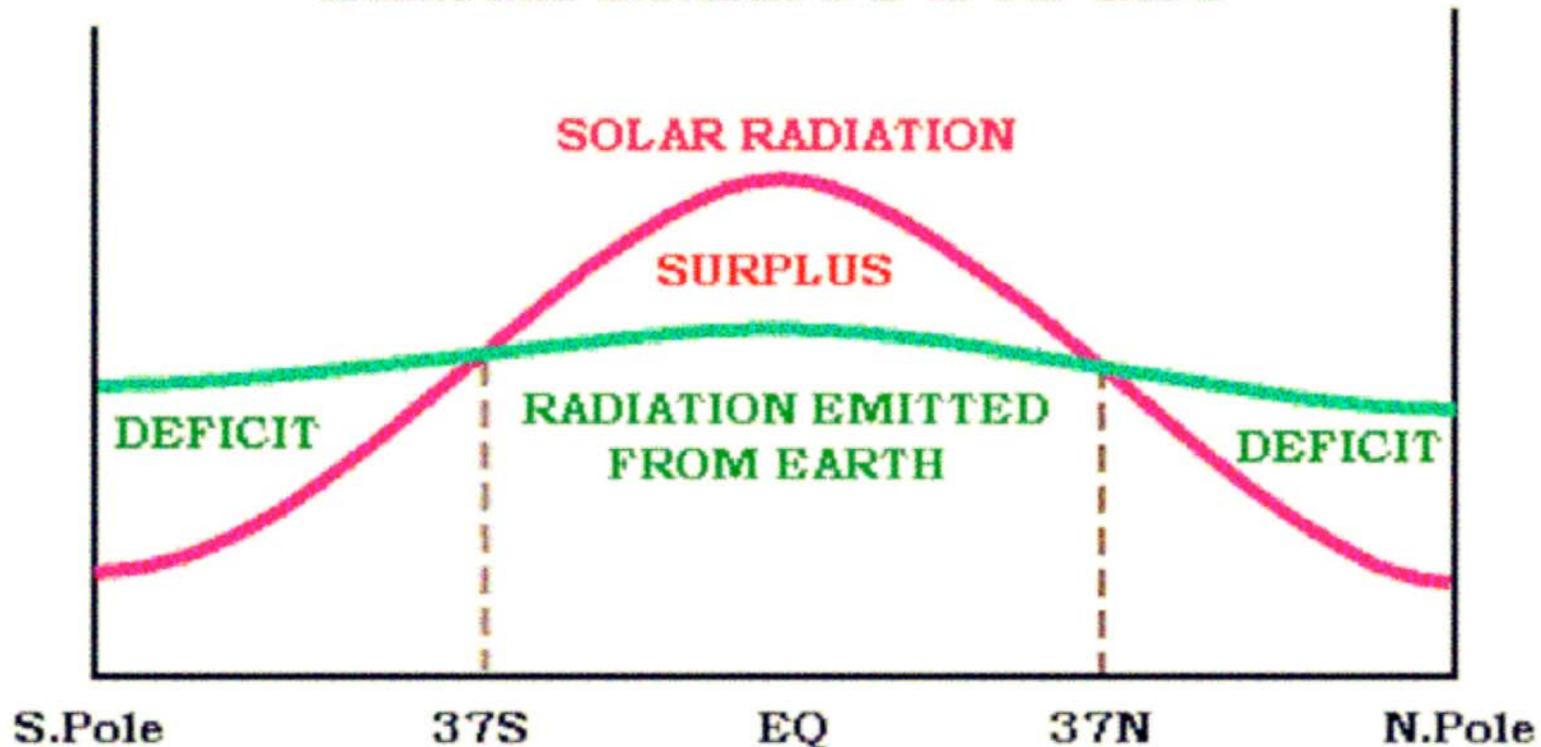


1 INFRARED COMPOSITE FROM 11 MAR 08 AT 03:00 UTC (SSEC:UW-MADISON) MIDAS

The more things change, the more they stay the same

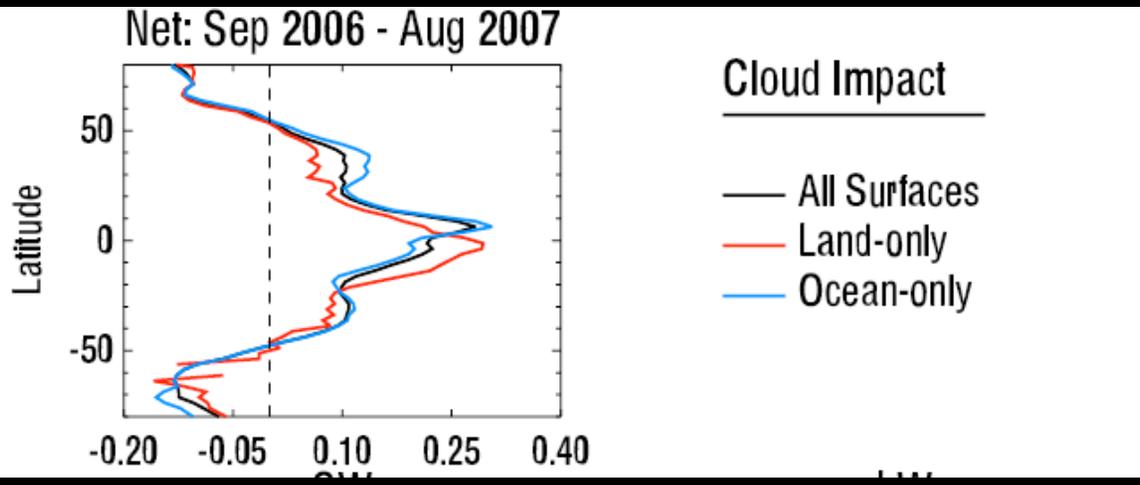
How do clouds evolve given
larger global constraints?

EARTH ENERGY BUDGET



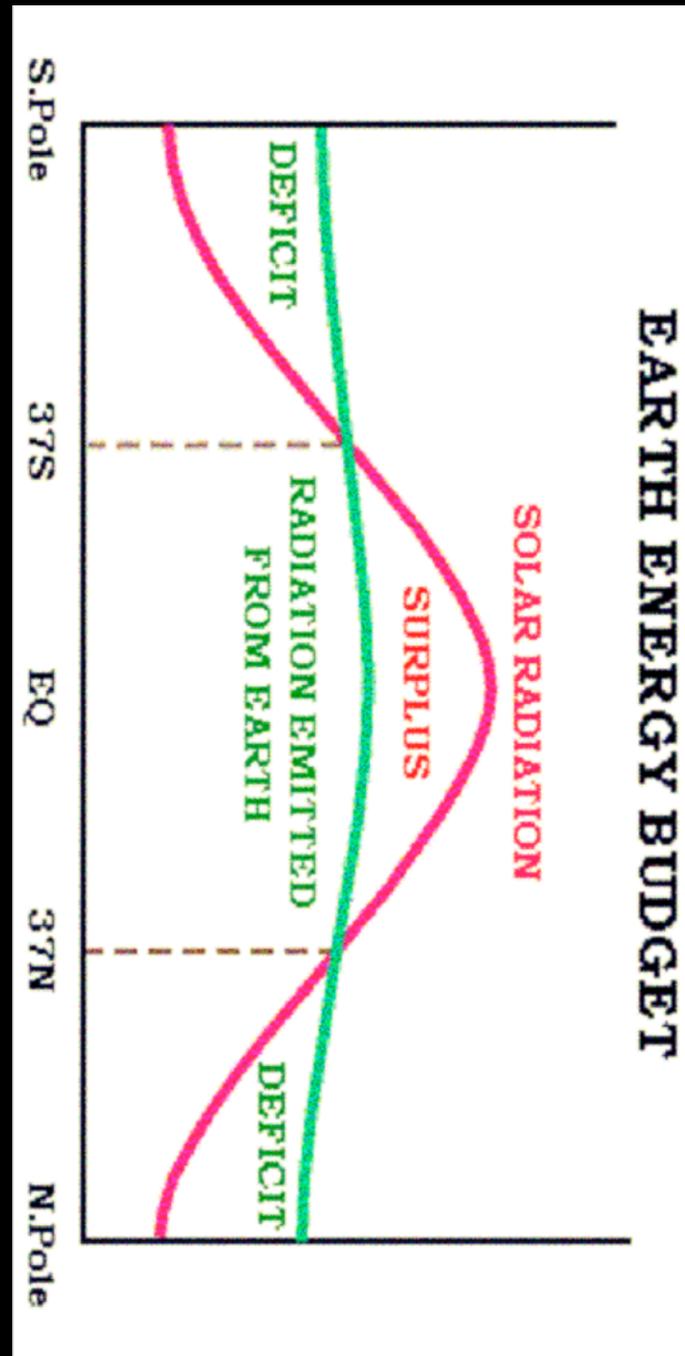
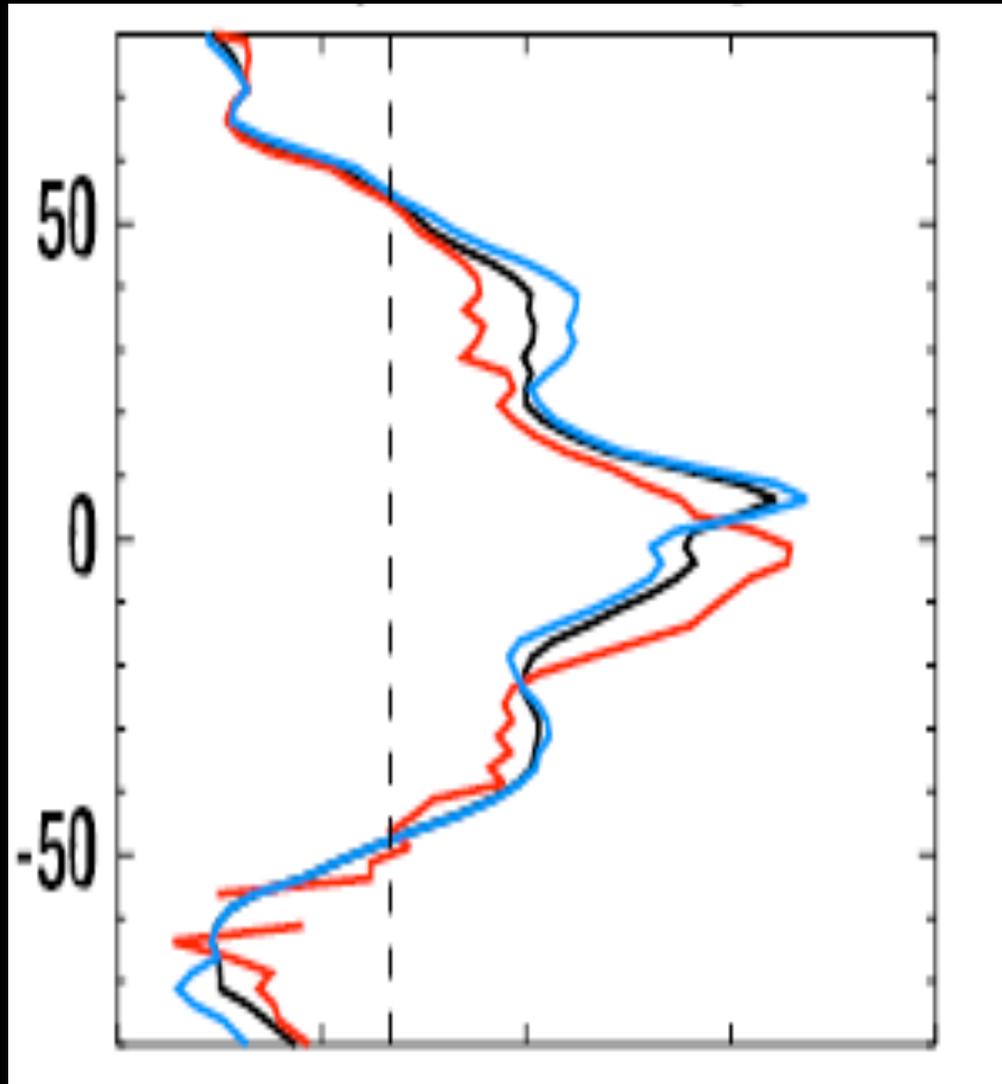
CloudSat latitudinal heating

Clouds consume energy



*Courtesy Tristan L'Ecuyer,
Colo. State*

Cloud heating or energy consumption



Thermodynamics is a funny subject. The first time you go through it, you don't understand it at all. The second time you go through it, you think you understand it, except for one or two small points. The third time you go through it, you know you don't understand it, but by that time you are so used to it, it doesn't bother you any more.

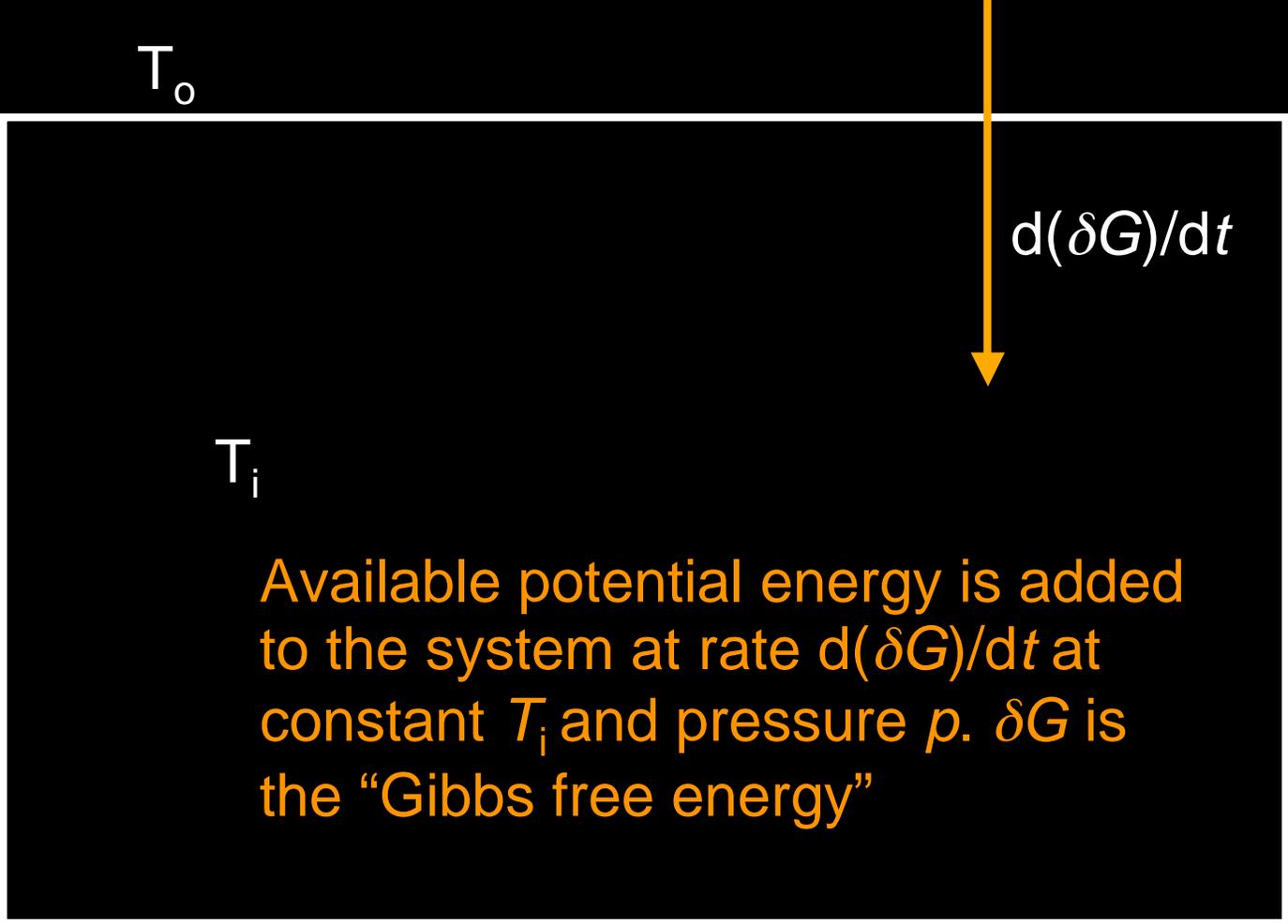
- Arnold Sommerfield (of the fine structure const.)

Outside at temperature T_o

System with internal temperature T_i

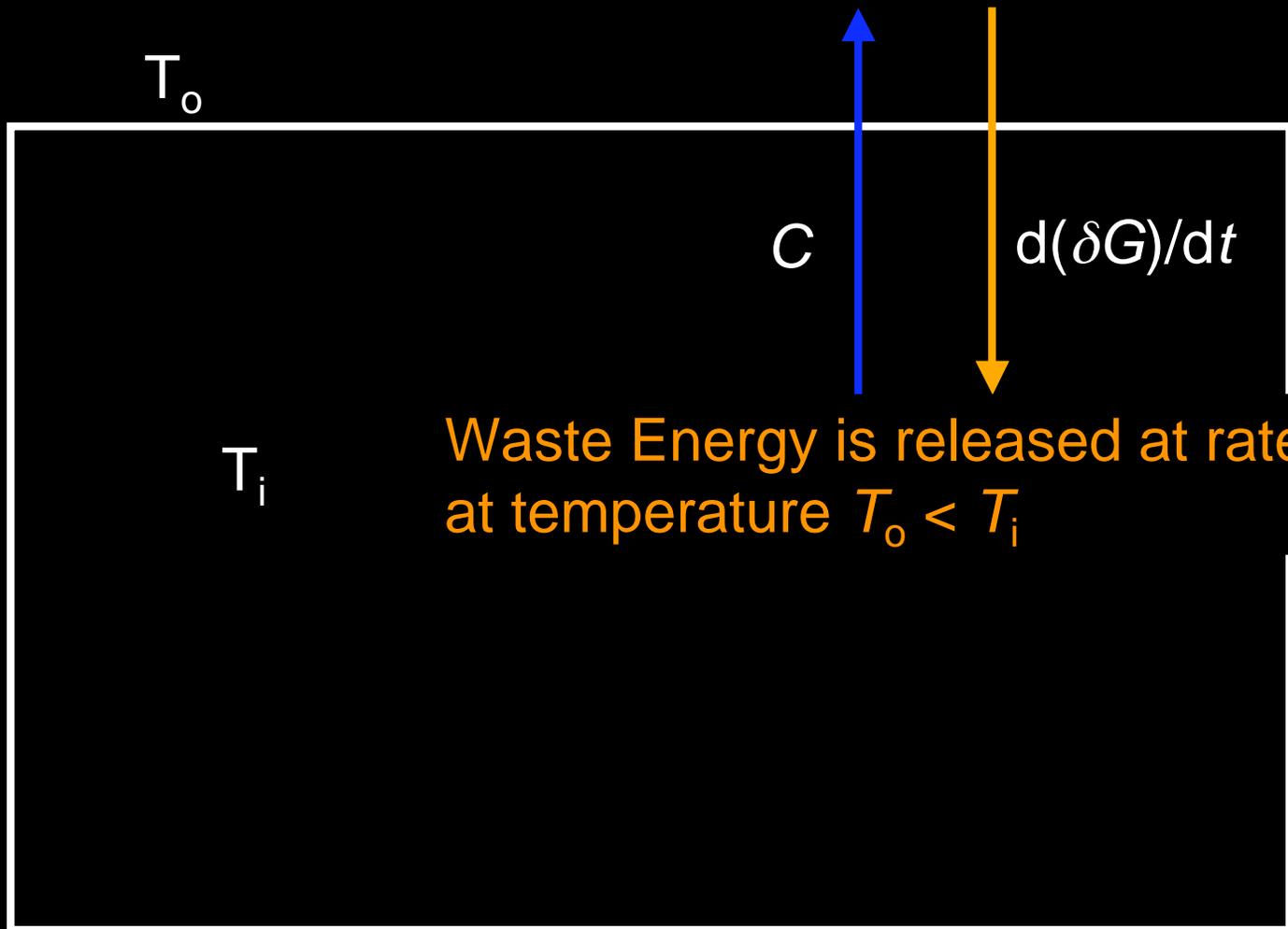
T_o

$d(\delta G)/dt$



T_i

Available potential energy is added to the system at rate $d(\delta G)/dt$ at constant T_i and pressure p . δG is the “Gibbs free energy”



Waste Energy is released at rate C
at temperature $T_o < T_i$

T_o

C

$d(\delta G)/dt$

T_i

Waste Energy is released at rate C
at temperature T_o

Energy is conserved
 $d(\delta G)/dt - C = 0$

T_o

C

$d(\delta G)/dt$

T_i

Waste Energy is released at rate C
at temperature T_o

Overall entropy production is positive

$$dS/dt = C/T_o > 0$$

Suppose T_0 is the temperature of the coldest reservoir at hand through conduction or radiation then

"Whenever an irreversible process takes place, the effect on the universe is the same as that which would be produced if a certain quantity of energy were converted from a form in which it is completely available for work into a form in which it is completely unavailable for work. The amount of energy δG is T_0 times the entropy change of the universe brought about by the irreversible process... δG is the energy that is rendered unavailable for work"

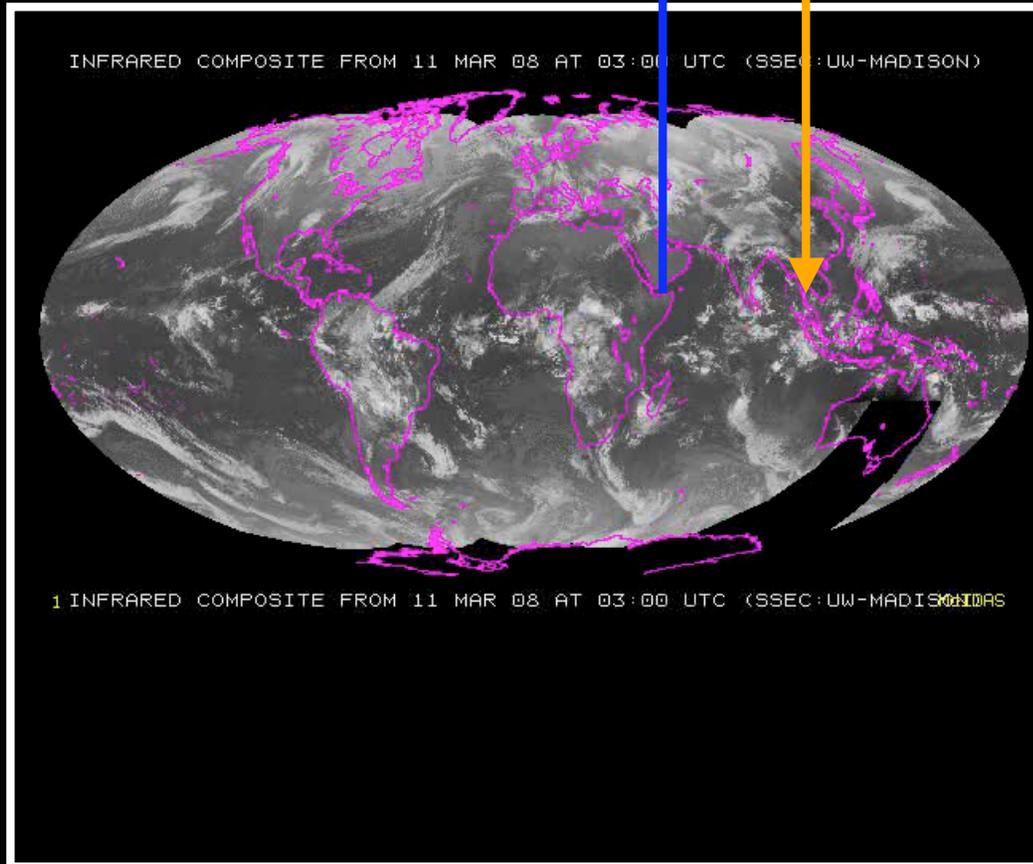
$$C = d(\delta G)/dt = T_0 dS/dt$$

Energy consumption = Entropy production (times T_0)

255 K

$C = 240 \text{ W m}^{-2}$

$d(\delta G)/dt = 240 \text{ W m}^{-2}$



$$\left. \frac{dS}{dt} \right|_{\theta} = \frac{C}{T_o} = \left. \frac{d(\delta G)/dt}{T_o} \right|_{\theta} \geq 0$$

Entropy production
is about $1 \text{ W K}^{-1} \text{ m}^{-2}$

What about the non-equilibrium consumption case?

This system evolves as energy consumption and entropy production increases

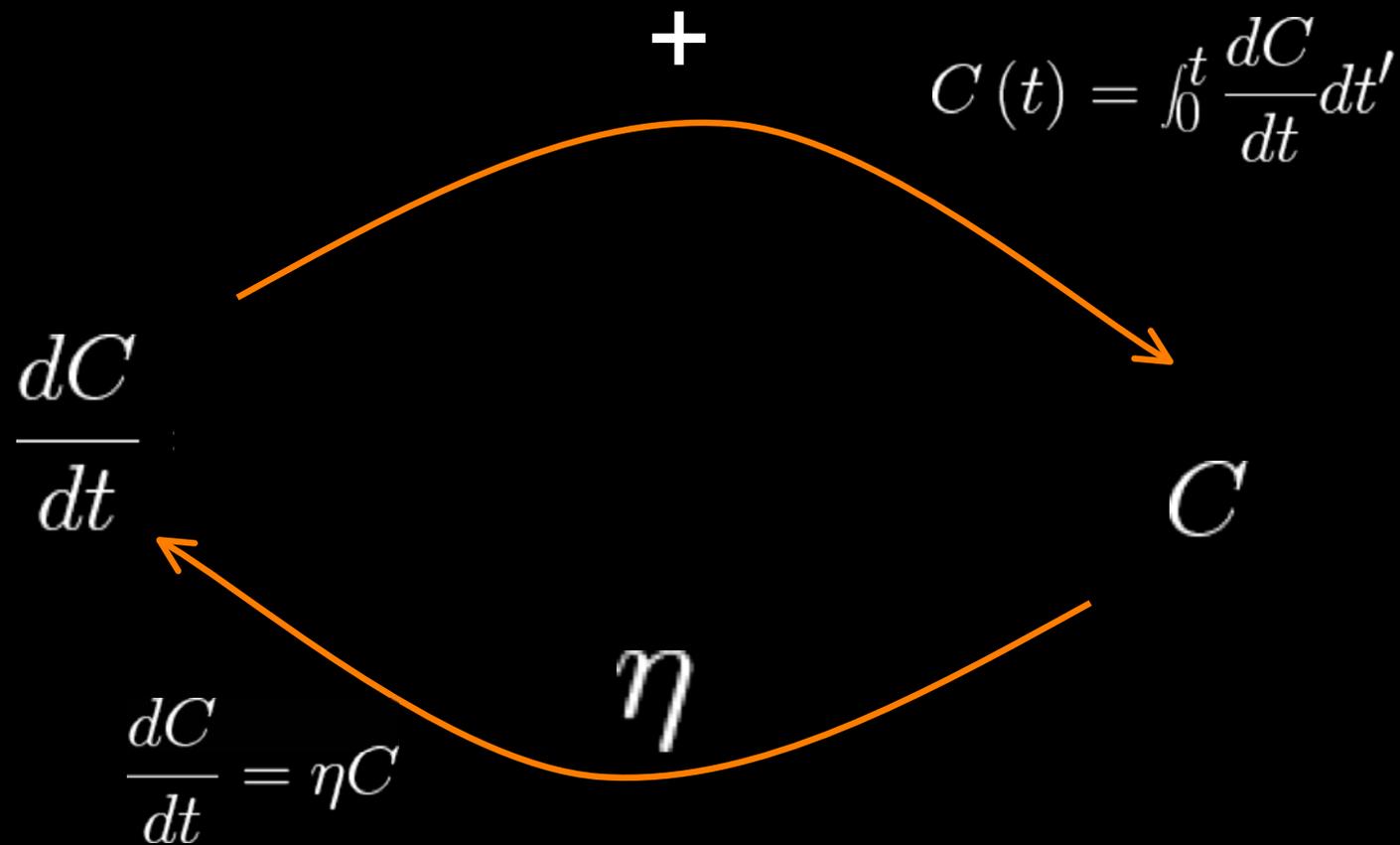
$$T_o = 255 \text{ K}$$

C

$$d(\delta G)/dt = 50 \text{ W}$$

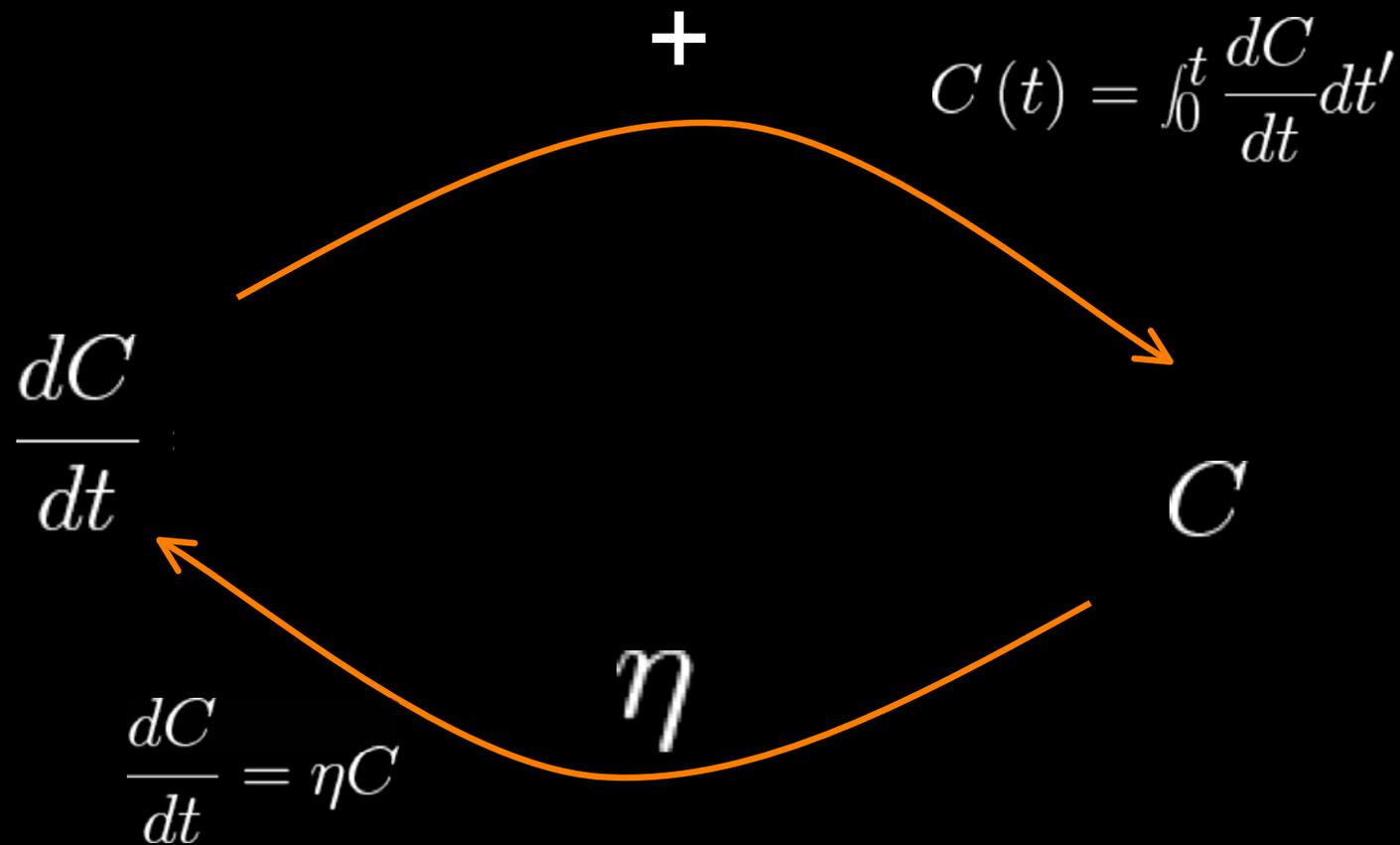


$$dS/dt = C/255 = 0.2 \text{ W/K and growing!}$$



η Feedback parameter. Units per time

C Energy consumption rate. Units power

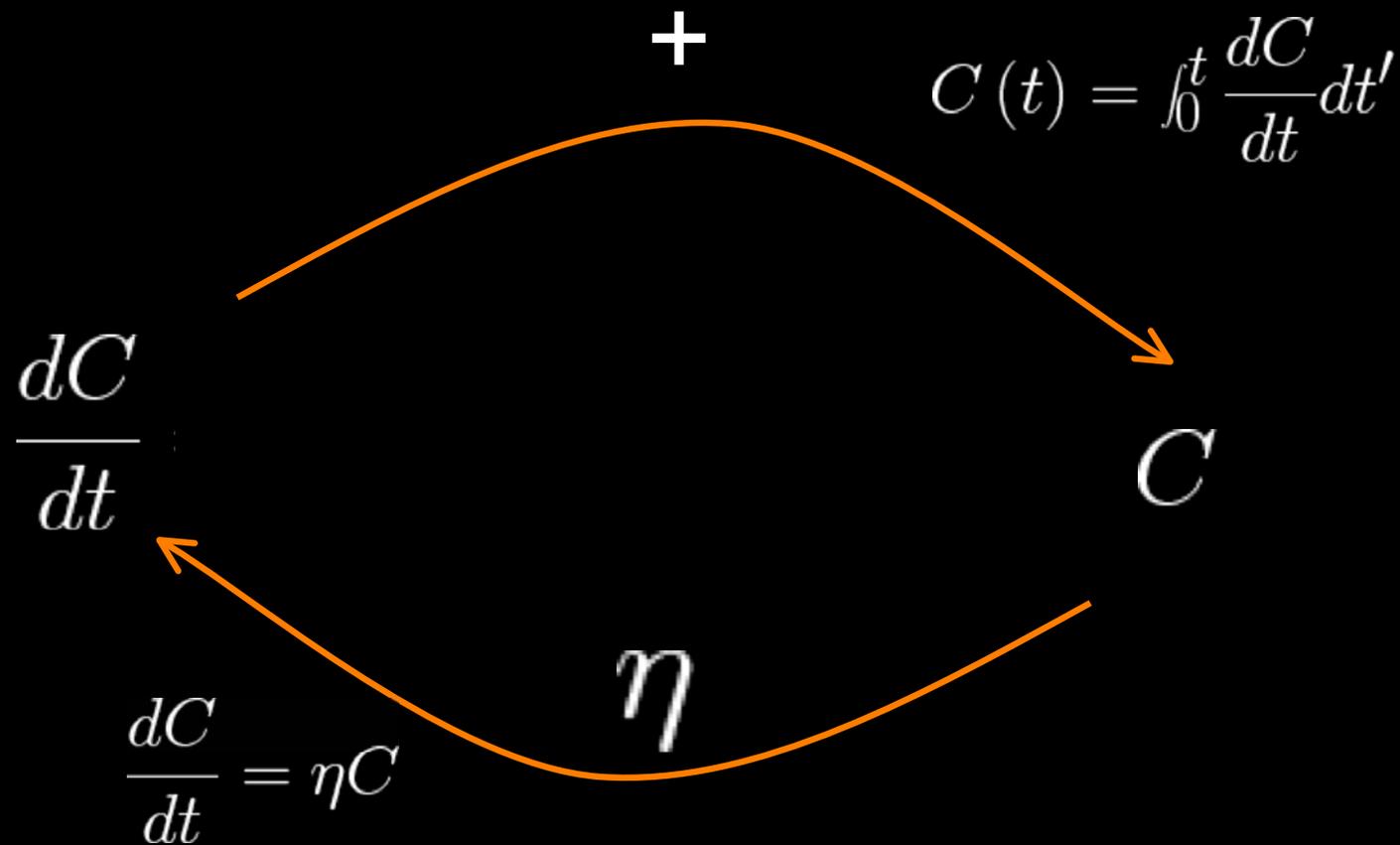


If equilibrium
consumption

$$\eta = 0 \quad C \neq 0$$

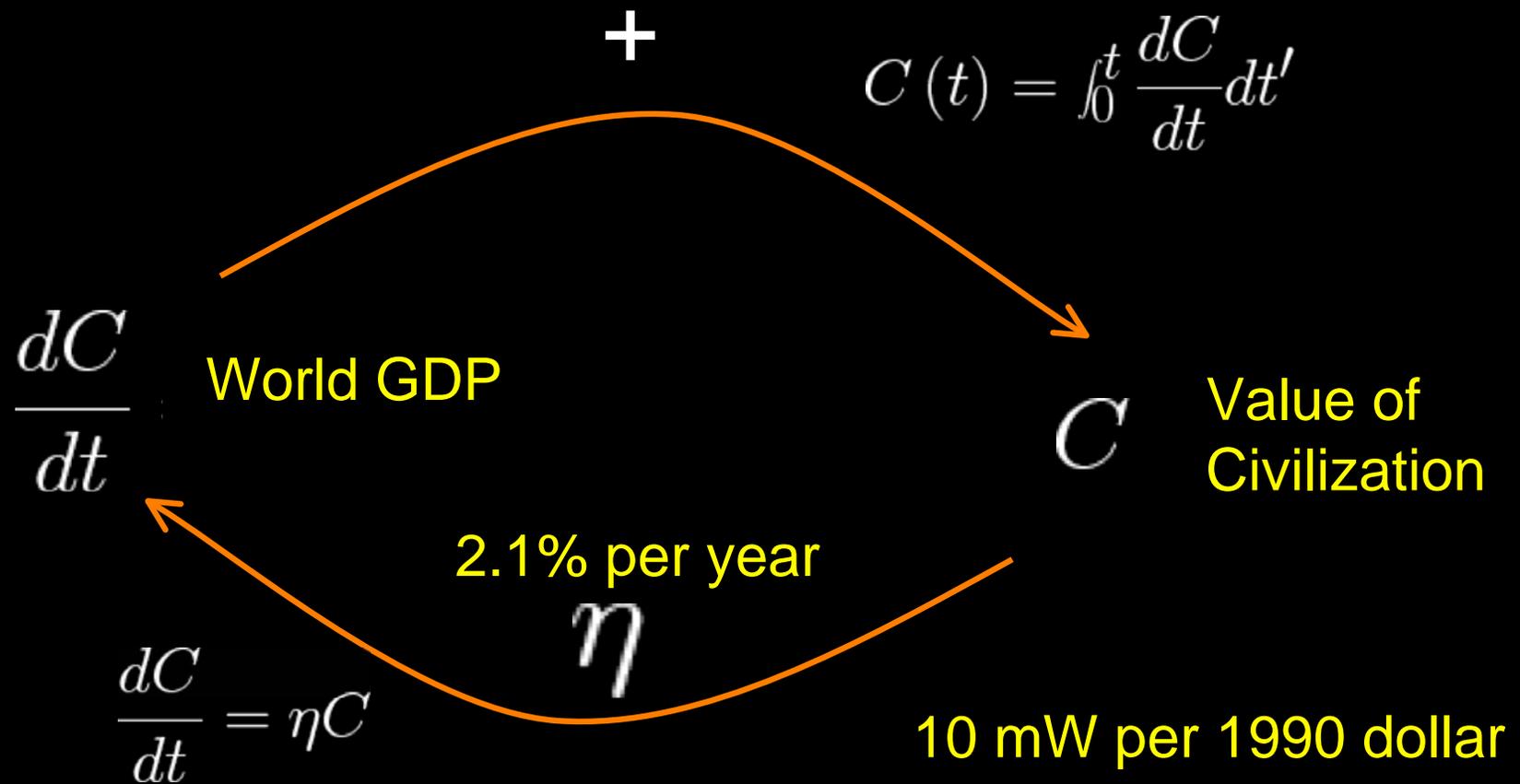
$$\frac{dC}{dt} = 0$$

Energy consumption is constant



If non-equilibrium consumption $\eta \neq 0$ $C \neq 0$

Energy consumption evolves



If non-equilibrium consumption $\eta \neq 0$ $C \neq 0$

Energy consumption evolves

First hypothesis

Non-equilibrium consumption:

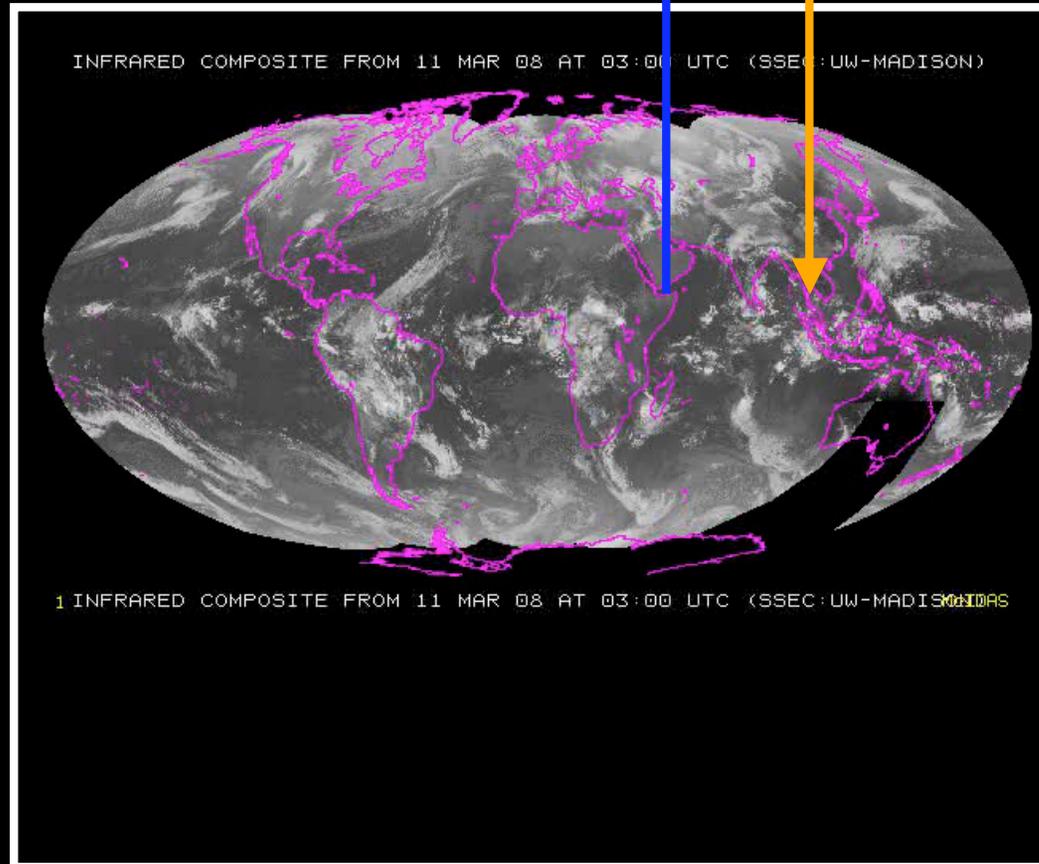
System evolution is defined by the most efficient departures from equilibrium consumption. These transitions are characterized by processes whose values of the feedback parameter η is large

$$\frac{dC}{dt} = \eta C$$

$$d(\delta G)/dt = 240 \text{ W m}^{-2}$$

255 K

C

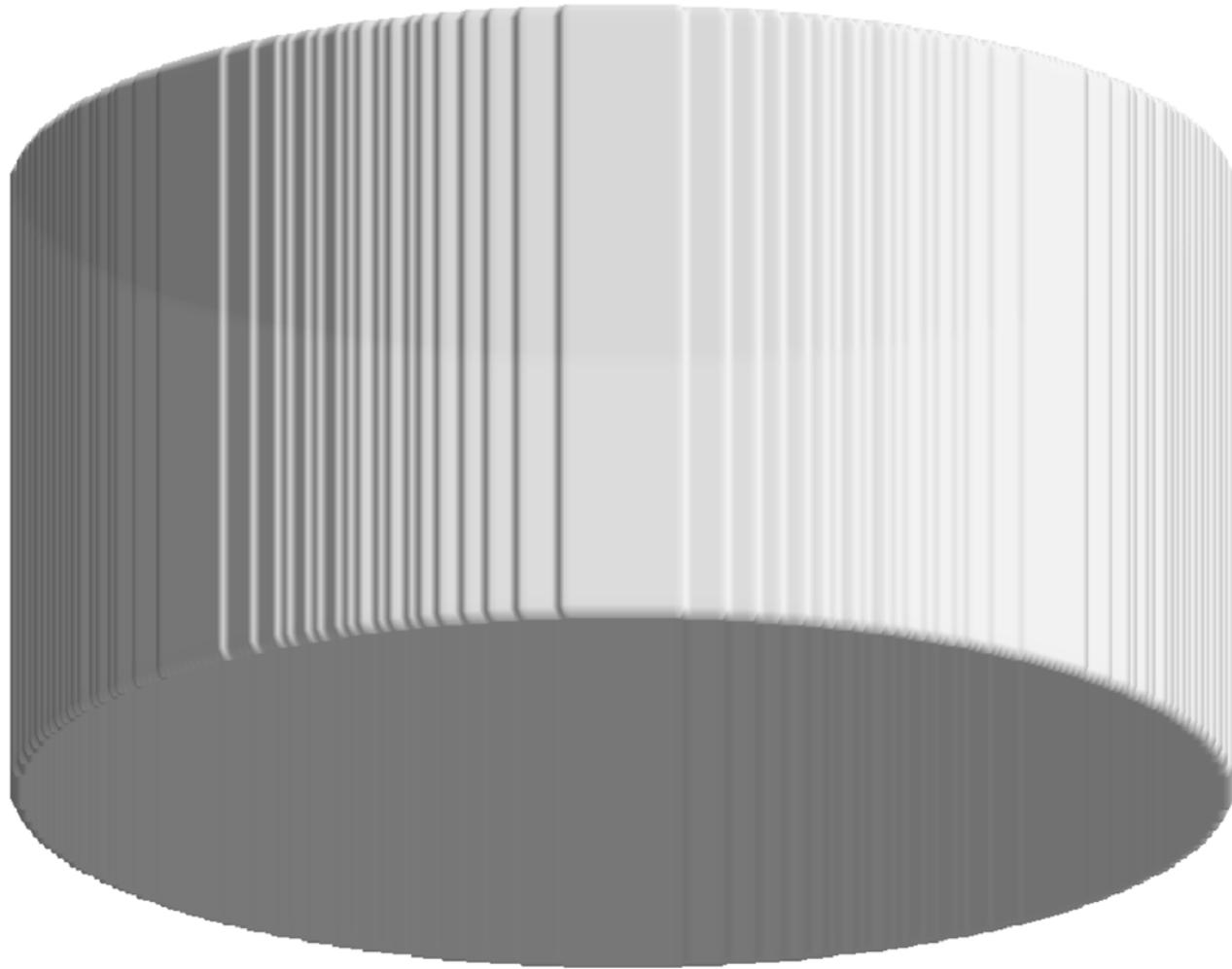


$$dS/dt = C/255$$

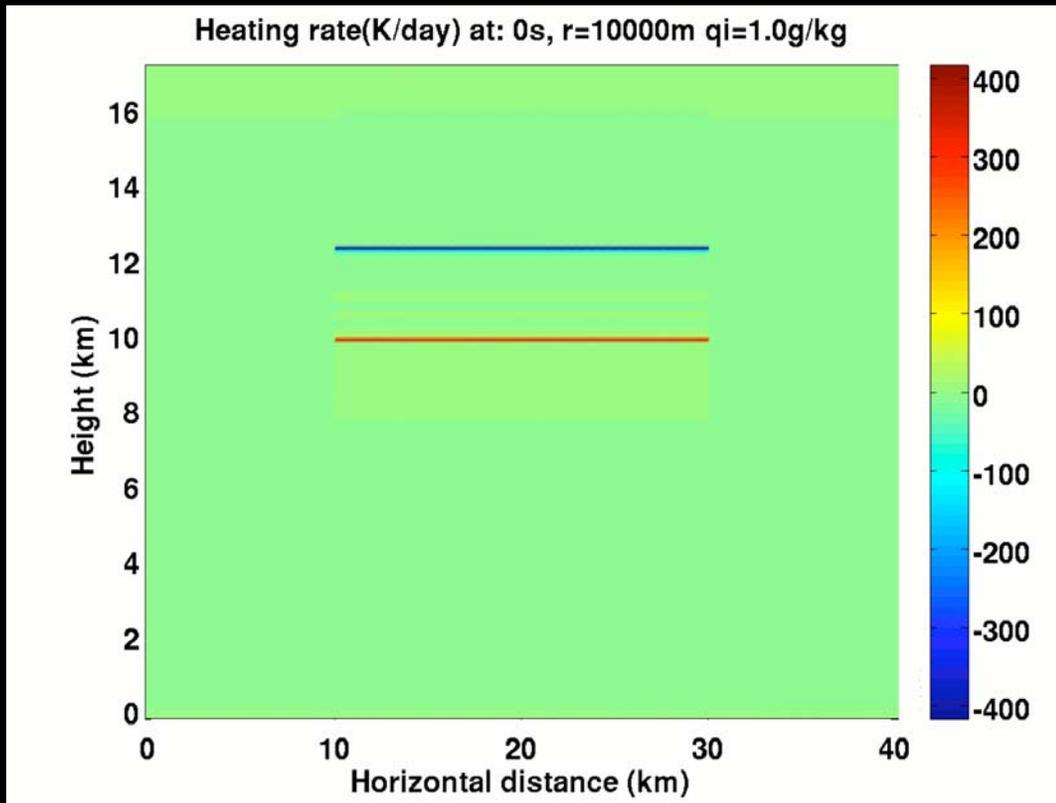




Initially neutrally buoyant cirrus cylinder, with no precip and no sun



Cloud evolves due to IR interactions only



Cross-section
of heating rates
in the cylindrical
cloud

$$C = \frac{d(\delta G)}{dt} \Big|_{\theta} = |\Delta F| A$$

Energy creation or consumption = IR flux deposition times
Area

$$\frac{dC_i}{dt} = \eta_i C_i$$

$$\frac{d \ln C_i}{dt} = \eta_i \quad \text{Feedback parameter}$$

$$C = \left. \frac{d(\delta G)}{dt} \right|_{\theta} = |\Delta F| A \quad \text{Total consumption} = \text{flux dep.} \times \text{area}$$

$$\frac{d \ln C}{dt} = \left. \frac{1}{\theta} \frac{d\theta}{dt} \right|_{\theta, A} + \left. \frac{1}{L} \frac{dL}{dt} \right|_{\theta, \Delta F} \quad \text{Specific feedbacks}$$

Cloud Evolution = Rate of change in cloud potential temperature + Rate of change of cloud radius

Evolution of energy consumption is determined by three feedback parameters: convective mixing, isentropic adjustment, and horizontal spreading

$$\frac{dC}{dt} = \sum_i \eta_i C_i = \left(\overbrace{\eta_{mix} C_{mix} + \eta_{adj} C_{adj}}^{\text{Temperature/ Flux dep. change}} + \overbrace{\eta_{spread} C_{spread}}^{\text{Area change}} \right)$$

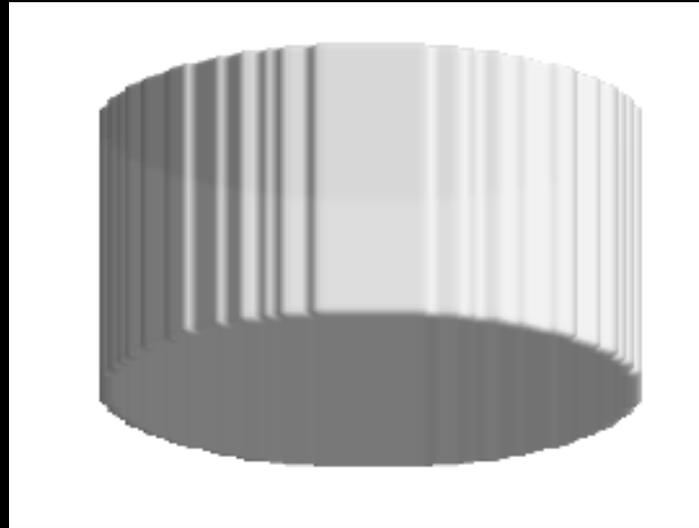
Change in consumption = sum of all possible feedbacks

$$\frac{dC}{dt} = \left(N C_{mix} + \frac{\mathcal{H}g}{\theta N^2 h} C_{adj} + \frac{Nh}{L} C_{spread} \right)$$

Case 1

- Tenuous cirrus blob
- 1 km radius, 2.5 km thick
- 0.01 g/kg
- Diffuse radiative energy deposition
- Low heating rates

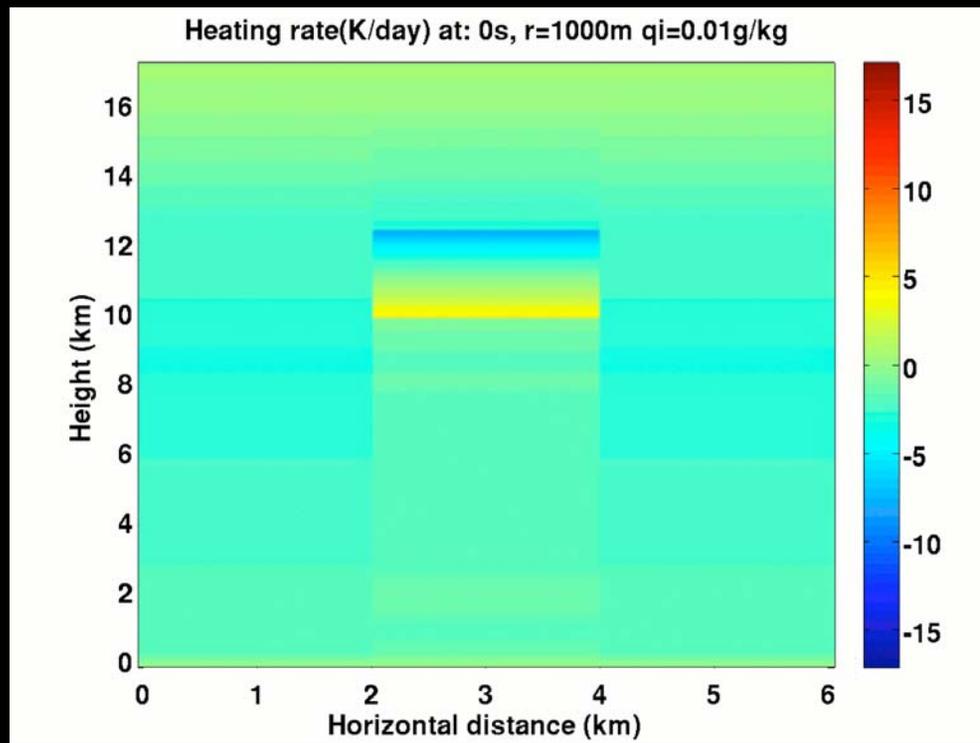




$$\eta_{spread} = 0.1 s^{-1}$$

$$\eta_{adj} = 2 \times 10^{-5} s^{-1}$$

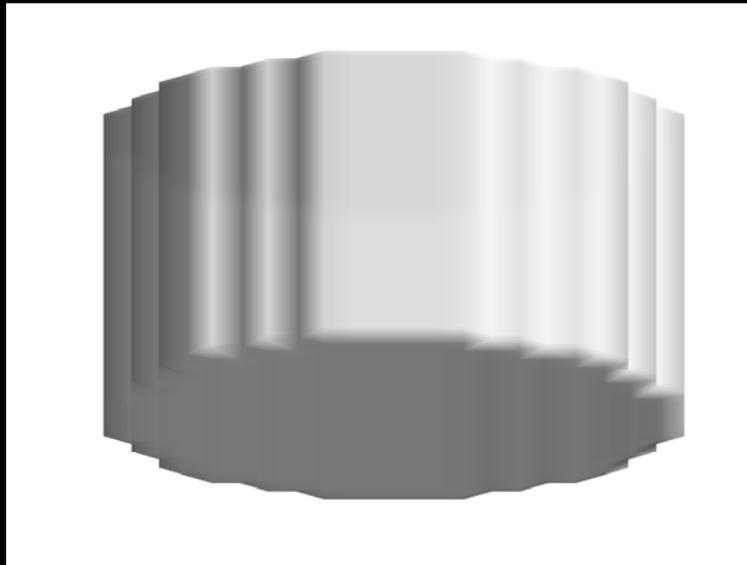
$$\eta_{mix} = 0.01 s^{-1}$$





Case 2

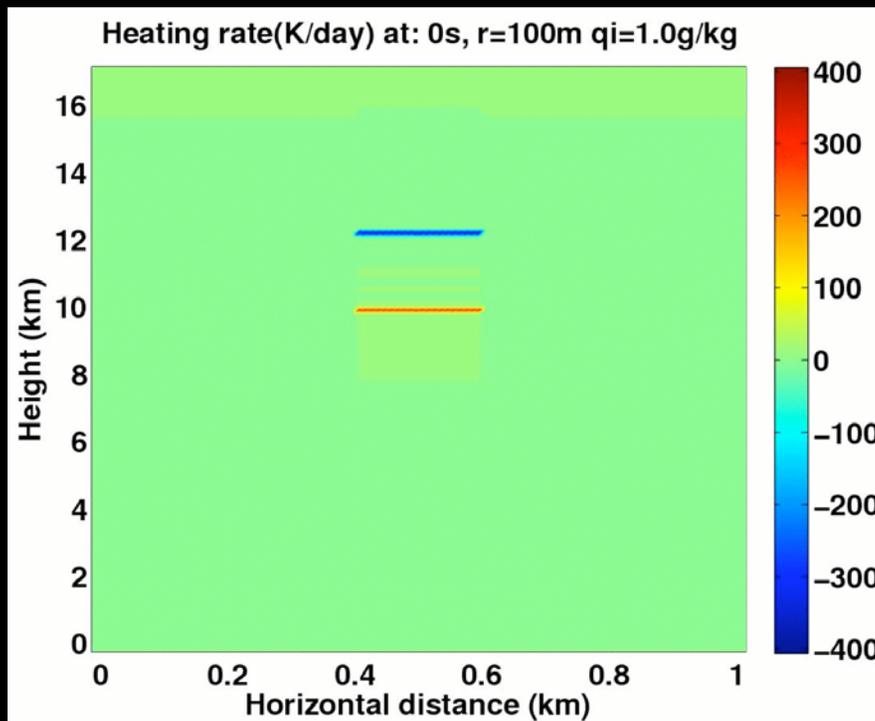
- Dense cirrus - weird cloud
- 100 m radius, 2.5 km thick
- 1 g/kg
- Concentrated radiative energy deposition
- Very high heating rates



$$\eta_{spread} = 3 \times 10^{-3} s^{-1}$$

$$\eta_{adj} = 0.04 s^{-1}$$

$$\eta_{mix} = 0.01 s^{-1}$$

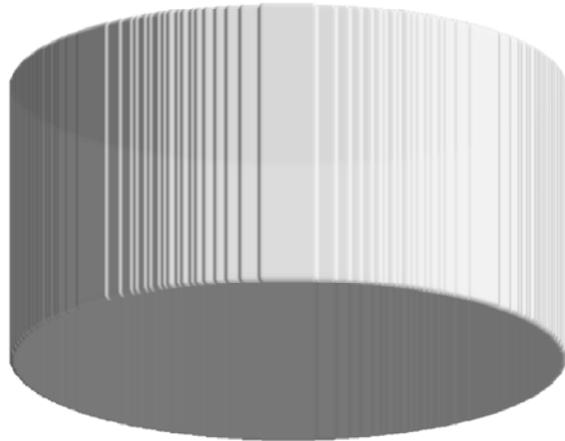


Case 3

- Thicker cirrus - aged anvil?
- 10 km radius
- 0.1 g/kg
- Medium radiative energy deposition
- Medium heating rates



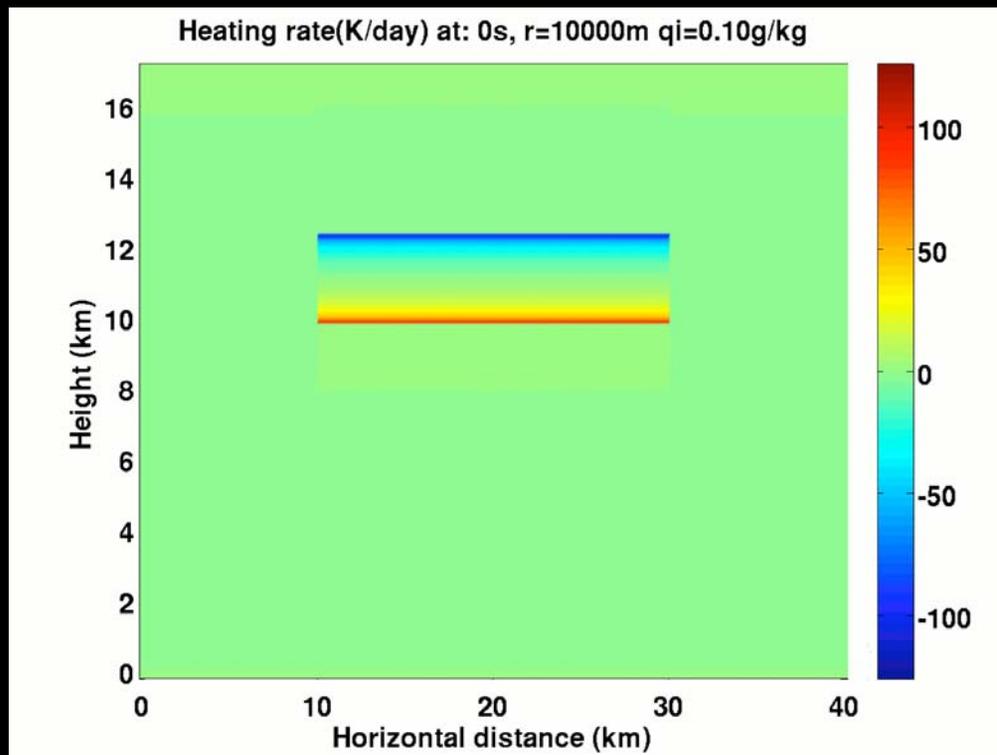




$$\eta_{spread} = 3 \times 10^{-4} \text{ s}^{-1}$$

$$\eta_{adj} = 1 \times 10^{-3} \text{ s}^{-1}$$

$$\eta_{mix} = 0.01 \text{ s}^{-1}$$



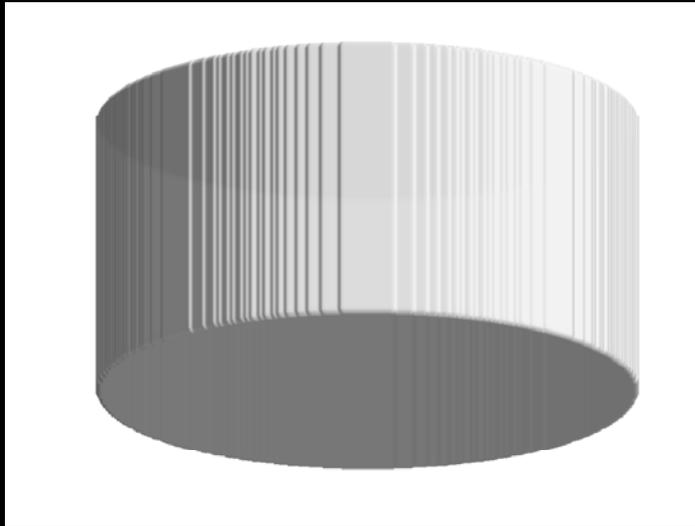




Case 4

- Dense newly formed cirrus anvil
- 10 km radius
- 1 g/kg
- High radiative energy deposition
- Very high heating rates

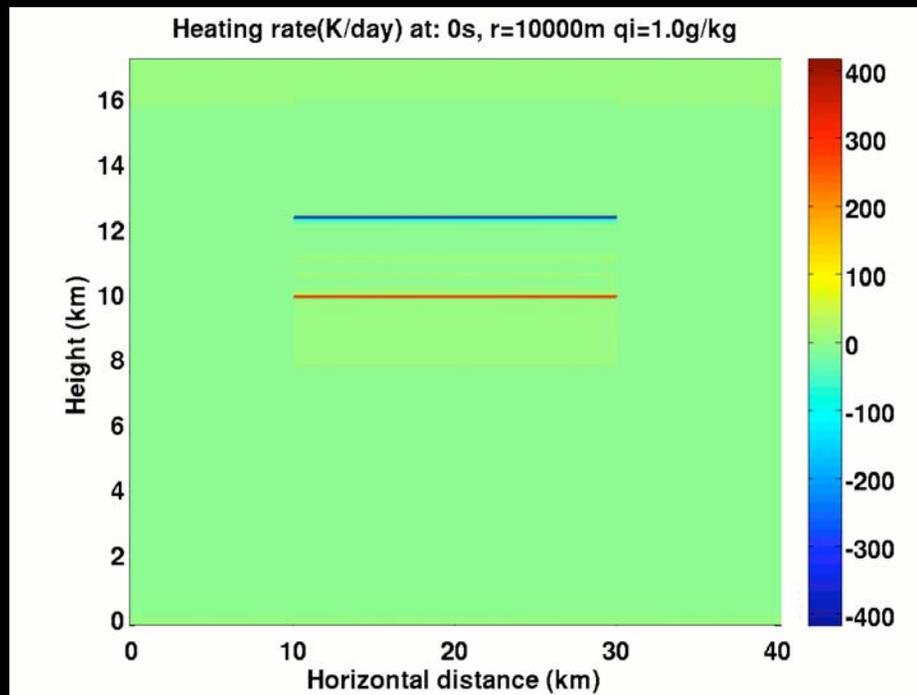


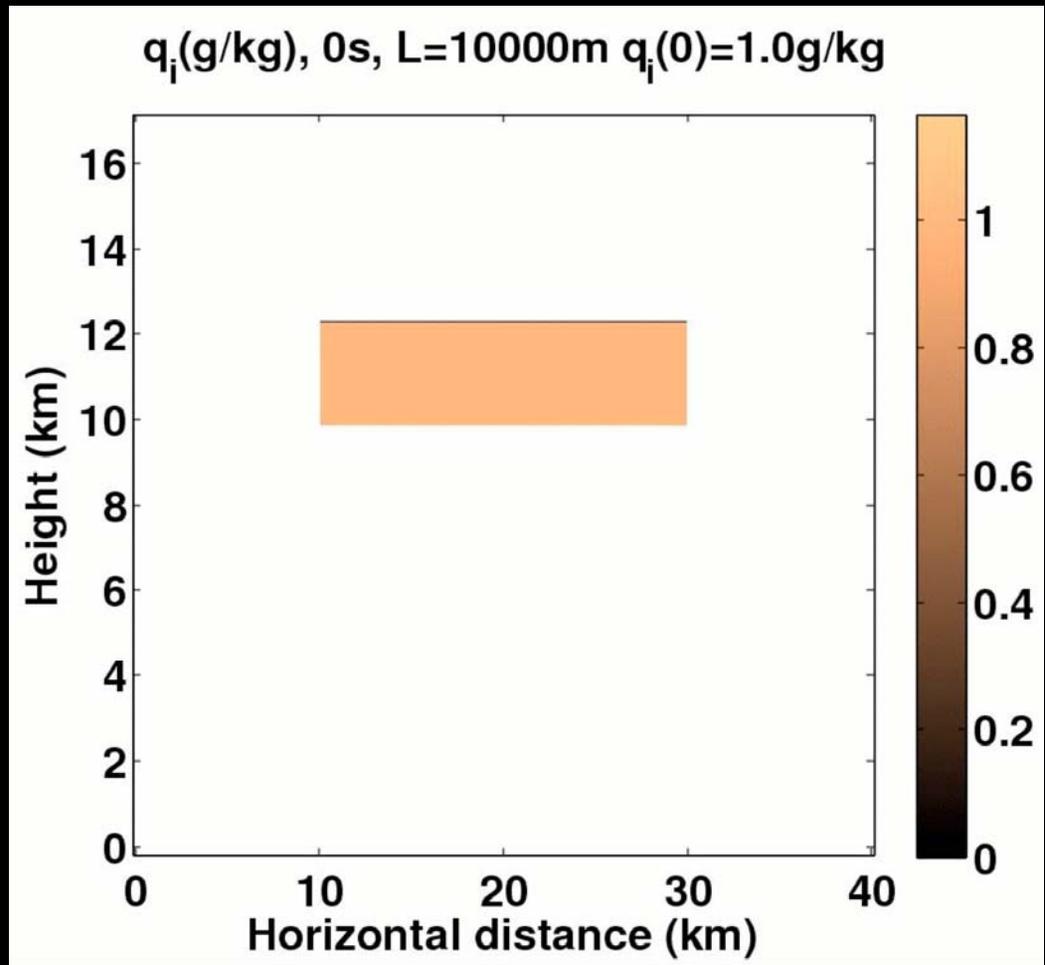
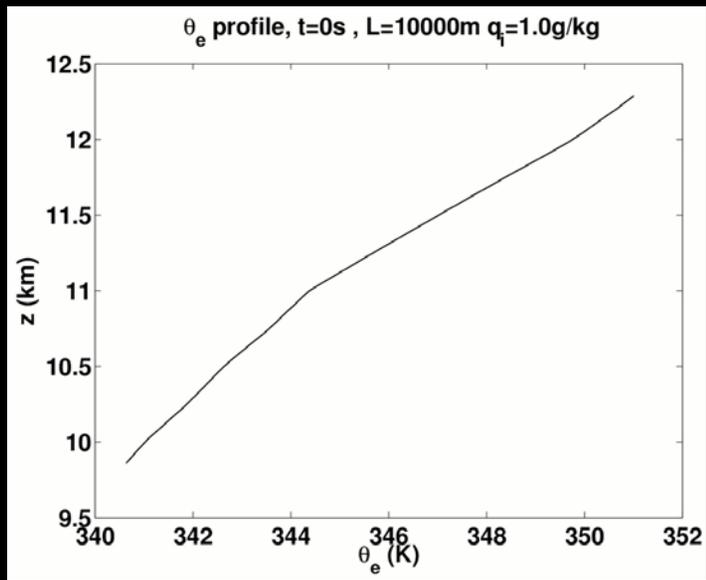


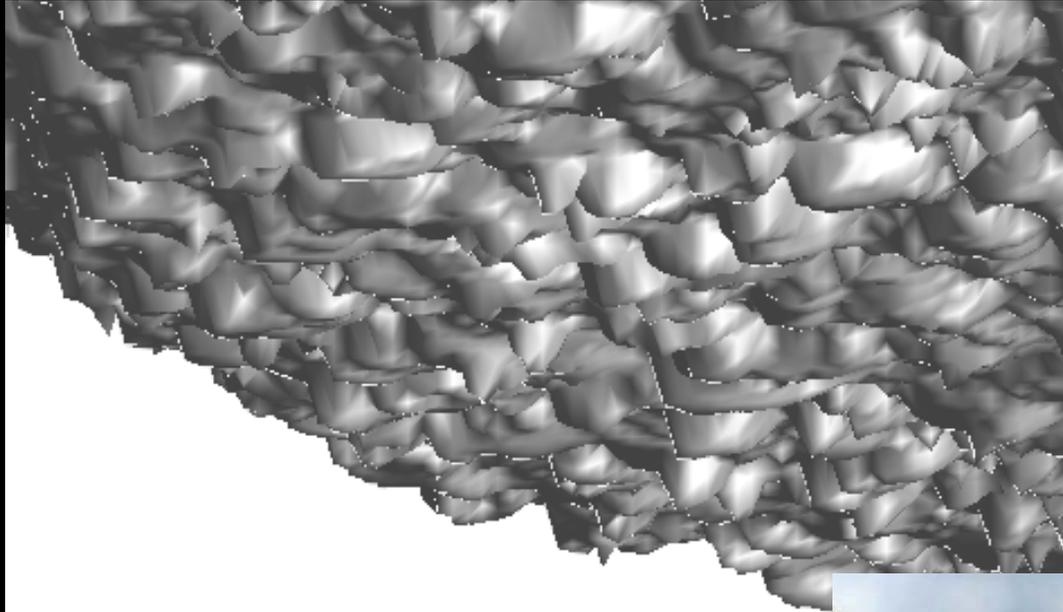
$$\eta_{spread} = 3 \times 10^{-5} s^{-1}$$

$$\eta_{adj} = 0.05 s^{-1}$$

$$\eta_{mix} = 0.01 s^{-1}$$







Mammatus associated with radiatively driven convective overturning near cloud base

$$\eta_{adj} = 0.05 \text{ s}^{-1}$$

$$\eta_{mix} = 0.01 \text{ s}^{-1}$$

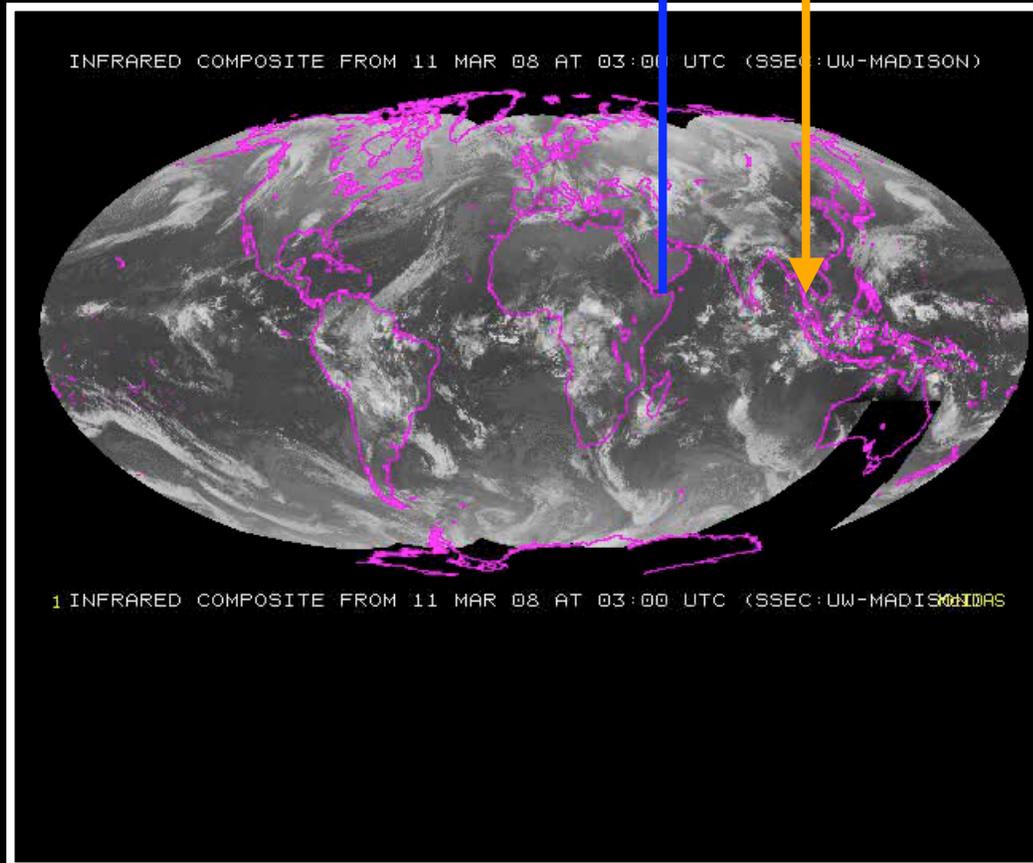
Overturning is “fitter” than isentropic adjustment for consumption of available potential energy supply



255 K

$C = 240 \text{ W m}^{-2}$

$d(\delta G)/dt = 240 \text{ W m}^{-2}$



$$\left. \frac{dS}{dt} \right|_{\theta} = \frac{C}{T_o} = \left. \frac{d(\delta G)/dt}{T_o} \right|_{\theta} \geq 0$$

Entropy production
is about $1 \text{ W K}^{-1} \text{ m}^{-2}$

Evolution is defined by many processes going on simultaneously

$$\sum_i C_i = C$$

Specific energy consumption is constrained by the total available: **competition**

$$\frac{dC}{dt} = \sum_i \eta_i C_i \rightarrow 0$$

Under quasi-equilibrium transitions, high specific energy consumption is favored by a low feedback parameter: **survival of the fittest?**

Summary

- Entropy production is a representation of the conversion of available potential energy to a lower temperature form - energy consumption

Summary

- Case 1: Consumption of available energy is constant; zero feedback η
- Example: Global/regional clouds and climate
- “The more things change, the more things stay the same”

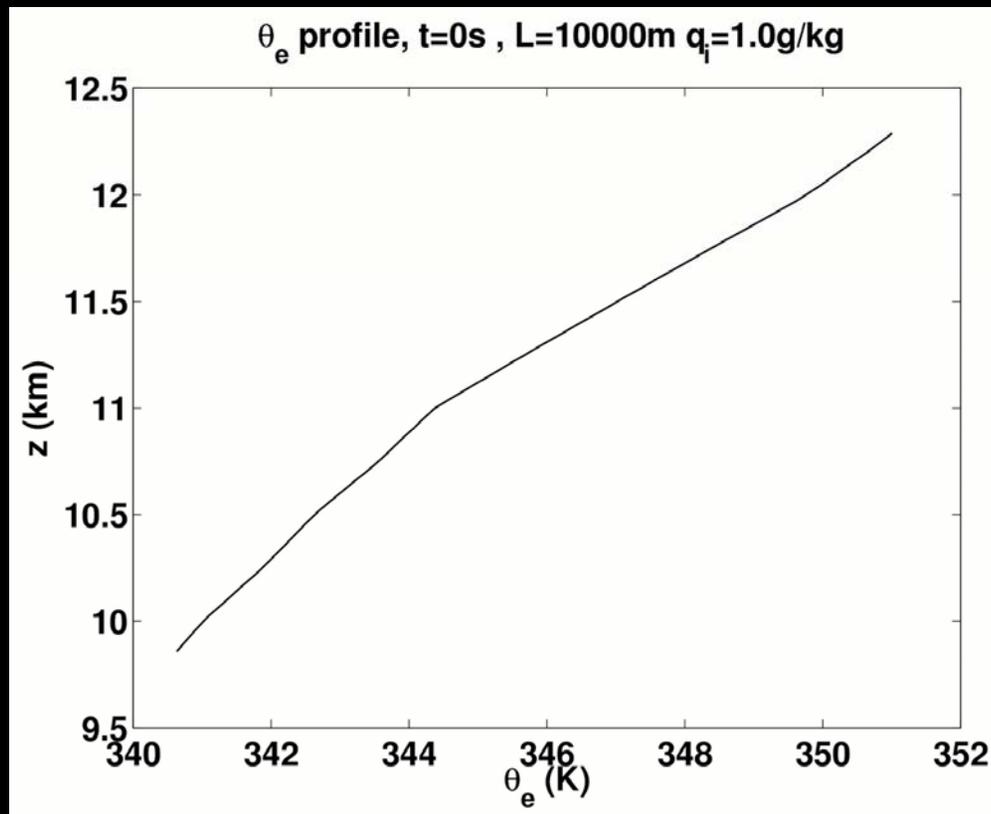
Summary

- Case 2: Consumption of available energy is evolving; non-zero feedback η
- Quasi-equilibrium transitions
- The process that dominates evolution is the one with the lowest feedback parameter - “Survival of the fittest” or “Haste makes waste?”
- Example: clouds on more local scales

Thankyou



Support from NASA New Investigator Program
Work done with Stina Sjostrom and Clint Schmidt



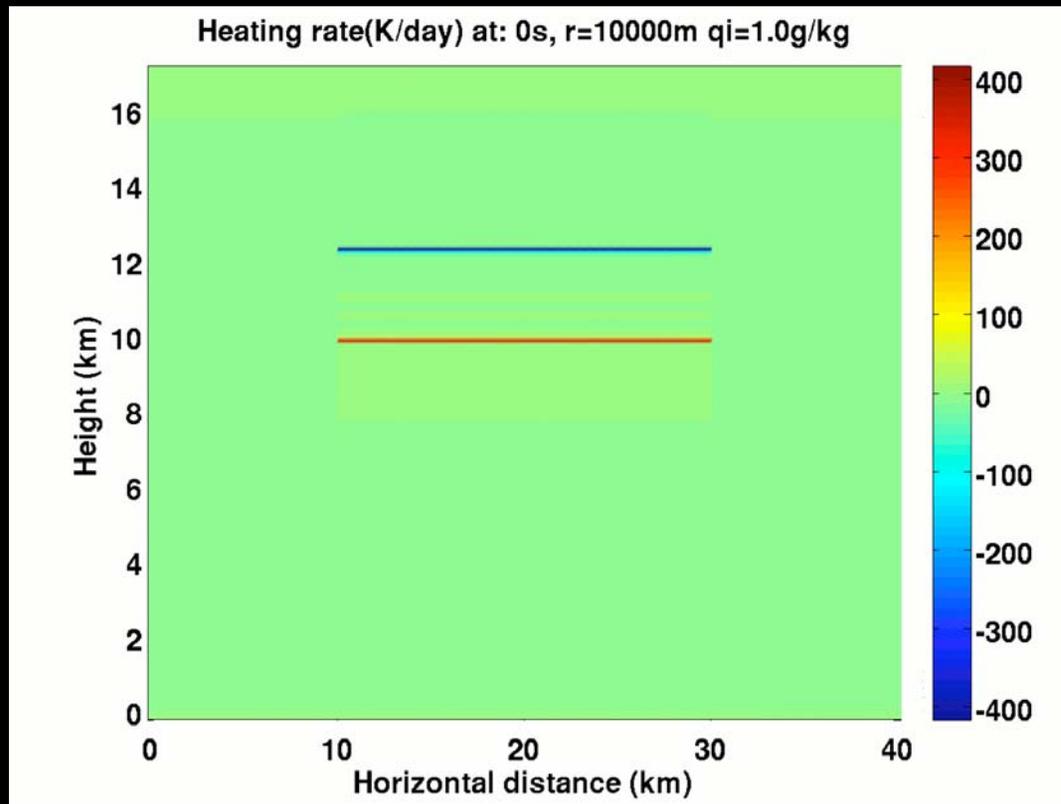
Potential temperature change due to convective mixing

$$\frac{1}{\theta} \frac{d\theta}{dt} \Big|_{\theta, A} \simeq \frac{1}{h} \frac{d(\delta z)}{dt}$$

$$\frac{1}{\theta} \frac{d\theta}{dt} \Big|_{\theta, A, z} \simeq \frac{1}{h} \frac{d(\delta z)}{dt} = \frac{w^*}{h} = \frac{Nh}{h} = N$$

$$\eta_{mix} = N$$

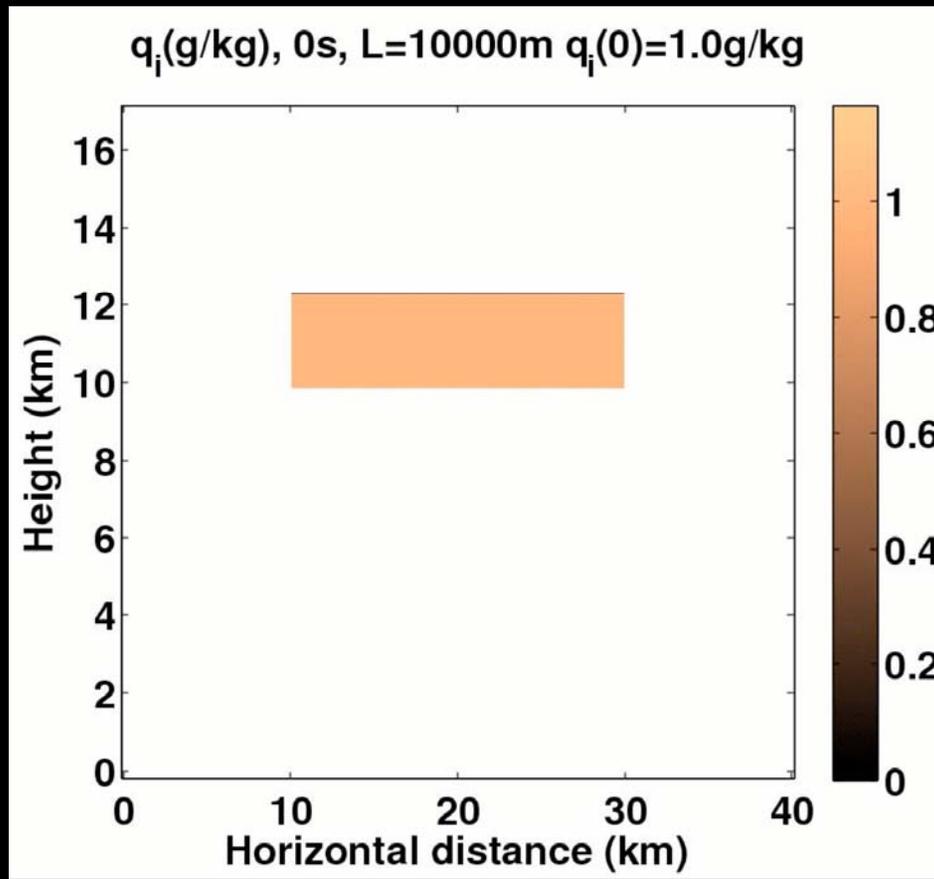
$$N = \left(\frac{g}{\theta} \frac{d\theta}{dz} \right)^{1/2}$$



Potential
temperature change
due to isentropic
adjustment

$$\frac{1}{\theta} \frac{d\theta}{dt} \Big|_{\theta, A, TKE} \simeq \frac{1}{h} \frac{d(\delta z)}{dt} = \frac{w_{strat}}{h} = \frac{1}{h} \frac{d\theta/dt}{d\theta/dz} = \frac{\mathcal{H}g}{\theta N^2 h}$$

$$\eta_{adj} = \frac{\mathcal{H}g}{\theta N^2 h}$$



Area change due to Spreading in density currents

$$u^{*2} \simeq N^2 h^2$$

$$h = \frac{1}{\gamma k(r_{ice}) m_i n_i}$$

$$\left. \frac{d \ln A}{dt} \right|_{\theta, \Delta F} \sim \left. \frac{d \ln L}{dt} \right|_{\theta, \Delta F} = \frac{Nh}{L}$$

$$\eta_{spread} = \frac{Nh}{L}$$

